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## Structure Reports

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## N-(4-Chlorophenylsulfonyl)-2-methylpropanamide

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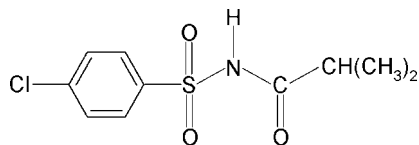
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Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.006$  Å;  $R$  factor = 0.060;  $wR$  factor = 0.126; data-to-parameter ratio = 16.7.

In the crystal structure of the title compound,  $\text{C}_{10}\text{H}_{12}\text{ClNO}_3\text{S}$ , the N–C bond in the C–SO<sub>2</sub>–NH–C segment has a *gauche* torsion with respect to the S=O bonds. The molecule is twisted at the S atom with a C–S–N–C torsion angle of  $-62.3(3)^\circ$ . The benzene ring and the SO<sub>2</sub>–NH–CO–C segment form a dihedral angle of  $89.3(1)^\circ$ . In the crystal, molecules are linked by pairs of N–H $\cdots$ O hydrogen bonds into inversion dimers.

### Related literature

For the sulfanilamide moiety in sulfonamide drugs, see: Maren (1976). For its ability to form hydrogen bonds in the solid state, see: Yang & Guillory (1972). For hydrogen-bonding modes of sulfonamides, see: Adsmoind & Grant (2001). For our studies on the effects of substituents on the structures and other aspects of *N*-(aryl)-amides, see: Arjunan *et al.* (2004), on *N*-(aryl)-methanesulfonamides, see: Gowda *et al.* (2007), on *N*-(aryl)-arylsulfonamides, see: Gowda *et al.* (2003) and on *N*-(arylsulfonyl)-amides, see: Gowda *et al.* (2008); Shakuntala *et al.* (2011).



### Experimental

#### Crystal data

$\text{C}_{10}\text{H}_{12}\text{ClNO}_3\text{S}$   
 $M_r = 261.72$

Triclinic,  $P\bar{1}$   
 $a = 6.207(1)$  Å

$b = 10.395(3)$  Å  
 $c = 10.497(3)$  Å  
 $\alpha = 70.150(2)^\circ$   
 $\beta = 79.160(2)^\circ$   
 $\gamma = 86.010(2)^\circ$   
 $V = 625.7(3)$  Å<sup>3</sup>

$Z = 2$   
Mo  $K\alpha$  radiation  
 $\mu = 0.46$  mm<sup>-1</sup>  
 $T = 293$  K  
 $0.46 \times 0.20 \times 0.08$  mm

#### Data collection

Oxford Diffraction Xcalibur diffractometer with a Sapphire CCD detector  
Absorption correction: multi-scan *CrysAlis RED* (Oxford)

Diffraction, 2009)  
 $T_{\min} = 0.815$ ,  $T_{\max} = 0.964$   
3846 measured reflections  
2476 independent reflections  
1842 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.014$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.060$   
 $wR(F^2) = 0.126$   
 $S = 1.19$   
2476 reflections  
148 parameters  
1 restraint

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.28$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.31$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1N}\cdots\text{O1}^i$	0.84 (2)	2.08 (2)	2.912 (4)	169 (3)

Symmetry code: (i)  $-x, -y + 1, -z + 2$ .

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2009); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2009); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NC2243).

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## supporting information

*Acta Cryst.* (2011). E67, o2274 [doi:10.1107/S1600536811031394]

## ***N*-(4-Chlorophenylsulfonyl)-2-methylpropanamide**

**P. G. Nirmala, Sabine Foro and B. Thimme Gowda**

### **S1. Comment**

The molecular structures of sulfonamide drugs contain the sulfanilamide moiety (Maren, 1976). The propensity for hydrogen bonding in the solid state, due to the presence of various hydrogen bond donors and acceptors gives rise to polymorphism (Yang & Guillory, 1972). The hydrogen bonding preferences of sulfonamides has also been studied (Adsmond & Grant, 2001). The nature and position of substituents play a significant role on the crystal structures of this class of compounds. As part of our work on the effects of substituents on the structures and other aspects of *N*-(aryl)-amides (Arjunan *et al.*, 2004), *N*-(aryl)-methanesulfonamides (Gowda *et al.*, 2007), *N*-(aryl)-arylsulfonamides (Gowda *et al.*, 2003) and *N*-(arylsulfonyl)-acetamides (Gowda *et al.*, 2008, Shakuntala *et al.*, 2011), in the present work, the crystal structure of *N*-(4-chlorophenylsulfonyl)-2,2-dimethylacetamide (I) has been determined. The N—C bond in the C—SO<sub>2</sub>—NH—C segment has *gauche* torsion with respect to the S=O bonds. The molecule is twisted at the S-atom with a C—S—N—C torsion angle of -62.3 (3)°, compared to the values of -72.5 (2)° in *N*-(4-chlorophenylsulfonyl)-2,2-dichloroacetamide (II) (Gowda *et al.*, 2008) and *N*-(2-chlorophenylsulfonyl)-2,2-dimethylacetamide (III) (Shakuntala *et al.*, 2011).

Further, the dihedral angle between the benzene ring and the SO<sub>2</sub>—NH—CO—C segment in (I) is 89.3 (1)°, compared to the values of 79.7 (1)° in (II) and 87.4 (1)° in (III).

In the crystal structure, the molecules are connected into centrosymmetrically dimers by intermolecular N—H...O hydrogen bonding (Table 1). Part of the crystal structure is shown in Fig. 2.

### **S2. Experimental**

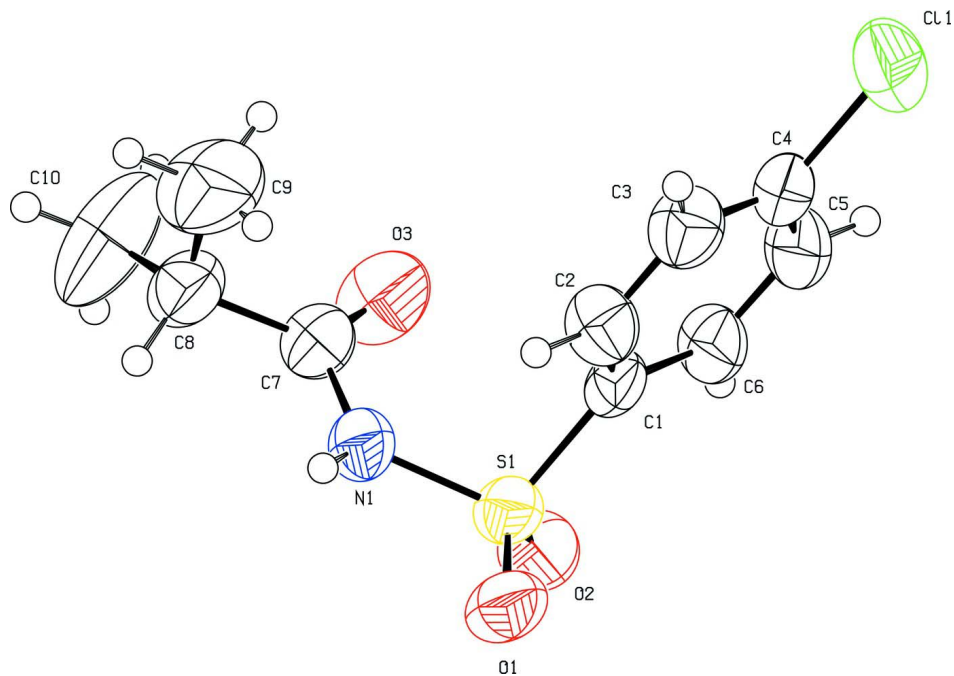
The title compound was prepared by refluxing 4-chlorobenzenesulfonamide (0.10 mole) with an excess of 2,2-dimethylacetyl chloride (0.20 mole) for about an hour on a water bath. The reaction mixture was cooled and poured into ice cold water. The resulting solid was separated, washed thoroughly with water and dissolved in warm dilute sodium hydrogen carbonate solution. The title compound was reprecipitated by acidifying the filtered solution with glacial acetic acid. It was filtered, dried and recrystallized from ethanol. The purity of the compound was checked by determining its melting point. It was further characterized by recording its infrared spectra.

Plate like colorless single crystals of the title compound used in X-ray diffraction studies were obtained from a slow evaporation of an ethanolic solution of the compound.

### **S3. Refinement**

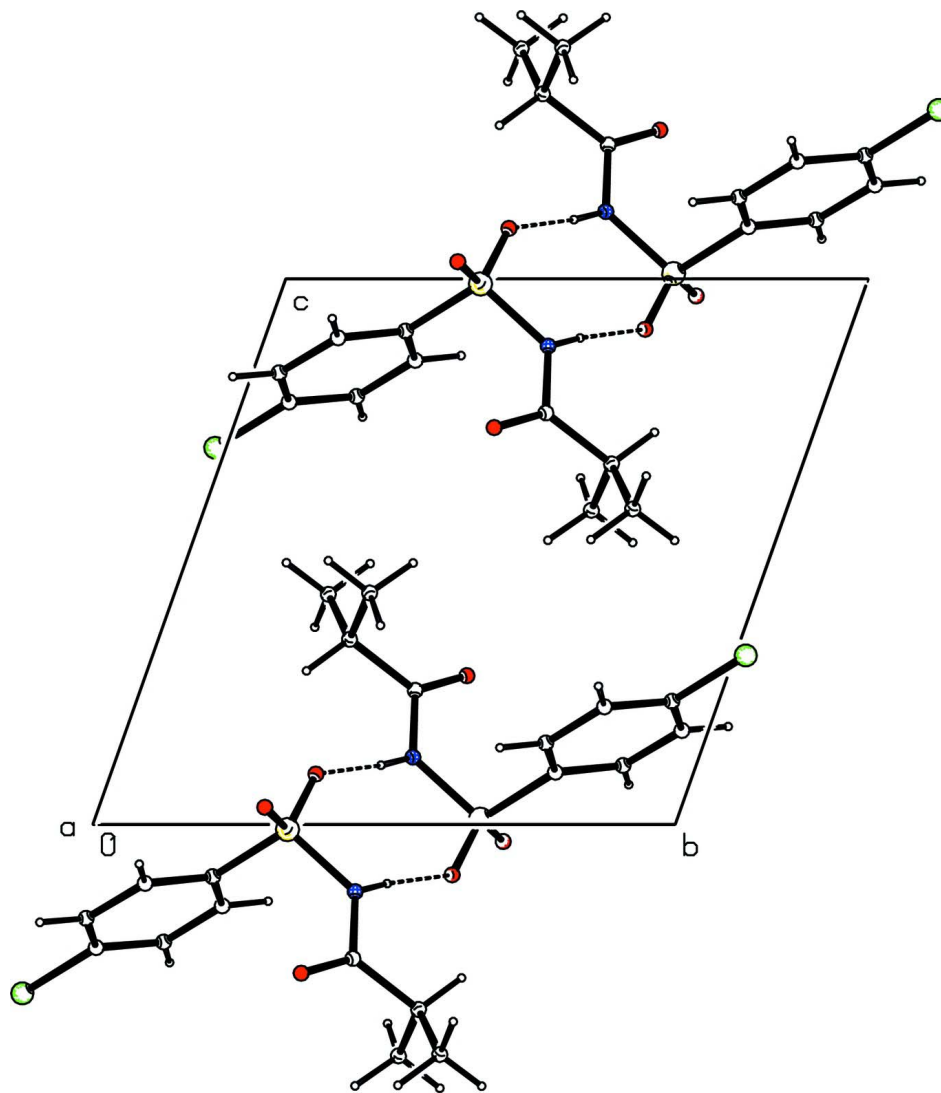
The H atom of the NH group was located in a difference map and later restrained to the distance N—H = 0.86 (2) Å. The other H atoms were positioned with idealized geometry using a riding model with the aromatic C—H = 0.93 Å, methyl C—H = 0.96 Å and methyne C—H = 0.98 Å.

All H atoms were refined with isotropic displacement parameters. The  $U_{\text{iso}}(\text{H})$  values were set at  $1.2U_{\text{eq}}(\text{C-aromatic, N})$  and  $1.5U_{\text{eq}}(\text{C-methyl})$ .



**Figure 1**

Molecular structure of the title compound, showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.



**Figure 2**

Crystal structure of the title compound showing the dimers. Hydrogen bonds are shown as dashed lines.

***N*-(4-Chlorophenylsulfonyl)-2-methylpropanamide**

*Crystal data*

$C_{10}H_{12}ClNO_3S$

$M_r = 261.72$

Triclinic,  $P\bar{1}$

Hall symbol:  $-P\ 1$

$a = 6.207\ (1)\ \text{\AA}$

$b = 10.395\ (3)\ \text{\AA}$

$c = 10.497\ (3)\ \text{\AA}$

$\alpha = 70.150\ (2)^\circ$

$\beta = 79.160\ (2)^\circ$

$\gamma = 86.010\ (2)^\circ$

$V = 625.7\ (3)\ \text{\AA}^3$

$Z = 2$

$F(000) = 272$

$D_x = 1.389\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 1240 reflections

$\theta = 3.3\text{--}27.8^\circ$

$\mu = 0.46\ \text{mm}^{-1}$

$T = 293\ \text{K}$

Plate, colourless

$0.46 \times 0.20 \times 0.08\ \text{mm}$

*Data collection*

Oxford Diffraction Xcalibur  
 diffractometer with a Sapphire CCD detector  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 Rotation method data acquisition using  $\omega$  and  $\varphi$   
 scans  
 Absorption correction: multi-scan  
*CrysAlis RED* (Oxford Diffraction, 2009)  
 $T_{\min} = 0.815$ ,  $T_{\max} = 0.964$

3846 measured reflections  
 2476 independent reflections  
 1842 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.014$   
 $\theta_{\max} = 26.4^\circ$ ,  $\theta_{\min} = 3.3^\circ$   
 $h = -7 \rightarrow 7$   
 $k = -12 \rightarrow 12$   
 $l = -13 \rightarrow 13$

*Refinement*

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.060$   
 $wR(F^2) = 0.126$   
 $S = 1.19$   
 2476 reflections  
 148 parameters  
 1 restraint  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H atoms treated by a mixture of independent  
 and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.0233P)^2 + 0.632P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.28 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.31 \text{ e } \text{\AA}^{-3}$

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.1905 (5)	0.2375 (3)	0.9043 (3)	0.0475 (7)
C2	-0.0001 (6)	0.2717 (4)	0.8506 (4)	0.0610 (9)
H2	-0.0818	0.3477	0.8593	0.073*
C3	-0.0683 (6)	0.1930 (4)	0.7844 (4)	0.0670 (10)
H3	-0.1957	0.2156	0.7468	0.080*
C4	0.0535 (7)	0.0805 (3)	0.7742 (4)	0.0638 (10)
C5	0.2423 (7)	0.0456 (4)	0.8278 (4)	0.0709 (11)
H5	0.3221	-0.0315	0.8207	0.085*
C6	0.3124 (6)	0.1255 (3)	0.8920 (4)	0.0612 (9)
H6	0.4421	0.1040	0.9271	0.073*
C7	0.4540 (6)	0.5278 (4)	0.7543 (4)	0.0578 (9)
C8	0.4194 (7)	0.6715 (4)	0.6608 (4)	0.0716 (11)
H8	0.3561	0.7263	0.7187	0.086*
C9	0.2542 (9)	0.6652 (6)	0.5756 (5)	0.1190 (19)
H9A	0.1202	0.6271	0.6350	0.143*

H9B	0.3103	0.6089	0.5208	0.143*
H9C	0.2265	0.7558	0.5165	0.143*
C10	0.6291 (10)	0.7353 (6)	0.5780 (6)	0.151 (3)
H10A	0.6959	0.6822	0.5219	0.182*
H10B	0.7258	0.7386	0.6383	0.182*
H10C	0.6015	0.8265	0.5202	0.182*
N1	0.3069 (5)	0.4914 (3)	0.8767 (3)	0.0560 (7)
H1N	0.200 (4)	0.541 (3)	0.892 (4)	0.067*
O1	0.0960 (4)	0.3525 (3)	1.0924 (2)	0.0729 (7)
O2	0.4794 (4)	0.2850 (3)	1.0314 (3)	0.0714 (7)
O3	0.5885 (4)	0.4482 (3)	0.7273 (3)	0.0863 (9)
Cl1	-0.0354 (3)	-0.01843 (12)	0.69033 (14)	0.1076 (5)
S1	0.27563 (15)	0.33738 (9)	0.99054 (9)	0.0559 (3)

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0524 (18)	0.0386 (16)	0.0429 (17)	0.0085 (14)	-0.0084 (14)	-0.0043 (13)
C2	0.055 (2)	0.056 (2)	0.074 (2)	0.0148 (16)	-0.0162 (18)	-0.0244 (18)
C3	0.063 (2)	0.063 (2)	0.072 (2)	0.0024 (18)	-0.0210 (19)	-0.014 (2)
C4	0.090 (3)	0.0428 (19)	0.053 (2)	-0.0118 (18)	-0.0140 (19)	-0.0066 (16)
C5	0.095 (3)	0.0419 (19)	0.077 (3)	0.0201 (19)	-0.026 (2)	-0.0180 (18)
C6	0.068 (2)	0.0471 (19)	0.070 (2)	0.0184 (17)	-0.0255 (19)	-0.0175 (17)
C7	0.052 (2)	0.059 (2)	0.059 (2)	0.0048 (17)	-0.0146 (17)	-0.0148 (17)
C8	0.086 (3)	0.062 (2)	0.057 (2)	0.007 (2)	-0.012 (2)	-0.0084 (18)
C9	0.146 (5)	0.114 (4)	0.095 (4)	0.039 (4)	-0.060 (4)	-0.020 (3)
C10	0.123 (5)	0.128 (5)	0.133 (5)	-0.026 (4)	-0.001 (4)	0.041 (4)
N1	0.0637 (19)	0.0447 (16)	0.0567 (17)	0.0105 (13)	-0.0102 (15)	-0.0155 (13)
O1	0.0929 (19)	0.0657 (16)	0.0489 (14)	0.0221 (14)	-0.0029 (13)	-0.0142 (12)
O2	0.0769 (17)	0.0681 (16)	0.0755 (17)	0.0171 (13)	-0.0381 (14)	-0.0221 (13)
O3	0.0719 (18)	0.087 (2)	0.0815 (19)	0.0296 (15)	-0.0026 (15)	-0.0156 (16)
Cl1	0.1677 (13)	0.0641 (7)	0.1053 (9)	-0.0180 (7)	-0.0502 (9)	-0.0294 (6)
S1	0.0676 (6)	0.0489 (5)	0.0494 (5)	0.0139 (4)	-0.0159 (4)	-0.0140 (4)

*Geometric parameters (Å, °)*

C1—C6	1.374 (4)	C7—C8	1.510 (5)
C1—C2	1.378 (4)	C8—C10	1.481 (6)
C1—S1	1.753 (3)	C8—C9	1.500 (6)
C2—C3	1.370 (5)	C8—H8	0.9800
C2—H2	0.9300	C9—H9A	0.9600
C3—C4	1.372 (5)	C9—H9B	0.9600
C3—H3	0.9300	C9—H9C	0.9600
C4—C5	1.368 (5)	C10—H10A	0.9600
C4—Cl1	1.736 (4)	C10—H10B	0.9600
C5—C6	1.368 (5)	C10—H10C	0.9600
C5—H5	0.9300	N1—S1	1.639 (3)
C6—H6	0.9300	N1—H1N	0.840 (18)

C7—O3	1.199 (4)	O1—S1	1.433 (2)
C7—N1	1.380 (4)	O2—S1	1.425 (2)
C6—C1—C2	120.7 (3)	C10—C8—H8	107.9
C6—C1—S1	119.9 (3)	C9—C8—H8	107.9
C2—C1—S1	119.4 (2)	C7—C8—H8	107.9
C3—C2—C1	119.5 (3)	C8—C9—H9A	109.5
C3—C2—H2	120.2	C8—C9—H9B	109.5
C1—C2—H2	120.2	H9A—C9—H9B	109.5
C2—C3—C4	119.2 (3)	C8—C9—H9C	109.5
C2—C3—H3	120.4	H9A—C9—H9C	109.5
C4—C3—H3	120.4	H9B—C9—H9C	109.5
C5—C4—C3	121.5 (3)	C8—C10—H10A	109.5
C5—C4—C11	119.8 (3)	C8—C10—H10B	109.5
C3—C4—C11	118.7 (3)	H10A—C10—H10B	109.5
C6—C5—C4	119.2 (3)	C8—C10—H10C	109.5
C6—C5—H5	120.4	H10A—C10—H10C	109.5
C4—C5—H5	120.4	H10B—C10—H10C	109.5
C5—C6—C1	119.8 (3)	C7—N1—S1	125.8 (2)
C5—C6—H6	120.1	C7—N1—H1N	122 (3)
C1—C6—H6	120.1	S1—N1—H1N	110 (3)
O3—C7—N1	121.2 (3)	O2—S1—O1	119.07 (16)
O3—C7—C8	125.4 (3)	O2—S1—N1	110.56 (16)
N1—C7—C8	113.3 (3)	O1—S1—N1	103.75 (15)
C10—C8—C9	113.5 (4)	O2—S1—C1	108.79 (15)
C10—C8—C7	111.6 (4)	O1—S1—C1	108.78 (17)
C9—C8—C7	107.7 (4)	N1—S1—C1	104.95 (15)
C6—C1—C2—C3	0.0 (5)	N1—C7—C8—C9	-87.3 (4)
S1—C1—C2—C3	-179.3 (3)	O3—C7—N1—S1	-7.0 (5)
C1—C2—C3—C4	0.7 (6)	C8—C7—N1—S1	170.6 (3)
C2—C3—C4—C5	-0.5 (6)	C7—N1—S1—O2	54.8 (3)
C2—C3—C4—C11	179.9 (3)	C7—N1—S1—O1	-176.4 (3)
C3—C4—C5—C6	-0.6 (6)	C7—N1—S1—C1	-62.3 (3)
C11—C4—C5—C6	179.1 (3)	C6—C1—S1—O2	2.9 (3)
C4—C5—C6—C1	1.4 (6)	C2—C1—S1—O2	-177.8 (3)
C2—C1—C6—C5	-1.1 (5)	C6—C1—S1—O1	-128.3 (3)
S1—C1—C6—C5	178.3 (3)	C2—C1—S1—O1	51.1 (3)
O3—C7—C8—C10	-35.0 (6)	C6—C1—S1—N1	121.2 (3)
N1—C7—C8—C10	147.5 (4)	C2—C1—S1—N1	-59.4 (3)
O3—C7—C8—C9	90.2 (5)		

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1N $\cdots$ O1 <sup>i</sup>	0.84 (2)	2.08 (2)	2.912 (4)	169 (3)

Symmetry code: (i)  $-x, -y+1, -z+2$ .