

Poly[[μ -1,4-bis(4,5-dihydro-1,3-oxazol-2-yl)benzene- κ^2 N:N']di- μ -bromido-cadmium]

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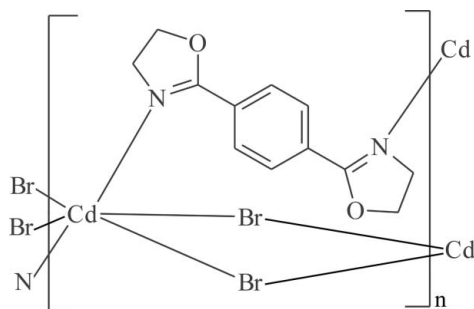
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.026; wR factor = 0.067; data-to-parameter ratio = 48.1.

In the title coordination polymer, $[\text{CdBr}_2(\text{C}_{12}\text{H}_{12}\text{N}_2\text{O}_2)]_n$, the Cd^{II} ion, situated on an inversion centre, is coordinated by four bridging Br atoms and two N atoms from two 1,4-bis(4,5-dihydro-1,3-oxazol-2-yl)benzene (*L*) ligands in a distorted octahedral geometry. The *L* ligand, which also lies across an inversion centre, bridges two Cd^{II} ions, forming layers parallel to (010).

Related literature

For background to coordination polymers with organic ligands, see: Chiang *et al.* (2008); Hsu *et al.* (2009); Kitagawa *et al.* (2004); Yeh *et al.* (2008, 2009). For Cd(II) coordination polymers, see: Suen & Wang (2007*a,b*). For related structures, see: Wang *et al.* (2008, 2011).



Experimental

Crystal data

 $[\text{CdBr}_2(\text{C}_{12}\text{H}_{12}\text{N}_2\text{O}_2)]$
 $M_r = 488.46$

Triclinic, $P\bar{1}$
 $a = 4.0595$ (2) Å
 $b = 8.1114$ (3) Å
 $c = 10.1132$ (4) Å
 $\alpha = 84.503$ (2)°
 $\beta = 81.963$ (2)°
 $\gamma = 84.898$ (2)°

$V = 327.26$ (2) Å³
 $Z = 1$
 Mo $K\alpha$ radiation
 $\mu = 7.77$ mm⁻¹
 $T = 296$ K
 $0.16 \times 0.06 \times 0.06$ mm

Data collection

Bruker APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2001)
 $T_{\text{min}} = 0.769$, $T_{\text{max}} = 0.971$

15213 measured reflections
 4234 independent reflections
 3127 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.081$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.026$
 $wR(F^2) = 0.067$
 $S = 0.96$
 4234 reflections

88 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.99$ e Å⁻³
 $\Delta\rho_{\text{min}} = -1.55$ e Å⁻³

Table 1

Selected bond lengths (Å).

Cd—N	2.5189 (12)	Cd—Br ⁱ	2.7901 (2)
Cd—Br	2.7085 (2)		

 Symmetry code: (i) $x - 1, y, z$.

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: DIAMOND (Brandenburg, 1999); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HY2448).

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supporting information

Acta Cryst. (2011). E67, m1099 [doi:10.1107/S1600536811027759]

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S1. Comment

The synthesis of metal coordination polymers has been a subject of intense research due to their interesting structural chemistry and potential applications in gas storage, separation, catalysis, magnetism, luminescence, and drug delivery (Kitagawa *et al.*, 2004). Roles of anions, solvents and ligand conformations in the self-assembly of coordination complexes containing polydentate nitrogen ligands are very interesting (Chiang *et al.*, 2008; Hsu *et al.*, 2009; Yeh *et al.*, 2008, 2009). The Cd(II) complexes containing polydentate ligands showing various types of frameworks are also reported (Suen & Wang, 2007a,b). The Ag(I) and Cu(II) complexes containing 1,4-bis(4,5-dihydro-2-oxazolyl)benzene (*L*) ligands have been reported, which show various one- and two-dimensional networks (Wang, Lee *et al.*, 2008; Wang, Yeh *et al.*, 2011).

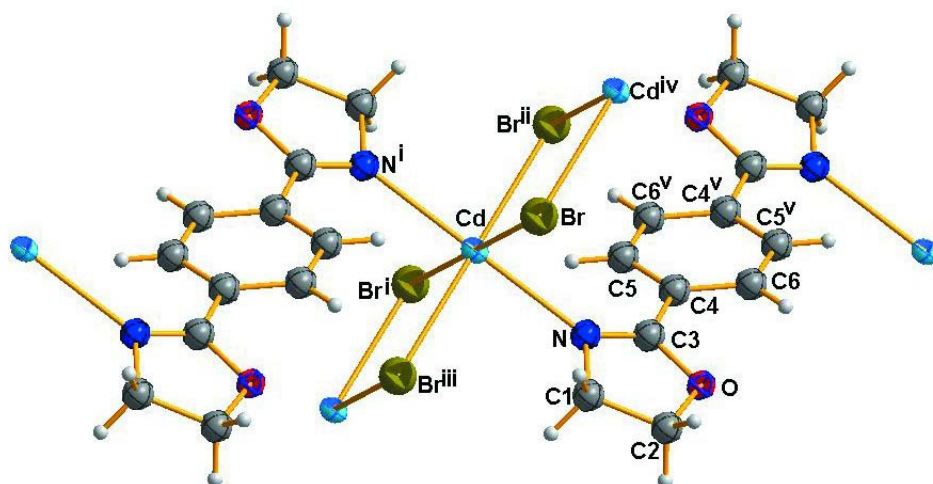
In the title complex, the Cd^{II} ion is six-coordinated with four Br atoms and two N atoms from two *L* ligands (Fig. 1, Table 1). The Cd \cdots Cd distances separated by the bridging *L* ligands and Br atoms are 10.3574 (4) and 4.0595 (2) Å. The ligand adopts an *anti* conformation in the structure (Fig. 2).

S2. Experimental

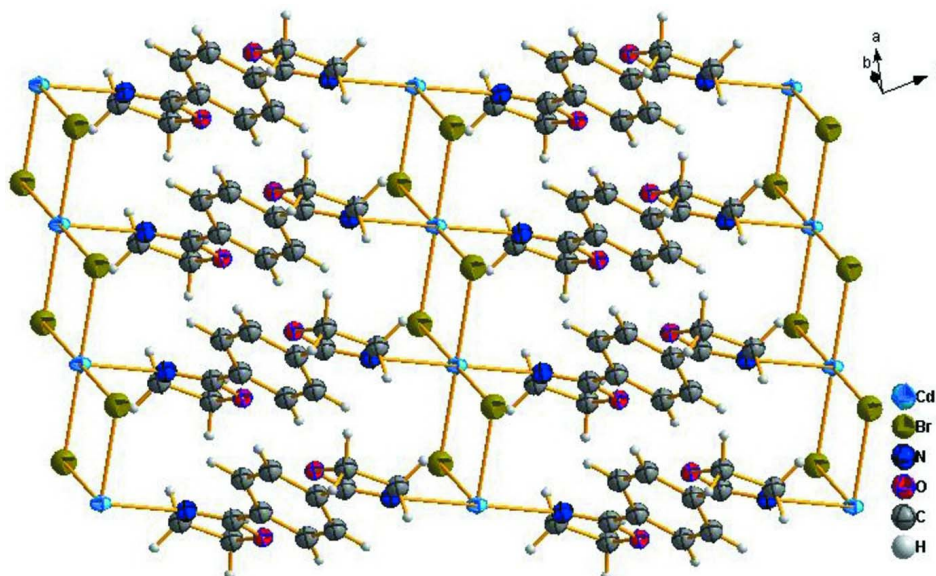
An aqueous solution (5.0 ml) of cadmium bromide (1.0 mmol) was layered carefully over a methanolic solution (5.0 ml) of 1,4-bis(4,5-dihydro-2-oxazolyl)benzene (1.0 mmol) in a tube. Colourless crystals were obtained after several weeks. These were washed with methanol and collected in 69.8% yield.

S3. Refinement

H atoms were positioned geometrically and refined as riding atoms, with C—H = 0.93 (phenyl) and 0.97 (methylene) Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.


Figure 1

A portion of the two-dimensional network in the title compound. Displacement ellipsoids are drawn at the 30% probability level. [Symmetry codes: (i) $-x + 1, -y + 1, -z$; (ii) $-x, -y + 1, -z$; (iii) $x + 1, y, z$; (iv) $x - 1, y, z$; (v) $-x, -y + 1, -z + 1$.]


Figure 2

A drawing of the two-dimensional network.

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Crystal data

[CdBr₂(C₁₂H₁₂N₂O₂)]

$M_r = 488.46$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 4.0595$ (2) Å

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$c = 10.1132$ (4) Å

$\alpha = 84.503$ (2)°

$\beta = 81.963$ (2)°

$\gamma = 84.898$ (2)°

$V = 327.26$ (2) Å³

$Z = 1$

$F(000) = 232$

$D_x = 2.478$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
 Cell parameters from 7163 reflections
 $\theta = 3.1\text{--}41.1^\circ$
 $\mu = 7.77 \text{ mm}^{-1}$

$T = 296 \text{ K}$
 Columnar, colourless
 $0.16 \times 0.06 \times 0.06 \text{ mm}$

Data collection

Bruker APEXII CCD
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 φ and ω scans
 Absorption correction: multi-scan
 (SADABS; Bruker, 2001)
 $T_{\min} = 0.769$, $T_{\max} = 0.971$

15213 measured reflections
 4234 independent reflections
 3127 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.081$
 $\theta_{\max} = 41.1^\circ$, $\theta_{\min} = 3.1^\circ$
 $h = -7 \rightarrow 6$
 $k = -14 \rightarrow 14$
 $l = -18 \rightarrow 18$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.026$
 $wR(F^2) = 0.067$
 $S = 0.96$
 4234 reflections
 88 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0224P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.99 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -1.55 \text{ e \AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Cd	0.5000	0.0000	0.0000	0.02232 (4)
N	0.4635 (3)	-0.21212 (15)	0.19936 (11)	0.0222 (2)
O	0.2887 (3)	-0.39410 (14)	0.37175 (11)	0.0361 (3)
C1	0.5970 (4)	-0.37792 (18)	0.15877 (14)	0.0282 (3)
H1B	0.8389	-0.3851	0.1436	0.034*
H1A	0.5149	-0.4007	0.0773	0.034*
C2	0.4731 (5)	-0.4991 (2)	0.27480 (16)	0.0333 (3)
H2B	0.3300	-0.5751	0.2468	0.040*
H2A	0.6583	-0.5627	0.3115	0.040*
C3	0.3062 (4)	-0.23541 (17)	0.31777 (13)	0.0212 (2)
C4	0.1447 (3)	-0.11180 (17)	0.40873 (12)	0.0200 (2)
C5	0.2347 (4)	0.05114 (18)	0.39301 (13)	0.0227 (2)
H5	0.3917	0.0853	0.3219	0.027*
C6	-0.0891 (4)	-0.16227 (18)	0.51607 (13)	0.0229 (2)
H6	-0.1479	-0.2714	0.5267	0.027*
Br	0.91841 (3)	0.179762 (17)	0.103470 (13)	0.02270 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cd	0.01836 (6)	0.02238 (7)	0.02671 (7)	-0.00089 (5)	-0.00254 (4)	-0.00583 (5)
N	0.0272 (5)	0.0202 (5)	0.0181 (5)	-0.0007 (4)	0.0012 (4)	-0.0038 (4)

O	0.0540 (8)	0.0203 (5)	0.0268 (5)	0.0041 (5)	0.0134 (5)	0.0000 (4)
C1	0.0369 (8)	0.0205 (6)	0.0237 (6)	0.0029 (5)	0.0056 (5)	-0.0034 (5)
C2	0.0432 (9)	0.0211 (7)	0.0304 (7)	0.0038 (6)	0.0093 (6)	-0.0029 (6)
C3	0.0253 (6)	0.0189 (6)	0.0187 (5)	-0.0003 (5)	-0.0008 (4)	-0.0026 (4)
C4	0.0234 (6)	0.0216 (6)	0.0149 (5)	0.0007 (5)	-0.0018 (4)	-0.0036 (4)
C5	0.0269 (6)	0.0241 (6)	0.0160 (5)	-0.0025 (5)	0.0020 (4)	-0.0019 (4)
C6	0.0288 (6)	0.0205 (6)	0.0188 (5)	-0.0031 (5)	0.0005 (4)	-0.0032 (4)
Br	0.02070 (7)	0.02520 (7)	0.02242 (7)	-0.00113 (5)	-0.00048 (4)	-0.00718 (5)

Geometric parameters (Å, °)

Cd—N	2.5189 (12)	C2—H2B	0.9700
Cd—Br	2.7085 (2)	C2—H2A	0.9700
Cd—Br ⁱ	2.7901 (2)	C3—C4	1.4739 (17)
N—C3	1.2813 (17)	C4—C5	1.3910 (19)
N—C1	1.4781 (18)	C4—C6	1.3940 (19)
O—C3	1.3530 (18)	C5—C6 ⁱⁱ	1.3855 (18)
O—C2	1.4451 (17)	C5—H5	0.9300
C1—C2	1.516 (2)	C6—C5 ⁱⁱ	1.3855 (18)
C1—H1B	0.9700	C6—H6	0.9300
C1—H1A	0.9700		
N—Cd—N ⁱⁱⁱ	180.00 (6)	N—C1—H1A	110.7
N—Cd—Br ⁱⁱⁱ	86.94 (3)	C2—C1—H1A	110.7
N ⁱⁱⁱ —Cd—Br ⁱⁱⁱ	93.06 (3)	H1B—C1—H1A	108.8
N—Cd—Br	93.06 (3)	O—C2—C1	103.97 (11)
N ⁱⁱⁱ —Cd—Br	86.94 (3)	O—C2—H2B	111.0
Br ⁱⁱⁱ —Cd—Br	180.000 (5)	C1—C2—H2B	111.0
N—Cd—Br ^{iv}	87.67 (3)	O—C2—H2A	111.0
N ⁱⁱⁱ —Cd—Br ^{iv}	92.33 (3)	C1—C2—H2A	111.0
Br ⁱⁱⁱ —Cd—Br ^{iv}	95.159 (5)	H2B—C2—H2A	109.0
Br—Cd—Br ^{iv}	84.841 (5)	N—C3—O	117.42 (12)
N—Cd—Br ⁱ	92.33 (3)	N—C3—C4	129.11 (13)
N ⁱⁱⁱ —Cd—Br ⁱ	87.67 (3)	O—C3—C4	113.44 (11)
Br ⁱⁱⁱ —Cd—Br ⁱ	84.841 (5)	C5—C4—C6	119.85 (12)
Br—Cd—Br ⁱ	95.159 (5)	C5—C4—C3	121.06 (11)
Br ^{iv} —Cd—Br ⁱ	180.000 (6)	C6—C4—C3	119.00 (12)
C3—N—C1	106.43 (12)	C6 ⁱⁱ —C5—C4	119.60 (12)
C3—N—Cd	140.65 (10)	C6 ⁱⁱ —C5—H5	120.2
C1—N—Cd	110.60 (8)	C4—C5—H5	120.2
C3—O—C2	106.97 (11)	C5 ⁱⁱ —C6—C4	120.55 (13)
N—C1—C2	105.15 (11)	C5 ⁱⁱ —C6—H6	119.7
N—C1—H1B	110.7	C4—C6—H6	119.7
C2—C1—H1B	110.7	Cd—Br—Cd ^v	95.159 (5)

Symmetry codes: (i) $x-1, y, z$; (ii) $-x, -y, -z+1$; (iii) $-x+1, -y, -z$; (iv) $-x+2, -y, -z$; (v) $x+1, y, z$.