

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

1-(2-Naphthyl)-3-phenyl-3-(4,5,6,7-tetrahydro-1,2,3-benzoselenadiazol-4-yl)-propan-1-one

J. Muthukumar,^a M. Nachiappan,^b S. Chitra,^c
 P. Manisankar,^c Suman Bhattacharya,^d S.
 Muthusubramanian,^e R. Krishna^{a‡} and J. Jeyakanthan^{b*}

^aCentre for Bioinformatics, School of Life Sciences, Pondicherry University, Puducherry 605 014, India, ^bDepartment of Bioinformatics, Alagappa University, Karaikudi 630 003, India, ^cDepartment of Industrial Chemistry, Alagappa University, Karaikudi 630 003, India, ^dDepartment of Chemistry, Pondicherry University, Puducherry 605 014, India, and ^eDepartment of Organic Chemistry, Madurai Kamaraj University, Madurai 625 021, India
 Correspondence e-mail: jjkanthan@gmail.com

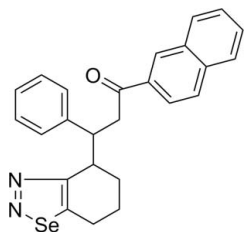
Received 11 May 2011; accepted 6 July 2011

Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.038; wR factor = 0.096; data-to-parameter ratio = 13.7.

In the title compound, $\text{C}_{25}\text{H}_{22}\text{N}_2\text{OSe}$, the fused six-membered cyclohexene ring of the 4,5,6,7-tetrahydro-1,2,3-benzoselenadiazole group adopts a near half-chair conformation and the five-membered 1,2,3-selenadiazole ring is essentially planar (r.m.s. deviation = 0.004 Å). There are weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\pi$ interactions in the crystal structure. Intermolecular $\pi-\pi$ stacking is also observed between the naphthyl units, with a centroid-centroid distance of 3.529 (15) Å.

Related literature

For the biological importance of 1,2,3-selenadiazole derivatives, see: Kuroda *et al.* (2001); El-Bahaie *et al.* (1990); El-Kashef *et al.* (1986); Plano *et al.* (2010). For ring puckering analysis, see: Cremer & Pople (1975). For synthetic procedures, see: Al Arab (1989); Li *et al.* (2003); Qian *et al.* (2008); Xu *et al.* (2009).



‡ Additional correspondence author, e-mail: krishstrucbio@gmail.com.

Experimental

Crystal data

$\text{C}_{25}\text{H}_{22}\text{N}_2\text{OSe}$
 $M_r = 445.41$
 Triclinic, $P\bar{1}$
 $a = 8.0600$ (4) Å
 $b = 10.1215$ (5) Å
 $c = 13.0827$ (6) Å
 $\alpha = 81.479$ (4)°
 $\beta = 76.478$ (4)°
 $\gamma = 82.087$ (4)°
 $V = 1020.33$ (9) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 1.86$ mm⁻¹
 $T = 293$ K
 $0.4 \times 0.3 \times 0.2$ mm

Data collection

Oxford Diffraction Xcalibur E diffractometer
 Absorption correction: multi-scan (*CrysAlis PRO*; Oxford Diffraction, 2009)
 $T_{\min} = 0.658$, $T_{\max} = 1.000$
 6762 measured reflections
 3594 independent reflections
 2770 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.025$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.096$
 $S = 1.01$
 3594 reflections
 262 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.29$ e Å⁻³
 $\Delta\rho_{\min} = -0.41$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg is the centroid of the C16–C21 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C20}-\text{H20}\cdots\text{O1}^i$	0.93	2.56	3.381 (3)	147
$\text{C22}-\text{H22}\cdots\text{O1}^i$	0.93	2.58	3.392 (4)	146
$\text{C4}-\text{H4A}\cdots\text{Cg}^{ii}$	0.97	2.63	3.584 (3)	167

Symmetry codes: (i) $x - 1, y, z$; (ii) $-x, -y + 1, -z + 1$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2009); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *PLATON*.

RK thanks the Centre for Bioinformatics [funded by the Department of Biotechnology (DBT) and the Department of Information Technology (DIT)], Pondicherry University, for providing computational facilities to carry out this research work. JJ thanks Dr Binoy Krishna Saha, Assistant Professor, Department of Chemistry, Pondicherry University, for providing the X-ray facilities. JM thanks the Council for Scientific and Industrial Research (CSIR) for Senior Research Fellowship (SRF).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: GK2375).

References

- Al Arab, M. M. (1989). *J. Chem. Soc. Pak.* **11**, 321–326.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- El-Bahaie, S., Assy, M. G. & Hassanien, M. M. (1990). *Pharmazie*, **45**, 791–793.
- El-Kashef, H. S., E-Bayoumy, B. & Aly, T. I. (1986). *Egypt. J. Pharm. Sci.* **27**, 27–30.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Kuroda, K., Uchikurohane, T., Tajima, S. & Tsubata, K. (2001). US Patent 6 166 054.
- Li, J. T., Chen, G.-F., Xu, W.-Z. & Li, T.-S. (2003). *Ultrason. Sonochem.* **10**, 115–118.
- Oxford Diffraction (2009). *CrysAlis PRO*. Oxford Diffraction Ltd, Yarnton, England.
- Plano, D., Moreno, E., Font, M., Encío, I., Palop, J. A. & Sanmartín, C. (2010). *Arch. Pharm.* **343**, 680–691.
- Qian, Y., Xiao, S., Liu, L. & Wang, Y. (2008). *Tetrahedron Asymmetry*, **19**, 1515–1518.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
- Xu, D.-Z., Shi, S., Liu, Y. & Wang, Y. (2009). *Tetrahedron*, **65**, 9344–9349.

supporting information

Acta Cryst. (2011). E67, o2010–o2011 [doi:10.1107/S1600536811027103]

1-(2-Naphthyl)-3-phenyl-3-(4,5,6,7-tetrahydro-1,2,3-benzoselenadiazol-4-yl)propan-1-one

J. Muthukumar, M. Nachiappan, S. Chitra, P. Manisankar, Suman Bhattacharya, S. Muthusubramanian, R. Krishna and J. Jeyakanthan

S1. Comment

Selenium containing heterocyclic compounds are of interest due to their biological applications. They possess various beneficial activities such as anti-fungal (Kuroda *et al.*, 2001), anti-bacterial (El-Kashef *et al.*, 1986), anti-microbial (El-Bahaie *et al.*, 1990), and anti-cancer (Plano *et al.*, 2010). Considering the importances of 1,2,3-selenadiazole derivatives, we report the structure of the title compound (Fig. 1). The five-membered 1,2,3-selenadiazole moiety (C1/N1/N2/Se1/C2) of the title compound adopts a planar conformation (r.m.s. deviation = 0.004 Å). Cremer & Pople puckering analysis (Cremer & Pople, 1975) cannot be performed to this moiety, for its weighted average absolute torsion angle is 0.62°, which is less than 5.0°. However, the fused six-membered ring (C1/C2/C3/C4/C5/C6) of 4,5,6,7-tetrahydrobenzo[*d*][1,2,3]selenadiazole group adopts a near to half-chair conformation with puckering parameters of $Q = 0.489$ (3) Å, $\theta = 48.9$ (4)° and $\Phi = 218.1$ (5)°. The stabilization of crystal packing is mainly governed by intermolecular hydrogen bonding interactions such as C20–H20⋯O1 ($x - 1, y, z$) and C22–H22⋯O1 (Fig. 2). The C–H⋯ π interaction (Fig. 3) is observed between C4–H4A⋯C_g ($-x, 1 - y, 1 - z$) [C_g is the centroid of C16–C21 ring, C⋯C_g distance: 3.584 (3) Å, H⋯C_g 2.63 Å, H-Perp: 2.61 Å], which also contribute to the crystal packing. The crystal packing also exhibits intermolecular π - π stacking interaction between naphthyl units with a centroid-centroid distance of 3.529 (15) Å.

S2. Experimental

Diastereomerically pure racemic cyclic 1,5-diketones, (2*S**, 1*R**)-2-[3-(2-naphthyl)-3-oxo-1-phenylpropyl]cyclohexan-1-one, were obtained via a simple Michael addition of cyclohexanone with substituted chalcones by a known method (Al Arab, 1989; Li *et al.*, 2003; Xu *et al.*, 2009; Qian *et al.*, 2008). A mixture of (2*S**, 1*R**)-2-[3-(2-naphthyl)-3-oxo-1-phenylpropyl]-1-cyclohexanone (1 mmol, 0.36 g) and semicarbazide hydrochloride (1 mmol, 0.11 g) in ethanol (10 ml) was refluxed for 3 h. After completion of the reaction as monitored by TLC, the mixture was poured into ice-cold water (50 ml) and the resulting mono-semicarbazone solid was filtered off. The mono-semicarbazone (1 mmol, 0.41 g) and SeO₂ (2 mmol, 0.44 g) in THF (10 ml) was refluxed on a water bath for 30 minutes and on completion of the reaction the mixture was filtered to remove selenium powder. The filtrate was concentrated under vacuum, and the residue was subjected to column chromatography using petroleum ether/ethyl acetate mixture (95:5; v/v) as eluent to afford the racemic mixture of the pure product (Yield: 70%, m.p.: 404–405 K). Dissolving the pure compound in 3:1 mixture of dichloromethane/ethyl acetate and on keeping for slow evaporation of the solvents provides crystals for X-ray analysis.

S3. Refinement

The hydrogen atoms were placed in calculated positions (C–H = 0.93–0.98 Å) and included in the refinement in riding-model approximation with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

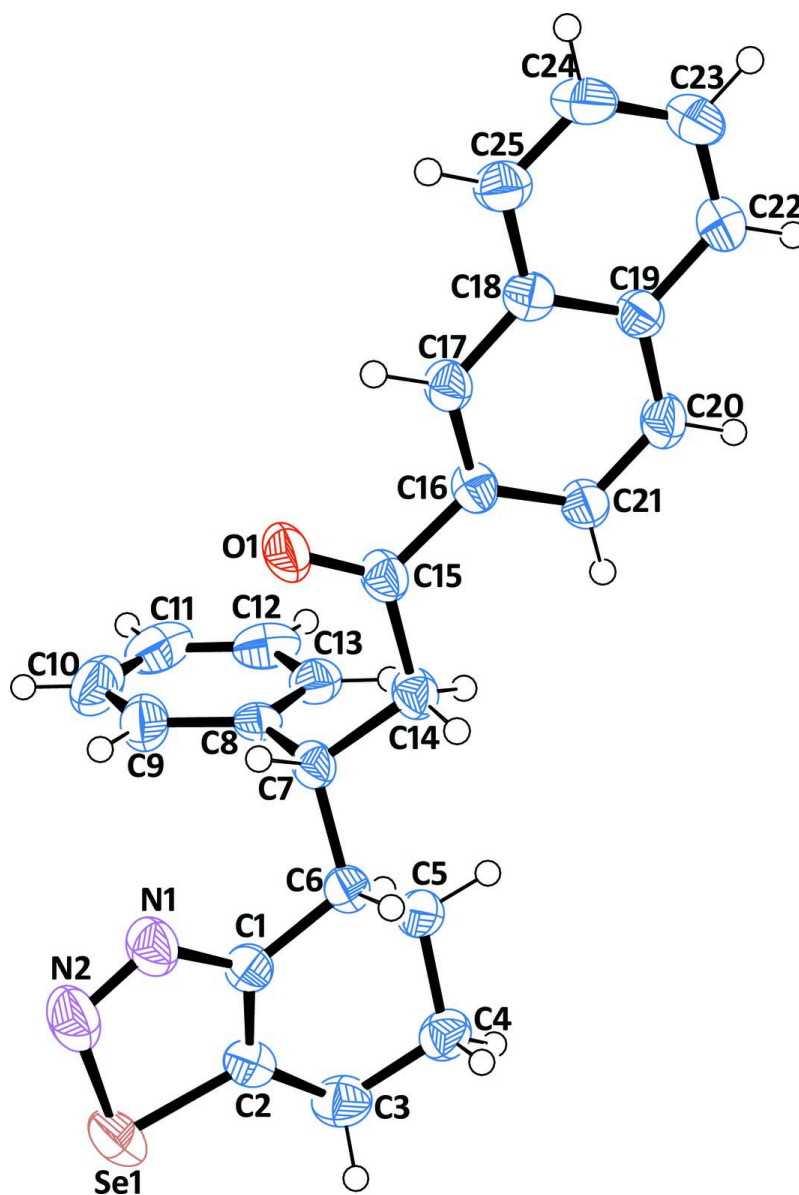


Figure 1

Molecular structure of the title compound, showing displacement ellipsoids drawn at the 50% probability level.

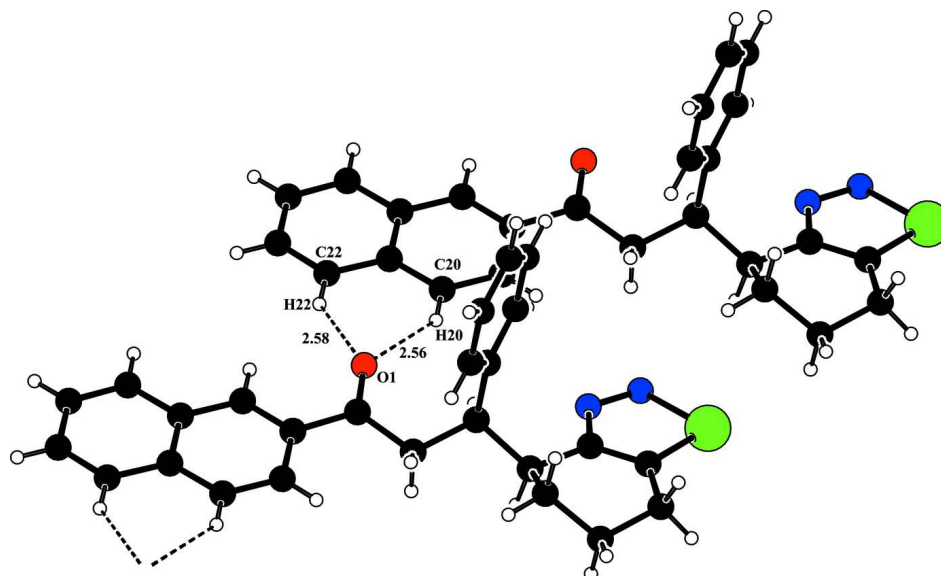
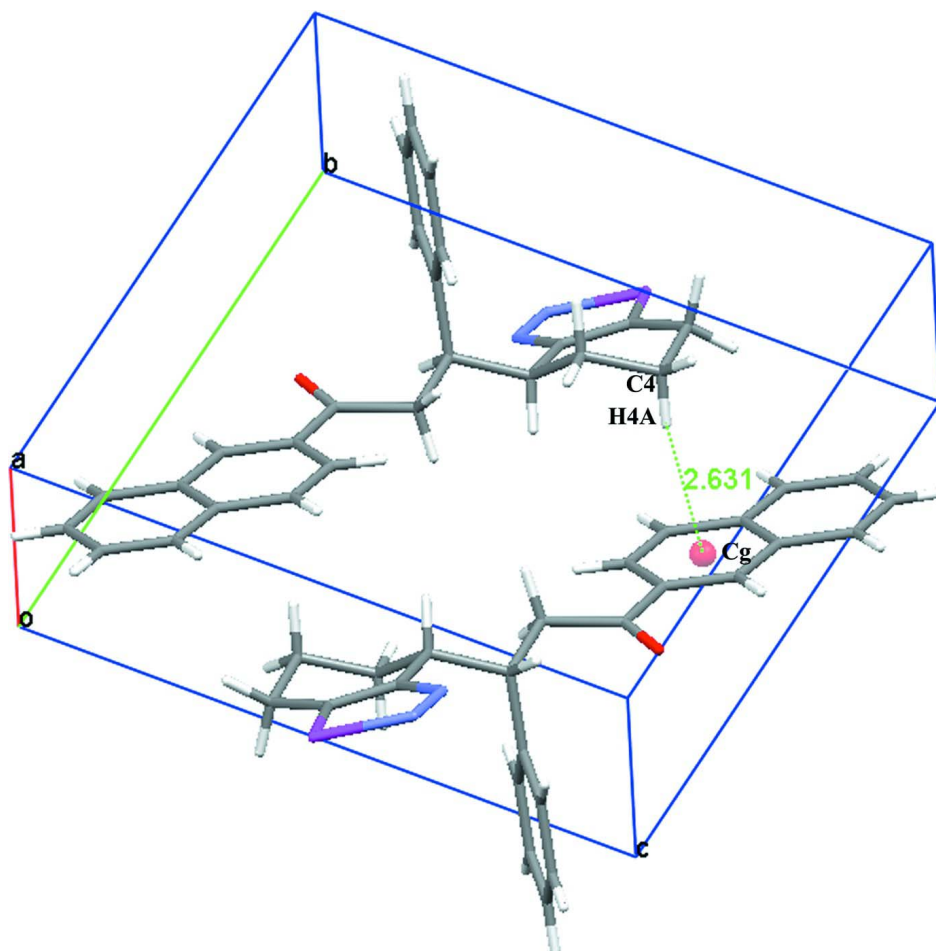


Figure 2

A weak intermolecular C—H...O interaction in title compound.

**Figure 3**

A weak intermolecular C—H... π interaction in title compound. *Cg* is the centroid of C16—C21 ring.

1-(2-Naphthyl)-3-phenyl-3-(4,5,6,7-tetrahydro-1,2,3-benzoselenadiazol-4-yl)propan-1-one

Crystal data

$C_{25}H_{22}N_2OSe$

$M_r = 445.41$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 8.0600$ (4) Å

$b = 10.1215$ (5) Å

$c = 13.0827$ (6) Å

$\alpha = 81.479$ (4)°

$\beta = 76.478$ (4)°

$\gamma = 82.087$ (4)°

$V = 1020.33$ (9) Å³

$Z = 2$

$F(000) = 456$

$D_x = 1.450$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 3065 reflections

$\theta = 2.6$ – 29.3 °

$\mu = 1.86$ mm⁻¹

$T = 293$ K

Block, blue

$0.4 \times 0.3 \times 0.2$ mm

Data collection

Oxford Diffraction Xcalibur E
diffractometer

Radiation source: fine-focus sealed tube
Graphite monochromator

Detector resolution: 15.9821 pixels mm⁻¹

ω scans

Absorption correction: multi-scan

(*CrysAlis PRO*; Oxford Diffraction, 2009)

$T_{\min} = 0.658$, $T_{\max} = 1.000$
 6762 measured reflections
 3594 independent reflections
 2770 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.025$

$\theta_{\max} = 25.0^\circ$, $\theta_{\min} = 2.6^\circ$
 $h = -9 \rightarrow 9$
 $k = -12 \rightarrow 12$
 $l = -12 \rightarrow 15$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.096$
 $S = 1.01$
 3594 reflections
 262 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.050P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.29 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.41 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Se1	0.62678 (4)	0.82915 (3)	0.61536 (3)	0.06072 (16)
O1	0.3614 (3)	0.5213 (2)	0.20388 (17)	0.0633 (6)
N1	0.5844 (3)	0.6746 (2)	0.48187 (19)	0.0476 (6)
N2	0.7071 (3)	0.7047 (3)	0.5160 (2)	0.0587 (7)
C1	0.4227 (3)	0.7366 (2)	0.5202 (2)	0.0344 (6)
C2	0.4117 (4)	0.8232 (3)	0.5918 (2)	0.0395 (7)
C3	0.2505 (4)	0.8991 (3)	0.6443 (2)	0.0485 (7)
H3A	0.2499	0.9934	0.6163	0.058*
H3B	0.2451	0.8916	0.7197	0.058*
C4	0.0939 (4)	0.8443 (3)	0.62557 (19)	0.0422 (7)
H4A	0.0714	0.7629	0.6736	0.051*
H4B	-0.0057	0.9098	0.6406	0.051*
C5	0.1215 (3)	0.8137 (3)	0.5122 (2)	0.0398 (6)
H5A	0.1464	0.8947	0.4642	0.048*
H5B	0.0169	0.7858	0.5016	0.048*
C6	0.2686 (3)	0.7034 (3)	0.48589 (19)	0.0339 (6)
H6	0.2328	0.6210	0.5303	0.041*
C7	0.3076 (3)	0.6729 (2)	0.36983 (19)	0.0331 (6)
H7	0.4113	0.6089	0.3604	0.040*
C8	0.3471 (3)	0.7946 (3)	0.28883 (18)	0.0334 (6)

C9	0.5157 (4)	0.8233 (3)	0.2486 (2)	0.0480 (7)
H9	0.6049	0.7671	0.2711	0.058*
C10	0.5519 (5)	0.9349 (3)	0.1752 (2)	0.0606 (9)
H10	0.6651	0.9530	0.1493	0.073*
C11	0.4227 (5)	1.0187 (3)	0.1405 (2)	0.0601 (9)
H11	0.4479	1.0933	0.0912	0.072*
C12	0.2555 (5)	0.9919 (3)	0.1790 (2)	0.0533 (8)
H12	0.1669	1.0484	0.1560	0.064*
C13	0.2194 (4)	0.8812 (3)	0.2518 (2)	0.0415 (7)
H13	0.1057	0.8639	0.2770	0.050*
C14	0.1634 (3)	0.6015 (3)	0.3511 (2)	0.0376 (6)
H14A	0.1320	0.5334	0.4105	0.045*
H14B	0.0635	0.6662	0.3483	0.045*
C15	0.2124 (4)	0.5363 (3)	0.2507 (2)	0.0375 (6)
C16	0.0773 (3)	0.4844 (2)	0.2104 (2)	0.0334 (6)
C17	0.1294 (3)	0.4041 (2)	0.13129 (19)	0.0358 (6)
H17	0.2463	0.3836	0.1048	0.043*
C18	0.0114 (3)	0.3514 (2)	0.08876 (19)	0.0340 (6)
C19	-0.1658 (3)	0.3833 (3)	0.12857 (19)	0.0352 (6)
C20	-0.2175 (3)	0.4677 (3)	0.2103 (2)	0.0410 (7)
H20	-0.3339	0.4906	0.2366	0.049*
C21	-0.1000 (3)	0.5157 (3)	0.2508 (2)	0.0367 (6)
H21	-0.1367	0.5694	0.3055	0.044*
C22	-0.2856 (4)	0.3321 (3)	0.0848 (2)	0.0485 (8)
H22	-0.4026	0.3538	0.1098	0.058*
C23	-0.2302 (4)	0.2516 (3)	0.0068 (2)	0.0550 (9)
H23	-0.3095	0.2182	-0.0214	0.066*
C24	-0.0545 (4)	0.2186 (3)	-0.0316 (2)	0.0532 (8)
H24	-0.0184	0.1623	-0.0845	0.064*
C25	0.0639 (4)	0.2670 (3)	0.0068 (2)	0.0441 (7)
H25	0.1802	0.2450	-0.0205	0.053*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Se1	0.0506 (2)	0.0778 (3)	0.0659 (2)	-0.01457 (18)	-0.02624 (18)	-0.01871 (18)
O1	0.0281 (12)	0.0930 (17)	0.0786 (15)	-0.0050 (11)	-0.0061 (11)	-0.0520 (13)
N1	0.0352 (14)	0.0520 (15)	0.0575 (15)	0.0042 (12)	-0.0141 (12)	-0.0141 (12)
N2	0.0365 (15)	0.0715 (18)	0.0714 (18)	0.0003 (13)	-0.0198 (14)	-0.0115 (14)
C1	0.0340 (16)	0.0341 (14)	0.0349 (14)	-0.0014 (12)	-0.0092 (12)	-0.0025 (12)
C2	0.0424 (17)	0.0410 (15)	0.0380 (15)	-0.0054 (13)	-0.0138 (13)	-0.0049 (12)
C3	0.056 (2)	0.0458 (17)	0.0457 (17)	-0.0006 (15)	-0.0107 (15)	-0.0160 (14)
C4	0.0400 (17)	0.0445 (16)	0.0394 (16)	0.0021 (14)	-0.0051 (14)	-0.0081 (13)
C5	0.0312 (15)	0.0499 (16)	0.0382 (15)	0.0044 (13)	-0.0075 (13)	-0.0130 (13)
C6	0.0320 (15)	0.0353 (14)	0.0352 (14)	-0.0048 (12)	-0.0076 (12)	-0.0053 (11)
C7	0.0261 (14)	0.0340 (14)	0.0404 (15)	0.0014 (11)	-0.0080 (12)	-0.0119 (12)
C8	0.0378 (16)	0.0356 (14)	0.0307 (13)	-0.0041 (12)	-0.0073 (12)	-0.0163 (11)
C9	0.0389 (17)	0.0519 (18)	0.0536 (18)	-0.0098 (14)	-0.0043 (15)	-0.0126 (15)

C10	0.061 (2)	0.065 (2)	0.0523 (19)	-0.028 (2)	0.0107 (18)	-0.0137 (17)
C11	0.088 (3)	0.051 (2)	0.0368 (17)	-0.017 (2)	0.0017 (19)	-0.0073 (15)
C12	0.078 (2)	0.0471 (18)	0.0356 (16)	-0.0009 (17)	-0.0170 (17)	-0.0052 (14)
C13	0.0439 (17)	0.0452 (16)	0.0364 (15)	-0.0020 (14)	-0.0086 (13)	-0.0110 (13)
C14	0.0313 (15)	0.0372 (15)	0.0463 (16)	-0.0039 (12)	-0.0066 (13)	-0.0138 (12)
C15	0.0301 (16)	0.0372 (15)	0.0474 (16)	-0.0004 (12)	-0.0108 (14)	-0.0114 (12)
C16	0.0299 (15)	0.0296 (13)	0.0413 (15)	-0.0013 (11)	-0.0090 (12)	-0.0064 (12)
C17	0.0289 (14)	0.0379 (15)	0.0413 (15)	0.0016 (12)	-0.0093 (12)	-0.0089 (12)
C18	0.0374 (16)	0.0317 (14)	0.0331 (14)	-0.0025 (12)	-0.0103 (12)	-0.0010 (11)
C19	0.0359 (16)	0.0373 (14)	0.0337 (14)	-0.0076 (12)	-0.0097 (12)	-0.0018 (12)
C20	0.0276 (15)	0.0471 (16)	0.0471 (16)	-0.0019 (13)	-0.0038 (13)	-0.0106 (13)
C21	0.0330 (15)	0.0361 (14)	0.0428 (15)	-0.0019 (12)	-0.0068 (13)	-0.0141 (12)
C22	0.0384 (17)	0.066 (2)	0.0453 (17)	-0.0154 (15)	-0.0092 (14)	-0.0096 (15)
C23	0.057 (2)	0.071 (2)	0.0461 (17)	-0.0244 (18)	-0.0162 (16)	-0.0134 (16)
C24	0.065 (2)	0.061 (2)	0.0414 (16)	-0.0106 (17)	-0.0159 (16)	-0.0187 (15)
C25	0.0448 (18)	0.0493 (17)	0.0398 (16)	-0.0006 (14)	-0.0099 (14)	-0.0135 (13)

Geometric parameters (Å, °)

Se1—C2	1.840 (3)	C20—C21	1.362 (3)
Se1—N2	1.888 (3)	C20—C19	1.417 (4)
O1—C15	1.214 (3)	C20—H20	0.9300
C1—C2	1.355 (4)	C5—H5A	0.9700
C1—N1	1.381 (3)	C5—H5B	0.9700
C1—C6	1.509 (3)	C13—C12	1.375 (4)
N1—N2	1.264 (3)	C13—H13	0.9300
C14—C15	1.503 (3)	C21—H21	0.9300
C14—C7	1.533 (3)	C18—C25	1.421 (4)
C14—H14A	0.9700	C18—C19	1.411 (4)
C14—H14B	0.9700	C25—C24	1.353 (4)
C15—C16	1.499 (3)	C25—H25	0.9300
C17—C16	1.365 (3)	C9—C10	1.385 (4)
C17—C18	1.404 (3)	C9—H9	0.9300
C17—H17	0.9300	C19—C22	1.419 (3)
C7—C8	1.518 (3)	C22—C23	1.356 (4)
C7—C6	1.546 (3)	C22—H22	0.9300
C7—H7	0.9800	C10—C11	1.369 (5)
C6—C5	1.526 (4)	C10—H10	0.9300
C6—H6	0.9800	C12—C11	1.373 (5)
C8—C13	1.386 (4)	C12—H12	0.9300
C8—C9	1.392 (4)	C3—H3A	0.9700
C16—C21	1.412 (4)	C3—H3B	0.9700
C2—C3	1.488 (4)	C23—C24	1.398 (5)
C4—C5	1.520 (3)	C23—H23	0.9300
C4—C3	1.529 (3)	C24—H24	0.9300
C4—H4A	0.9700	C11—H11	0.9300
C4—H4B	0.9700		

C2—Se1—N2	86.70 (11)	C4—C5—C6	112.0 (2)
C2—C1—N1	116.2 (2)	C4—C5—H5A	109.2
C2—C1—C6	123.1 (2)	C6—C5—H5A	109.2
N1—C1—C6	120.6 (2)	C4—C5—H5B	109.2
N2—N1—C1	117.3 (2)	C6—C5—H5B	109.2
C15—C14—C7	113.2 (2)	H5A—C5—H5B	107.9
C15—C14—H14A	108.9	C12—C13—C8	122.1 (3)
C7—C14—H14A	108.9	C12—C13—H13	119.0
C15—C14—H14B	108.9	C8—C13—H13	119.0
C7—C14—H14B	108.9	C20—C21—C16	120.3 (2)
H14A—C14—H14B	107.8	C20—C21—H21	119.8
O1—C15—C16	119.7 (2)	C16—C21—H21	119.8
O1—C15—C14	120.5 (2)	C17—C18—C25	122.4 (2)
C16—C15—C14	119.8 (2)	C17—C18—C19	119.0 (2)
C16—C17—C18	121.8 (2)	C25—C18—C19	118.6 (2)
C16—C17—H17	119.1	C24—C25—C18	120.3 (3)
C18—C17—H17	119.1	C24—C25—H25	119.9
C8—C7—C14	112.5 (2)	C18—C25—H25	119.9
C8—C7—C6	113.86 (19)	C10—C9—C8	120.6 (3)
C14—C7—C6	110.17 (19)	C10—C9—H9	119.7
C8—C7—H7	106.6	C8—C9—H9	119.7
C14—C7—H7	106.6	C22—C19—C20	122.3 (3)
C6—C7—H7	106.6	C22—C19—C18	119.3 (2)
N1—N2—Se1	110.7 (2)	C20—C19—C18	118.4 (2)
C1—C6—C5	108.8 (2)	C23—C22—C19	120.3 (3)
C1—C6—C7	113.7 (2)	C23—C22—H22	119.9
C5—C6—C7	114.0 (2)	C19—C22—H22	119.9
C1—C6—H6	106.6	C11—C10—C9	120.7 (3)
C5—C6—H6	106.6	C11—C10—H10	119.6
C7—C6—H6	106.6	C9—C10—H10	119.7
C13—C8—C9	117.2 (3)	C11—C12—C13	119.8 (3)
C13—C8—C7	122.2 (2)	C11—C12—H12	120.1
C9—C8—C7	120.6 (3)	C13—C12—H12	120.1
C17—C16—C21	119.2 (2)	C2—C3—C4	110.6 (2)
C17—C16—C15	118.1 (2)	C2—C3—H3A	109.5
C21—C16—C15	122.7 (2)	C4—C3—H3A	109.5
C1—C2—C3	125.2 (2)	C2—C3—H3B	109.5
C1—C2—Se1	109.2 (2)	C4—C3—H3B	109.5
C3—C2—Se1	125.65 (19)	H3A—C3—H3B	108.1
C5—C4—C3	111.4 (2)	C22—C23—C24	120.4 (3)
C5—C4—H4A	109.3	C22—C23—H23	119.8
C3—C4—H4A	109.3	C24—C23—H23	119.8
C5—C4—H4B	109.3	C25—C24—C23	121.2 (3)
C3—C4—H4B	109.3	C25—C24—H24	119.4
H4A—C4—H4B	108.0	C23—C24—H24	119.4
C21—C20—C19	121.2 (3)	C10—C11—C12	119.6 (3)
C21—C20—H20	119.4	C10—C11—H11	120.2
C19—C20—H20	119.4	C12—C11—H11	120.2

C2—C1—N1—N2	-0.6 (4)	C3—C4—C5—C6	63.3 (3)
C6—C1—N1—N2	-178.3 (2)	C1—C6—C5—C4	-49.3 (3)
C7—C14—C15—O1	13.6 (4)	C7—C6—C5—C4	-177.3 (2)
C7—C14—C15—C16	-168.8 (2)	C9—C8—C13—C12	-0.5 (4)
C15—C14—C7—C8	67.1 (3)	C7—C8—C13—C12	179.5 (2)
C15—C14—C7—C6	-164.7 (2)	C19—C20—C21—C16	1.3 (4)
C1—N1—N2—Se1	0.9 (3)	C17—C16—C21—C20	-0.9 (4)
C2—Se1—N2—N1	-0.8 (2)	C15—C16—C21—C20	179.0 (2)
C2—C1—C6—C5	19.6 (3)	C16—C17—C18—C25	-179.6 (2)
N1—C1—C6—C5	-162.9 (2)	C16—C17—C18—C19	0.4 (4)
C2—C1—C6—C7	147.8 (2)	C17—C18—C25—C24	179.9 (3)
N1—C1—C6—C7	-34.7 (3)	C19—C18—C25—C24	-0.1 (4)
C8—C7—C6—C1	-69.0 (3)	C13—C8—C9—C10	0.5 (4)
C14—C7—C6—C1	163.5 (2)	C7—C8—C9—C10	-179.5 (2)
C8—C7—C6—C5	56.5 (3)	C21—C20—C19—C22	-179.9 (2)
C14—C7—C6—C5	-71.0 (3)	C21—C20—C19—C18	-0.9 (4)
C14—C7—C8—C13	39.2 (3)	C17—C18—C19—C22	179.1 (2)
C6—C7—C8—C13	-87.1 (3)	C25—C18—C19—C22	-0.9 (4)
C14—C7—C8—C9	-140.8 (2)	C17—C18—C19—C20	0.0 (3)
C6—C7—C8—C9	92.9 (3)	C25—C18—C19—C20	-179.9 (2)
C18—C17—C16—C21	0.0 (4)	C20—C19—C22—C23	-179.9 (3)
C18—C17—C16—C15	-179.9 (2)	C18—C19—C22—C23	1.0 (4)
O1—C15—C16—C17	9.8 (4)	C8—C9—C10—C11	-0.3 (4)
C14—C15—C16—C17	-167.9 (2)	C8—C13—C12—C11	0.3 (4)
O1—C15—C16—C21	-170.1 (3)	C1—C2—C3—C4	12.4 (4)
C14—C15—C16—C21	12.3 (4)	Se1—C2—C3—C4	-166.41 (18)
N1—C1—C2—C3	-179.1 (2)	C5—C4—C3—C2	-41.7 (3)
C6—C1—C2—C3	-1.5 (4)	C19—C22—C23—C24	-0.2 (4)
N1—C1—C2—Se1	-0.1 (3)	C18—C25—C24—C23	1.0 (4)
C6—C1—C2—Se1	177.50 (18)	C22—C23—C24—C25	-0.8 (5)
N2—Se1—C2—C1	0.47 (19)	C9—C10—C11—C12	0.1 (5)
N2—Se1—C2—C3	179.4 (2)	C13—C12—C11—C10	-0.1 (4)

Hydrogen-bond geometry (\AA , $^\circ$)

Cg is the centroid of the C16—C21 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C20—H20 \cdots O1 ⁱ	0.93	2.56	3.381 (3)	147
C22—H22 \cdots O1 ⁱ	0.93	2.58	3.392 (4)	146
C4—H4A \cdots Cg ⁱⁱ	0.97	2.63	3.584 (3)	167

Symmetry codes: (i) $x-1, y, z$; (ii) $-x, -y+1, -z+1$.