

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

1-Ethyl-1-methyl-3-(2-nitrobenzoyl)thiourea

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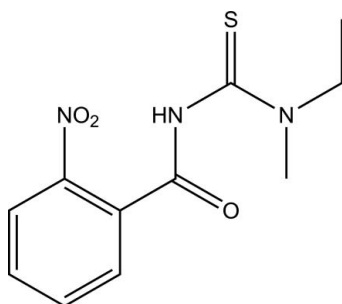
Received 20 June 2011; accepted 22 June 2011

Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.043; wR factor = 0.124; data-to-parameter ratio = 13.6.

In the title compound, $\text{C}_{11}\text{H}_{13}\text{N}_3\text{O}_3\text{S}$, the benzene ring is twisted relative to the amidic fragment, forming a dihedral angle of 27.26 (9)°. The thiono and carbonyl groups are *trans* with respect to the C–N bond. Intermolecular N–H···S and C–H···O hydrogen bonds link the molecules in the crystal structure.

Related literature

For the synthesis, see: Al-abbasi *et al.* (2010). For related structures and background references, see: Shanmuga Sundara Raj *et al.* (1999); Arslan *et al.* (2003); Al-abbasi & Kassim (2011). For standard bond lengths, see: Allen *et al.* (1987) and for bond lengths in other substituted thioureas, see: Nasir *et al.* (2011); Pérez *et al.* (2011).



Experimental

Crystal data

$\text{C}_{11}\text{H}_{13}\text{N}_3\text{O}_3\text{S}$
 $M_r = 267.30$
 Monoclinic, $P2_1/n$
 $a = 11.447$ (2) Å
 $b = 7.8664$ (15) Å
 $c = 15.159$ (3) Å
 $\beta = 107.128$ (4)°

$V = 1304.5$ (4) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.25$ mm⁻¹
 $T = 298$ K
 $0.55 \times 0.38 \times 0.21$ mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2000)
 $T_{\min} = 0.874$, $T_{\max} = 0.949$

7105 measured reflections
 2294 independent reflections
 1971 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.020$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.043$
 $wR(F^2) = 0.124$
 $S = 1.06$
 2294 reflections
 169 parameters
 1 restraint

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.33$ e Å⁻³
 $\Delta\rho_{\min} = -0.19$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--|----------|-------------|-------------|---------------|
| $\text{N1}-\text{H1A}\cdots\text{S1}^i$ | 0.85 (2) | 2.55 (2) | 3.3828 (18) | 167 (2) |
| $\text{C6}-\text{H6}\cdots\text{O3}^{\text{ii}}$ | 0.93 | 2.41 | 3.317 (3) | 164 |

Symmetry codes: (i) $-x + 1, -y, -z$; (ii) $x, y + 1, z$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2009).

The authors thank Universiti Kebangsaan Malaysia for UKM-GUP-BTT-07-30-190 and UKM-OUP-TK-16-73/2010 grants and sabbatical leave for MBK, and the Kementerian Pengajian Tinggi, Malaysia, for the UKM-ST-06-FRGS0111-2009 research fund. The authors acknowledge B. M. Yamin for the data collection and AAA also thanks the Libyan Ministry of Higher Education and Sabha University for her PhD scholarship.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: JH2299).

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supporting information

Acta Cryst. (2011). E67, o1840 [doi:10.1107/S1600536811024652]

1-Ethyl-1-methyl-3-(2-nitrobenzoyl)thiourea

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S1. Comment

The title compound, I, is a thiourea derivative analogous to our previously reported compounds (Al-abbasi & Kassim, 2011). Bond distances are similar to those usually found in other substituted thioureas [Nasir *et al.* (2011) & Pérez *et al.* (2011)]. The C–S and C–O exhibited the expected double-bond character. However, the C–N bond lengths are intermediate between a single and double, indicating a partial electron delocalization in the O1/C7/N1/C8/S1 fragment.

The phenyl ring is twisted due to the presence of the nitro group (O2O3N3) in *ortho* position. A rotation around C1—C7 bond makes the oxygen atom (O1) perpendicular to the phenyl ring mean planes and the torsion angles of C2C1C7O1 and C6C1C7O1 are $-95.5(2)$ and $86.5(2)^\circ$, respectively. The dihedral angle between the mean planes of the thiourea (S1/N1/N2/C8/C9) and the phenyl ring (C1/C2/C3/C4/C5/C6) plane is $27.56(10)^\circ$. Other bond lengths and angles are in normal ranges (Allen *et al.* 1987).

The crystal structure is stabilized by the intermolecular N1—H1A \cdots S1 and C5—H5A \cdots O3 hydrogen bonds linking the molecules into a dimer resulting in a channel along [101] (Fig. 2).

S2. Experimental

The title compound was prepared according to a previously reported procedure (Al-abbasi *et al.*, 2010). A very pale brown colour crystal, suitable for X-ray crystallography, was obtained by a slow evaporation from ethanol solution at room temperature (yield 78%).

S3. Refinement

Hydrogen atom of the amide group was determined from the difference Fourier map and N—H was initially fixed at $0.86(0.01)$ Å and allowed to be refined on the parent N atom with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N})$. All other H atoms were positioned geometrically with C—H bond lengths in the range $0.93 - 0.97$ Å and refined in the riding model approximation with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$, except for methyl group where $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$.

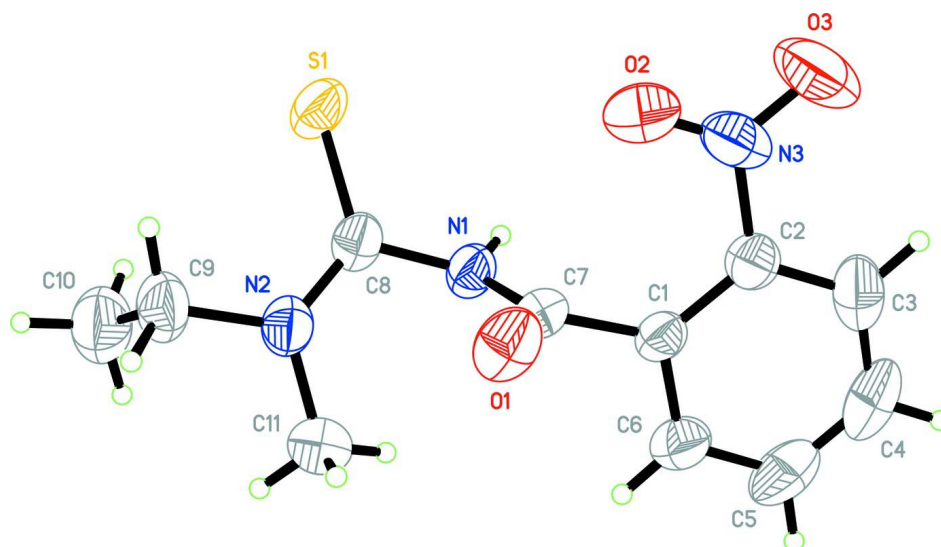


Figure 1

The molecular structure of the title compound with displacement ellipsoids drawn at the 50% probability level.

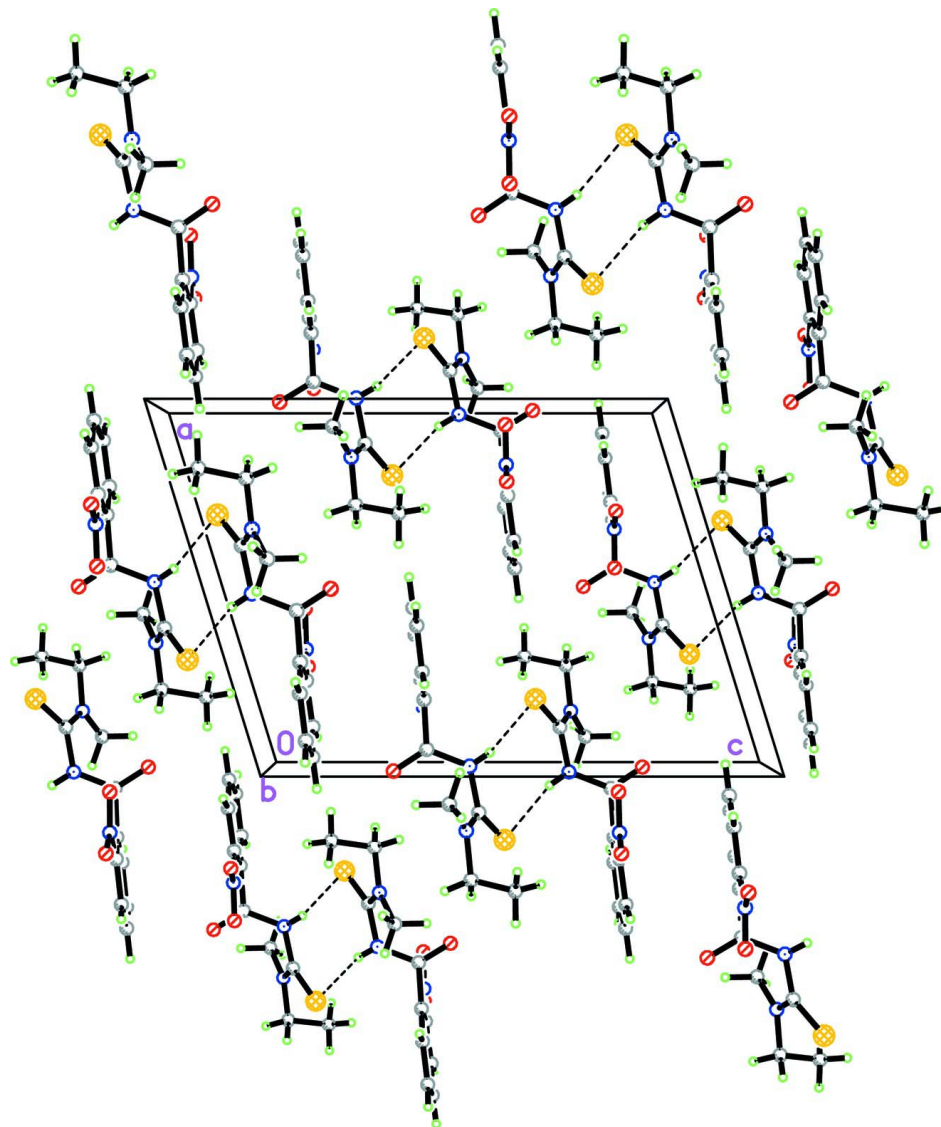


Figure 2

A packing diagram of the title compound viewed down the *a*-axis showing the intermolecular hydrogen bonds N1—H1A···S1 ($-x + 1, -y, -z$) and C6—H6···O3 ($x, y + 1, z$).

1-Ethyl-1-methyl-3-(2-nitrobenzoyl)thiourea

Crystal data

$C_{11}H_{13}N_3O_3S$

$M_r = 267.30$

Monoclinic, $P2_1/n$

Hall symbol: $-P 2_1n$

$a = 11.447 (2) \text{ \AA}$

$b = 7.8664 (15) \text{ \AA}$

$c = 15.159 (3) \text{ \AA}$

$\beta = 107.128 (4)^\circ$

$V = 1304.5 (4) \text{ \AA}^3$

$Z = 4$

$F(000) = 560$

$D_x = 1.361 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 4015 reflections

$\theta = 2.0\text{--}25.0^\circ$

$\mu = 0.25 \text{ mm}^{-1}$

$T = 298 \text{ K}$

Block, brown

$0.55 \times 0.38 \times 0.21 \text{ mm}$

Data collection

Bruker SMART APEX CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2000)

$T_{\min} = 0.874$, $T_{\max} = 0.949$

7105 measured reflections

2294 independent reflections

1971 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.020$

$\theta_{\max} = 25.0^\circ$, $\theta_{\min} = 2.0^\circ$

$h = -13 \rightarrow 13$

$k = -7 \rightarrow 9$

$l = -15 \rightarrow 18$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.043$

$wR(F^2) = 0.124$

$S = 1.06$

2294 reflections

169 parameters

1 restraint

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H atoms treated by a mixture of independent
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.065P)^2 + 0.531P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.33 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|--------------|-------------|--------------|----------------------------------|
| S1 | 0.69163 (5) | 0.06431 (8) | 0.07382 (4) | 0.0575 (2) |
| O1 | 0.49702 (13) | 0.2128 (2) | 0.24893 (10) | 0.0608 (4) |
| O2 | 0.44396 (18) | -0.1541 (2) | 0.17973 (16) | 0.0811 (6) |
| O3 | 0.2858 (2) | -0.3156 (2) | 0.14610 (17) | 0.0984 (7) |
| N1 | 0.48480 (14) | 0.1814 (2) | 0.09706 (11) | 0.0435 (4) |
| N2 | 0.65025 (15) | 0.3643 (2) | 0.14053 (13) | 0.0528 (5) |
| N3 | 0.3344 (2) | -0.1771 (2) | 0.15678 (14) | 0.0612 (5) |
| C1 | 0.30487 (17) | 0.1334 (2) | 0.14682 (13) | 0.0401 (4) |
| C2 | 0.25371 (19) | -0.0265 (3) | 0.14295 (14) | 0.0459 (5) |
| C3 | 0.1301 (2) | -0.0513 (4) | 0.12710 (16) | 0.0636 (7) |
| H3 | 0.0984 | -0.1604 | 0.1258 | 0.076* |
| C4 | 0.0545 (2) | 0.0879 (4) | 0.11328 (18) | 0.0731 (8) |
| H4 | -0.0291 | 0.0735 | 0.1026 | 0.088* |
| C5 | 0.1024 (2) | 0.2479 (4) | 0.11526 (19) | 0.0728 (8) |
| H5 | 0.0507 | 0.3418 | 0.1050 | 0.087* |

| | | | | |
|------|--------------|------------|--------------|------------|
| C6 | 0.2264 (2) | 0.2709 (3) | 0.13230 (16) | 0.0558 (6) |
| H6 | 0.2577 | 0.3803 | 0.1341 | 0.067* |
| C7 | 0.43942 (17) | 0.1743 (2) | 0.17127 (14) | 0.0433 (5) |
| C8 | 0.60936 (17) | 0.2138 (3) | 0.10694 (13) | 0.0441 (5) |
| C9 | 0.7795 (2) | 0.4114 (3) | 0.16004 (17) | 0.0607 (6) |
| H9A | 0.8292 | 0.3093 | 0.1710 | 0.073* |
| H9B | 0.8039 | 0.4801 | 0.2156 | 0.073* |
| C10 | 0.8020 (3) | 0.5099 (4) | 0.0808 (2) | 0.0793 (8) |
| H10A | 0.7897 | 0.4366 | 0.0282 | 0.119* |
| H10B | 0.8844 | 0.5518 | 0.0987 | 0.119* |
| H10C | 0.7461 | 0.6038 | 0.0651 | 0.119* |
| C11 | 0.5729 (2) | 0.5026 (3) | 0.1556 (2) | 0.0737 (8) |
| H11A | 0.4905 | 0.4853 | 0.1175 | 0.111* |
| H11B | 0.6028 | 0.6090 | 0.1400 | 0.111* |
| H11C | 0.5747 | 0.5042 | 0.2193 | 0.111* |
| H1A | 0.4498 (18) | 0.123 (3) | 0.0494 (11) | 0.051 (6)* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|-------------|--------------|
| S1 | 0.0395 (3) | 0.0708 (4) | 0.0647 (4) | 0.0004 (2) | 0.0194 (3) | -0.0241 (3) |
| O1 | 0.0506 (9) | 0.0857 (12) | 0.0461 (9) | -0.0004 (8) | 0.0142 (7) | -0.0123 (8) |
| O2 | 0.0695 (12) | 0.0584 (11) | 0.1181 (16) | 0.0213 (9) | 0.0318 (11) | 0.0114 (10) |
| O3 | 0.134 (2) | 0.0423 (10) | 0.1218 (18) | -0.0117 (11) | 0.0429 (15) | -0.0064 (10) |
| N1 | 0.0362 (8) | 0.0513 (10) | 0.0453 (9) | -0.0046 (7) | 0.0156 (7) | -0.0126 (8) |
| N2 | 0.0415 (9) | 0.0572 (11) | 0.0618 (11) | -0.0084 (8) | 0.0185 (8) | -0.0138 (9) |
| N3 | 0.0866 (15) | 0.0377 (10) | 0.0639 (12) | 0.0017 (10) | 0.0295 (11) | 0.0017 (8) |
| C1 | 0.0402 (10) | 0.0417 (10) | 0.0428 (10) | 0.0035 (8) | 0.0190 (8) | -0.0007 (8) |
| C2 | 0.0504 (11) | 0.0470 (11) | 0.0449 (11) | 0.0001 (9) | 0.0209 (9) | 0.0002 (8) |
| C3 | 0.0608 (14) | 0.0764 (17) | 0.0602 (14) | -0.0244 (13) | 0.0279 (12) | -0.0070 (12) |
| C4 | 0.0389 (12) | 0.118 (2) | 0.0658 (16) | -0.0025 (14) | 0.0213 (11) | -0.0029 (15) |
| C5 | 0.0510 (14) | 0.090 (2) | 0.0805 (18) | 0.0280 (14) | 0.0242 (12) | 0.0086 (14) |
| C6 | 0.0536 (12) | 0.0485 (12) | 0.0698 (14) | 0.0113 (10) | 0.0252 (11) | 0.0041 (10) |
| C7 | 0.0405 (10) | 0.0429 (11) | 0.0497 (11) | 0.0049 (8) | 0.0181 (9) | -0.0035 (8) |
| C8 | 0.0367 (10) | 0.0560 (12) | 0.0400 (10) | -0.0039 (9) | 0.0122 (8) | -0.0071 (8) |
| C9 | 0.0448 (12) | 0.0725 (15) | 0.0638 (14) | -0.0165 (11) | 0.0145 (10) | -0.0187 (12) |
| C10 | 0.0656 (16) | 0.0856 (19) | 0.090 (2) | -0.0134 (14) | 0.0287 (14) | -0.0004 (16) |
| C11 | 0.0675 (16) | 0.0533 (14) | 0.107 (2) | -0.0033 (12) | 0.0356 (15) | -0.0200 (14) |

Geometric parameters (Å, °)

| | | | |
|--------|------------|--------|-----------|
| S1—C8 | 1.674 (2) | C3—H3 | 0.9300 |
| O1—C7 | 1.206 (2) | C4—C5 | 1.370 (4) |
| O2—N3 | 1.212 (3) | C4—H4 | 0.9300 |
| O3—N3 | 1.212 (3) | C5—C6 | 1.378 (3) |
| N1—C7 | 1.372 (2) | C5—H5 | 0.9300 |
| N1—C8 | 1.412 (2) | C6—H6 | 0.9300 |
| N1—H1A | 0.849 (10) | C9—C10 | 1.514 (4) |

| | | | |
|-------------|-------------|---------------|--------------|
| N2—C8 | 1.319 (3) | C9—H9A | 0.9700 |
| N2—C11 | 1.462 (3) | C9—H9B | 0.9700 |
| N2—C9 | 1.469 (3) | C10—H10A | 0.9600 |
| N3—C2 | 1.479 (3) | C10—H10B | 0.9600 |
| C1—C2 | 1.381 (3) | C10—H10C | 0.9600 |
| C1—C6 | 1.382 (3) | C11—H11A | 0.9600 |
| C1—C7 | 1.509 (3) | C11—H11B | 0.9600 |
| C2—C3 | 1.378 (3) | C11—H11C | 0.9600 |
| C3—C4 | 1.372 (4) | | |
| | | | |
| C7—N1—C8 | 122.31 (16) | C1—C6—H6 | 119.6 |
| C7—N1—H1A | 118.6 (15) | O1—C7—N1 | 124.04 (18) |
| C8—N1—H1A | 113.3 (15) | O1—C7—C1 | 121.22 (17) |
| C8—N2—C11 | 124.45 (18) | N1—C7—C1 | 114.38 (17) |
| C8—N2—C9 | 121.75 (18) | N2—C8—N1 | 115.81 (17) |
| C11—N2—C9 | 113.66 (19) | N2—C8—S1 | 125.42 (15) |
| O3—N3—O2 | 124.6 (2) | N1—C8—S1 | 118.75 (15) |
| O3—N3—C2 | 117.3 (2) | N2—C9—C10 | 111.5 (2) |
| O2—N3—C2 | 118.12 (18) | N2—C9—H9A | 109.3 |
| C2—C1—C6 | 117.29 (18) | C10—C9—H9A | 109.3 |
| C2—C1—C7 | 126.44 (17) | N2—C9—H9B | 109.3 |
| C6—C1—C7 | 116.16 (18) | C10—C9—H9B | 109.3 |
| C3—C2—C1 | 122.5 (2) | H9A—C9—H9B | 108.0 |
| C3—C2—N3 | 118.5 (2) | C9—C10—H10A | 109.5 |
| C1—C2—N3 | 118.97 (18) | C9—C10—H10B | 109.5 |
| C4—C3—C2 | 118.9 (2) | H10A—C10—H10B | 109.5 |
| C4—C3—H3 | 120.6 | C9—C10—H10C | 109.5 |
| C2—C3—H3 | 120.6 | H10A—C10—H10C | 109.5 |
| C5—C4—C3 | 120.0 (2) | H10B—C10—H10C | 109.5 |
| C5—C4—H4 | 120.0 | N2—C11—H11A | 109.5 |
| C3—C4—H4 | 120.0 | N2—C11—H11B | 109.5 |
| C4—C5—C6 | 120.5 (2) | H11A—C11—H11B | 109.5 |
| C4—C5—H5 | 119.8 | N2—C11—H11C | 109.5 |
| C6—C5—H5 | 119.8 | H11A—C11—H11C | 109.5 |
| C5—C6—C1 | 120.9 (2) | H11B—C11—H11C | 109.5 |
| C5—C6—H6 | 119.6 | | |
| | | | |
| C6—C1—C2—C3 | -1.2 (3) | C8—N1—C7—O1 | 8.5 (3) |
| C7—C1—C2—C3 | 174.71 (19) | C8—N1—C7—C1 | -178.37 (17) |
| C6—C1—C2—N3 | 179.13 (19) | C2—C1—C7—O1 | -95.5 (3) |
| C7—C1—C2—N3 | -5.0 (3) | C6—C1—C7—O1 | 80.5 (3) |
| O3—N3—C2—C3 | 5.7 (3) | C2—C1—C7—N1 | 91.2 (2) |
| O2—N3—C2—C3 | -172.8 (2) | C6—C1—C7—N1 | -92.9 (2) |
| O3—N3—C2—C1 | -174.6 (2) | C11—N2—C8—N1 | -8.5 (3) |
| O2—N3—C2—C1 | 6.9 (3) | C9—N2—C8—N1 | 176.06 (19) |
| C1—C2—C3—C4 | 1.0 (3) | C11—N2—C8—S1 | 170.1 (2) |
| N3—C2—C3—C4 | -179.3 (2) | C9—N2—C8—S1 | -5.4 (3) |
| C2—C3—C4—C5 | 0.1 (4) | C7—N1—C8—N2 | -63.8 (3) |

| | | | |
|-------------|------------|---------------|-------------|
| C3—C4—C5—C6 | -0.9 (4) | C7—N1—C8—S1 | 117.53 (18) |
| C4—C5—C6—C1 | 0.6 (4) | C8—N2—C9—C10 | 96.1 (3) |
| C2—C1—C6—C5 | 0.4 (3) | C11—N2—C9—C10 | -79.8 (3) |
| C7—C1—C6—C5 | -175.9 (2) | | |

Hydrogen-bond geometry (Å, °)

| <i>D—H...A</i> | <i>D—H</i> | <i>H...A</i> | <i>D...A</i> | <i>D—H...A</i> |
|--------------------------|------------|--------------|--------------|----------------|
| N1—H1A...S1 ⁱ | 0.85 (2) | 2.55 (2) | 3.3828 (18) | 167 (2) |
| C6—H6...O3 ⁱⁱ | 0.93 | 2.41 | 3.317 (3) | 164 |

Symmetry codes: (i) $-x+1, -y, -z$; (ii) $x, y+1, z$.