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## Structure Reports

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# (E)-1-(4-Methylphenyl)-3-[(1-phenyl-ethylidene)amino]thiourea

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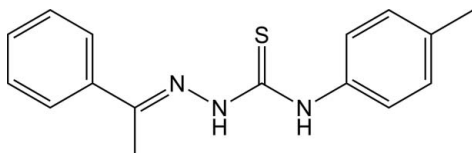
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 Key indicators: single-crystal X-ray study;  $T = 291$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.044;  $wR$  factor = 0.126; data-to-parameter ratio = 15.4.

In the title compound,  $\text{C}_{16}\text{H}_{17}\text{N}_3\text{S}$ , the aminothiurea unit is nearly planar (r.m.s. deviation =  $0.0425$  Å), and is twisted with respect to the tolyl and phenyl rings by  $57.84$  (7) and  $15.88$  (14)°, respectively; the tolyl and phenyl rings are twisted by  $65.64$  (11)° to each other. Intermolecular  $\text{N}-\text{H}\cdots\text{S}$  and weak  $\text{C}-\text{H}\cdots\text{S}$  hydrogen bonds are present in the crystal structure.

## Related literature

The title compound is a derivative of thiosemicarbazone. For applications of thiosemicarbazones in the biological field, see: Hu *et al.* (2006).



## Experimental

### Crystal data

$\text{C}_{16}\text{H}_{17}\text{N}_3\text{S}$   
 $M_r = 283.39$   
 Monoclinic,  $P2_1/c$   
 $a = 10.5881$  (3) Å  
 $b = 5.7355$  (2) Å  
 $c = 26.9746$  (7) Å  
 $\beta = 108.670$  (2)°

$V = 1551.91$  (8) Å<sup>3</sup>  
 $Z = 4$   
 Cu  $K\alpha$  radiation  
 $\mu = 1.79$  mm<sup>-1</sup>  
 $T = 291$  K  
 $0.25 \times 0.18 \times 0.18$  mm

### Data collection

Oxford Diffraction Xcalibur Eos  
 Gemini diffractometer  
 Absorption correction: multi-scan  
 (*CrysAlis PRO*; Oxford  
 Diffraction, 2010)  
 $T_{\min} = 0.60$ ,  $T_{\max} = 0.73$

15835 measured reflections  
 2945 independent reflections  
 2587 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.041$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$   
 $wR(F^2) = 0.126$   
 $S = 1.06$   
 2945 reflections  
 191 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.23$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.24$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N2}-\text{H2}\cdots\text{S1}^i$	0.90 (2)	2.86 (2)	3.7456 (16)	167.8 (19)
$\text{C16}-\text{H16B}\cdots\text{S1}^i$	0.96	2.74	3.471 (2)	133

 Symmetry code: (i)  $-x + 2, -y, -z + 1$ .

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *OLEX2* (Dolomanov *et al.*, 2009); software used to prepare material for publication: *OLEX2*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5214).

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## supporting information

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**(E)-1-(4-Methylphenyl)-3-[(1-phenylethylidene)amino]thiourea**

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**S1. Comment**

Thiosemicarbazones have attracted our attention because of their biological applications (Hu *et al.*, 2006). A few single-crystal structures were reported. For understanding their anticancer activity, it is necessary to have detailed information on their geometries.

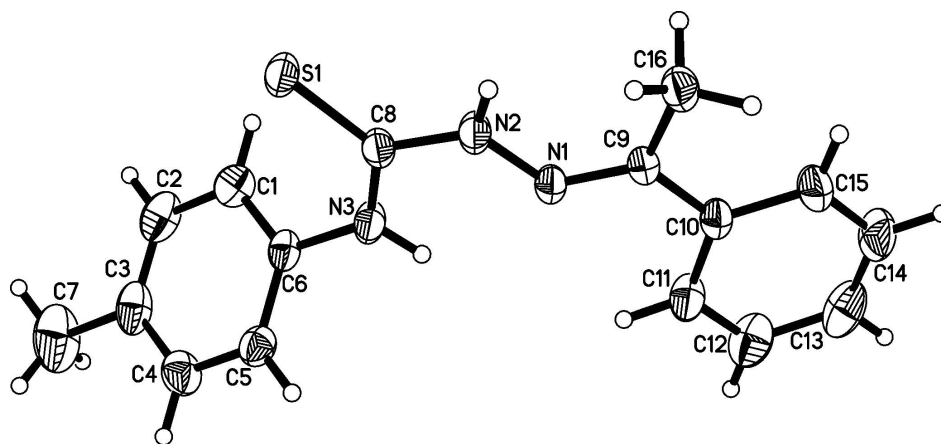
The molecular structure of (I) is shown in Fig 1. The molecules reveal an E configuration. The dihedral angles formed by the tolyl and phenyl rings with the almost planar aminothiourea unit (r.m.s. deviation = 0.0425 Å) are 57.84 (7) and 15.88 (14)°, respectively. Intermolecular N—H···S and C—H···S interactions occur in the crystal structure (Table 1).

**S2. Experimental**

*N*-(*p*-Tolyl)thiosemicarbazide (2.7 g, 15 mmol) and acetophenone (1.8 g, 15 mmol) was dissolved in 95% ethanol (20 ml), and the solution was refluxed for 3 h. Fine colorless crystals appeared on cooling. They were filtered and washed by 95% ethanol to give 2.9 g of the title compound in 69.0% yield. Single crystals suitable for X-ray measurements were obtained from ether by slow evaporation at room temperature.

**S3. Refinement**

Imino H atoms were located in a difference Fourier map and refined isotropically. Other H atoms were placed in calculated positions with C—H = 0.93–0.96 and refined using a riding model,  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl H atoms and  $1.2U_{\text{eq}}(\text{C})$  for the others.



**Figure 1**

The molecular structure of the title compound, with displacement ellipsoids drawn at 30% probability level.

**(E)-1-(4-Methylphenyl)-3-[(1-phenylethylidene)amino]thiourea***Crystal data*C<sub>16</sub>H<sub>17</sub>N<sub>3</sub>S $M_r = 283.39$ Monoclinic,  $P2_1/c$  $a = 10.5881$  (3) Å $b = 5.7355$  (2) Å $c = 26.9746$  (7) Å $\beta = 108.670$  (2)° $V = 1551.91$  (8) Å<sup>3</sup> $Z = 4$  $F(000) = 600$  $D_x = 1.213$  Mg m<sup>-3</sup>Cu  $K\alpha$  radiation,  $\lambda = 1.5418$  Å

Cell parameters from 6724 reflections

 $\theta = 3.5$ – $70.9$ ° $\mu = 1.79$  mm<sup>-1</sup> $T = 291$  K

Prismatic, colorless

 $0.25 \times 0.18 \times 0.18$  mm*Data collection*Oxford Diffraction Xcalibur Eos Gemini  
diffractometer

Radiation source: Enhance (Cu) X-ray Source

Graphite monochromator

Detector resolution: 16.2312 pixels mm<sup>-1</sup> $\omega$  scans

Absorption correction: multi-scan

(Crys.Alis PRO; Oxford Diffraction, 2010)

 $T_{\min} = 0.60$ ,  $T_{\max} = 0.73$ 

15835 measured reflections

2945 independent reflections

2587 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.041$  $\theta_{\max} = 70.9$ °,  $\theta_{\min} = 3.5$ ° $h = -12 \rightarrow 12$  $k = -7 \rightarrow 6$  $l = -28 \rightarrow 31$ *Refinement*Refinement on  $F^2$ 

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.044$  $wR(F^2) = 0.126$  $S = 1.06$ 

2945 reflections

191 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methodsSecondary atom site location: difference Fourier  
mapHydrogen site location: inferred from  
neighbouring sitesH atoms treated by a mixture of independent  
and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0693P)^2 + 0.2489P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\max} = 0.001$  $\Delta\rho_{\max} = 0.23$  e Å<sup>-3</sup> $\Delta\rho_{\min} = -0.24$  e Å<sup>-3</sup>*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>)*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	1.08154 (5)	0.02598 (9)	0.436028 (18)	0.05334 (18)
N1	0.82552 (14)	0.5213 (3)	0.42960 (6)	0.0465 (3)
N2	0.90859 (14)	0.3341 (3)	0.44695 (6)	0.0499 (4)

N3	0.99833 (16)	0.4238 (3)	0.38311 (6)	0.0510 (4)
C1	1.06088 (19)	0.2091 (4)	0.31610 (7)	0.0559 (5)
H1	1.0044	0.0876	0.3179	0.067*
C2	1.1318 (2)	0.1997 (4)	0.28098 (7)	0.0656 (6)
H2A	1.1229	0.0699	0.2595	0.079*
C3	1.2149 (2)	0.3777 (5)	0.27725 (7)	0.0651 (6)
C4	1.2278 (2)	0.5681 (4)	0.30964 (8)	0.0630 (5)
H4	1.2841	0.6896	0.3077	0.076*
C5	1.15789 (19)	0.5810 (4)	0.34519 (7)	0.0520 (4)
H5	1.1672	0.7109	0.3667	0.062*
C6	1.07490 (16)	0.4014 (3)	0.34853 (6)	0.0443 (4)
C7	1.2928 (3)	0.3658 (8)	0.23886 (12)	0.1137 (13)
H7A	1.2491	0.4593	0.2087	0.171*
H7B	1.2971	0.2070	0.2283	0.171*
H7C	1.3814	0.4237	0.2553	0.171*
C8	0.99248 (15)	0.2719 (3)	0.42007 (6)	0.0440 (4)
C9	0.74014 (16)	0.5714 (3)	0.45287 (7)	0.0437 (4)
C10	0.64955 (15)	0.7680 (3)	0.43048 (6)	0.0452 (4)
C11	0.6772 (2)	0.9235 (4)	0.39573 (7)	0.0584 (5)
H11	0.7547	0.9058	0.3869	0.070*
C12	0.5910 (3)	1.1032 (5)	0.37429 (9)	0.0761 (7)
H12	0.6102	1.2057	0.3510	0.091*
C13	0.4749 (2)	1.1320 (5)	0.38739 (9)	0.0783 (8)
H13	0.4162	1.2526	0.3726	0.094*
C14	0.4478 (2)	0.9828 (5)	0.42192 (12)	0.0773 (7)
H14	0.3707	1.0024	0.4309	0.093*
C15	0.53386 (19)	0.8031 (4)	0.44367 (10)	0.0665 (6)
H15	0.5146	0.7037	0.4675	0.080*
C16	0.7269 (2)	0.4397 (4)	0.49895 (8)	0.0570 (5)
H16A	0.6858	0.5377	0.5182	0.086*
H16B	0.8137	0.3932	0.5212	0.086*
H16C	0.6728	0.3038	0.4869	0.086*
H2	0.904 (2)	0.230 (4)	0.4714 (8)	0.058 (6)*
H3	0.960 (2)	0.548 (4)	0.3831 (8)	0.053 (6)*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0594 (3)	0.0488 (3)	0.0626 (3)	0.01893 (19)	0.0347 (2)	0.01199 (19)
N1	0.0428 (7)	0.0442 (9)	0.0593 (8)	0.0093 (6)	0.0257 (6)	0.0052 (6)
N2	0.0502 (8)	0.0488 (9)	0.0607 (8)	0.0150 (7)	0.0319 (6)	0.0117 (7)
N3	0.0550 (8)	0.0492 (10)	0.0578 (8)	0.0193 (7)	0.0305 (7)	0.0099 (7)
C1	0.0614 (10)	0.0575 (12)	0.0464 (8)	0.0029 (9)	0.0138 (7)	-0.0037 (8)
C2	0.0828 (13)	0.0722 (15)	0.0404 (8)	0.0195 (11)	0.0177 (8)	-0.0086 (8)
C3	0.0717 (12)	0.0870 (17)	0.0436 (9)	0.0292 (12)	0.0282 (8)	0.0116 (9)
C4	0.0648 (11)	0.0709 (14)	0.0624 (11)	0.0067 (10)	0.0333 (9)	0.0157 (10)
C5	0.0624 (10)	0.0480 (11)	0.0518 (9)	0.0075 (8)	0.0268 (8)	0.0018 (7)
C6	0.0450 (8)	0.0503 (10)	0.0395 (7)	0.0137 (7)	0.0162 (6)	0.0063 (6)

C7	0.130 (3)	0.159 (3)	0.0798 (16)	0.037 (2)	0.0733 (18)	0.0119 (19)
C8	0.0398 (7)	0.0478 (10)	0.0487 (8)	0.0052 (7)	0.0201 (6)	0.0017 (7)
C9	0.0392 (7)	0.0441 (10)	0.0535 (8)	0.0035 (7)	0.0229 (6)	0.0005 (7)
C10	0.0375 (7)	0.0506 (10)	0.0490 (8)	0.0058 (7)	0.0159 (6)	-0.0033 (7)
C11	0.0623 (11)	0.0632 (13)	0.0556 (10)	0.0202 (9)	0.0273 (8)	0.0082 (9)
C12	0.0911 (16)	0.0768 (17)	0.0577 (11)	0.0305 (13)	0.0201 (10)	0.0146 (11)
C13	0.0655 (12)	0.0806 (18)	0.0706 (13)	0.0360 (12)	-0.0040 (10)	-0.0071 (12)
C14	0.0433 (10)	0.0823 (18)	0.1062 (18)	0.0183 (10)	0.0237 (11)	-0.0094 (14)
C15	0.0480 (10)	0.0686 (15)	0.0930 (14)	0.0100 (9)	0.0367 (10)	0.0057 (12)
C16	0.0595 (10)	0.0586 (12)	0.0638 (10)	0.0104 (9)	0.0351 (8)	0.0095 (9)

*Geometric parameters (Å, °)*

S1—C8	1.6746 (18)	C7—H7A	0.9600
N1—N2	1.372 (2)	C7—H7B	0.9600
N1—C9	1.287 (2)	C7—H7C	0.9600
N2—C8	1.362 (2)	C9—C10	1.478 (2)
N2—H2	0.90 (2)	C9—C16	1.499 (2)
N3—C6	1.424 (2)	C10—C11	1.391 (3)
N3—C8	1.340 (2)	C10—C15	1.395 (2)
N3—H3	0.82 (2)	C11—H11	0.9300
C1—H1	0.9300	C11—C12	1.375 (3)
C1—C2	1.386 (3)	C12—H12	0.9300
C1—C6	1.386 (3)	C12—C13	1.393 (4)
C2—H2A	0.9300	C13—H13	0.9300
C2—C3	1.373 (4)	C13—C14	1.361 (4)
C3—C4	1.378 (4)	C14—H14	0.9300
C3—C7	1.517 (3)	C14—C15	1.376 (3)
C4—H4	0.9300	C15—H15	0.9300
C4—C5	1.388 (3)	C16—H16A	0.9600
C5—H5	0.9300	C16—H16B	0.9600
C5—C6	1.376 (3)	C16—H16C	0.9600
C9—N1—N2	118.83 (15)	N2—C8—S1	119.57 (13)
N1—N2—H2	126.2 (14)	N3—C8—S1	125.62 (12)
C8—N2—N1	118.67 (15)	N3—C8—N2	114.78 (16)
C8—N2—H2	114.3 (14)	N1—C9—C10	115.87 (15)
C6—N3—H3	117.8 (16)	N1—C9—C16	123.94 (16)
C8—N3—C6	126.88 (15)	C10—C9—C16	120.17 (14)
C8—N3—H3	115.0 (16)	C11—C10—C9	121.03 (15)
C2—C1—H1	120.3	C11—C10—C15	118.04 (18)
C2—C1—C6	119.4 (2)	C15—C10—C9	120.93 (17)
C6—C1—H1	120.3	C10—C11—H11	119.7
C1—C2—H2A	119.3	C12—C11—C10	120.6 (2)
C3—C2—C1	121.4 (2)	C12—C11—H11	119.7
C3—C2—H2A	119.3	C11—C12—H12	119.9
C2—C3—C4	118.58 (18)	C11—C12—C13	120.2 (2)
C2—C3—C7	121.2 (3)	C13—C12—H12	119.9

C4—C3—C7	120.2 (3)	C12—C13—H13	120.2
C3—C4—H4	119.5	C14—C13—C12	119.7 (2)
C3—C4—C5	121.0 (2)	C14—C13—H13	120.2
C5—C4—H4	119.5	C13—C14—H14	119.8
C4—C5—H5	120.1	C13—C14—C15	120.5 (2)
C6—C5—C4	119.86 (19)	C15—C14—H14	119.8
C6—C5—H5	120.1	C10—C15—H15	119.5
C1—C6—N3	121.13 (18)	C14—C15—C10	121.0 (2)
C5—C6—N3	119.00 (17)	C14—C15—H15	119.5
C5—C6—C1	119.75 (16)	C9—C16—H16A	109.5
C3—C7—H7A	109.5	C9—C16—H16B	109.5
C3—C7—H7B	109.5	C9—C16—H16C	109.5
C3—C7—H7C	109.5	H16A—C16—H16B	109.5
H7A—C7—H7B	109.5	H16A—C16—H16C	109.5
H7A—C7—H7C	109.5	H16B—C16—H16C	109.5
H7B—C7—H7C	109.5		
N1—N2—C8—S1	-172.72 (13)	C6—C1—C2—C3	-0.6 (3)
N1—N2—C8—N3	9.3 (2)	C7—C3—C4—C5	-179.6 (2)
N1—C9—C10—C11	-15.8 (3)	C8—N3—C6—C1	56.4 (3)
N1—C9—C10—C15	164.35 (19)	C8—N3—C6—C5	-127.6 (2)
N2—N1—C9—C10	-177.12 (15)	C9—N1—N2—C8	175.59 (16)
N2—N1—C9—C16	1.2 (3)	C9—C10—C11—C12	178.7 (2)
C1—C2—C3—C4	0.4 (3)	C9—C10—C15—C14	-178.5 (2)
C1—C2—C3—C7	179.7 (2)	C10—C11—C12—C13	0.3 (4)
C2—C1—C6—N3	176.51 (16)	C11—C10—C15—C14	1.6 (3)
C2—C1—C6—C5	0.6 (3)	C11—C12—C13—C14	0.6 (4)
C2—C3—C4—C5	-0.3 (3)	C12—C13—C14—C15	-0.4 (4)
C3—C4—C5—C6	0.3 (3)	C13—C14—C15—C10	-0.8 (4)
C4—C5—C6—N3	-176.47 (16)	C15—C10—C11—C12	-1.4 (3)
C4—C5—C6—C1	-0.4 (3)	C16—C9—C10—C11	165.88 (19)
C6—N3—C8—S1	3.6 (3)	C16—C9—C10—C15	-14.0 (3)
C6—N3—C8—N2	-178.56 (17)		

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2—H2 $\cdots$ S1 <sup>i</sup>	0.90 (2)	2.86 (2)	3.7456 (16)	167.8 (19)
C16—H16B $\cdots$ S1 <sup>i</sup>	0.96	2.74	3.471 (2)	133

Symmetry code: (i)  $-x+2, -y, -z+1$ .