

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

Diethyl 2-[[[(5-oxo-5*H*-thiochromeno-[2,3-*b*]pyridin-7-yl)amino]methylidene]-propanedioate

Muhammad Naeem Khan,^{a,b} M. Nawaz Tahir,^{c*}
 Misbahul Ain Khan,^a Munawar Ali Munawar^d and
 Abdul Qayyum Ather^b

^aDepartment of Chemistry, Islamia University, Bahawalpur, Pakistan, ^bApplied Chemistry Research Center, PCSIR Laboratories complex, Lahore 54600, Pakistan, ^cUniversity of Sargodha, Department of Physics, Sargodha, Pakistan, and ^dInstitute of Chemistry, University of the Punjab, Lahore, Pakistan
 Correspondence e-mail: dmntahir_uos@yahoo.com

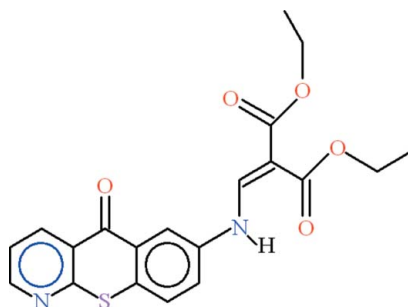
Received 17 April 2011; accepted 29 April 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; disorder in main residue; R factor = 0.054; wR factor = 0.130; data-to-parameter ratio = 12.3.

In the title compound, $\text{C}_{15}\text{H}_{14}\text{O}_5\text{S}$, the three fused rings are roughly coplanar, the largest deviation from the mean plane being 0.1285 (13) Å for the S atom. An intramolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bond generates an S_6 ring. In the crystal, intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds form $R_2^2(14)$, $R_2^2(13)$ and $R_3^2(17)$ ring motifs, building a layer parallel to (100).

Related literature

For related structures, see: Khan *et al.* (2008*a,b*); Lokaj *et al.* (1994); Lynch & McClenaghan (2003); For graph-set notation, see: Etter *et al.* (1990); Bernstein *et al.* (1995).



Experimental

Crystal data

$\text{C}_{20}\text{H}_{18}\text{N}_2\text{O}_5\text{S}$
 $M_r = 398.42$
 Monoclinic, $C2/c$
 $a = 13.8013$ (7) Å

$b = 7.5180$ (3) Å
 $c = 36.2743$ (15) Å
 $\beta = 94.330$ (3)°
 $V = 3753.0$ (3) Å³

$Z = 8$
 Mo $K\alpha$ radiation
 $\mu = 0.21$ mm⁻¹

$T = 296$ K
 $0.35 \times 0.18 \times 0.15$ mm

Data collection

Bruker Kappa APEXII CCD diffractometer
 Absorption correction: multi-scan (*SADABS*; Bruker, 2005)
 $T_{\min} = 0.968$, $T_{\max} = 0.985$

13274 measured reflections
 3314 independent reflections
 1776 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.077$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.130$
 $S = 0.97$
 3314 reflections
 269 parameters

10 restraints
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.18$ e Å⁻³

Table 1
 Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N2}-\text{H2}\cdots\text{O3}$	0.86	2.06	2.683 (3)	129
$\text{C11}-\text{H11}\cdots\text{O5}^i$	0.93	2.49	3.403 (4)	167
$\text{C12}-\text{H12}\cdots\text{O1}^i$	0.93	2.45	3.322 (4)	157
$\text{C17A}-\text{H17C}\cdots\text{O3}^{ii}$	0.96	2.58	3.516 (8)	165
$\text{C19A}-\text{H19B}\cdots\text{O3}^{iii}$	0.97	2.47	3.231 (11)	135

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+\frac{1}{2}, y+\frac{1}{2}, -z+\frac{1}{2}$; (iii) $x, y+1, z$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-III* (Burnett & Johnson, 1996), *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

The authors acknowledge the provision of funds for the purchase of the diffractometer and encouragement by Dr Muhammad Akram Chaudhary, Ex-Vice Chancellor, University of Sargodha, Pakistan.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: DN2678).

References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
 Bruker (2005). *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (2009). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Burnett, M. N. & Johnson, C. K. (1996). *ORTEP-III*. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.
 Etter, M. C., MacDonald, J. C. & Bernstein, J. (1990). *Acta Cryst.* **B46**, 256–262.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Khan, M. N., Tahir, M. N., Khan, M. A., Khan, I. U. & Arshad, M. N. (2008*a*). *Acta Cryst.* **E64**, o730.
 Khan, M. N., Tahir, M. N., Khan, M. A., Khan, I. U. & Arshad, M. N. (2008*b*). *Acta Cryst.* **E64**, o1704.
 Lokaj, J., Kettmann, V., Vrabel, V., Ilavský, D. & Milata, V. (1994). *Acta Cryst.* **C50**, 1784–1786.
 Lynch, D. E. & McClenaghan, I. (2003). *Acta Cryst.* **E59**, o242–o243.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

supporting information

Acta Cryst. (2011). E67, o1348 [doi:10.1107/S1600536811016291]

Diethyl 2-[(5-oxo-5*H*-thiochromeno[2,3-*b*]pyridin-7-yl)amino]methylidene}propanedioate

Muhammad Naeem Khan, M. Nawaz Tahir, Misbahul Ain Khan, Munawar Ali Munawar and Abdul Qayyum Ather

S1. Comment

We reported the crystal structures of (II) *i.e.* 7-nitro-5*H*-thiochromeno[2,3-*b*]pyridin-5-one (Khan *et al.*, 2008*a*) and (III) *i.e.* 5*H*-thiochromeno[2,3-*b*]pyridin-5-one (Khan *et al.*, 2008*b*). The title compound (Fig. 1) is in continuation of our work to synthesize the derivatives of 5*H*-thiochromeno[2,3-*b*]pyridin-5-one. The crystal structures of (IV) *i.e.*, diethyl 2-(2,3-diphenylquinoxalin-6-ylaminomethylene)malonate (Lokaj *et al.*, 1994) and (V) *i.e.*, diethyl (4-*tert*-butyl-1,3-thiazol-2-ylaminomethylene)malonate (Lynch & McClenaghan, 2003) have been published. Both contain the diethyl (amino-methylidene)propanedioate moiety which is also present in (I).

In (I), the central heterocyclic ring A (C1/C2/C6/S1/C7/C8), the pyridinic group B (C2/C3/C4/C5/N1/C6) and benzene ring C (C7—C12) are planar with r. m. s. deviation of 0.0388, 0.0053 and 0.0049 Å, respectively. The carbonyl O-atom is at a distance of 0.1532 (35) Å from its parent ring A. The dihedral angle between A/B, A/C and B/C is 3.68 (12), 3.50 (9) and 7.19 (12)°, respectively. A strong intramolecular H-bonding of N—H···O type completes an S(6) (Etter *et al.*, 1990; Bernstein *et al.*, 1995) ring motif. Intermolecular C—H···O hydrogen bonds complete $R_2^2(14)$, $R_2^2(13)$ and $R_3^2(17)$ ring motifs building a layer parallel to the (1 0 0) plane (Table 1, Fig 2).

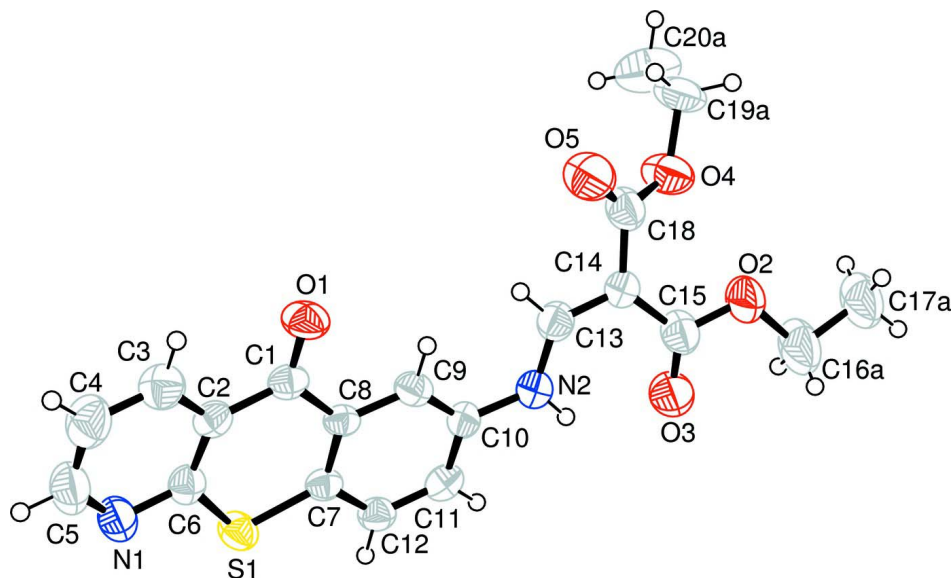
S2. Experimental

Diethyl (ethoxymethylene)malonate (0.473 g, 2.1 mmol), 7-amino-5*H*-thiochromeno[2,3-*b*]pyridin-5-one (500 mg, 2.1 mmol) and 8.0 ml of ethyl alcohol was heated under reflux on water bath for 3 h. Completion of reaction was monitored by TLC. The product precipitated by the addition of n-hexane was filtered, washed, dried and recrystallized from chloroform to give the yellow needles of (I).

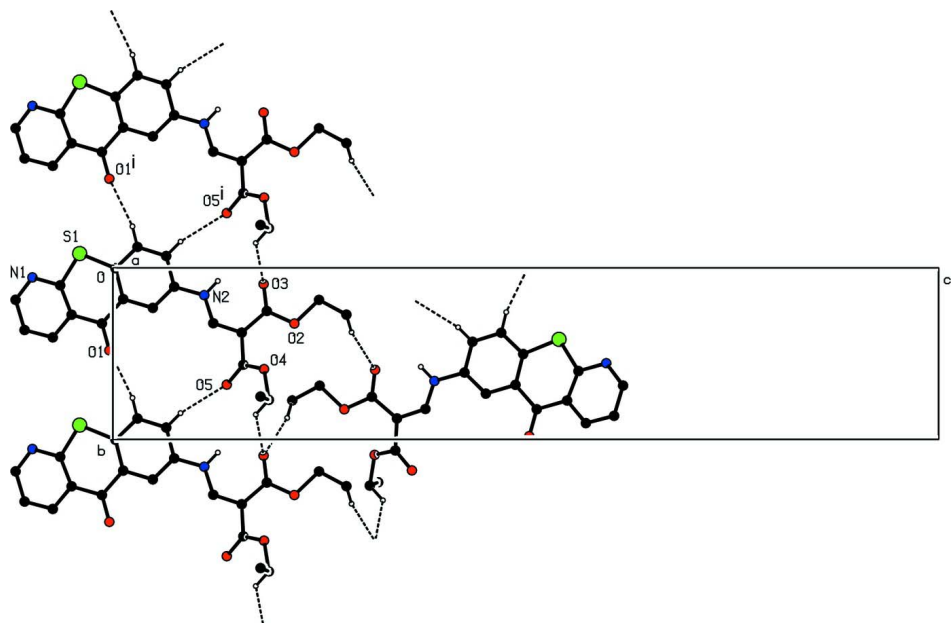
S3. Refinement

The diethyl groups are disordered over two set of sites with occupancy ratio of 0.75:0.25 and 0.53:0.47. The disordered positions were refined using O-C and C-C bond restraints to maintain chemically reasonable geometry. The disordered groups were refined anisotropically with however equal thermal parameters for the C-atoms.

The H-atoms were positioned geometrically (N—H = 0.86, C—H = 0.93–0.97 Å) and treated as riding with $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C}, \text{O})$, where $x = 1.5$ for methyl and $x = 1.2$ for aryl H-atoms.

**Figure 1**

View of the title compound with the atom numbering scheme. The thermal displacements are drawn at the 50% probability level. H-atoms are shown as small spheres of arbitrary radii. The disordered minority ethyl groups are omitted for clarity.

**Figure 2**

Partial packing view showing the formation of different ring motifs through C-H...O hydrogen bonds. Only the major component of the disordered moieties are represented and H atoms not involved in hydrogen bonding have been omitted for clarity. [Symmetry code: (i) $x, y-1, z$.]

Diethyl 2-[(5-oxo-5H-thiochromeno[2,3-b]pyridin-7-yl)amino]methylidene}propanedioate

Crystal data

C₂₀H₁₈N₂O₅S $M_r = 398.42$

Monoclinic, C2/c

Hall symbol: -C 2yc

 $a = 13.8013 (7) \text{ \AA}$ $b = 7.5180 (3) \text{ \AA}$ $c = 36.2743 (15) \text{ \AA}$ $\beta = 94.330 (3)^\circ$ $V = 3753.0 (3) \text{ \AA}^3$ $Z = 8$ $F(000) = 1664$ $D_x = 1.410 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 1789 reflections

 $\theta = 2.3\text{--}25.3^\circ$ $\mu = 0.21 \text{ mm}^{-1}$ $T = 296 \text{ K}$

Needle, yellow

 $0.35 \times 0.18 \times 0.15 \text{ mm}$

Data collection

Bruker Kappa APEXII CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 8.10 pixels mm^{-1} ω scans

Absorption correction: multi-scan

(SADABS; Bruker, 2005)

 $T_{\min} = 0.968$, $T_{\max} = 0.985$

13274 measured reflections

3314 independent reflections

1776 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.077$ $\theta_{\max} = 25.0^\circ$, $\theta_{\min} = 2.3^\circ$ $h = -16 \rightarrow 16$ $k = 0 \rightarrow 8$ $l = 0 \rightarrow 42$

Refinement

Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.054$ $wR(F^2) = 0.130$ $S = 0.97$

3314 reflections

269 parameters

10 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0533P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.20 \text{ e \AA}^{-3}$ $\Delta\rho_{\min} = -0.18 \text{ e \AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.10364 (6)	-0.08366 (11)	-0.04024 (2)	0.0529 (3)	
O1	0.11162 (17)	0.4807 (3)	-0.00428 (6)	0.0604 (7)	
O2	0.3194 (2)	0.3243 (3)	0.21973 (6)	0.0911 (9)	
C16A	0.3217 (7)	0.1867 (9)	0.24963 (15)	0.1062 (19)	0.75

H16A	0.2563	0.1492	0.2540	0.127*	0.75
H16B	0.3585	0.0834	0.2428	0.127*	0.75
C17A	0.3688 (6)	0.2726 (9)	0.28278 (13)	0.1062 (19)	0.75
H17A	0.4333	0.3086	0.2780	0.159*	0.75
H17B	0.3718	0.1901	0.3030	0.159*	0.75
H17C	0.3318	0.3751	0.2890	0.159*	0.75
C16B	0.2928 (14)	0.259 (3)	0.2569 (5)	0.122 (7)	0.25
H16C	0.2441	0.1659	0.2556	0.147*	0.25
H16D	0.2762	0.3534	0.2736	0.147*	0.25
C17B	0.3965 (14)	0.192 (3)	0.2636 (8)	0.122 (7)	0.25
H17D	0.4131	0.1221	0.2429	0.184*	0.25
H17E	0.4019	0.1198	0.2855	0.184*	0.25
H17F	0.4401	0.2912	0.2668	0.184*	0.25
O3	0.2852 (2)	0.0944 (4)	0.18250 (6)	0.0853 (9)	
O4	0.40074 (19)	0.5902 (3)	0.18304 (7)	0.0793 (8)	
C19A	0.4388 (7)	0.7695 (14)	0.1889 (4)	0.105 (3)	0.53
H19A	0.4366	0.8060	0.2145	0.126*	0.53
H19B	0.4031	0.8548	0.1730	0.126*	0.53
C20A	0.5401 (7)	0.7499 (13)	0.1787 (3)	0.105 (3)	0.53
H20A	0.5726	0.6611	0.1941	0.157*	0.53
H20B	0.5734	0.8614	0.1822	0.157*	0.53
H20C	0.5399	0.7146	0.1533	0.157*	0.53
C19B	0.4500 (8)	0.7644 (17)	0.1791 (5)	0.108 (4)	0.47
H19C	0.4098	0.8599	0.1874	0.130*	0.47
H19D	0.4615	0.7854	0.1534	0.130*	0.47
C20B	0.5450 (7)	0.7572 (16)	0.2024 (3)	0.108 (4)	0.47
H20D	0.5325	0.7451	0.2280	0.163*	0.47
H20E	0.5808	0.8647	0.1990	0.163*	0.47
H20F	0.5822	0.6571	0.1950	0.163*	0.47
O5	0.2939 (2)	0.6807 (4)	0.13788 (7)	0.0874 (9)	
N1	0.0155 (2)	0.0557 (4)	-0.09743 (7)	0.0591 (8)	
N2	0.23819 (19)	0.1593 (4)	0.11064 (7)	0.0544 (8)	
H2	0.2432	0.0765	0.1270	0.065*	
C1	0.1051 (2)	0.3242 (5)	-0.01327 (8)	0.0440 (8)	
C2	0.0601 (2)	0.2734 (4)	-0.05003 (8)	0.0446 (8)	
C3	0.0182 (2)	0.4052 (5)	-0.07279 (9)	0.0618 (10)	
H3	0.0195	0.5229	-0.0649	0.074*	
C4	-0.0251 (3)	0.3623 (6)	-0.10695 (10)	0.0737 (12)	
H4	-0.0542	0.4491	-0.1223	0.088*	
C5	-0.0242 (3)	0.1861 (6)	-0.11782 (9)	0.0719 (11)	
H5	-0.0533	0.1574	-0.1410	0.086*	
C6	0.0558 (2)	0.1015 (4)	-0.06379 (8)	0.0460 (8)	
C7	0.1410 (2)	0.0021 (4)	0.00313 (8)	0.0407 (8)	
C8	0.1395 (2)	0.1820 (4)	0.01247 (8)	0.0390 (8)	
C9	0.1711 (2)	0.2333 (4)	0.04837 (8)	0.0452 (8)	
H9	0.1691	0.3528	0.0549	0.054*	
C10	0.2052 (2)	0.1102 (5)	0.07425 (8)	0.0448 (8)	
C11	0.2065 (2)	-0.0693 (4)	0.06457 (8)	0.0507 (9)	

H11	0.2294	-0.1534	0.0819	0.061*
C12	0.1743 (2)	-0.1219 (4)	0.02973 (8)	0.0468 (8)
H12	0.1746	-0.2421	0.0237	0.056*
C13	0.2620 (2)	0.3227 (5)	0.12156 (8)	0.0525 (9)
H13	0.2555	0.4106	0.1035	0.063*
C14	0.2949 (2)	0.3776 (4)	0.15601 (8)	0.0503 (9)
C15	0.3008 (3)	0.2515 (6)	0.18640 (9)	0.0634 (10)
C18	0.3268 (3)	0.5646 (5)	0.15800 (9)	0.0608 (10)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0609 (6)	0.0445 (6)	0.0529 (5)	0.0014 (5)	0.0003 (4)	-0.0085 (4)
O1	0.0728 (18)	0.0357 (15)	0.0714 (15)	0.0011 (13)	-0.0036 (12)	-0.0025 (12)
O2	0.150 (3)	0.077 (2)	0.0444 (14)	-0.0195 (18)	-0.0106 (15)	0.0048 (13)
C16A	0.175 (6)	0.088 (4)	0.050 (2)	0.010 (4)	-0.023 (3)	0.004 (2)
C17A	0.175 (6)	0.088 (4)	0.050 (2)	0.010 (4)	-0.023 (3)	0.004 (2)
C16B	0.20 (2)	0.056 (13)	0.107 (13)	-0.034 (13)	0.004 (15)	0.009 (9)
C17B	0.20 (2)	0.056 (13)	0.107 (13)	-0.034 (13)	0.004 (15)	0.009 (9)
O3	0.135 (3)	0.0529 (19)	0.0666 (16)	-0.0019 (18)	0.0001 (15)	0.0044 (14)
O4	0.085 (2)	0.0608 (19)	0.0880 (18)	-0.0109 (15)	-0.0233 (15)	-0.0001 (14)
C19A	0.127 (9)	0.055 (5)	0.129 (7)	-0.022 (5)	-0.010 (6)	0.006 (4)
C20A	0.127 (9)	0.055 (5)	0.129 (7)	-0.022 (5)	-0.010 (6)	0.006 (4)
C19B	0.080 (8)	0.102 (8)	0.138 (8)	-0.009 (5)	-0.026 (5)	-0.044 (6)
C20B	0.080 (8)	0.102 (8)	0.138 (8)	-0.009 (5)	-0.026 (5)	-0.044 (6)
O5	0.115 (2)	0.063 (2)	0.0792 (18)	0.0021 (16)	-0.0280 (16)	0.0106 (15)
N1	0.058 (2)	0.072 (2)	0.0474 (17)	-0.0035 (16)	0.0008 (14)	-0.0064 (15)
N2	0.065 (2)	0.053 (2)	0.0452 (16)	-0.0068 (16)	0.0044 (13)	0.0012 (13)
C1	0.034 (2)	0.044 (2)	0.054 (2)	-0.0031 (17)	0.0088 (15)	0.0006 (17)
C2	0.038 (2)	0.046 (2)	0.0496 (19)	0.0001 (16)	0.0032 (15)	0.0073 (16)
C3	0.060 (2)	0.059 (3)	0.065 (2)	0.003 (2)	0.0019 (19)	0.003 (2)
C4	0.075 (3)	0.081 (3)	0.063 (3)	0.007 (2)	-0.008 (2)	0.016 (2)
C5	0.075 (3)	0.091 (4)	0.049 (2)	-0.001 (3)	-0.0056 (19)	0.000 (2)
C6	0.039 (2)	0.054 (2)	0.0446 (18)	-0.0004 (17)	0.0057 (15)	0.0009 (17)
C7	0.035 (2)	0.039 (2)	0.0479 (18)	-0.0044 (16)	0.0062 (15)	0.0002 (15)
C8	0.0351 (19)	0.037 (2)	0.0452 (18)	-0.0044 (15)	0.0061 (14)	-0.0006 (15)
C9	0.044 (2)	0.042 (2)	0.0507 (19)	-0.0033 (16)	0.0073 (16)	-0.0075 (16)
C10	0.047 (2)	0.050 (2)	0.0386 (17)	-0.0050 (17)	0.0066 (14)	0.0010 (16)
C11	0.053 (2)	0.046 (2)	0.053 (2)	-0.0014 (18)	0.0071 (17)	0.0101 (17)
C12	0.051 (2)	0.037 (2)	0.052 (2)	-0.0022 (16)	0.0077 (16)	-0.0007 (15)
C13	0.051 (2)	0.060 (3)	0.047 (2)	-0.0038 (19)	0.0060 (16)	0.0018 (17)
C14	0.059 (2)	0.049 (2)	0.0430 (18)	-0.0047 (18)	0.0061 (16)	0.0007 (16)
C15	0.072 (3)	0.066 (3)	0.051 (2)	0.003 (2)	-0.0017 (19)	-0.005 (2)
C18	0.067 (3)	0.064 (3)	0.050 (2)	0.001 (2)	-0.0044 (19)	-0.0032 (19)

Geometric parameters (Å, °)

S1—C6	1.737 (3)	C20B—H20D	0.9600
S1—C7	1.742 (3)	C20B—H20E	0.9600
O1—C1	1.222 (3)	C20B—H20F	0.9600
O2—C15	1.334 (4)	O5—C18	1.204 (4)
O2—C16A	1.497 (6)	N1—C5	1.321 (4)
O2—C16B	1.506 (14)	N1—C6	1.347 (4)
C16A—C17A	1.472 (7)	N2—C13	1.324 (4)
C16A—H16A	0.9700	N2—C10	1.413 (4)
C16A—H16B	0.9700	N2—H2	0.8600
C17A—H17A	0.9600	C1—C8	1.474 (4)
C17A—H17B	0.9600	C1—C2	1.478 (4)
C17A—H17C	0.9600	C2—C6	1.385 (4)
C16B—C17B	1.520 (10)	C2—C3	1.388 (4)
C16B—H16C	0.9700	C3—C4	1.373 (4)
C16B—H16D	0.9700	C3—H3	0.9300
C17B—H17D	0.9600	C4—C5	1.382 (5)
C17B—H17E	0.9600	C4—H4	0.9300
C17B—H17F	0.9600	C5—H5	0.9300
O3—C15	1.206 (4)	C7—C12	1.395 (4)
O4—C18	1.328 (4)	C7—C8	1.395 (4)
O4—C19A	1.457 (10)	C8—C9	1.396 (4)
O4—C19B	1.487 (12)	C9—C10	1.375 (4)
C19A—C20A	1.480 (9)	C9—H9	0.9300
C19A—H19A	0.9700	C10—C11	1.395 (4)
C19A—H19B	0.9700	C11—C12	1.366 (4)
C20A—H20A	0.9600	C11—H11	0.9300
C20A—H20B	0.9600	C12—H12	0.9300
C20A—H20C	0.9600	C13—C14	1.361 (4)
C19B—C20B	1.506 (10)	C13—H13	0.9300
C19B—H19C	0.9700	C14—C15	1.452 (4)
C19B—H19D	0.9700	C14—C18	1.473 (4)
C6—S1—C7	102.90 (15)	C6—C2—C1	124.7 (3)
C15—O2—C16A	111.4 (4)	C3—C2—C1	118.7 (3)
C15—O2—C16B	129.4 (11)	C4—C3—C2	120.2 (3)
C16A—O2—C16B	28.4 (8)	C4—C3—H3	119.9
C17A—C16A—O2	105.8 (5)	C2—C3—H3	119.9
C17A—C16A—H16A	110.6	C3—C4—C5	118.0 (4)
O2—C16A—H16A	110.6	C3—C4—H4	121.0
C17A—C16A—H16B	110.6	C5—C4—H4	121.0
O2—C16A—H16B	110.6	N1—C5—C4	124.3 (4)
H16A—C16A—H16B	108.7	N1—C5—H5	117.9
O2—C16B—C17B	87.8 (14)	C4—C5—H5	117.9
O2—C16B—H16C	114.0	N1—C6—C2	124.6 (3)
C17B—C16B—H16C	114.0	N1—C6—S1	110.9 (3)
O2—C16B—H16D	114.0	C2—C6—S1	124.5 (2)

C17B—C16B—H16D	114.0	C12—C7—C8	119.4 (3)
H16C—C16B—H16D	111.2	C12—C7—S1	115.9 (2)
C16B—C17B—H17D	109.5	C8—C7—S1	124.7 (2)
C16B—C17B—H17E	109.5	C7—C8—C9	119.0 (3)
H17D—C17B—H17E	109.5	C7—C8—C1	124.0 (3)
C16B—C17B—H17F	109.5	C9—C8—C1	117.0 (3)
H17D—C17B—H17F	109.5	C10—C9—C8	121.2 (3)
H17E—C17B—H17F	109.5	C10—C9—H9	119.4
C18—O4—C19A	118.9 (6)	C8—C9—H9	119.4
C18—O4—C19B	113.3 (6)	C9—C10—C11	119.3 (3)
C19A—O4—C19B	15.6 (11)	C9—C10—N2	122.1 (3)
O4—C19A—C20A	102.0 (8)	C11—C10—N2	118.6 (3)
O4—C19A—H19A	111.4	C12—C11—C10	120.2 (3)
C20A—C19A—H19A	111.4	C12—C11—H11	119.9
O4—C19A—H19B	111.4	C10—C11—H11	119.9
C20A—C19A—H19B	111.4	C11—C12—C7	120.9 (3)
H19A—C19A—H19B	109.2	C11—C12—H12	119.5
O4—C19B—C20B	107.4 (10)	C7—C12—H12	119.5
O4—C19B—H19C	110.2	N2—C13—C14	127.9 (3)
C20B—C19B—H19C	110.2	N2—C13—H13	116.1
O4—C19B—H19D	110.2	C14—C13—H13	116.1
C20B—C19B—H19D	110.2	C13—C14—C15	119.7 (3)
H19C—C19B—H19D	108.5	C13—C14—C18	114.3 (3)
C5—N1—C6	116.3 (3)	C15—C14—C18	125.9 (3)
C13—N2—C10	125.3 (3)	O3—C15—O2	121.9 (3)
C13—N2—H2	117.4	O3—C15—C14	123.4 (3)
C10—N2—H2	117.4	O2—C15—C14	114.6 (4)
O1—C1—C8	121.0 (3)	O5—C18—O4	123.0 (4)
O1—C1—C2	120.4 (3)	O5—C18—C14	124.4 (4)
C8—C1—C2	118.5 (3)	O4—C18—C14	112.5 (3)
C6—C2—C3	116.6 (3)		
C15—O2—C16A—C17A	164.6 (6)	O1—C1—C8—C9	5.8 (4)
C16B—O2—C16A—C17A	-60.2 (19)	C2—C1—C8—C9	-172.7 (2)
C15—O2—C16B—C17B	101.0 (17)	C7—C8—C9—C10	1.2 (4)
C16A—O2—C16B—C17B	42.9 (14)	C1—C8—C9—C10	-179.5 (3)
C18—O4—C19A—C20A	-118.4 (9)	C8—C9—C10—C11	-1.1 (4)
C19B—O4—C19A—C20A	-46 (3)	C8—C9—C10—N2	179.5 (3)
C18—O4—C19B—C20B	-166.4 (9)	C13—N2—C10—C9	-16.8 (5)
C19A—O4—C19B—C20B	79 (3)	C13—N2—C10—C11	163.8 (3)
O1—C1—C2—C6	174.9 (3)	C9—C10—C11—C12	0.0 (4)
C8—C1—C2—C6	-6.5 (4)	N2—C10—C11—C12	179.4 (3)
O1—C1—C2—C3	-5.4 (4)	C10—C11—C12—C7	1.0 (4)
C8—C1—C2—C3	173.2 (3)	C8—C7—C12—C11	-0.9 (4)
C6—C2—C3—C4	0.4 (5)	S1—C7—C12—C11	179.2 (2)
C1—C2—C3—C4	-179.3 (3)	C10—N2—C13—C14	-179.4 (3)
C2—C3—C4—C5	-0.9 (5)	N2—C13—C14—C15	-4.2 (5)
C6—N1—C5—C4	0.9 (5)	N2—C13—C14—C18	172.5 (3)

C3—C4—C5—N1	0.2 (6)	C16A—O2—C15—O3	2.2 (6)
C5—N1—C6—C2	-1.5 (5)	C16B—O2—C15—O3	-23.5 (11)
C5—N1—C6—S1	179.4 (2)	C16A—O2—C15—C14	178.4 (5)
C3—C2—C6—N1	0.9 (4)	C16B—O2—C15—C14	152.7 (10)
C1—C2—C6—N1	-179.4 (3)	C13—C14—C15—O3	6.0 (5)
C3—C2—C6—S1	179.8 (2)	C18—C14—C15—O3	-170.3 (4)
C1—C2—C6—S1	-0.5 (4)	C13—C14—C15—O2	-170.2 (3)
C7—S1—C6—N1	-174.8 (2)	C18—C14—C15—O2	13.5 (5)
C7—S1—C6—C2	6.2 (3)	C19A—O4—C18—O5	5.6 (8)
C6—S1—C7—C12	173.8 (2)	C19B—O4—C18—O5	-10.6 (8)
C6—S1—C7—C8	-6.1 (3)	C19A—O4—C18—C14	-178.4 (6)
C12—C7—C8—C9	-0.2 (4)	C19B—O4—C18—C14	165.4 (7)
S1—C7—C8—C9	179.7 (2)	C13—C14—C18—O5	29.7 (5)
C12—C7—C8—C1	-179.4 (3)	C15—C14—C18—O5	-153.8 (4)
S1—C7—C8—C1	0.5 (4)	C13—C14—C18—O4	-146.2 (3)
O1—C1—C8—C7	-175.0 (3)	C15—C14—C18—O4	30.3 (5)
C2—C1—C8—C7	6.5 (4)		

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2—H2 \cdots O3	0.86	2.06	2.683 (3)	129
C11—H11 \cdots O5 ⁱ	0.93	2.49	3.403 (4)	167
C12—H12 \cdots O1 ⁱ	0.93	2.45	3.322 (4)	157
C17A—H17C \cdots O3 ⁱⁱ	0.96	2.58	3.516 (8)	165
C19A—H19B \cdots O3 ⁱⁱⁱ	0.97	2.47	3.231 (11)	135

Symmetry codes: (i) $x, y-1, z$; (ii) $-x+1/2, y+1/2, -z+1/2$; (iii) $x, y+1, z$.