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(Z)-N-{3-[(6-Chloropyridin-3-yl)methyl]-1,3-thiazolidin-2-ylidene}cyanamideJin-Sheng Gao,^{a,b*} Jun Qiao,^a Ying-Hui Yu^a and Guang-Feng Hou^b^aCollege of Chemistry and Materials Science, Heilongjiang University, Harbin 150080, People's Republic of China, and ^bEngineering Research Center of Pesticide of Heilongjiang Province, Heilongjiang University, Harbin 150080, People's Republic of China

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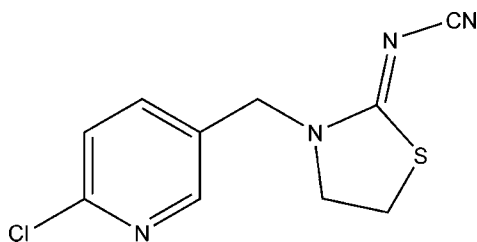
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.048; wR factor = 0.137; data-to-parameter ratio = 17.8.

The asymmetric unit of the title compound, $\text{C}_{10}\text{H}_9\text{ClN}_4\text{S}$, common name thiacloprid, comprises two molecules. In both molecules, the thiazolidine rings are almost planar (with r.m.s. deviations of 0.016 and 0.065 Å) and form dihedral angles of 73.36 (6) and 70.25 (8)° with the 2-chloropyridine rings. In the crystal, intermolecular $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds links the molecules into chains propagating in $[\bar{1}01]$.

Related literature

For background to the title compound, a member of the neonicotinoide class of insecticides, see Maienfisch *et al.* (2003). For the synthesis, see Ishimitsu *et al.*, (1991)



Experimental

Crystal data

 $\text{C}_{10}\text{H}_9\text{ClN}_4\text{S}$
 $M_r = 252.73$ Monoclinic, $P2_1/c$
 $a = 7.1294$ (14) Å $b = 35.469$ (7) Å
 $c = 9.0211$ (18) Å
 $\beta = 97.80$ (3)°
 $V = 2260.1$ (8) Å³
 $Z = 8$ Mo $K\alpha$ radiation
 $\mu = 0.50$ mm⁻¹
 $T = 293$ K
 $0.31 \times 0.29 \times 0.20$ mm

Data collection

Rigaku R-AXIS RAPID
diffractometer
Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)
 $T_{\min} = 0.863$, $T_{\max} = 0.909$ 21869 measured reflections
5151 independent reflections
3505 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.137$
 $S = 1.08$
5151 reflections289 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.21$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.29$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C3}-\text{H3}\cdots\text{N5}^i$	0.93	2.55	3.459 (4)	167
$\text{C13}-\text{H13}\cdots\text{N1}^{ii}$	0.93	2.49	3.408 (4)	169

Symmetry codes: (i) $x - 1, y, z$; (ii) $x, y, z + 1$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalClear* (Rigaku/MSK, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: KP2319).

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supporting information

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(Z)-N-{3-[(6-Chloropyridin-3-yl)methyl]-1,3-thiazolidin-2-ylidene}cyanamide**Jin-Sheng Gao, Jun Qiao, Ying-Hui Yu and Guang-Feng Hou****S1. Comment**

Thiacloprid is the common name of the title compound, which is neonicotinoide class of insecticide. High efficacy and flexible application methods make it well suited for modern integrated pest management programs in many cropping systems (Ishimitsu *et al.*, 1991; Maienfisch *et al.*, 2003). We report here the synthesis and crystal structure of thiacloprid.

The asymmetric unit comprises two molecules; the thiazolidine rings are almost planar, and form the dihedral angles with 2-chloropyridine rings of 73.36 (6) and 70.25 (8)°, respectively (Fig. 1).

In the crystal, the intermolecular C—H···N hydrogen bonds link the molecules to form a chain (Fig. 2, Table 1).

S2. Experimental

The title compound was synthesised according the reference (Ishimitsu *et al.*, 1991). A mixture of 2-cyanoiminothiazolidine (12.7 g, 0.1 mol), 2-chloro-5-pyridylmethyl-chloride (17.4 g, 0.1 mol), and K₂CO₃ (41.4 g, 0.3 mol) in 150 mL of DMF was heated to 323 K and kept stirring for 7 h. After filtered under reduced pressure, the DMF solution was distilled off. A total of 20.2 g (80.2%) title compound was obtained after the recrystallisation from ethyl acetate (15 mL). The suitable colourless block crystal was picked out for the single crystal X-ray diffraction measurement.

S3. Refinement

H atoms bound to C atoms were placed in calculated positions and treated as riding on their parent atoms, with C—H = 0.93 Å (aromatic); C—H = 0.97 Å (methylene), and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

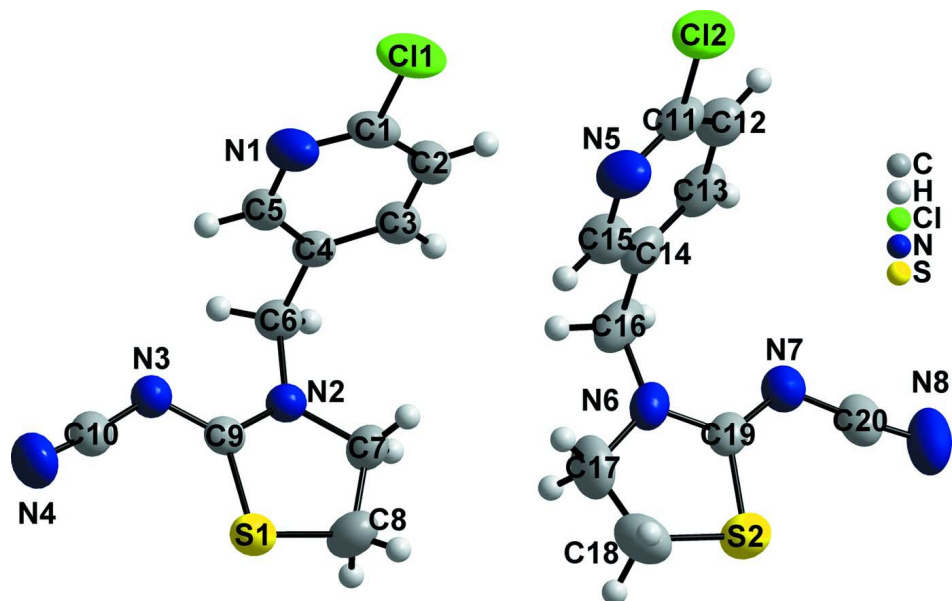


Figure 1

The molecular structure of the title compound showing displacement ellipsoids at the 50% probability level for non-H atoms.

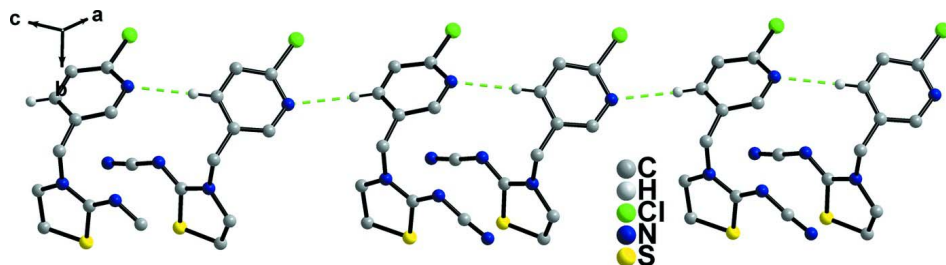


Figure 2

A partial packing view, showing the hydrogen bonding chain. Dashed lines indicate the hydrogen bonds, no involving H atoms have been omitted for clarity.

(Z)-N-[3-[(6-Chloropyridin-3-yl)methyl]-1,3-thiazolidin-2-ylidene]cyanamide

Crystal data

$C_{10}H_9ClN_4S$

$M_r = 252.73$

Monoclinic, $P2_1/c$

Hall symbol: $-P 2_1/c$

$a = 7.1294 (14) \text{ \AA}$

$b = 35.469 (7) \text{ \AA}$

$c = 9.0211 (18) \text{ \AA}$

$\beta = 97.80 (3)^\circ$

$V = 2260.1 (8) \text{ \AA}^3$

$Z = 8$

$F(000) = 1040$

$D_x = 1.486 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 12991 reflections

$\theta = 3.1\text{--}27.5^\circ$

$\mu = 0.50 \text{ mm}^{-1}$

$T = 293 \text{ K}$

Block, colourless

$0.31 \times 0.29 \times 0.20 \text{ mm}$

Data collection

Rigaku R-AXIS RAPID diffractometer	21869 measured reflections
Radiation source: fine-focus sealed tube	5151 independent reflections
Graphite monochromator	3505 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.046$
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)	$\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 3.1^\circ$
$T_{\text{min}} = 0.863$, $T_{\text{max}} = 0.909$	$h = -8 \rightarrow 9$
	$k = -45 \rightarrow 45$
	$l = -11 \rightarrow 11$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.048$	H-atom parameters constrained
$wR(F^2) = 0.137$	$w = 1/[\sigma^2(F_o^2) + (0.0595P)^2 + 0.4995P]$
$S = 1.08$	where $P = (F_o^2 + 2F_c^2)/3$
5151 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$
289 parameters	$\Delta\rho_{\text{max}} = 0.21 \text{ e } \text{\AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.29 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.8652 (3)	0.04805 (6)	0.2994 (3)	0.0528 (6)
C2	0.7778 (4)	0.05431 (7)	0.4241 (3)	0.0564 (6)
H2	0.8315	0.0461	0.5183	0.068*
C3	0.6087 (4)	0.07309 (7)	0.4038 (3)	0.0541 (6)
H3	0.5443	0.0777	0.4850	0.065*
C4	0.5334 (3)	0.08520 (5)	0.2624 (3)	0.0417 (5)
C5	0.6313 (4)	0.07628 (6)	0.1459 (3)	0.0525 (6)
H5	0.5795	0.0835	0.0500	0.063*
C6	0.3545 (3)	0.10860 (6)	0.2362 (3)	0.0486 (5)
H6A	0.2856	0.1025	0.1390	0.058*
H6B	0.2744	0.1024	0.3115	0.058*
C7	0.4452 (4)	0.16785 (7)	0.3852 (3)	0.0586 (6)
H7A	0.5677	0.1592	0.4331	0.070*
H7B	0.3516	0.1622	0.4507	0.070*
C8	0.4506 (6)	0.20899 (8)	0.3572 (4)	0.0831 (10)
H8A	0.5573	0.2202	0.4200	0.100*

H8B	0.3354	0.2207	0.3811	0.100*
C9	0.4189 (3)	0.16902 (6)	0.1222 (3)	0.0422 (5)
C10	0.4177 (3)	0.17589 (7)	-0.1287 (3)	0.0545 (6)
C11	1.3337 (3)	0.05498 (7)	0.7709 (3)	0.0558 (6)
C12	1.2138 (4)	0.04665 (8)	0.8736 (3)	0.0618 (6)
H12	1.2387	0.0268	0.9406	0.074*
C13	1.0549 (4)	0.06898 (8)	0.8737 (3)	0.0609 (7)
H13	0.9712	0.0646	0.9423	0.073*
C14	1.0214 (3)	0.09783 (7)	0.7712 (3)	0.0523 (6)
C15	1.1509 (4)	0.10300 (8)	0.6735 (3)	0.0665 (7)
H15	1.1280	0.1222	0.6033	0.080*
C16	0.8483 (3)	0.12283 (9)	0.7672 (4)	0.0678 (8)
H16A	0.7818	0.1234	0.6660	0.081*
H16B	0.7636	0.1121	0.8315	0.081*
C17	0.8939 (5)	0.19245 (10)	0.7102 (4)	0.0840 (10)
H17A	0.7664	0.1963	0.6595	0.101*
H17B	0.9753	0.1866	0.6353	0.101*
C18	0.9605 (6)	0.22673 (9)	0.7904 (4)	0.0853 (10)
H18A	0.8799	0.2478	0.7556	0.102*
H18B	1.0887	0.2325	0.7728	0.102*
C19	0.9155 (3)	0.17075 (7)	0.9595 (3)	0.0464 (5)
C20	0.9267 (3)	0.15741 (8)	1.2059 (3)	0.0593 (6)
Cl1	1.08715 (10)	0.02601 (2)	0.31970 (11)	0.0811 (3)
Cl2	1.53924 (11)	0.02813 (2)	0.76846 (10)	0.0809 (2)
N1	0.7967 (3)	0.05776 (6)	0.1620 (3)	0.0592 (5)
N2	0.3963 (2)	0.14899 (5)	0.2425 (2)	0.0432 (4)
N3	0.4000 (3)	0.15412 (5)	-0.0121 (2)	0.0504 (5)
N4	0.4288 (4)	0.19221 (7)	-0.2375 (3)	0.0777 (7)
N5	1.3080 (3)	0.08234 (7)	0.6720 (3)	0.0658 (6)
N6	0.8957 (3)	0.16118 (6)	0.8155 (2)	0.0540 (5)
N7	0.9051 (3)	0.14604 (6)	1.0652 (2)	0.0551 (5)
N8	0.9439 (4)	0.16423 (9)	1.3319 (3)	0.0864 (8)
S1	0.47264 (10)	0.216369 (16)	0.16494 (8)	0.05802 (19)
S2	0.95422 (12)	0.21907 (2)	0.98475 (9)	0.0698 (2)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0496 (12)	0.0361 (10)	0.0750 (18)	0.0031 (10)	0.0159 (12)	0.0066 (11)
C2	0.0653 (14)	0.0536 (13)	0.0499 (15)	0.0087 (12)	0.0069 (12)	0.0066 (11)
C3	0.0656 (14)	0.0505 (12)	0.0483 (14)	0.0063 (11)	0.0159 (11)	0.0021 (11)
C4	0.0476 (11)	0.0336 (9)	0.0458 (13)	-0.0015 (9)	0.0138 (9)	0.0020 (8)
C5	0.0664 (14)	0.0486 (12)	0.0450 (14)	0.0081 (11)	0.0160 (11)	0.0053 (10)
C6	0.0465 (11)	0.0424 (11)	0.0580 (15)	-0.0045 (9)	0.0113 (10)	0.0002 (10)
C7	0.0753 (16)	0.0539 (13)	0.0462 (14)	-0.0019 (12)	0.0061 (12)	-0.0049 (11)
C8	0.133 (3)	0.0574 (16)	0.060 (2)	0.0008 (18)	0.0185 (19)	-0.0117 (14)
C9	0.0367 (10)	0.0416 (10)	0.0476 (13)	0.0033 (8)	0.0029 (9)	0.0007 (9)
C10	0.0555 (13)	0.0566 (13)	0.0501 (15)	0.0046 (11)	0.0032 (11)	0.0001 (12)

C11	0.0568 (13)	0.0556 (13)	0.0579 (16)	-0.0027 (11)	0.0177 (12)	-0.0152 (12)
C12	0.0708 (16)	0.0605 (15)	0.0575 (17)	-0.0074 (13)	0.0214 (13)	-0.0025 (12)
C13	0.0601 (14)	0.0736 (16)	0.0541 (16)	-0.0178 (13)	0.0257 (12)	-0.0177 (13)
C14	0.0495 (12)	0.0619 (14)	0.0465 (14)	-0.0090 (11)	0.0106 (10)	-0.0205 (11)
C15	0.0728 (17)	0.0727 (17)	0.0591 (18)	0.0088 (14)	0.0279 (14)	0.0006 (13)
C16	0.0482 (13)	0.0861 (19)	0.0685 (19)	-0.0033 (13)	0.0062 (12)	-0.0297 (15)
C17	0.096 (2)	0.112 (3)	0.0459 (17)	0.009 (2)	0.0141 (15)	0.0145 (17)
C18	0.106 (2)	0.0733 (19)	0.080 (2)	0.0222 (18)	0.0239 (19)	0.0220 (17)
C19	0.0389 (10)	0.0567 (13)	0.0438 (13)	0.0051 (10)	0.0068 (9)	-0.0056 (10)
C20	0.0484 (12)	0.0786 (17)	0.0525 (16)	0.0114 (12)	0.0125 (11)	0.0098 (13)
C11	0.0568 (4)	0.0582 (4)	0.1311 (8)	0.0141 (3)	0.0232 (4)	0.0113 (4)
C12	0.0785 (5)	0.0756 (5)	0.0938 (6)	0.0180 (4)	0.0307 (4)	-0.0044 (4)
N1	0.0684 (13)	0.0531 (11)	0.0623 (15)	0.0100 (10)	0.0310 (11)	0.0084 (10)
N2	0.0454 (9)	0.0407 (9)	0.0437 (11)	0.0020 (8)	0.0068 (8)	-0.0008 (8)
N3	0.0559 (11)	0.0496 (10)	0.0450 (12)	0.0029 (9)	0.0045 (9)	-0.0002 (9)
N4	0.0968 (18)	0.0831 (17)	0.0537 (15)	0.0063 (14)	0.0119 (13)	0.0126 (13)
N5	0.0701 (14)	0.0680 (13)	0.0662 (15)	0.0064 (11)	0.0349 (12)	-0.0018 (11)
N6	0.0504 (10)	0.0702 (13)	0.0413 (11)	0.0095 (10)	0.0058 (8)	-0.0082 (9)
N7	0.0581 (11)	0.0601 (12)	0.0485 (13)	0.0040 (10)	0.0129 (9)	0.0025 (9)
N8	0.0813 (17)	0.133 (2)	0.0470 (15)	0.0174 (16)	0.0147 (12)	0.0044 (15)
S1	0.0681 (4)	0.0420 (3)	0.0615 (4)	-0.0052 (3)	-0.0003 (3)	0.0019 (3)
S2	0.0892 (5)	0.0566 (4)	0.0649 (5)	0.0004 (3)	0.0145 (4)	-0.0053 (3)

Geometric parameters (Å, °)

C1—N1	1.314 (3)	C11—N5	1.314 (4)
C1—C2	1.376 (4)	C11—C12	1.375 (4)
C1—C11	1.752 (2)	C11—C12	1.750 (3)
C2—C3	1.368 (3)	C12—C13	1.383 (4)
C2—H2	0.9300	C12—H12	0.9300
C3—C4	1.383 (3)	C13—C14	1.378 (4)
C3—H3	0.9300	C13—H13	0.9300
C4—C5	1.375 (3)	C14—C15	1.373 (4)
C4—C6	1.513 (3)	C14—C16	1.516 (4)
C5—N1	1.340 (3)	C15—N5	1.340 (4)
C5—H5	0.9300	C15—H15	0.9300
C6—N2	1.463 (3)	C16—N6	1.454 (3)
C6—H6A	0.9700	C16—H16A	0.9700
C6—H6B	0.9700	C16—H16B	0.9700
C7—N2	1.451 (3)	C17—N6	1.459 (4)
C7—C8	1.482 (4)	C17—C18	1.461 (5)
C7—H7A	0.9700	C17—H17A	0.9700
C7—H7B	0.9700	C17—H17B	0.9700
C8—S1	1.782 (3)	C18—S2	1.781 (4)
C8—H8A	0.9700	C18—H18A	0.9700
C8—H8B	0.9700	C18—H18B	0.9700
C9—N3	1.312 (3)	C19—N7	1.305 (3)
C9—N2	1.325 (3)	C19—N6	1.331 (3)

C9—S1	1.754 (2)	C19—S2	1.746 (2)
C10—N4	1.152 (3)	C20—N8	1.152 (4)
C10—N3	1.324 (3)	C20—N7	1.321 (3)
N1—C1—C2	125.2 (2)	C14—C13—C12	119.5 (2)
N1—C1—C11	115.5 (2)	C14—C13—H13	120.2
C2—C1—C11	119.3 (2)	C12—C13—H13	120.2
C3—C2—C1	117.2 (2)	C15—C14—C13	117.3 (2)
C3—C2—H2	121.4	C15—C14—C16	121.5 (3)
C1—C2—H2	121.4	C13—C14—C16	121.1 (3)
C2—C3—C4	120.0 (2)	N5—C15—C14	124.7 (3)
C2—C3—H3	120.0	N5—C15—H15	117.6
C4—C3—H3	120.0	C14—C15—H15	117.6
C5—C4—C3	117.4 (2)	N6—C16—C14	112.62 (19)
C5—C4—C6	120.7 (2)	N6—C16—H16A	109.1
C3—C4—C6	121.9 (2)	C14—C16—H16A	109.1
N1—C5—C4	124.1 (2)	N6—C16—H16B	109.1
N1—C5—H5	118.0	C14—C16—H16B	109.1
C4—C5—H5	118.0	H16A—C16—H16B	107.8
N2—C6—C4	111.60 (17)	N6—C17—C18	109.7 (3)
N2—C6—H6A	109.3	N6—C17—H17A	109.7
C4—C6—H6A	109.3	C18—C17—H17A	109.7
N2—C6—H6B	109.3	N6—C17—H17B	109.7
C4—C6—H6B	109.3	C18—C17—H17B	109.7
H6A—C6—H6B	108.0	H17A—C17—H17B	108.2
N2—C7—C8	108.2 (2)	C17—C18—S2	108.1 (2)
N2—C7—H7A	110.1	C17—C18—H18A	110.1
C8—C7—H7A	110.1	S2—C18—H18A	110.1
N2—C7—H7B	110.1	C17—C18—H18B	110.1
C8—C7—H7B	110.1	S2—C18—H18B	110.1
H7A—C7—H7B	108.4	H18A—C18—H18B	108.4
C7—C8—S1	108.5 (2)	N7—C19—N6	122.2 (2)
C7—C8—H8A	110.0	N7—C19—S2	125.99 (19)
S1—C8—H8A	110.0	N6—C19—S2	111.81 (18)
C7—C8—H8B	110.0	N8—C20—N7	174.3 (3)
S1—C8—H8B	110.0	C1—N1—C5	116.1 (2)
H8A—C8—H8B	108.4	C9—N2—C7	116.01 (19)
N3—C9—N2	122.1 (2)	C9—N2—C6	122.68 (19)
N3—C9—S1	125.52 (18)	C7—N2—C6	120.61 (19)
N2—C9—S1	112.34 (17)	C9—N3—C10	119.3 (2)
N4—C10—N3	174.2 (3)	C11—N5—C15	115.9 (2)
N5—C11—C12	125.0 (2)	C19—N6—C16	121.3 (2)
N5—C11—C12	115.7 (2)	C19—N6—C17	115.6 (2)
C12—C11—C12	119.3 (2)	C16—N6—C17	122.3 (2)
C11—C12—C13	117.5 (3)	C19—N7—C20	119.1 (2)
C11—C12—H12	121.3	C9—S1—C8	91.61 (12)
C13—C12—H12	121.3	C19—S2—C18	92.62 (14)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C3—H3 \cdots N5 ⁱ	0.93	2.55	3.459 (4)	167
C13—H13 \cdots N1 ⁱⁱ	0.93	2.49	3.408 (4)	169

Symmetry codes: (i) $x-1, y, z$; (ii) $x, y, z+1$.