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4-Amino-1-(2-benzoyl-1-phenylethyl)-3-phenyl-1*H*-1,2,4-triazol-5(4*H*)-thione

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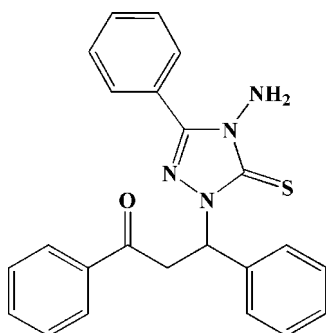
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 Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.037; wR factor = 0.121; data-to-parameter ratio = 13.2.

In the title compound, $\text{C}_{23}\text{H}_{20}\text{N}_4\text{OS}$, the two phenyl rings of the diphenylpropanone fragment form a dihedral angle of $86.8(1)^\circ$, and the third phenyl ring attached to the triazole ring is twisted from the latter at $40.1(1)^\circ$. In the crystal, molecules are paired into centrosymmetric dimers *via* pairs of intermolecular $\text{N}-\text{H}\cdots\text{O}$ and $\text{N}-\text{H}\cdots\text{S}$ hydrogen bonds.

Related literature

For the crystal structures of related 1,2,4-triazole-5(4*H*)-thione derivatives, see: Al-Tamimi *et al.* (2010); Fun *et al.* (2009); Tan *et al.* (2010); Wang *et al.* (2011).



Experimental

Crystal data

 $\text{C}_{23}\text{H}_{20}\text{N}_4\text{OS}$
 $M_r = 400.49$

 Triclinic, $P\bar{1}$
 $a = 9.4625(19)$ Å
 $b = 11.340(2)$ Å
 $c = 11.655(2)$ Å
 $\alpha = 111.80(3)^\circ$
 $\beta = 111.01(3)^\circ$
 $\gamma = 98.91(3)^\circ$
 $V = 1022.4(4)$ Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.18$ mm⁻¹
 $T = 113$ K
 $0.20 \times 0.18 \times 0.10$ mm

Data collection

 Rigaku Saturn CCD area-detector diffractometer
 Absorption correction: multi-scan (*CrystalClear*; Rigaku/MS, 2005)
 $T_{\min} = 0.965$, $T_{\max} = 0.982$

 9308 measured reflections
 3590 independent reflections
 2686 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.121$
 $S = 1.04$
 3590 reflections
 271 parameters

 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.35$ e Å⁻³
 $\Delta\rho_{\min} = -0.29$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|---|--------------|--------------------|-------------|----------------------|
| $\text{N4}-\text{H4A}\cdots\text{O1}^i$ | 0.91 (2) | 2.39 (2) | 2.873 (3) | 114 (2) |
| $\text{N4}-\text{H4B}\cdots\text{S1}^i$ | 0.91 (2) | 2.66 (2) | 3.490 (2) | 151 (2) |

 Symmetry code: (i) $-x + 1, -y, -z$.

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5022).

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supporting information

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4-Amino-1-(2-benzoyl-1-phenylethyl)-3-phenyl-1*H*-1,2,4-triazol-5(4*H*)-thione

Xiao-qiu Song, Lin Ye and He-wen Wang

S1. Comment

In continuation of structural study of 1,2,4-triazole-5(4*H*)-thione derivatives in our group (Wang *et al.*, 2011), we present here the crystal structure of the title compound, (I).

In (I) (Fig. 1), the bond lengths and angles are found to have normal values comparable with those observed in the related 1,2,4-triazole-5(4*H*)-thione derivatives (Al-Tamimi *et al.*, 2010; Fun *et al.*, 2009; Tan *et al.*, 2010; Wang *et al.*, 2011). There are three phenyl rings in the molecule. Phenyl ring A (C1—C6) attached in the triazole ring makes the dihedral angles of 61.0 (1) and 70.9 (1)° with the phenyl ring B (C10—C15) and C (C18—C23), respectively. Rings B and C form a dihedral angle of 86.8 (1)°.

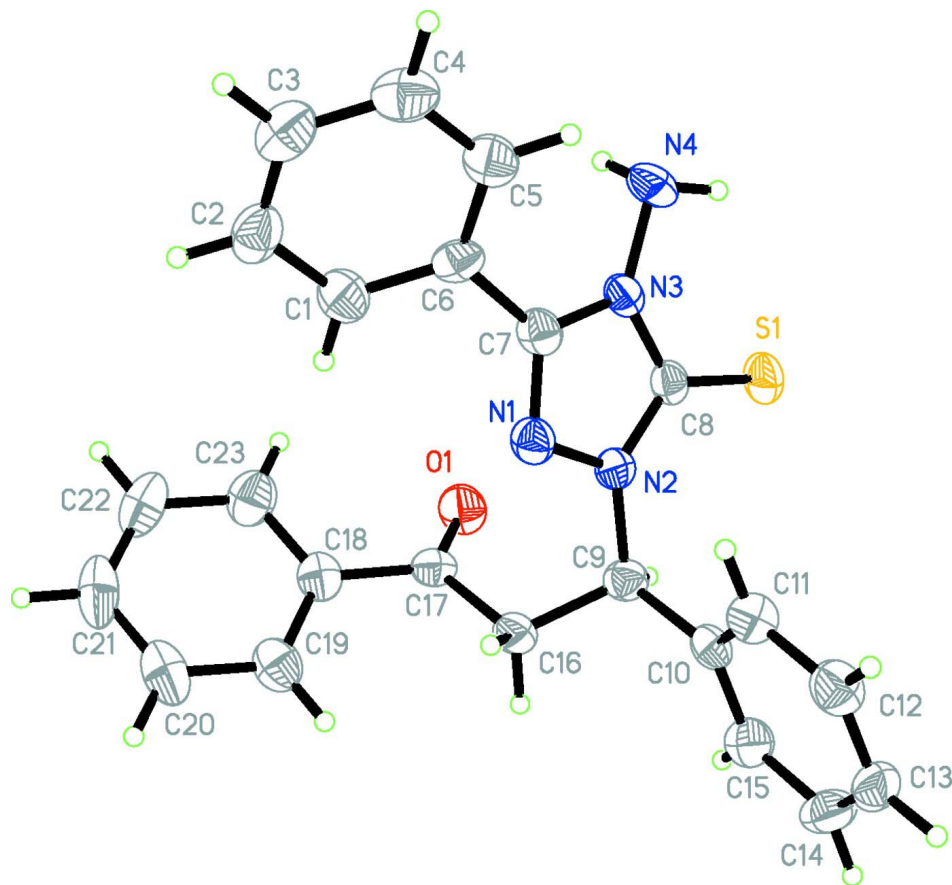
Intermolecular N—H···S and N—H···O hydrogen bonds (Table 1) link the adjacent molecules into centrosymmetric dimers.

S2. Experimental

The title compound was synthesized by the reaction of the chalcone (2.0 mmol) with 4-amino-3-phenyl-4*H*-1,2,4-triazole-5-thiol (2.0 mmol) by refluxing in ethanol for 24 h. The reaction progress was monitored *via* TLC. The resulting precipitate was filtered off, washed with cold ethanol, dried and purified to give the target product as colorless solid in 87% yield. Crystals of (I) suitable for single-crystal X-ray analysis were grown by slow evaporation of a solution in chloroform-ethanol (1:1).

S3. Refinement

The H atoms attached to N atoms were located on a difference map, and isotropically refined using bond length restraint N—H = 0.91 (2) Å. C-bound H atoms were positioned geometrically (C—H = 0.95–1.00 Å), and refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$.

**Figure 1**

View of the molecule of (I) showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 60% probability level.

4-Amino-1-(2-benzoyl-1-phenylethyl)-3-phenyl-1*H*-1,2,4-triazol-5(4*H*)-thione

Crystal data

$C_{23}H_{20}N_4OS$
 $M_r = 400.49$
 Triclinic, $P\bar{1}$
 Hall symbol: -P 1
 $a = 9.4625$ (19) Å
 $b = 11.340$ (2) Å
 $c = 11.655$ (2) Å
 $\alpha = 111.80$ (3)°
 $\beta = 111.01$ (3)°
 $\gamma = 98.91$ (3)°
 $V = 1022.4$ (4) Å³

$Z = 2$
 $F(000) = 420$
 $D_x = 1.301$ Mg m⁻³
 Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
 Cell parameters from 3363 reflections
 $\theta = 2.1$ – 27.8 °
 $\mu = 0.18$ mm⁻¹
 $T = 113$ K
 Prism, colourless
 $0.20 \times 0.18 \times 0.10$ mm

Data collection

Rigaku Saturn CCD area-detector
 diffractometer
 Radiation source: rotating anode
 Multilayer monochromator
 Detector resolution: 7.31 pixels mm⁻¹

φ and ω scans
 Absorption correction: multi-scan
 (*CrystalClear*; Rigaku/MSK, 2005)
 $T_{\min} = 0.965$, $T_{\max} = 0.982$
 9308 measured reflections

3590 independent reflections
 2686 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.039$
 $\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 2.1^\circ$

$h = -11 \rightarrow 11$
 $k = -13 \rightarrow 13$
 $l = -13 \rightarrow 12$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.121$
 $S = 1.04$
 3590 reflections
 271 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods
 Secondary atom site location: difference Fourier
 map

Hydrogen site location: inferred from
 neighbouring sites
 H atoms treated by a mixture of independent
 and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.071P)^2 + 0.1941P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.35 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.29 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXL97* (Sheldrick,
 2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
 Extinction coefficient: 0.107 (8)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|---------------|--------------|----------------------------------|
| S1 | 0.54562 (6) | -0.11721 (5) | 0.13685 (5) | 0.02158 (19) |
| O1 | 0.45937 (17) | 0.25822 (14) | 0.24735 (14) | 0.0237 (4) |
| N1 | 0.82240 (19) | 0.25821 (16) | 0.32190 (16) | 0.0182 (4) |
| N2 | 0.72008 (19) | 0.15086 (16) | 0.31246 (16) | 0.0163 (4) |
| N3 | 0.75632 (19) | 0.07328 (15) | 0.13230 (15) | 0.0159 (4) |
| N4 | 0.7447 (2) | -0.01242 (17) | 0.00269 (17) | 0.0213 (4) |
| H4A | 0.703 (2) | -0.0962 (13) | -0.010 (2) | 0.036 (7)* |
| H4B | 0.675 (2) | 0.005 (2) | -0.062 (2) | 0.055 (9)* |
| C1 | 0.9302 (2) | 0.4216 (2) | 0.2059 (2) | 0.0231 (5) |
| H1 | 0.8614 | 0.4514 | 0.2443 | 0.028* |
| C2 | 1.0231 (3) | 0.5072 (2) | 0.1812 (2) | 0.0252 (5) |
| H2 | 1.0168 | 0.5949 | 0.2010 | 0.030* |
| C3 | 1.1260 (2) | 0.4637 (2) | 0.1269 (2) | 0.0257 (5) |
| H3 | 1.1913 | 0.5223 | 0.1111 | 0.031* |
| C4 | 1.1330 (3) | 0.3347 (2) | 0.0962 (2) | 0.0260 (5) |
| H4 | 1.2038 | 0.3059 | 0.0600 | 0.031* |
| C5 | 1.0376 (2) | 0.2475 (2) | 0.1179 (2) | 0.0221 (5) |
| H5 | 1.0408 | 0.1586 | 0.0943 | 0.026* |
| C6 | 0.9371 (2) | 0.2915 (2) | 0.17470 (19) | 0.0176 (4) |

| | | | | |
|------|------------|--------------|--------------|------------|
| C7 | 0.8422 (2) | 0.20849 (19) | 0.21041 (19) | 0.0176 (5) |
| C8 | 0.6737 (2) | 0.03559 (19) | 0.19571 (19) | 0.0160 (4) |
| C9 | 0.6600 (2) | 0.17172 (19) | 0.41591 (19) | 0.0178 (5) |
| H9 | 0.5473 | 0.1088 | 0.3687 | 0.021* |
| C10 | 0.7610 (2) | 0.14453 (18) | 0.53269 (19) | 0.0176 (4) |
| C11 | 0.9151 (2) | 0.1418 (2) | 0.5597 (2) | 0.0222 (5) |
| H11 | 0.9606 | 0.1555 | 0.5032 | 0.027* |
| C12 | 1.0040 (3) | 0.1191 (2) | 0.6688 (2) | 0.0245 (5) |
| H12 | 1.1093 | 0.1169 | 0.6861 | 0.029* |
| C13 | 0.9392 (3) | 0.0995 (2) | 0.7523 (2) | 0.0237 (5) |
| H13 | 0.9991 | 0.0824 | 0.8259 | 0.028* |
| C14 | 0.7863 (3) | 0.1051 (2) | 0.7281 (2) | 0.0250 (5) |
| H14 | 0.7426 | 0.0939 | 0.7866 | 0.030* |
| C15 | 0.6970 (2) | 0.1268 (2) | 0.6188 (2) | 0.0215 (5) |
| H15 | 0.5921 | 0.1297 | 0.6023 | 0.026* |
| C16 | 0.6580 (2) | 0.31677 (19) | 0.47288 (19) | 0.0187 (5) |
| H16A | 0.6083 | 0.3303 | 0.5364 | 0.022* |
| H16B | 0.7697 | 0.3799 | 0.5269 | 0.022* |
| C17 | 0.5659 (2) | 0.3486 (2) | 0.35832 (19) | 0.0174 (4) |
| C18 | 0.6095 (2) | 0.4905 (2) | 0.3827 (2) | 0.0188 (5) |
| C19 | 0.6976 (2) | 0.5982 (2) | 0.5161 (2) | 0.0229 (5) |
| H19 | 0.7328 | 0.5816 | 0.5941 | 0.028* |
| C20 | 0.7336 (3) | 0.7291 (2) | 0.5348 (2) | 0.0278 (5) |
| H20 | 0.7929 | 0.8020 | 0.6258 | 0.033* |
| C21 | 0.6842 (3) | 0.7544 (2) | 0.4223 (3) | 0.0298 (6) |
| H21 | 0.7084 | 0.8445 | 0.4361 | 0.036* |
| C22 | 0.5990 (3) | 0.6483 (2) | 0.2889 (2) | 0.0317 (6) |
| H22 | 0.5672 | 0.6657 | 0.2114 | 0.038* |
| C23 | 0.5603 (3) | 0.5168 (2) | 0.2688 (2) | 0.0261 (5) |
| H23 | 0.5003 | 0.4443 | 0.1775 | 0.031* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|-------------|-------------|-------------|-------------|-------------|-------------|
| S1 | 0.0236 (3) | 0.0133 (3) | 0.0240 (3) | 0.0040 (2) | 0.0089 (2) | 0.0076 (2) |
| O1 | 0.0232 (8) | 0.0199 (8) | 0.0169 (7) | 0.0047 (6) | 0.0019 (6) | 0.0057 (6) |
| N1 | 0.0185 (9) | 0.0149 (9) | 0.0190 (9) | 0.0036 (7) | 0.0084 (7) | 0.0066 (7) |
| N2 | 0.0188 (9) | 0.0134 (8) | 0.0154 (8) | 0.0037 (7) | 0.0078 (7) | 0.0061 (7) |
| N3 | 0.0171 (9) | 0.0136 (8) | 0.0144 (8) | 0.0062 (7) | 0.0056 (7) | 0.0049 (7) |
| N4 | 0.0235 (10) | 0.0178 (9) | 0.0152 (9) | 0.0066 (8) | 0.0075 (8) | 0.0017 (7) |
| C1 | 0.0231 (11) | 0.0187 (11) | 0.0211 (11) | 0.0038 (9) | 0.0079 (9) | 0.0061 (9) |
| C2 | 0.0282 (12) | 0.0217 (11) | 0.0192 (11) | 0.0033 (10) | 0.0051 (10) | 0.0106 (9) |
| C3 | 0.0215 (11) | 0.0267 (12) | 0.0212 (11) | -0.0007 (9) | 0.0052 (9) | 0.0114 (9) |
| C4 | 0.0206 (11) | 0.0295 (12) | 0.0226 (11) | 0.0032 (10) | 0.0104 (10) | 0.0084 (10) |
| C5 | 0.0217 (11) | 0.0198 (11) | 0.0170 (10) | 0.0039 (9) | 0.0057 (9) | 0.0052 (9) |
| C6 | 0.0154 (10) | 0.0195 (10) | 0.0105 (9) | 0.0006 (8) | 0.0029 (8) | 0.0049 (8) |
| C7 | 0.0161 (10) | 0.0146 (10) | 0.0158 (10) | 0.0026 (8) | 0.0045 (8) | 0.0046 (8) |
| C8 | 0.0166 (10) | 0.0142 (10) | 0.0161 (10) | 0.0064 (8) | 0.0057 (9) | 0.0070 (8) |

| | | | | | | |
|-----|-------------|-------------|-------------|-------------|-------------|-------------|
| C9 | 0.0185 (10) | 0.0168 (10) | 0.0156 (10) | 0.0038 (8) | 0.0075 (9) | 0.0062 (8) |
| C10 | 0.0200 (10) | 0.0105 (9) | 0.0166 (10) | 0.0027 (8) | 0.0068 (9) | 0.0032 (8) |
| C11 | 0.0235 (11) | 0.0216 (11) | 0.0215 (11) | 0.0076 (9) | 0.0106 (9) | 0.0096 (9) |
| C12 | 0.0238 (11) | 0.0226 (11) | 0.0225 (11) | 0.0090 (9) | 0.0078 (10) | 0.0080 (9) |
| C13 | 0.0308 (12) | 0.0195 (11) | 0.0153 (10) | 0.0067 (9) | 0.0053 (10) | 0.0082 (9) |
| C14 | 0.0314 (12) | 0.0247 (11) | 0.0183 (11) | 0.0041 (10) | 0.0130 (10) | 0.0098 (9) |
| C15 | 0.0206 (11) | 0.0191 (10) | 0.0216 (11) | 0.0032 (9) | 0.0090 (9) | 0.0081 (9) |
| C16 | 0.0213 (10) | 0.0152 (10) | 0.0158 (10) | 0.0036 (8) | 0.0079 (9) | 0.0048 (8) |
| C17 | 0.0177 (10) | 0.0196 (10) | 0.0142 (10) | 0.0065 (9) | 0.0074 (9) | 0.0069 (8) |
| C18 | 0.0143 (10) | 0.0194 (11) | 0.0241 (11) | 0.0062 (8) | 0.0097 (9) | 0.0103 (9) |
| C19 | 0.0196 (11) | 0.0195 (11) | 0.0253 (11) | 0.0043 (9) | 0.0094 (9) | 0.0077 (9) |
| C20 | 0.0234 (11) | 0.0180 (11) | 0.0388 (13) | 0.0067 (9) | 0.0149 (10) | 0.0094 (10) |
| C21 | 0.0252 (12) | 0.0222 (12) | 0.0519 (15) | 0.0120 (10) | 0.0196 (11) | 0.0232 (11) |
| C22 | 0.0323 (13) | 0.0343 (13) | 0.0386 (13) | 0.0124 (11) | 0.0151 (11) | 0.0273 (11) |
| C23 | 0.0238 (11) | 0.0267 (12) | 0.0271 (12) | 0.0071 (10) | 0.0089 (10) | 0.0149 (10) |

Geometric parameters (Å, °)

| | | | |
|----------|-------------|-------------|-----------|
| S1—C8 | 1.677 (2) | C10—C11 | 1.386 (3) |
| O1—C17 | 1.225 (2) | C10—C15 | 1.400 (3) |
| N1—C7 | 1.305 (3) | C11—C12 | 1.391 (3) |
| N1—N2 | 1.375 (2) | C11—H11 | 0.9500 |
| N2—C8 | 1.351 (2) | C12—C13 | 1.385 (3) |
| N2—C9 | 1.468 (3) | C12—H12 | 0.9500 |
| N3—C8 | 1.373 (3) | C13—C14 | 1.387 (3) |
| N3—C7 | 1.374 (3) | C13—H13 | 0.9500 |
| N3—N4 | 1.413 (2) | C14—C15 | 1.387 (3) |
| N4—H4A | 0.91 (2) | C14—H14 | 0.9500 |
| N4—H4B | 0.91 (2) | C15—H15 | 0.9500 |
| C1—C2 | 1.387 (3) | C16—C17 | 1.514 (3) |
| C1—C6 | 1.400 (3) | C16—H16A | 0.9900 |
| C1—H1 | 0.9500 | C16—H16B | 0.9900 |
| C2—C3 | 1.396 (3) | C17—C18 | 1.488 (3) |
| C2—H2 | 0.9500 | C18—C19 | 1.396 (3) |
| C3—C4 | 1.389 (3) | C18—C23 | 1.400 (3) |
| C3—H3 | 0.9500 | C19—C20 | 1.385 (3) |
| C4—C5 | 1.389 (3) | C19—H19 | 0.9500 |
| C4—H4 | 0.9500 | C20—C21 | 1.379 (3) |
| C5—C6 | 1.395 (3) | C20—H20 | 0.9500 |
| C5—H5 | 0.9500 | C21—C22 | 1.388 (3) |
| C6—C7 | 1.471 (3) | C21—H21 | 0.9500 |
| C9—C10 | 1.527 (3) | C22—C23 | 1.386 (3) |
| C9—C16 | 1.534 (3) | C22—H22 | 0.9500 |
| C9—H9 | 1.0000 | C23—H23 | 0.9500 |
| C7—N1—N2 | 104.69 (16) | C10—C11—C12 | 120.6 (2) |
| C8—N2—N1 | 113.11 (16) | C10—C11—H11 | 119.7 |
| C8—N2—C9 | 126.58 (17) | C12—C11—H11 | 119.7 |

| | | | |
|-------------|--------------|-----------------|--------------|
| N1—N2—C9 | 120.00 (15) | C13—C12—C11 | 120.2 (2) |
| C8—N3—C7 | 109.01 (16) | C13—C12—H12 | 119.9 |
| C8—N3—N4 | 124.81 (16) | C11—C12—H12 | 119.9 |
| C7—N3—N4 | 126.05 (17) | C12—C13—C14 | 119.66 (19) |
| N3—N4—H4A | 104.0 (15) | C12—C13—H13 | 120.2 |
| N3—N4—H4B | 106.7 (17) | C14—C13—H13 | 120.2 |
| H4A—N4—H4B | 111.0 (14) | C15—C14—C13 | 120.3 (2) |
| C2—C1—C6 | 120.4 (2) | C15—C14—H14 | 119.9 |
| C2—C1—H1 | 119.8 | C13—C14—H14 | 119.9 |
| C6—C1—H1 | 119.8 | C14—C15—C10 | 120.3 (2) |
| C1—C2—C3 | 119.5 (2) | C14—C15—H15 | 119.9 |
| C1—C2—H2 | 120.2 | C10—C15—H15 | 119.9 |
| C3—C2—H2 | 120.2 | C17—C16—C9 | 112.01 (16) |
| C4—C3—C2 | 120.0 (2) | C17—C16—H16A | 109.2 |
| C4—C3—H3 | 120.0 | C9—C16—H16A | 109.2 |
| C2—C3—H3 | 120.0 | C17—C16—H16B | 109.2 |
| C3—C4—C5 | 120.7 (2) | C9—C16—H16B | 109.2 |
| C3—C4—H4 | 119.6 | H16A—C16—H16B | 107.9 |
| C5—C4—H4 | 119.6 | O1—C17—C18 | 121.08 (18) |
| C4—C5—C6 | 119.5 (2) | O1—C17—C16 | 120.19 (18) |
| C4—C5—H5 | 120.3 | C18—C17—C16 | 118.73 (17) |
| C6—C5—H5 | 120.3 | C19—C18—C23 | 119.19 (19) |
| C5—C6—C1 | 119.83 (19) | C19—C18—C17 | 121.70 (18) |
| C5—C6—C7 | 122.41 (19) | C23—C18—C17 | 119.11 (18) |
| C1—C6—C7 | 117.70 (18) | C20—C19—C18 | 120.1 (2) |
| N1—C7—N3 | 110.27 (18) | C20—C19—H19 | 120.0 |
| N1—C7—C6 | 122.88 (17) | C18—C19—H19 | 120.0 |
| N3—C7—C6 | 126.76 (17) | C21—C20—C19 | 120.5 (2) |
| N2—C8—N3 | 102.87 (16) | C21—C20—H20 | 119.7 |
| N2—C8—S1 | 130.11 (16) | C19—C20—H20 | 119.7 |
| N3—C8—S1 | 127.01 (14) | C20—C21—C22 | 120.1 (2) |
| N2—C9—C10 | 112.15 (16) | C20—C21—H21 | 120.0 |
| N2—C9—C16 | 107.96 (16) | C22—C21—H21 | 120.0 |
| C10—C9—C16 | 111.00 (16) | C23—C22—C21 | 120.0 (2) |
| N2—C9—H9 | 108.6 | C23—C22—H22 | 120.0 |
| C10—C9—H9 | 108.6 | C21—C22—H22 | 120.0 |
| C16—C9—H9 | 108.6 | C22—C23—C18 | 120.2 (2) |
| C11—C10—C15 | 118.93 (18) | C22—C23—H23 | 119.9 |
| C11—C10—C9 | 122.58 (18) | C18—C23—H23 | 119.9 |
| C15—C10—C9 | 118.44 (18) | | |
| C7—N1—N2—C8 | -1.1 (2) | C8—N2—C9—C16 | -145.64 (18) |
| C7—N1—N2—C9 | -175.02 (17) | N1—N2—C9—C16 | 27.4 (2) |
| C6—C1—C2—C3 | -1.1 (3) | N2—C9—C10—C11 | 18.4 (3) |
| C1—C2—C3—C4 | 1.0 (3) | C16—C9—C10—C11 | -102.5 (2) |
| C2—C3—C4—C5 | 0.4 (3) | N2—C9—C10—C15 | -164.09 (17) |
| C3—C4—C5—C6 | -1.8 (3) | C16—C9—C10—C15 | 75.1 (2) |
| C4—C5—C6—C1 | 1.6 (3) | C15—C10—C11—C12 | 1.3 (3) |

| | | | |
|--------------|--------------|-----------------|--------------|
| C4—C5—C6—C7 | -175.21 (18) | C9—C10—C11—C12 | 178.81 (18) |
| C2—C1—C6—C5 | -0.2 (3) | C10—C11—C12—C13 | -0.3 (3) |
| C2—C1—C6—C7 | 176.77 (18) | C11—C12—C13—C14 | -1.1 (3) |
| N2—N1—C7—N3 | -0.5 (2) | C12—C13—C14—C15 | 1.5 (3) |
| N2—N1—C7—C6 | 176.48 (17) | C13—C14—C15—C10 | -0.5 (3) |
| C8—N3—C7—N1 | 1.8 (2) | C11—C10—C15—C14 | -0.9 (3) |
| N4—N3—C7—N1 | 177.66 (16) | C9—C10—C15—C14 | -178.48 (17) |
| C8—N3—C7—C6 | -175.02 (18) | N2—C9—C16—C17 | 53.6 (2) |
| N4—N3—C7—C6 | 0.9 (3) | C10—C9—C16—C17 | 176.93 (16) |
| C5—C6—C7—N1 | 140.4 (2) | C9—C16—C17—O1 | 26.2 (3) |
| C1—C6—C7—N1 | -36.5 (3) | C9—C16—C17—C18 | -152.79 (18) |
| C5—C6—C7—N3 | -43.1 (3) | O1—C17—C18—C19 | 162.63 (19) |
| C1—C6—C7—N3 | 139.9 (2) | C16—C17—C18—C19 | -18.4 (3) |
| N1—N2—C8—N3 | 2.1 (2) | O1—C17—C18—C23 | -17.1 (3) |
| C9—N2—C8—N3 | 175.54 (16) | C16—C17—C18—C23 | 161.87 (18) |
| N1—N2—C8—S1 | -177.12 (14) | C23—C18—C19—C20 | 0.8 (3) |
| C9—N2—C8—S1 | -3.6 (3) | C17—C18—C19—C20 | -178.92 (18) |
| C7—N3—C8—N2 | -2.2 (2) | C18—C19—C20—C21 | -0.5 (3) |
| N4—N3—C8—N2 | -178.20 (15) | C19—C20—C21—C22 | -0.7 (3) |
| C7—N3—C8—S1 | 176.97 (14) | C20—C21—C22—C23 | 1.5 (3) |
| N4—N3—C8—S1 | 1.0 (3) | C21—C22—C23—C18 | -1.2 (3) |
| C8—N2—C9—C10 | 91.8 (2) | C19—C18—C23—C22 | 0.0 (3) |
| N1—N2—C9—C10 | -95.16 (19) | C17—C18—C23—C22 | 179.7 (2) |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|-----------------------------------|-------------|---------------|-----------------------|-------------------------|
| N4—H4 <i>A</i> ...O1 ⁱ | 0.91 (2) | 2.39 (2) | 2.873 (3) | 114 (2) |
| N4—H4 <i>A</i> ...S1 | 0.91 (2) | 2.69 (2) | 3.194 (2) | 116 (2) |
| N4—H4 <i>B</i> ...S1 ⁱ | 0.91 (2) | 2.66 (2) | 3.490 (2) | 151 (2) |

Symmetry code: (i) $-x+1, -y, -z$.