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Guanidinium L-glutamate

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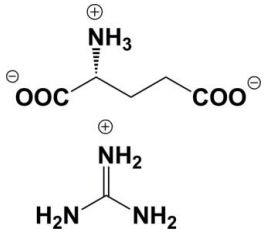
Key indicators: single-crystal X-ray study; T = 150 K; mean σ(C–C) = 0.003 Å; R factor = 0.031; wR factor = 0.081; data-to-parameter ratio = 8.7.

In the title compound, $CH_6N_3^+\cdot C_5H_8NO_4^-$, there are two independent cations and two independent anions in the asymmetric unit. In the crystal structure, cations and anions are linked by intermolecular N—H···O hydrogen bonds into a three-dimensional network.

Related literature

For an early report of salts formed from amino acids and guanidines, see: Armstrong (1956).



Experimental

Crystal data

$CH_6N_3^+\cdot C_5H_8NO_4^-$
M_r = 206.21
Monoclinic, $P2_1$
a = 8.7793 (7) Å
b = 10.8729 (10) Å
c = 10.0801 (9) Å
β = 104.552 (1)°
V = 931.34 (14) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.12 mm^{-1}
T = 150 K
0.42 × 0.26 × 0.20 mm

Data collection

Bruker SMART APEX diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.950, T_{max} = 0.976
5501 measured reflections
2220 independent reflections
2087 reflections with $I > 2\sigma(I)$
R_{int} = 0.021

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.031
$wR(F^2)$ = 0.081
S = 1.06
2220 reflections
255 parameters
1 restraint
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.30 e Å^{-3}
$\Delta\rho_{min}$ = −0.23 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1A···O8	0.91	1.89	2.795 (2)	179
N1—H1B···O4^{i}	0.91	1.84	2.738 (2)	170
N1—H1C···O2^{i}	0.91	2.13	3.017 (2)	165
N2—H2A···O2^{ii}	0.91	2.09	2.998 (2)	173
N2—H2B···O7^{iii}	0.91	2.16	2.740 (2)	120
N2—H2C···O5^{iii}	0.91	1.92	2.817 (3)	170
N3—H3A···O2^{i}	0.88	2.08	2.900 (3)	154
N3—H3B···O3	0.88	2.08	2.841 (3)	145
N4—H4A···O3^{iv}	0.88	1.95	2.826 (2)	173
N4—H4B···O1^{i}	0.88	2.22	3.095 (2)	170
N5—H5A···O4^{iv}	0.88	1.96	2.831 (2)	172
N5—H5B···O6	0.88	2.35	3.092 (3)	142
N6—H6A···O6	0.88	2.04	2.897 (2)	165
N6—H6B···O8^{v}	0.88	1.97	2.824 (2)	164
N7—H7A···O5	0.88	2.00	2.851 (2)	163
N7—H7B···O8^{vi}	0.88	2.02	2.775 (3)	143
N8—H8A···O7^{v}	0.88	2.08	2.954 (3)	170
N8—H8B···O1^{vi}	0.88	2.23	2.953 (3)	140

Symmetry codes: (i) $-x+1, y+\frac{1}{2}, -z$; (ii) $x, y, z+1$; (iii) $-x+1, y-\frac{1}{2}, -z+1$; (iv) $-x+2, y+\frac{1}{2}, -z+1$; (v) $x+1, y, z+1$; (vi) $-x+1, y+\frac{1}{2}, -z+1$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5125).

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Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
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supporting information

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Guanidinium L-glutamate

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S1. Comment

To better understand the formation of complex salts between a guanidine compounds and amino acids we carried out the crystal structure determination of the title compound. The asymmetric unit of the title compound is shown in Fig. 1. There are two independent cations and two indpendent anions in the asymmetric unit. In the crystal structure, cations and anions are linked by intramolecular N—H···O hydrogen bonds into a three-dimensional network (see Fig. 2).

S2. Experimental

L-Glutamic acid (1.47 g.) and guanidine carbonate (0.90 g) were suspended in 10 ml of water. When the evolution of CO_2 had ceased the solution was diluted with 20 ml of acetone, and evaporated to a clear syrup. The syrup was dissolved in 30 ml of absolute methanol to yield a clear solution, and was allowed to stand overnight at room temperature. This solution was then placed in a fume hood for another day, whereupon the crystals of the title compound were collected and dried.

S3. Refinement

In the absence of significant anomalous dispersion effects Friedel pairs were merged. The absolute configuation is known from the starting material. H atoms were placed in calculated positions (C—H = 0.99 or 1.00 Å, N—H = 0.88 or 0.91 Å) and were refined as riding, with U_{iso}(H) = 1.2U_{eq}(C,*N*) or 1.5$_{eq}$(N) for –NH_3 groups.

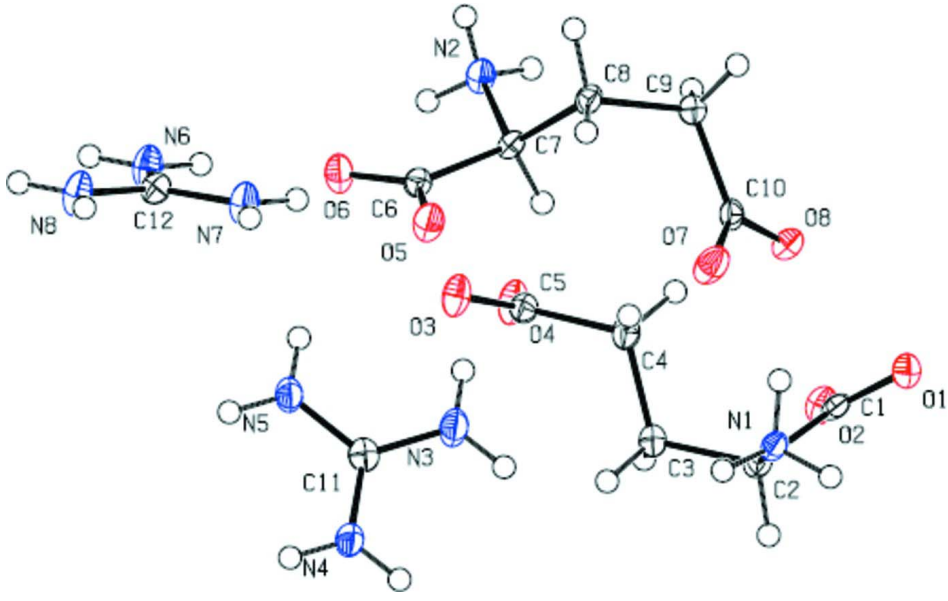

Figure 1

The asymmetric unit of the title compound with displacement ellipsoids drawn at the 30% probability level.

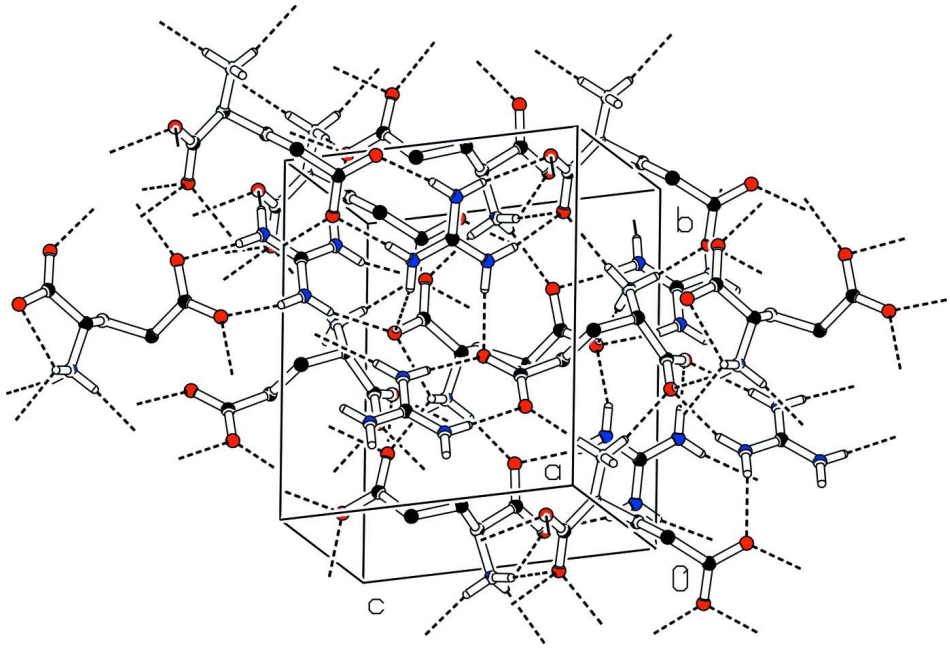

Figure 2

Part of the crystal structure of the title compound with hydrogen bonds shown as dashed lines.

bis(carbamimidoylazanium) (2R)-2-aminopentanedioate

Crystal data

$CH_6N_3^+ \cdot C_5H_8NO_4^-$
$M_r = 206.21$
Monoclinic, $P2_1$
Hall symbol: P 2yb
$a = 8.7793\ (7)$ Å
$b = 10.8729\ (10)$ Å
$c = 10.0801\ (9)$ Å
$\beta = 104.552\ (1)°$
$V = 931.34\ (14)$ Å^3
$Z = 4$
$F(000) = 440$
$D_x = 1.471$ Mg m^{-3}
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 2748 reflections
$\theta = 2.4–27.5°$
$\mu = 0.12$ mm^{-1}
$T = 150$ K
Prism, colourless
0.42 × 0.26 × 0.20 mm

Data collection

Bruker SMART APEX diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
φ and ω scans
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
$T_{min} = 0.950$, $T_{max} = 0.976$
5501 measured reflections
2220 independent reflections
2087 reflections with $I > 2\sigma(I)$
$R_{int} = 0.021$
$\theta_{max} = 27.5°$, $\theta_{min} = 2.4°$
$h = -11 \rightarrow 10$
$k = -14 \rightarrow 9$
$l = -8 \rightarrow 13$

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.031$
$wR(F^2) = 0.081$
$S = 1.06$
2220 reflections
255 parameters
1 restraint
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.044P)^2 + 0.2149P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max} = 0.30$ e Å^{-3}
$\Delta\rho_{min} = -0.23$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	*y*	*z*	U_{iso}*/U_{eq}
C1	0.3633 (2)	0.4498 (2)	−0.1328 (2)	0.0157 (4)
C2	0.4808 (2)	0.5516 (2)	−0.06507 (19)	0.0152 (4)
H2	0.5234	0.5892	−0.1389	0.018*
C3	0.6207 (2)	0.5030 (2)	0.0442 (2)	0.0175 (4)
H3C	0.6924	0.5723	0.0803	0.021*
H3D	0.6793	0.4434	0.0016	0.021*
C4	0.5727 (2)	0.4405 (2)	0.1632 (2)	0.0185 (4)
H4C	0.4937	0.3761	0.1264	0.022*
H4D	0.5227	0.5020	0.2111	0.022*
C5	0.7121 (2)	0.3823 (2)	0.2661 (2)	0.0161 (4)
C6	0.6284 (3)	0.5883 (2)	0.6253 (2)	0.0170 (4)
C7	0.5001 (2)	0.5088 (2)	0.5315 (2)	0.0142 (4)
H7	0.5153	0.5159	0.4368	0.017*
C8	0.3304 (2)	0.5466 (2)	0.52340 (19)	0.0164 (4)
H8C	0.3259	0.6367	0.5357	0.020*
H8D	0.2954	0.5067	0.5990	0.020*
C9	0.2181 (2)	0.5106 (2)	0.3863 (2)	0.0171 (4)
H9A	0.2379	0.4240	0.3655	0.020*
H9B	0.1083	0.5165	0.3946	0.020*
C10	0.2368 (2)	0.5920 (2)	0.2679 (2)	0.0156 (4)
C11	0.9131 (2)	0.7443 (2)	0.4492 (2)	0.0176 (4)
C12	0.9193 (3)	0.7368 (2)	0.9641 (2)	0.0179 (4)
N1	0.3944 (2)	0.65067 (18)	−0.01100 (17)	0.0155 (4)
H1A	0.3351	0.6164	0.0415	0.023*
H1B	0.3307	0.6918	−0.0823	0.023*

H1C	0.4648	0.7040	0.0407	0.023*
N2	0.5301 (2)	0.37747 (18)	0.57395 (19)	0.0183 (4)
H2A	0.4960	0.3633	0.6507	0.027*
H2B	0.6351	0.3617	0.5918	0.027*
H2C	0.4774	0.3275	0.5052	0.027*
N3	0.7857 (2)	0.6972 (2)	0.3634 (2)	0.0240 (4)
H3A	0.7411	0.7361	0.2870	0.029*
H3B	0.7461	0.6272	0.3832	0.029*
N4	0.9733 (2)	0.84922 (19)	0.42024 (19)	0.0204 (4)
H4A	1.0578	0.8797	0.4770	0.024*
H4B	0.9288	0.8887	0.3441	0.024*
N5	0.9797 (2)	0.68433 (19)	0.5639 (2)	0.0215 (4)
H5A	1.0643	0.7145	0.6209	0.026*
H5B	0.9394	0.6144	0.5829	0.026*
N6	0.9603 (2)	0.62410 (19)	0.9389 (2)	0.0242 (4)
H6A	0.9036	0.5832	0.8682	0.029*
H6B	1.0442	0.5899	0.9926	0.029*
N7	0.7924 (2)	0.78798 (19)	0.88260 (19)	0.0222 (4)
H7A	0.7361	0.7467	0.8121	0.027*
H7B	0.7648	0.8631	0.8991	0.027*
N8	1.0012 (2)	0.8000 (2)	1.0716 (2)	0.0248 (5)
H8A	1.0844	0.7668	1.1274	0.030*
H8B	0.9721	0.8751	1.0868	0.030*
O1	0.22034 (18)	0.47085 (14)	−0.15334 (15)	0.0197 (3)
O2	0.42560 (19)	0.35290 (15)	−0.16612 (15)	0.0206 (3)
O3	0.76473 (19)	0.43703 (16)	0.37806 (16)	0.0221 (4)
O4	0.76678 (19)	0.28334 (16)	0.23297 (16)	0.0236 (4)
O5	0.6030 (2)	0.70163 (15)	0.62702 (17)	0.0234 (4)
O6	0.75073 (18)	0.53379 (16)	0.68837 (16)	0.0225 (4)
O7	0.2657 (2)	0.70271 (16)	0.28796 (17)	0.0277 (4)
O8	0.21550 (17)	0.54149 (15)	0.15054 (14)	0.0184 (3)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0218 (10)	0.0148 (10)	0.0098 (9)	−0.0013 (8)	0.0027 (8)	0.0026 (8)
C2	0.0177 (9)	0.0146 (10)	0.0126 (8)	−0.0017 (8)	0.0025 (7)	−0.0002 (8)
C3	0.0153 (9)	0.0193 (10)	0.0165 (10)	0.0003 (9)	0.0013 (8)	−0.0003 (8)
C4	0.0159 (10)	0.0222 (12)	0.0161 (10)	0.0029 (9)	0.0017 (8)	−0.0005 (9)
C5	0.0153 (9)	0.0163 (10)	0.0158 (10)	0.0001 (8)	0.0024 (8)	0.0010 (8)
C6	0.0180 (10)	0.0187 (11)	0.0135 (9)	−0.0030 (9)	0.0024 (8)	0.0002 (8)
C7	0.0177 (9)	0.0134 (10)	0.0110 (9)	0.0014 (8)	0.0023 (7)	0.0009 (8)
C8	0.0167 (9)	0.0203 (10)	0.0115 (9)	0.0013 (8)	0.0024 (7)	0.0006 (8)
C9	0.0166 (9)	0.0184 (10)	0.0151 (9)	−0.0024 (8)	0.0021 (7)	0.0000 (8)
C10	0.0126 (9)	0.0167 (10)	0.0155 (10)	0.0014 (8)	−0.0005 (7)	0.0014 (8)
C11	0.0169 (10)	0.0180 (11)	0.0182 (10)	0.0029 (8)	0.0048 (8)	−0.0029 (8)
C12	0.0186 (10)	0.0176 (11)	0.0169 (10)	0.0009 (8)	0.0035 (8)	0.0015 (8)
N1	0.0186 (8)	0.0142 (8)	0.0122 (8)	−0.0007 (7)	0.0008 (6)	0.0010 (7)

N2	0.0202 (9)	0.0136 (9)	0.0184 (9)	0.0008 (7)	−0.0003 (7)	−0.0010 (7)
N3	0.0246 (10)	0.0203 (10)	0.0214 (9)	−0.0026 (8)	−0.0052 (8)	0.0023 (8)
N4	0.0205 (9)	0.0211 (10)	0.0172 (9)	−0.0025 (8)	0.0001 (7)	0.0001 (8)
N5	0.0189 (9)	0.0211 (10)	0.0206 (9)	−0.0030 (8)	−0.0024 (7)	0.0020 (8)
N6	0.0237 (10)	0.0209 (11)	0.0233 (10)	0.0065 (8)	−0.0028 (8)	−0.0031 (8)
N7	0.0240 (10)	0.0171 (9)	0.0209 (9)	0.0043 (8)	−0.0032 (8)	−0.0035 (8)
N8	0.0284 (10)	0.0203 (11)	0.0195 (9)	0.0056 (8)	−0.0057 (8)	−0.0024 (8)
O1	0.0183 (7)	0.0212 (9)	0.0175 (7)	−0.0006 (6)	0.0010 (6)	−0.0013 (6)
O2	0.0253 (8)	0.0183 (8)	0.0167 (8)	0.0026 (7)	0.0024 (6)	−0.0035 (6)
O3	0.0235 (8)	0.0218 (9)	0.0171 (8)	0.0053 (7)	−0.0020 (6)	−0.0046 (7)
O4	0.0253 (8)	0.0208 (8)	0.0199 (8)	0.0068 (7)	−0.0032 (6)	−0.0048 (7)
O5	0.0280 (9)	0.0137 (8)	0.0232 (8)	−0.0024 (7)	−0.0033 (7)	0.0003 (6)
O6	0.0191 (8)	0.0201 (9)	0.0234 (8)	0.0000 (7)	−0.0035 (6)	−0.0009 (7)
O7	0.0405 (10)	0.0178 (8)	0.0195 (8)	−0.0069 (8)	−0.0025 (7)	0.0026 (7)
O8	0.0220 (7)	0.0185 (8)	0.0143 (7)	0.0015 (6)	0.0039 (6)	0.0016 (6)

Geometric parameters (Å, º)

C1—O1	1.241 (3)	C10—O7	1.237 (3)
C1—O2	1.271 (3)	C10—O8	1.275 (3)
C1—C2	1.549 (3)	C11—N4	1.321 (3)
C2—N1	1.497 (3)	C11—N5	1.328 (3)
C2—C3	1.524 (3)	C11—N3	1.332 (3)
C2—H2	1.0000	C12—N6	1.319 (3)
C3—C4	1.527 (3)	C12—N7	1.329 (3)
C3—H3C	0.9900	C12—N8	1.331 (3)
C3—H3D	0.9900	N1—H1A	0.9100
C4—C5	1.529 (3)	N1—H1B	0.9100
C4—H4C	0.9900	N1—H1C	0.9100
C4—H4D	0.9900	N2—H2A	0.9100
C5—O4	1.257 (3)	N2—H2B	0.9100
C5—O3	1.257 (3)	N2—H2C	0.9100
C6—O6	1.251 (3)	N3—H3A	0.8800
C6—O5	1.253 (3)	N3—H3B	0.8800
C6—C7	1.541 (3)	N4—H4A	0.8800
C7—N2	1.495 (3)	N4—H4B	0.8800
C7—C8	1.528 (3)	N5—H5A	0.8800
C7—H7	1.0000	N5—H5B	0.8800
C8—C9	1.533 (3)	N6—H6A	0.8800
C8—H8C	0.9900	N6—H6B	0.8800
C8—H8D	0.9900	N7—H7A	0.8800
C9—C10	1.527 (3)	N7—H7B	0.8800
C9—H9A	0.9900	N8—H8A	0.8800
C9—H9B	0.9900	N8—H8B	0.8800
O1—C1—O2	126.4 (2)	C10—C9—H9B	109.1
O1—C1—C2	118.4 (2)	C8—C9—H9B	109.1
O2—C1—C2	115.21 (18)	H9A—C9—H9B	107.8

N1—C2—C3	112.11 (16)	O7—C10—O8	123.2 (2)
N1—C2—C1	109.42 (16)	O7—C10—C9	119.6 (2)
C3—C2—C1	113.44 (18)	O8—C10—C9	117.14 (19)
N1—C2—H2	107.2	N4—C11—N5	120.2 (2)
C3—C2—H2	107.2	N4—C11—N3	120.4 (2)
C1—C2—H2	107.2	N5—C11—N3	119.4 (2)
C2—C3—C4	112.99 (17)	N6—C12—N7	119.7 (2)
C2—C3—H3C	109.0	N6—C12—N8	121.3 (2)
C4—C3—H3C	109.0	N7—C12—N8	119.0 (2)
C2—C3—H3D	109.0	C2—N1—H1A	109.5
C4—C3—H3D	109.0	C2—N1—H1B	109.5
H3C—C3—H3D	107.8	H1A—N1—H1B	109.5
C3—C4—C5	112.67 (17)	C2—N1—H1C	109.5
C3—C4—H4C	109.1	H1A—N1—H1C	109.5
C5—C4—H4C	109.1	H1B—N1—H1C	109.5
C3—C4—H4D	109.1	C7—N2—H2A	109.5
C5—C4—H4D	109.1	C7—N2—H2B	109.5
H4C—C4—H4D	107.8	H2A—N2—H2B	109.5
O4—C5—O3	124.40 (19)	C7—N2—H2C	109.5
O4—C5—C4	117.95 (18)	H2A—N2—H2C	109.5
O3—C5—C4	117.6 (2)	H2B—N2—H2C	109.5
O6—C6—O5	126.2 (2)	C11—N3—H3A	120.0
O6—C6—C7	116.6 (2)	C11—N3—H3B	120.0
O5—C6—C7	117.06 (19)	H3A—N3—H3B	120.0
N2—C7—C8	111.79 (18)	C11—N4—H4A	120.0
N2—C7—C6	108.12 (16)	C11—N4—H4B	120.0
C8—C7—C6	115.78 (18)	H4A—N4—H4B	120.0
N2—C7—H7	106.9	C11—N5—H5A	120.0
C8—C7—H7	106.9	C11—N5—H5B	120.0
C6—C7—H7	106.9	H5A—N5—H5B	120.0
C7—C8—C9	112.17 (17)	C12—N6—H6A	120.0
C7—C8—H8C	109.2	C12—N6—H6B	120.0
C9—C8—H8C	109.2	H6A—N6—H6B	120.0
C7—C8—H8D	109.2	C12—N7—H7A	120.0
C9—C8—H8D	109.2	C12—N7—H7B	120.0
H8C—C8—H8D	107.9	H7A—N7—H7B	120.0
C10—C9—C8	112.69 (17)	C12—N8—H8A	120.0
C10—C9—H9A	109.1	C12—N8—H8B	120.0
C8—C9—H9A	109.1	H8A—N8—H8B	120.0

O1—C1—C2—N1	11.3 (3)	O6—C6—C7—N2	18.9 (3)
O2—C1—C2—N1	−170.89 (17)	O5—C6—C7—N2	−163.9 (2)
O1—C1—C2—C3	137.29 (19)	O6—C6—C7—C8	145.18 (19)
O2—C1—C2—C3	−44.9 (2)	O5—C6—C7—C8	−37.6 (3)
N1—C2—C3—C4	63.7 (2)	N2—C7—C8—C9	−83.4 (2)
C1—C2—C3—C4	−60.8 (2)	C6—C7—C8—C9	152.21 (18)
C2—C3—C4—C5	174.99 (19)	C7—C8—C9—C10	−73.5 (2)
C3—C4—C5—O4	−75.2 (3)	C8—C9—C10—O7	−36.4 (3)

C3—C4—C5—O3	104.4 (2)	C8—C9—C10—O8	146.12 (19)

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N1—H1*A*···O8	0.91	1.89	2.795 (2)	179
N1—H1*B*···O4^{i}	0.91	1.84	2.738 (2)	170
N1—H1*C*···O2^{i}	0.91	2.13	3.017 (2)	165
N2—H2*A*···O2^{ii}	0.91	2.09	2.998 (2)	173
N2—H2*B*···O7^{iii}	0.91	2.16	2.740 (2)	120
N2—H2*C*···O5^{iii}	0.91	1.92	2.817 (3)	170
N3—H3*A*···O2^{i}	0.88	2.08	2.900 (3)	154
N3—H3*B*···O3	0.88	2.08	2.841 (3)	145
N4—H4*A*···O3^{iv}	0.88	1.95	2.826 (2)	173
N4—H4*B*···O1^{i}	0.88	2.22	3.095 (2)	170
N5—H5*A*···O4^{iv}	0.88	1.96	2.831 (2)	172
N5—H5*B*···O6	0.88	2.35	3.092 (3)	142
N6—H6*A*···O6	0.88	2.04	2.897 (2)	165
N6—H6*B*···O8^{v}	0.88	1.97	2.824 (2)	164
N7—H7*A*···O5	0.88	2.00	2.851 (2)	163
N7—H7*B*···O8^{vi}	0.88	2.02	2.775 (3)	143
N8—H8*A*···O7^{v}	0.88	2.08	2.954 (3)	170
N8—H8*B*···O1^{vi}	0.88	2.23	2.953 (3)	140

Symmetry codes: (i) −*x*+1, *y*+1/2, −*z*; (ii) *x*, *y*, *z*+1; (iii) −*x*+1, *y*−1/2, −*z*+1; (iv) −*x*+2, *y*+1/2, −*z*+1; (v) *x*+1, *y*, *z*+1; (vi) −*x*+1, *y*+1/2, −*z*+1.