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(E)-N'-[4-(Dimethylamino)benzylidene]-2-methoxybenzohydrazide

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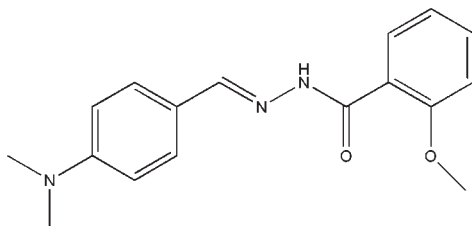
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 Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.039; wR factor = 0.106; data-to-parameter ratio = 10.4.

In the title compound, $\text{C}_{17}\text{H}_{19}\text{N}_3\text{O}_2$, the two benzene rings form a dihedral angle of $89.2(2)^\circ$. In the crystal structure, molecules are linked through $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds, forming $C(4)$ chains running along the c axis.

Related literature

For the medicinal applications of hydrazone compounds, see: Hillmer *et al.* (2010); Zhu *et al.* (2009); Jimenez-Pulido *et al.* (2008); Raj *et al.* (2007); Zhong *et al.* (2007). For hydrazones we have reported previously, see: Liu & You (2010*a,b,c*). For the crystal structures of similar hydrazone compounds, see: Khaledi *et al.* (2009); Warad *et al.* (2009); Back *et al.* (2009); Vijayakumar *et al.* (2009). For other related structures, see: Cao (2009); Xu *et al.* (2009); Shafiq *et al.* (2009).



Experimental

Crystal data

 $\text{C}_{17}\text{H}_{19}\text{N}_3\text{O}_2$
 $M_r = 297.35$
 Orthorhombic, *Pbca*
 $a = 24.726(4)$ Å
 $b = 15.385(3)$ Å
 $c = 8.2700(15)$ Å

 $V = 3146.0(10)$ Å³
 $Z = 8$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 298$ K
 $0.20 \times 0.17 \times 0.13$ mm

Data collection

 Bruker SMART CCD area-detector diffractometer
 Absorption correction: multi-scan (*SADABS*; Bruker, 2001)
 $T_{\min} = 0.983$, $T_{\max} = 0.989$
 12277 measured reflections
 2096 independent reflections
 1545 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.045$
 $\theta_{\max} = 22.7^\circ$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.106$
 $S = 1.04$
 2096 reflections
 202 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.11$ e Å⁻³
 $\Delta\rho_{\min} = -0.15$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O2}^i$	0.90	2.03	2.914 (2)	165

 Symmetry code: (i) $x, -y + \frac{1}{2}, z - \frac{1}{2}$

Data collection: *SMART* (Bruker, 2007); cell refinement: *SAINTE* (Bruker, 2007); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SJ5024).

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supporting information

Acta Cryst. (2010). E66, o1775 [doi:10.1107/S1600536810023937]

(E)-N'-[4-(Dimethylamino)benzylidene]-2-methoxybenzohydrazide**Shi-Yong Liu and Xiaoling Wang****S1. Comment**

Considerable attention has been focused on hydrazones and their medicinal applications (Hillmer *et al.*, 2010; Zhu *et al.*, 2009; Jimenez-Pulido *et al.*, 2008; Raj *et al.*, 2007; Zhong *et al.*, 2007). The study on the crystal structures of such compounds is of particular interest (Khaledi *et al.*, 2009; Warad *et al.*, 2009; Back *et al.*, 2009; Vijayakumar *et al.*, 2009). As a continuation of our work on such compounds (Liu & You, 2010a,b,c), we report herein the crystal structure of the title new hydrazone.

The molecular structure of the title compound is shown in Fig. 1. The dihedral angle between the C1—C6 and C10—C15 benzene rings is 89.2 (2)°, indicating they are nearly perpendicular to each other. All the bond lengths are comparable to those observed in related structures (Cao, 2009; Xu *et al.*, 2009; Shafiq *et al.*, 2009) and those we reported previously.

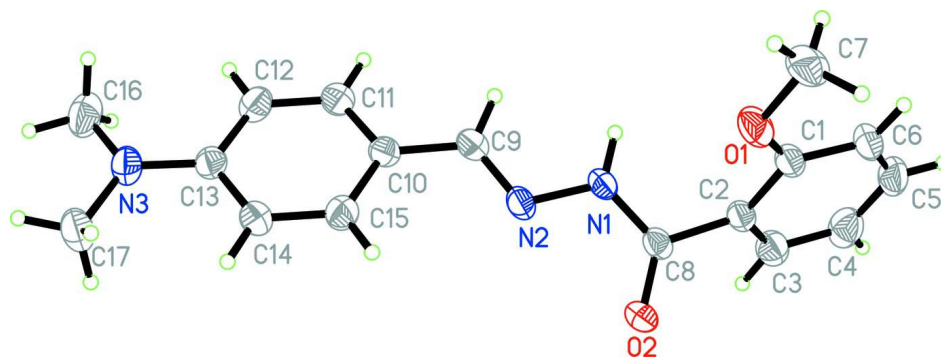
In the crystal structure, molecules are linked through N—H···O hydrogen bonds, to form one-dimensional chains running along the *c* axis (Fig. 2 and Table 1).

S2. Experimental

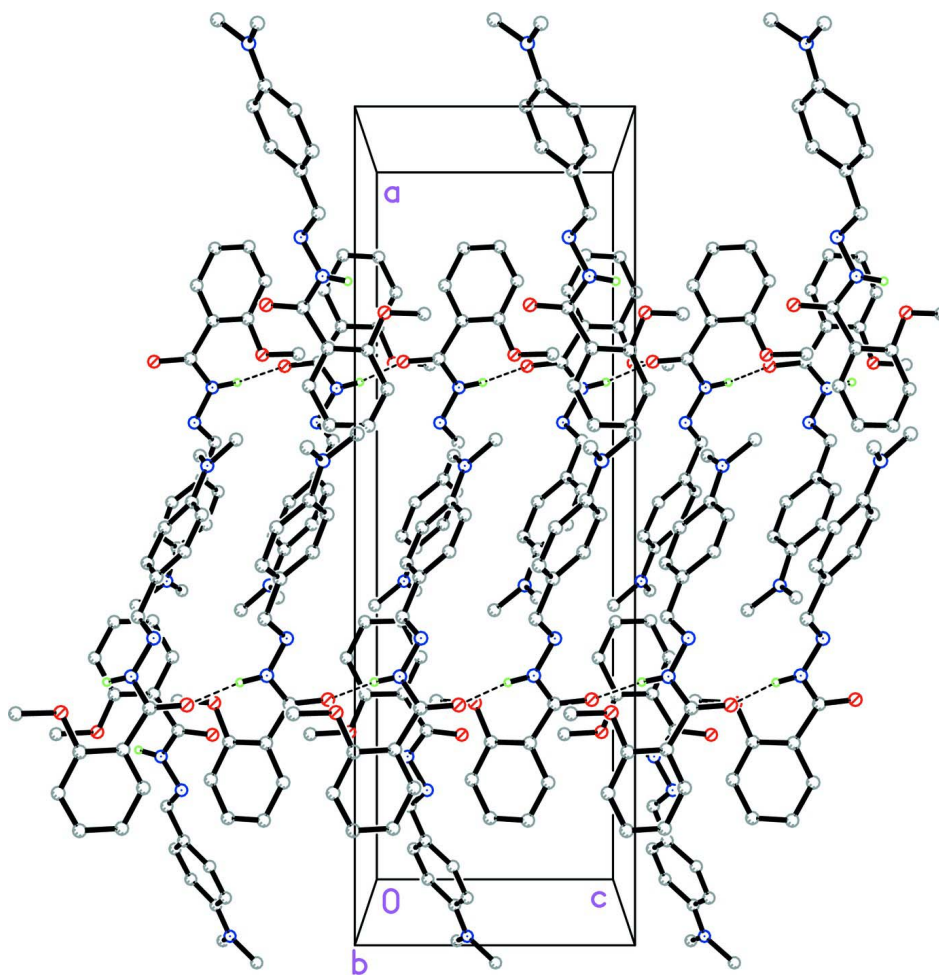
The title compound was prepared by the condensation reaction of 4-dimethylaminobenzaldehyde (0.05 mol, 7.5 g) and 2-methoxybenzohydrazide (0.05 mol, 8.3 g) in anhydrous methanol (200 ml) at ambient temperature. Colourless block-shaped single crystals suitable for X-ray structural determination were obtained by slow evaporation of the solution for a period of 13 d.

S3. Refinement

H atoms were positioned geometrically and constrained to ride on their parent atoms, with C—H distances of 0.93–0.96 Å, and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ and $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$. In the absence of significant anomalous dispersion effects, Friedel pairs were averaged.

**Figure 1**

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 30% probability level. Hydrogen atoms are shown as spheres of arbitrary radius.

**Figure 2**

The molecular packing of the title compound, viewed along the *b* axis. Hydrogen bonds are shown as dashed lines. Hydrogen atoms not involved in hydrogen bonding have been omitted.

(E)-N'-[4-(Dimethylamino)benzylidene]-2-methoxybenzohydrazide*Crystal data*C₁₇H₁₉N₃O₂ $M_r = 297.35$ Orthorhombic, *Pbca* $a = 24.726$ (4) Å $b = 15.385$ (3) Å $c = 8.2700$ (15) Å $V = 3146.0$ (10) Å³ $Z = 8$ $F(000) = 1264$ $D_x = 1.256$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 1881 reflections

 $\theta = 2.5$ – 24.0° $\mu = 0.08$ mm⁻¹ $T = 298$ K

Block, colourless

 $0.20 \times 0.17 \times 0.13$ mm*Data collection*Bruker SMART CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 ω scans

Absorption correction: multi-scan

(SADABS; Bruker, 2001)

 $T_{\min} = 0.983$, $T_{\max} = 0.989$

12277 measured reflections

2096 independent reflections

1545 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.045$ $\theta_{\max} = 22.7^\circ$, $\theta_{\min} = 1.7^\circ$ $h = -26 \rightarrow 26$ $k = -13 \rightarrow 16$ $l = -8 \rightarrow 8$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.039$ $wR(F^2) = 0.106$ $S = 1.04$

2096 reflections

202 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0545P)^2 + 0.2933P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.11$ e Å⁻³ $\Delta\rho_{\min} = -0.15$ e Å⁻³*Special details*

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.22678 (6)	0.40089 (10)	-0.07729 (18)	0.0681 (5)
O2	0.22007 (6)	0.33552 (10)	0.36694 (18)	0.0635 (5)
N1	0.18746 (6)	0.26778 (11)	0.1458 (2)	0.0497 (5)
H1	0.1951	0.2438	0.0491	0.060*
N2	0.14034 (6)	0.24198 (11)	0.2235 (2)	0.0485 (5)

N3	-0.07797 (7)	0.02776 (13)	0.3833 (2)	0.0649 (5)
C1	0.27573 (9)	0.37736 (13)	-0.0152 (3)	0.0529 (6)
C2	0.27502 (8)	0.33335 (12)	0.1323 (2)	0.0456 (5)
C3	0.32355 (9)	0.30874 (14)	0.2011 (3)	0.0611 (6)
H3	0.3234	0.2808	0.3009	0.073*
C4	0.37239 (10)	0.32468 (17)	0.1251 (4)	0.0754 (8)
H4	0.4047	0.3059	0.1707	0.090*
C5	0.37218 (11)	0.36868 (17)	-0.0186 (4)	0.0784 (8)
H5	0.4048	0.3805	-0.0699	0.094*
C6	0.32491 (11)	0.39565 (17)	-0.0884 (3)	0.0700 (7)
H6	0.3258	0.4264	-0.1852	0.084*
C7	0.22616 (11)	0.44770 (17)	-0.2266 (3)	0.0836 (8)
H7A	0.2432	0.4134	-0.3092	0.125*
H7B	0.2454	0.5014	-0.2138	0.125*
H7C	0.1894	0.4597	-0.2570	0.125*
C8	0.22464 (8)	0.31366 (12)	0.2249 (3)	0.0454 (5)
C9	0.11475 (8)	0.18206 (15)	0.1511 (3)	0.0511 (6)
H9	0.1286	0.1607	0.0545	0.061*
C10	0.06471 (8)	0.14545 (13)	0.2125 (2)	0.0477 (5)
C11	0.04784 (8)	0.06446 (15)	0.1597 (3)	0.0575 (6)
H11	0.0683	0.0359	0.0817	0.069*
C12	0.00212 (8)	0.02449 (15)	0.2177 (3)	0.0595 (6)
H12	-0.0070	-0.0307	0.1808	0.071*
C13	-0.03087 (8)	0.06541 (14)	0.3311 (2)	0.0505 (6)
C14	-0.01414 (8)	0.14694 (15)	0.3850 (3)	0.0578 (6)
H14	-0.0349	0.1759	0.4618	0.069*
C15	0.03229 (8)	0.18573 (14)	0.3275 (3)	0.0559 (6)
H15	0.0421	0.2402	0.3666	0.067*
C16	-0.09234 (10)	-0.05974 (17)	0.3378 (3)	0.0803 (8)
H16A	-0.0627	-0.0981	0.3608	0.120*
H16B	-0.1004	-0.0616	0.2243	0.120*
H16C	-0.1235	-0.0778	0.3981	0.120*
C17	-0.11115 (9)	0.07019 (18)	0.5026 (3)	0.0823 (8)
H17A	-0.1181	0.1290	0.4696	0.124*
H17B	-0.0927	0.0703	0.6048	0.124*
H17C	-0.1448	0.0396	0.5131	0.124*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0796 (12)	0.0712 (11)	0.0536 (10)	-0.0085 (9)	0.0061 (8)	0.0234 (8)
O2	0.0784 (11)	0.0708 (10)	0.0414 (9)	-0.0189 (8)	0.0174 (8)	-0.0065 (8)
N1	0.0541 (10)	0.0541 (11)	0.0409 (10)	-0.0082 (9)	0.0165 (8)	-0.0036 (9)
N2	0.0496 (10)	0.0499 (11)	0.0459 (11)	-0.0043 (9)	0.0139 (8)	0.0023 (9)
N3	0.0521 (12)	0.0772 (14)	0.0653 (13)	-0.0155 (10)	0.0051 (10)	0.0058 (11)
C1	0.0655 (15)	0.0453 (13)	0.0479 (14)	-0.0098 (11)	0.0155 (12)	-0.0028 (11)
C2	0.0550 (14)	0.0359 (11)	0.0458 (13)	-0.0035 (10)	0.0138 (10)	-0.0035 (10)
C3	0.0631 (16)	0.0563 (14)	0.0639 (15)	-0.0020 (12)	0.0095 (13)	-0.0019 (12)

C4	0.0549 (15)	0.0767 (18)	0.095 (2)	-0.0044 (13)	0.0095 (14)	-0.0162 (17)
C5	0.0687 (18)	0.0812 (19)	0.085 (2)	-0.0273 (15)	0.0342 (16)	-0.0231 (17)
C6	0.0837 (19)	0.0681 (16)	0.0581 (16)	-0.0265 (14)	0.0266 (15)	-0.0032 (13)
C7	0.125 (2)	0.0708 (17)	0.0551 (16)	-0.0141 (16)	0.0011 (15)	0.0205 (14)
C8	0.0575 (13)	0.0362 (11)	0.0423 (13)	-0.0011 (10)	0.0121 (11)	0.0031 (10)
C9	0.0520 (13)	0.0587 (14)	0.0425 (13)	-0.0002 (11)	0.0080 (10)	-0.0007 (11)
C10	0.0466 (12)	0.0553 (14)	0.0413 (12)	0.0003 (10)	0.0009 (10)	-0.0011 (11)
C11	0.0540 (14)	0.0641 (15)	0.0545 (14)	-0.0009 (12)	0.0057 (11)	-0.0100 (12)
C12	0.0548 (14)	0.0567 (14)	0.0671 (16)	-0.0072 (12)	-0.0045 (12)	-0.0058 (12)
C13	0.0440 (12)	0.0620 (15)	0.0455 (13)	-0.0029 (11)	-0.0064 (10)	0.0062 (12)
C14	0.0497 (13)	0.0691 (16)	0.0545 (14)	-0.0016 (12)	0.0086 (11)	-0.0074 (12)
C15	0.0547 (13)	0.0534 (13)	0.0597 (14)	-0.0066 (11)	0.0042 (11)	-0.0074 (12)
C16	0.0687 (16)	0.0782 (19)	0.0941 (19)	-0.0233 (14)	-0.0054 (14)	0.0165 (16)
C17	0.0606 (15)	0.114 (2)	0.0728 (17)	-0.0136 (15)	0.0145 (14)	0.0077 (17)

Geometric parameters (Å, °)

O1—C1	1.363 (2)	C7—H7A	0.9600
O1—C7	1.429 (3)	C7—H7B	0.9600
O2—C8	1.227 (2)	C7—H7C	0.9600
N1—C8	1.331 (2)	C9—C10	1.451 (3)
N1—N2	1.389 (2)	C9—H9	0.9300
N1—H1	0.9002	C10—C11	1.385 (3)
N2—C9	1.269 (2)	C10—C15	1.390 (3)
N3—C13	1.370 (3)	C11—C12	1.373 (3)
N3—C17	1.440 (3)	C11—H11	0.9300
N3—C16	1.442 (3)	C12—C13	1.393 (3)
C1—C6	1.387 (3)	C12—H12	0.9300
C1—C2	1.396 (3)	C13—C14	1.394 (3)
C2—C3	1.381 (3)	C14—C15	1.379 (3)
C2—C8	1.493 (3)	C14—H14	0.9300
C3—C4	1.384 (3)	C15—H15	0.9300
C3—H3	0.9300	C16—H16A	0.9600
C4—C5	1.367 (4)	C16—H16B	0.9600
C4—H4	0.9300	C16—H16C	0.9600
C5—C6	1.368 (4)	C17—H17A	0.9600
C5—H5	0.9300	C17—H17B	0.9600
C6—H6	0.9300	C17—H17C	0.9600
C1—O1—C7	117.94 (17)	N1—C8—C2	115.55 (18)
C8—N1—N2	120.22 (16)	N2—C9—C10	122.83 (19)
C8—N1—H1	120.6	N2—C9—H9	118.6
N2—N1—H1	118.0	C10—C9—H9	118.6
C9—N2—N1	114.06 (16)	C11—C10—C15	116.31 (19)
C13—N3—C17	120.6 (2)	C11—C10—C9	119.73 (19)
C13—N3—C16	121.5 (2)	C15—C10—C9	123.93 (19)
C17—N3—C16	117.47 (19)	C12—C11—C10	122.7 (2)
O1—C1—C6	124.1 (2)	C12—C11—H11	118.6

O1—C1—C2	116.54 (18)	C10—C11—H11	118.6
C6—C1—C2	119.4 (2)	C11—C12—C13	121.0 (2)
C3—C2—C1	118.84 (19)	C11—C12—H12	119.5
C3—C2—C8	117.25 (19)	C13—C12—H12	119.5
C1—C2—C8	123.87 (19)	N3—C13—C12	121.2 (2)
C2—C3—C4	121.5 (2)	N3—C13—C14	122.1 (2)
C2—C3—H3	119.2	C12—C13—C14	116.62 (19)
C4—C3—H3	119.2	C15—C14—C13	121.8 (2)
C5—C4—C3	118.6 (2)	C15—C14—H14	119.1
C5—C4—H4	120.7	C13—C14—H14	119.1
C3—C4—H4	120.7	C14—C15—C10	121.6 (2)
C4—C5—C6	121.3 (2)	C14—C15—H15	119.2
C4—C5—H5	119.3	C10—C15—H15	119.2
C6—C5—H5	119.3	N3—C16—H16A	109.5
C5—C6—C1	120.2 (2)	N3—C16—H16B	109.5
C5—C6—H6	119.9	H16A—C16—H16B	109.5
C1—C6—H6	119.9	N3—C16—H16C	109.5
O1—C7—H7A	109.5	H16A—C16—H16C	109.5
O1—C7—H7B	109.5	H16B—C16—H16C	109.5
H7A—C7—H7B	109.5	N3—C17—H17A	109.5
O1—C7—H7C	109.5	N3—C17—H17B	109.5
H7A—C7—H7C	109.5	H17A—C17—H17B	109.5
H7B—C7—H7C	109.5	N3—C17—H17C	109.5
O2—C8—N1	123.55 (18)	H17A—C17—H17C	109.5
O2—C8—C2	120.81 (19)	H17B—C17—H17C	109.5

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1...O2 ⁱ	0.90	2.03	2.914 (2)	165

Symmetry code: (i) *x*, $-\gamma+1/2$, *z*-1/2.