

# Methyl 4-phenyl-1,2,3,3a,4,4a,5,12c-octahydronaphtho[1',2':3,2]furo[5,4-b]-pyrrolizine-4a-carboxylate

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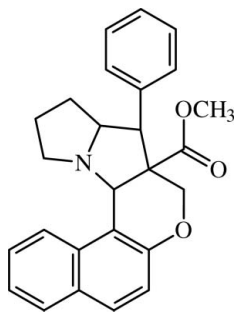
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Key indicators: single-crystal X-ray study;  $T = 292$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.047;  $wR$  factor = 0.134; data-to-parameter ratio = 17.6.

In the title compound,  $\text{C}_{26}\text{H}_{25}\text{NO}_3$ , both pyrrolidine rings adopt envelope conformations, whereas the dihydropyran ring adopts a half-chair conformation. The phenyl ring is oriented at an angle of  $27.9$  (1)° with respect to the naphthalene ring system. An intramolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bond is observed. The crystal packing is stabilized by weak intermolecular  $\text{C}-\text{H}\cdots\pi$  interactions.

## Related literature

For general background to pyrrolizine derivatives, see: Barsoum & Nawar (2003); Abbas *et al.* (2010); Anderson & Corey (1977); Makoni & Sugden (1980); Laufer *et al.* (1997). For a related structure, see: Nirmala *et al.* (2009). For ring-puckering and asymmetry parameters, see: Cremer & Pople (1975); Nardelli (1983).



## Experimental

### Crystal data

$\text{C}_{26}\text{H}_{25}\text{NO}_3$   
 $M_r = 399.47$

Orthorhombic,  $Pbca$   
 $a = 15.0117$  (12) Å

$b = 13.3421$  (11) Å  
 $c = 20.0242$  (16) Å  
 $V = 4010.6$  (6) Å<sup>3</sup>  
 $Z = 8$

Mo  $K\alpha$  radiation  
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 292$  K  
 $0.24 \times 0.22 \times 0.20$  mm

### Data collection

Bruker SMART APEX CCD area-detector diffractometer  
43738 measured reflections  
4779 independent reflections  
3892 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.030$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.047$   
 $wR(F^2) = 0.134$   
 $S = 1.04$   
4779 reflections  
272 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.33$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.14$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

Cg1 and Cg2 are the centroids of the C3–C8 and C19–C24 rings, respectively.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C24–H24 $\cdots$ O2	0.93	2.59	3.388 (2)	145
C7–H7 $\cdots$ Cg2 <sup>i</sup>	0.93	2.82	3.6435 (17)	148
C17–H17A $\cdots$ Cg2 <sup>ii</sup>	0.97	2.78	3.7065 (17)	159
C21–H21 $\cdots$ Cg1 <sup>iii</sup>	0.93	2.49	3.4174 (18)	175
C26–H26B $\cdots$ Cg1 <sup>iv</sup>	0.96	2.58	3.477 (2)	156

Symmetry codes: (i)  $-x + \frac{1}{2}, -y, z + \frac{1}{2}$ ; (ii)  $-x + \frac{1}{2}, y - \frac{1}{2}, z$ ; (iii)  $x, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (iv)  $-x + \frac{1}{2}, y + \frac{1}{2}, z$ .

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97* and *PLATON*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI5092).

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## supporting information

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## Methyl 4-phenyl-1,2,3,3a,4,4a,5,12c-octahydronaphtho[1',2':3,2]furo[5,4-b]pyrrolizine-4a-carboxylate

S. Selvanayagam, B. Sridhar, K. Ravikumar, S. Kathiravan and R. Raghunathan

### S1. Comment

Pyrrolizine derivatives possess antimicrobial (Barsoum & Nawar, 2003), anti-inflammatory (Abbas *et al.*, 2010) and antileukemic (Anderson & Corey, 1977) activities. These derivatives are used as inhibitors of blood platelet aggregation (Makoni & Sugden, 1980), dual cyclooxygenase-1 and 5-lipoxygenase (Laufer *et al.*, 1997). In view of these importance, we have undertaken the crystal structure determination of the title compound, a pyrrolizine derivative, and the results are presented here.

The geometry of the pyrrolizine and naphthalene ring system is comparable with that observed in a related structure (Nirmala *et al.*, 2009). The sum of the angles ( $337.1^\circ$ ) around atom N1 is in accordance with  $sp^3$  hybridization. There is a short contact (2.19 Å) between atoms H12A and H24, which results in the widening of the C24—C19—C14 angle [ $124.1(1)^\circ$ ] from the ideal value of  $120^\circ$ .

The naphthalene ring system (C2—C11) and the phenyl ring (C19—C24) are oriented at an angle of  $27.9(1)^\circ$ . In the pyrrolizine ring system, both the pyrrolidine rings N1/C1/C13—C15 and N1/C15—C18 adopt envelope conformations; the puckering parameters (Cremer & Pople, 1975) are:  $q_2 = 0.396(1)$  Å and  $\varphi = -107.2(2)^\circ$  for N1/C1/C13—C15 ring, and  $q_2 = 0.383(2)$  Å and  $\varphi = -79.1(2)^\circ$  for N1/C15—C18 ring. In the N1/C1/C13—C15 ring, atom C13 deviates by  $0.620(1)$  Å from the least-squares plane through the remaining four atoms, whereas in the ring N1/C15—C18, atom C17 deviates by  $-0.583(2)$  Å from the plane through the remaining four atoms. The dihydropyran ring of the chromene unit adopts a half-chair conformation, with the lowest asymmetry parameter  $\Delta C_2(C2—C11)$  of  $3.8(2)^\circ$  (Nardelli, 1983).

The molecular structure is influenced by an intramolecular C—H $\cdots$ O hydrogen bond. The crystal packing is stabilized by weak intermolecular C—H $\cdots$  $\pi$  interactions ((Table 1).

### S2. Experimental

A mixture of (*Z*)-methyl-2[(1-formylnaphthalen-2-yloxy)methyl]-3-(4-phenyl) acrylate (20 mmol) and proline (30 mmol) was refluxed in benzene for 20 h and the solvent was removed under reduced pressure. The crude product was subjected to column chromatography to get the pure product. Single crystals were grown by slow evaporation of a chloroform-methanol (1:1) solution.

### S3. Refinement

H atoms were placed in idealized positions and allowed to ride on their parent atoms, with C—H = 0.93, 0.98, 0.97 and 0.96 Å for aromatic, methine, methylene and methyl H respectively, and  $U_{eq}(H) = 1.5U_{eq}(C)$  for methyl H and  $U_{eq}(H) = 1.2U_{eq}(C)$  for all other H atoms.

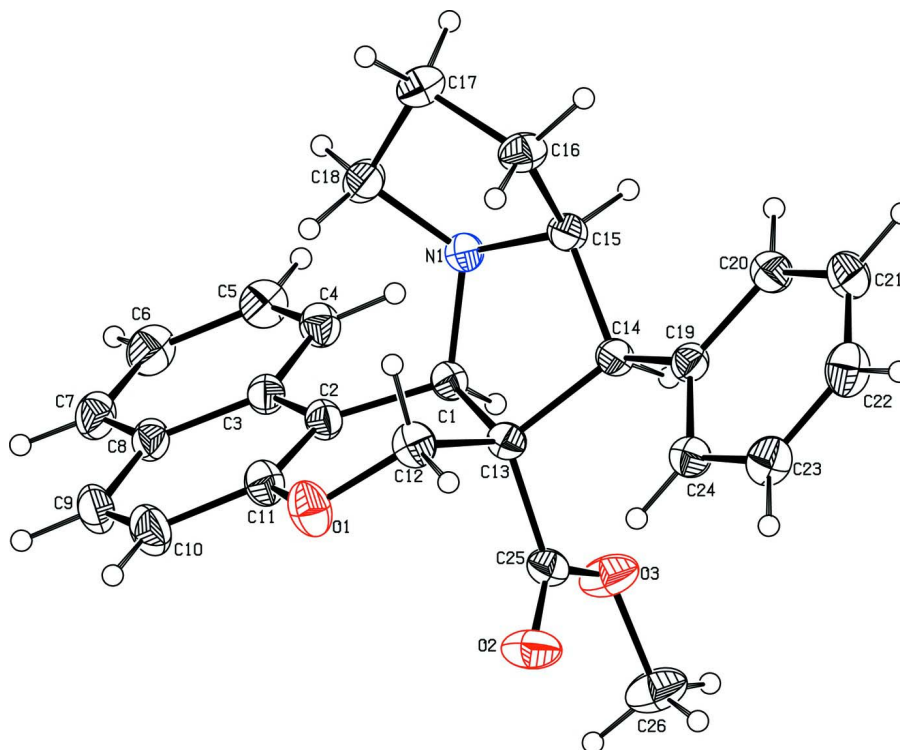


Figure 1

The molecular structure of the title compound, showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level

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#### Crystal data

$C_{26}H_{25}NO_3$

$M_r = 399.47$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 15.0117$  (12) Å

$b = 13.3421$  (11) Å

$c = 20.0242$  (16) Å

$V = 4010.6$  (6) Å<sup>3</sup>

$Z = 8$

$F(000) = 1696$

$D_x = 1.323$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 31182 reflections

$\theta = 2.1$ – $27.3^\circ$

$\mu = 0.09$  mm<sup>-1</sup>

$T = 292$  K

Block, colourless

$0.24 \times 0.22 \times 0.20$  mm

#### Data collection

Bruker SMART APEX CCD area-detector  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\omega$  scans

43738 measured reflections

4779 independent reflections

3892 reflections with  $I > 2\sigma(I)$

$R_{int} = 0.030$

$\theta_{max} = 28.0^\circ$ ,  $\theta_{min} = 2.0^\circ$

$h = -19 \rightarrow 19$

$k = -17 \rightarrow 17$

$l = -26 \rightarrow 26$

*Refinement*Refinement on  $F^2$ 

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.047$  $wR(F^2) = 0.134$  $S = 1.04$ 

4779 reflections

272 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methodsSecondary atom site location: difference Fourier  
mapHydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0727P)^2 + 0.9686P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\max} = 0.001$  $\Delta\rho_{\max} = 0.33 \text{ e } \text{\AA}^{-3}$  $\Delta\rho_{\min} = -0.14 \text{ e } \text{\AA}^{-3}$ *Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.10550 (8)	-0.01484 (8)	0.07769 (6)	0.0520 (3)
O2	0.00482 (7)	0.17668 (10)	0.09040 (6)	0.0580 (3)
O3	0.10217 (7)	0.27234 (10)	0.14444 (6)	0.0557 (3)
C1	0.23953 (8)	0.13141 (9)	0.11484 (6)	0.0320 (3)
H1	0.2427	0.1849	0.1483	0.038*
C11	0.15871 (10)	-0.02986 (10)	0.13242 (7)	0.0416 (3)
C10	0.13883 (12)	-0.11691 (11)	0.17021 (9)	0.0519 (4)
H10	0.0920	-0.1583	0.1574	0.062*
C9	0.18806 (12)	-0.13999 (11)	0.22501 (8)	0.0498 (4)
H9	0.1741	-0.1971	0.2495	0.060*
C8	0.26021 (10)	-0.07888 (10)	0.24562 (7)	0.0403 (3)
C7	0.31301 (12)	-0.10157 (12)	0.30224 (7)	0.0488 (4)
H7	0.3009	-0.1592	0.3267	0.059*
C6	0.38112 (12)	-0.04072 (13)	0.32157 (8)	0.0524 (4)
H6	0.4144	-0.0562	0.3594	0.063*
C5	0.40078 (11)	0.04538 (12)	0.28425 (8)	0.0477 (4)
H5	0.4479	0.0864	0.2971	0.057*
C4	0.35132 (10)	0.06990 (10)	0.22891 (7)	0.0400 (3)
H4	0.3655	0.1272	0.2047	0.048*
C3	0.27917 (9)	0.00946 (9)	0.20814 (6)	0.0346 (3)
C2	0.22530 (9)	0.03416 (9)	0.15139 (6)	0.0348 (3)
C13	0.16187 (8)	0.15725 (9)	0.06619 (6)	0.0330 (3)
C12	0.13359 (10)	0.06083 (11)	0.03088 (7)	0.0415 (3)
H12A	0.0850	0.0755	0.0004	0.050*
H12B	0.1831	0.0354	0.0048	0.050*

N1	0.32208 (7)	0.13550 (8)	0.07478 (5)	0.0347 (3)
C18	0.36312 (10)	0.04152 (11)	0.05115 (8)	0.0438 (3)
H18A	0.3189	-0.0111	0.0471	0.053*
H18B	0.4095	0.0195	0.0815	0.053*
C17	0.40179 (10)	0.06837 (12)	-0.01673 (8)	0.0450 (3)
H17A	0.4072	0.0095	-0.0449	0.054*
H17B	0.4598	0.0996	-0.0122	0.054*
C16	0.33354 (10)	0.14184 (12)	-0.04488 (7)	0.0431 (3)
H16A	0.2827	0.1069	-0.0637	0.052*
H16B	0.3598	0.1844	-0.0789	0.052*
C15	0.30691 (9)	0.20250 (10)	0.01671 (6)	0.0349 (3)
H15	0.3468	0.2604	0.0205	0.042*
C14	0.20917 (8)	0.23942 (10)	0.02349 (6)	0.0327 (3)
H14	0.2124	0.2983	0.0527	0.039*
C25	0.08026 (9)	0.20086 (10)	0.10127 (6)	0.0362 (3)
C26	0.02975 (12)	0.32140 (15)	0.17845 (10)	0.0615 (5)
H26A	-0.0140	0.3423	0.1464	0.092*
H26B	0.0519	0.3789	0.2020	0.092*
H26C	0.0031	0.2757	0.2096	0.092*
C19	0.16797 (9)	0.27750 (9)	-0.04106 (6)	0.0336 (3)
C24	0.07883 (9)	0.26452 (10)	-0.05833 (7)	0.0398 (3)
H24	0.0424	0.2252	-0.0314	0.048*
C23	0.04346 (11)	0.30931 (11)	-0.11511 (8)	0.0469 (4)
H23	-0.0165	0.3008	-0.1253	0.056*
C22	0.09672 (12)	0.36633 (12)	-0.15645 (8)	0.0505 (4)
H22	0.0733	0.3952	-0.1949	0.061*
C21	0.18522 (11)	0.38012 (12)	-0.14008 (8)	0.0493 (4)
H21	0.2215	0.4187	-0.1676	0.059*
C20	0.22020 (10)	0.33690 (11)	-0.08297 (7)	0.0415 (3)
H20	0.2797	0.3477	-0.0723	0.050*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0578 (7)	0.0420 (5)	0.0561 (6)	-0.0168 (5)	-0.0144 (5)	0.0024 (5)
O2	0.0359 (6)	0.0730 (8)	0.0650 (7)	-0.0104 (5)	0.0103 (5)	-0.0186 (6)
O3	0.0384 (5)	0.0666 (7)	0.0622 (7)	0.0043 (5)	0.0008 (5)	-0.0281 (6)
C1	0.0346 (6)	0.0304 (6)	0.0311 (6)	-0.0025 (5)	-0.0002 (5)	0.0001 (5)
C11	0.0473 (8)	0.0343 (7)	0.0432 (8)	-0.0044 (6)	-0.0007 (6)	-0.0009 (6)
C10	0.0575 (9)	0.0356 (7)	0.0628 (10)	-0.0137 (7)	0.0014 (8)	0.0002 (7)
C9	0.0633 (10)	0.0324 (7)	0.0539 (9)	-0.0039 (6)	0.0109 (8)	0.0079 (6)
C8	0.0518 (8)	0.0320 (6)	0.0369 (7)	0.0056 (6)	0.0111 (6)	0.0022 (5)
C7	0.0648 (10)	0.0430 (8)	0.0387 (7)	0.0134 (7)	0.0112 (7)	0.0103 (6)
C6	0.0618 (10)	0.0594 (10)	0.0359 (7)	0.0139 (8)	-0.0014 (7)	0.0064 (7)
C5	0.0513 (9)	0.0525 (8)	0.0394 (7)	0.0014 (7)	-0.0040 (6)	-0.0010 (6)
C4	0.0466 (7)	0.0383 (7)	0.0351 (7)	-0.0005 (6)	0.0011 (6)	0.0016 (5)
C3	0.0422 (7)	0.0312 (6)	0.0304 (6)	0.0028 (5)	0.0058 (5)	0.0002 (5)
C2	0.0402 (7)	0.0302 (6)	0.0339 (6)	-0.0019 (5)	0.0032 (5)	-0.0004 (5)

C13	0.0328 (6)	0.0348 (6)	0.0315 (6)	-0.0014 (5)	0.0021 (5)	0.0005 (5)
C12	0.0456 (8)	0.0404 (7)	0.0386 (7)	-0.0044 (6)	-0.0037 (6)	-0.0041 (6)
N1	0.0342 (6)	0.0349 (5)	0.0349 (6)	0.0015 (4)	0.0014 (4)	0.0029 (4)
C18	0.0453 (8)	0.0398 (7)	0.0463 (8)	0.0088 (6)	0.0031 (6)	0.0005 (6)
C17	0.0402 (7)	0.0503 (8)	0.0444 (8)	0.0085 (6)	0.0025 (6)	-0.0061 (6)
C16	0.0393 (7)	0.0547 (8)	0.0352 (7)	0.0064 (6)	0.0037 (6)	-0.0012 (6)
C15	0.0321 (6)	0.0379 (7)	0.0347 (6)	-0.0006 (5)	0.0012 (5)	0.0031 (5)
C14	0.0322 (6)	0.0326 (6)	0.0334 (6)	-0.0015 (5)	0.0018 (5)	0.0005 (5)
C25	0.0346 (7)	0.0416 (7)	0.0322 (6)	-0.0029 (5)	0.0038 (5)	0.0030 (5)
C26	0.0476 (9)	0.0748 (12)	0.0619 (10)	0.0107 (8)	0.0039 (8)	-0.0265 (9)
C19	0.0362 (7)	0.0320 (6)	0.0326 (6)	0.0033 (5)	0.0018 (5)	-0.0001 (5)
C24	0.0375 (7)	0.0402 (7)	0.0418 (7)	-0.0004 (5)	0.0008 (6)	0.0021 (6)
C23	0.0458 (8)	0.0465 (8)	0.0483 (8)	0.0014 (6)	-0.0112 (6)	-0.0031 (6)
C22	0.0647 (10)	0.0488 (8)	0.0379 (7)	0.0054 (7)	-0.0101 (7)	0.0055 (6)
C21	0.0567 (9)	0.0477 (8)	0.0436 (8)	0.0007 (7)	0.0060 (7)	0.0121 (6)
C20	0.0390 (7)	0.0426 (7)	0.0429 (7)	-0.0002 (6)	0.0021 (6)	0.0055 (6)

*Geometric parameters (Å, °)*

O1—C11	1.3708 (18)	C12—H12B	0.97
O1—C12	1.4408 (18)	N1—C18	1.4751 (18)
O2—C25	1.1974 (17)	N1—C15	1.4842 (16)
O3—C25	1.3285 (17)	C18—C17	1.521 (2)
O3—C26	1.4402 (18)	C18—H18A	0.97
C1—N1	1.4771 (16)	C18—H18B	0.97
C1—C2	1.5049 (17)	C17—C16	1.526 (2)
C1—C13	1.5578 (18)	C17—H17A	0.97
C1—H1	0.98	C17—H17B	0.97
C11—C2	1.3686 (19)	C16—C15	1.5283 (19)
C11—C10	1.418 (2)	C16—H16A	0.97
C10—C9	1.358 (2)	C16—H16B	0.97
C10—H10	0.93	C15—C14	1.5538 (18)
C9—C8	1.417 (2)	C15—H15	0.98
C9—H9	0.93	C14—C19	1.5202 (17)
C8—C7	1.416 (2)	C14—H14	0.98
C8—C3	1.4261 (18)	C26—H26A	0.96
C7—C6	1.362 (3)	C26—H26B	0.96
C7—H7	0.93	C26—H26C	0.96
C6—C5	1.402 (2)	C19—C24	1.3931 (19)
C6—H6	0.93	C19—C20	1.3953 (19)
C5—C4	1.373 (2)	C24—C23	1.390 (2)
C5—H5	0.93	C24—H24	0.93
C4—C3	1.413 (2)	C23—C22	1.380 (2)
C4—H4	0.93	C23—H23	0.93
C3—C2	1.4331 (19)	C22—C21	1.381 (2)
C13—C25	1.5274 (18)	C22—H22	0.93
C13—C12	1.5281 (18)	C21—C20	1.384 (2)
C13—C14	1.5612 (17)	C21—H21	0.93

C12—H12A	0.97	C20—H20	0.93
C11—O1—C12	116.86 (11)	C17—C18—H18A	110.9
C25—O3—C26	116.55 (12)	N1—C18—H18B	110.9
N1—C1—C2	114.52 (10)	C17—C18—H18B	110.9
N1—C1—C13	106.26 (10)	H18A—C18—H18B	108.9
C2—C1—C13	112.87 (10)	C18—C17—C16	103.01 (11)
N1—C1—H1	107.6	C18—C17—H17A	111.2
C2—C1—H1	107.6	C16—C17—H17A	111.2
C13—C1—H1	107.6	C18—C17—H17B	111.2
C2—C11—O1	123.80 (13)	C16—C17—H17B	111.2
C2—C11—C10	121.12 (14)	H17A—C17—H17B	109.1
O1—C11—C10	115.07 (13)	C17—C16—C15	102.57 (11)
C9—C10—C11	120.15 (15)	C17—C16—H16A	111.3
C9—C10—H10	119.9	C15—C16—H16A	111.3
C11—C10—H10	119.9	C17—C16—H16B	111.3
C10—C9—C8	121.38 (13)	C15—C16—H16B	111.3
C10—C9—H9	119.3	H16A—C16—H16B	109.2
C8—C9—H9	119.3	N1—C15—C16	105.85 (11)
C7—C8—C9	122.55 (13)	N1—C15—C14	105.50 (10)
C7—C8—C3	119.10 (14)	C16—C15—C14	119.04 (11)
C9—C8—C3	118.34 (13)	N1—C15—H15	108.7
C6—C7—C8	121.36 (14)	C16—C15—H15	108.7
C6—C7—H7	119.3	C14—C15—H15	108.7
C8—C7—H7	119.3	C19—C14—C15	114.57 (10)
C7—C6—C5	119.67 (15)	C19—C14—C13	121.02 (10)
C7—C6—H6	120.2	C15—C14—C13	104.76 (10)
C5—C6—H6	120.2	C19—C14—H14	105.0
C4—C5—C6	120.76 (15)	C15—C14—H14	105.0
C4—C5—H5	119.6	C13—C14—H14	105.0
C6—C5—H5	119.6	O2—C25—O3	123.08 (13)
C5—C4—C3	121.07 (13)	O2—C25—C13	124.91 (13)
C5—C4—H4	119.5	O3—C25—C13	111.97 (11)
C3—C4—H4	119.5	O3—C26—H26A	109.5
C4—C3—C8	118.02 (12)	O3—C26—H26B	109.5
C4—C3—C2	122.33 (12)	H26A—C26—H26B	109.5
C8—C3—C2	119.66 (13)	O3—C26—H26C	109.5
C11—C2—C3	119.25 (12)	H26A—C26—H26C	109.5
C11—C2—C1	120.45 (12)	H26B—C26—H26C	109.5
C3—C2—C1	120.25 (12)	C24—C19—C20	117.45 (12)
C25—C13—C12	108.10 (11)	C24—C19—C14	124.08 (12)
C25—C13—C1	113.39 (10)	C20—C19—C14	118.20 (12)
C12—C13—C1	108.12 (10)	C23—C24—C19	121.11 (13)
C25—C13—C14	110.43 (10)	C23—C24—H24	119.4
C12—C13—C14	117.65 (11)	C19—C24—H24	119.4
C1—C13—C14	99.06 (9)	C22—C23—C24	120.44 (14)
O1—C12—C13	111.73 (11)	C22—C23—H23	119.8
O1—C12—H12A	109.3	C24—C23—H23	119.8

C13—C12—H12A	109.3	C23—C22—C21	119.25 (14)
O1—C12—H12B	109.3	C23—C22—H22	120.4
C13—C12—H12B	109.3	C21—C22—H22	120.4
H12A—C12—H12B	107.9	C22—C21—C20	120.38 (14)
C18—N1—C1	119.55 (11)	C22—C21—H21	119.8
C18—N1—C15	108.94 (10)	C20—C21—H21	119.8
C1—N1—C15	108.61 (10)	C21—C20—C19	121.36 (14)
N1—C18—C17	104.24 (11)	C21—C20—H20	119.3
N1—C18—H18A	110.9	C19—C20—H20	119.3
C12—O1—C11—C2	13.1 (2)	C2—C1—N1—C15	149.91 (11)
C12—O1—C11—C10	-168.32 (13)	C13—C1—N1—C15	24.60 (13)
C2—C11—C10—C9	-2.3 (3)	C1—N1—C18—C17	145.27 (12)
O1—C11—C10—C9	179.13 (15)	C15—N1—C18—C17	19.65 (15)
C11—C10—C9—C8	-0.4 (3)	N1—C18—C17—C16	-36.07 (15)
C10—C9—C8—C7	-179.52 (15)	C18—C17—C16—C15	38.40 (15)
C10—C9—C8—C3	1.3 (2)	C18—N1—C15—C16	4.57 (14)
C9—C8—C7—C6	-178.96 (15)	C1—N1—C15—C16	-127.17 (11)
C3—C8—C7—C6	0.2 (2)	C18—N1—C15—C14	131.61 (11)
C8—C7—C6—C5	-1.1 (2)	C1—N1—C15—C14	-0.13 (13)
C7—C6—C5—C4	0.9 (2)	C17—C16—C15—N1	-26.71 (14)
C6—C5—C4—C3	0.2 (2)	C17—C16—C15—C14	-145.10 (12)
C5—C4—C3—C8	-1.2 (2)	N1—C15—C14—C19	-159.08 (10)
C5—C4—C3—C2	178.77 (14)	C16—C15—C14—C19	-40.51 (16)
C7—C8—C3—C4	0.96 (19)	N1—C15—C14—C13	-24.14 (13)
C9—C8—C3—C4	-179.87 (13)	C16—C15—C14—C13	94.43 (13)
C7—C8—C3—C2	-178.97 (12)	C25—C13—C14—C19	-72.42 (14)
C9—C8—C3—C2	0.20 (19)	C12—C13—C14—C19	52.29 (16)
O1—C11—C2—C3	-177.76 (13)	C1—C13—C14—C19	168.34 (11)
C10—C11—C2—C3	3.8 (2)	C25—C13—C14—C15	156.27 (11)
O1—C11—C2—C1	4.7 (2)	C12—C13—C14—C15	-79.02 (14)
C10—C11—C2—C1	-173.79 (13)	C1—C13—C14—C15	37.03 (12)
C4—C3—C2—C11	177.36 (13)	C26—O3—C25—O2	-0.3 (2)
C8—C3—C2—C11	-2.7 (2)	C26—O3—C25—C13	177.64 (14)
C4—C3—C2—C1	-5.09 (19)	C12—C13—C25—O2	-15.40 (19)
C8—C3—C2—C1	174.83 (11)	C1—C13—C25—O2	-135.26 (15)
N1—C1—C2—C11	-110.51 (14)	C14—C13—C25—O2	114.60 (16)
C13—C1—C2—C11	11.26 (18)	C12—C13—C25—O3	166.74 (12)
N1—C1—C2—C3	71.97 (15)	C1—C13—C25—O3	46.88 (15)
C13—C1—C2—C3	-166.26 (11)	C14—C13—C25—O3	-63.26 (14)
N1—C1—C13—C25	-154.82 (10)	C15—C14—C19—C24	144.48 (13)
C2—C1—C13—C25	78.87 (13)	C13—C14—C19—C24	17.48 (19)
N1—C1—C13—C12	85.34 (12)	C15—C14—C19—C20	-41.57 (16)
C2—C1—C13—C12	-40.98 (14)	C13—C14—C19—C20	-168.57 (12)
N1—C1—C13—C14	-37.80 (12)	C20—C19—C24—C23	-0.1 (2)
C2—C1—C13—C14	-164.12 (10)	C14—C19—C24—C23	173.86 (13)
C11—O1—C12—C13	-45.96 (17)	C19—C24—C23—C22	1.2 (2)
C25—C13—C12—O1	-64.37 (14)	C24—C23—C22—C21	-1.3 (2)



C1—C13—C12—O1	58.76 (15)	C23—C22—C21—C20	0.2 (2)
C14—C13—C12—O1	169.77 (11)	C22—C21—C20—C19	0.9 (2)
C2—C1—N1—C18	24.13 (16)	C24—C19—C20—C21	-0.9 (2)
C13—C1—N1—C18	-101.18 (13)	C14—C19—C20—C21	-175.28 (13)

*Hydrogen-bond geometry (Å, °)*

Cg1 and Cg2 are the centroids of the C3–C8 and C19–C24 rings, respectively.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C24—H24...O2	0.93	2.59	3.388 (2)	145
C7—H7...Cg2 <sup>i</sup>	0.93	2.82	3.6435 (17)	148
C17—H17 <i>A</i> ...Cg2 <sup>ii</sup>	0.97	2.78	3.7065 (17)	159
C21—H21...Cg1 <sup>iii</sup>	0.93	2.49	3.4174 (18)	175
C26—H26 <i>B</i> ...Cg1 <sup>iv</sup>	0.96	2.58	3.477 (2)	156

Symmetry codes: (i)  $-x+1/2, -y, z+1/2$ ; (ii)  $-x+1/2, y-1/2, z$ ; (iii)  $x, -y+1/2, z-1/2$ ; (iv)  $-x+1/2, y+1/2, z$ .