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## Structure Reports

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## Methyl 4-chloro-3,5-dinitrobenzoate

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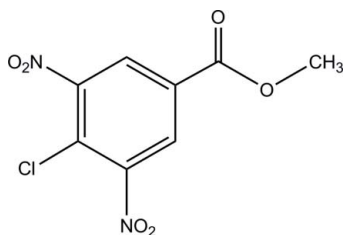
Received 26 November 2009; accepted 1 December 2009

Key indicators: single-crystal X-ray study;  $T = 93$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.047;  $wR$  factor = 0.142; data-to-parameter ratio = 14.1.

In the molecule of the title compound,  $\text{C}_8\text{H}_5\text{ClN}_2\text{O}_6$ , the two nitro groups and the ester group make dihedral angles of  $29.6$  ( $1^\circ$ ),  $82.3$  ( $1^\circ$ ) and  $13.7$  ( $1^\circ$ ), respectively, with the benzene ring. In the crystal structure weak  $\text{C}-\text{H}\cdots\text{O}$  interactions are present.

## Related literature

For the use of the title compound as a herbicide, see: Akira *et al.* (1978); Ferenc *et al.* (1984).



## Experimental

## Crystal data

 $\text{C}_8\text{H}_5\text{ClN}_2\text{O}_6$  $M_r = 260.59$ Triclinic,  $P\bar{1}$  $a = 4.8579$  (10) Å $b = 9.4438$  (19) Å $c = 11.369$  (2) Å $\alpha = 73.36$  ( $3^\circ$ ) $\beta = 88.09$  ( $3^\circ$ ) $\gamma = 87.47$  ( $3^\circ$ )  
 $V = 499.14$  (18) Å<sup>3</sup>  
 $Z = 2$   
Mo  $K\alpha$  radiation $\mu = 0.40$  mm<sup>-1</sup>  
 $T = 93$  K  
 $0.50 \times 0.33 \times 0.17$  mm

## Data collection

Rigaku SPIDER diffractometer  
Absorption correction: multi-scan  
(North *et al.*, 1968)  
 $T_{\text{min}} = 0.824$ ,  $T_{\text{max}} = 0.936$   
3935 measured reflections2193 independent reflections  
1607 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.029$   
Standard reflections: 0

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.047$   
 $wR(F^2) = 0.142$   
 $S = 1.02$   
2193 reflections155 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.85$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.32$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C1}-\text{H1}\cdots\text{O1}^{\text{i}}$	0.95	2.51	3.329 (3)	145
$\text{C5}-\text{H5}\cdots\text{O6}^{\text{ii}}$	0.95	2.34	3.143 (3)	142

Symmetry codes: (i)  $-x + 2, -y + 1, -z + 1$ ; (ii)  $-x + 1, -y + 2, -z$ .

Data collection: *RAPID-AUTO* (Rigaku 2004); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

The authors acknowledge financial support from Jiangsu Institute of Nuclear Medicine, China.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2697).

## References

- Akira, S., Shoji, K. & Kenichi, S. (1978). Japan Patent No. 53101528  
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
Ferenc, B., Gyoery, K. & Mihaly, N. (1984). German Patent No. 3410566  
North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.  
Rigaku (2004). *RAPID-AUTO*. Rigaku Corporation, Tokyo, Japan.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

## supporting information

*Acta Cryst.* (2010). E66, o62 [doi:10.1107/S1600536809051630]

**Methyl 4-chloro-3,5-dinitrobenzoate**

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**S1. Comment**

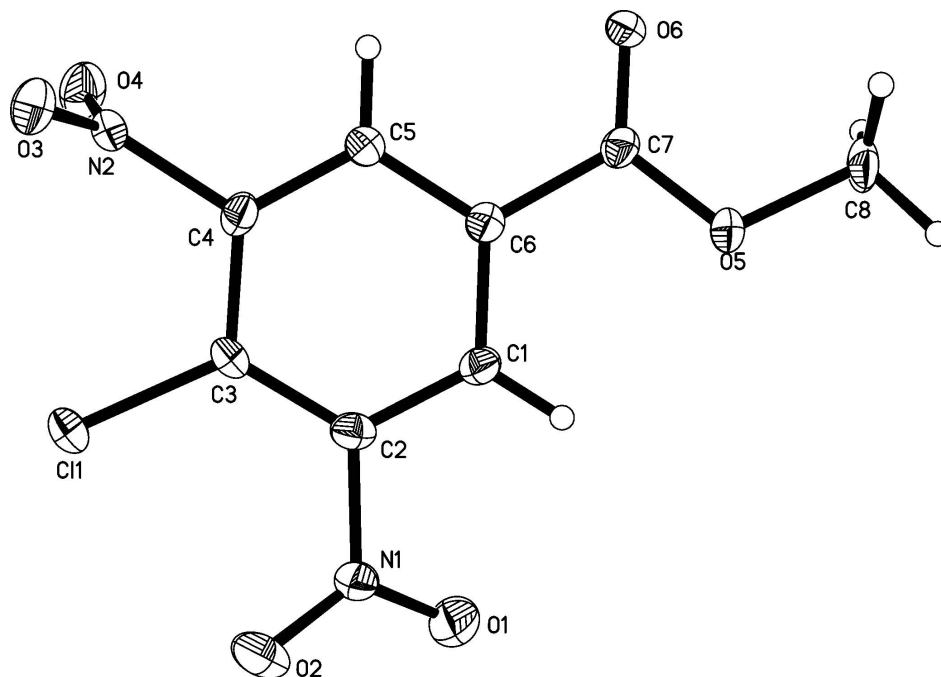
The title compound (Fig.1) is useful as a herbicide (Akira *et al.*,1978; Ferenc *et al.*, 1984). It give good result as seed-dressing fungicide for sunflower, corn and flax. We report here the crystal structure of the title compound. Two nitro groups (O1/N1/O2 and O3/N2/O4) attached at C2 and C4, the ester group(O5/C7/O6) attached at C6 form dihedral angles of 150.4 (1)°, 97.7 (1)° and 166.3 (1)° with the mean plane of the C1-benzene ring, respectively. In the crystal structure, adjacent molecules are linked through weak C—H···O hydrogen interactions (Table 1).

**S2. Experimental**

A sample of commercial methyl 4-chloro-3,5-dinitrobenzoate (Aldrich) was crystalized by slow evaporation of a solution in methanol: colourless chunk-shaped crystals were formed after several days.

**S3. Refinement**

H atoms are positioned geometrically, with C—H = 0.95 and 0.98 Å for benzene and methyl H atoms respectively, and are allowed to ride on the C atoms to which they are bonded, with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl H atoms and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for the aromatic H atoms.

**Figure 1**

A view of the title compound with the atomic numbering scheme. Displacement ellipsoids were drawn at the 30% probability level.

### methyl 4-chloro-3,5-dinitrobenzoate

#### Crystal data

$C_8H_5ClN_2O_6$

$M_r = 260.59$

Triclinic,  $P\bar{1}$

Hall symbol: -P 1

$a = 4.8579$  (10) Å

$b = 9.4438$  (19) Å

$c = 11.369$  (2) Å

$\alpha = 73.36$  (3)°

$\beta = 88.09$  (3)°

$\gamma = 87.47$  (3)°

$V = 499.14$  (18) Å<sup>3</sup>

$Z = 2$

$F(000) = 264$

$D_x = 1.734$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 1306 reflections

$\theta = 3.7$ – $27.5$ °

$\mu = 0.40$  mm<sup>-1</sup>

$T = 93$  K

Chunk, colorless

$0.50 \times 0.33 \times 0.17$  mm

#### Data collection

Rigaku SPIDER  
diffractometer

Radiation source: Rotating Anode  
Graphite monochromator

$\omega$  scans

Absorption correction: multi-scan  
(North *et al.*, 1968)

$T_{\min} = 0.824$ ,  $T_{\max} = 0.936$

3935 measured reflections

2193 independent reflections

1607 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.029$

$\theta_{\max} = 27.5$ °,  $\theta_{\min} = 3.3$ °

$h = -6 \rightarrow 6$

$k = -9 \rightarrow 12$

$l = -13 \rightarrow 14$

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.047$   
 $wR(F^2) = 0.142$   
 $S = 1.02$   
 2193 reflections  
 155 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0772P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.85 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.32 \text{ e } \text{\AA}^{-3}$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	1.27729 (13)	1.05900 (7)	0.37558 (6)	0.0216 (2)
O1	0.9997 (5)	0.6244 (2)	0.56066 (17)	0.0317 (5)
O2	1.3643 (4)	0.7546 (2)	0.51374 (19)	0.0345 (5)
O3	1.1854 (4)	1.2028 (2)	0.07808 (17)	0.0249 (5)
O4	0.8170 (4)	1.2738 (2)	0.16400 (18)	0.0278 (5)
O5	0.4877 (4)	0.58546 (19)	0.21655 (17)	0.0220 (4)
O6	0.3199 (4)	0.79731 (19)	0.08858 (15)	0.0194 (4)
N1	1.1375 (5)	0.7234 (2)	0.4913 (2)	0.0222 (5)
N2	0.9797 (5)	1.1792 (2)	0.1458 (2)	0.0191 (5)
C1	0.8221 (5)	0.7323 (3)	0.3272 (2)	0.0172 (5)
H1	0.7881	0.6315	0.3672	0.021*
C2	1.0070 (5)	0.8080 (3)	0.3755 (2)	0.0175 (5)
C3	1.0622 (5)	0.9551 (3)	0.3184 (2)	0.0164 (5)
C4	0.9246 (5)	1.0237 (3)	0.2102 (2)	0.0172 (5)
C5	0.7370 (5)	0.9523 (3)	0.1612 (2)	0.0171 (5)
H5	0.6432	1.0033	0.0884	0.021*
C6	0.6874 (5)	0.8061 (3)	0.2193 (2)	0.0168 (5)
C7	0.4771 (5)	0.7304 (3)	0.1662 (2)	0.0168 (5)
C8	0.2922 (6)	0.5029 (3)	0.1694 (3)	0.0244 (6)
H8A	0.1036	0.5346	0.1860	0.037*
H8B	0.3196	0.3969	0.2100	0.037*
H8C	0.3221	0.5219	0.0807	0.037*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0217 (4)	0.0228 (4)	0.0222 (4)	-0.0051 (3)	-0.0061 (3)	-0.0083 (3)
O1	0.0396 (13)	0.0269 (11)	0.0254 (11)	-0.0063 (9)	-0.0043 (9)	-0.0013 (9)
O2	0.0279 (12)	0.0400 (13)	0.0325 (12)	-0.0039 (10)	-0.0142 (10)	-0.0035 (10)
O3	0.0229 (10)	0.0239 (10)	0.0253 (10)	-0.0084 (8)	0.0037 (8)	-0.0025 (8)
O4	0.0286 (11)	0.0177 (10)	0.0380 (12)	0.0006 (8)	-0.0006 (9)	-0.0093 (9)
O5	0.0251 (10)	0.0143 (9)	0.0260 (10)	-0.0062 (7)	-0.0078 (8)	-0.0035 (8)
O6	0.0213 (10)	0.0192 (9)	0.0165 (9)	-0.0052 (7)	-0.0055 (8)	-0.0017 (8)
N1	0.0279 (13)	0.0192 (12)	0.0189 (12)	-0.0012 (10)	-0.0085 (10)	-0.0037 (10)
N2	0.0213 (12)	0.0165 (11)	0.0196 (11)	-0.0053 (9)	-0.0058 (9)	-0.0042 (9)
C1	0.0209 (13)	0.0150 (12)	0.0145 (13)	-0.0007 (10)	-0.0014 (10)	-0.0020 (10)
C2	0.0191 (13)	0.0188 (13)	0.0148 (13)	0.0025 (10)	-0.0032 (10)	-0.0053 (10)
C3	0.0145 (12)	0.0191 (13)	0.0174 (13)	-0.0022 (10)	-0.0033 (10)	-0.0074 (11)
C4	0.0173 (13)	0.0136 (12)	0.0192 (13)	-0.0030 (10)	0.0014 (10)	-0.0022 (10)
C5	0.0159 (12)	0.0168 (12)	0.0179 (13)	-0.0018 (10)	-0.0025 (10)	-0.0035 (10)
C6	0.0167 (12)	0.0152 (12)	0.0177 (13)	-0.0015 (10)	-0.0013 (10)	-0.0034 (10)
C7	0.0166 (12)	0.0152 (12)	0.0167 (13)	-0.0039 (10)	0.0009 (11)	-0.0012 (10)
C8	0.0252 (15)	0.0154 (13)	0.0345 (16)	-0.0085 (11)	-0.0045 (12)	-0.0082 (12)

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

Cl1—C3	1.725 (3)	C1—C6	1.393 (4)
O1—N1	1.242 (3)	C1—H1	0.9500
O2—N1	1.207 (3)	C2—C3	1.388 (4)
O3—N2	1.229 (3)	C3—C4	1.393 (4)
O4—N2	1.224 (3)	C4—C5	1.377 (3)
O5—C7	1.323 (3)	C5—C6	1.378 (3)
O5—C8	1.461 (3)	C5—H5	0.9500
O6—C7	1.202 (3)	C6—C7	1.507 (3)
N1—C2	1.477 (3)	C8—H8A	0.9800
N2—C4	1.473 (3)	C8—H8B	0.9800
C1—C2	1.390 (3)	C8—H8C	0.9800
C7—O5—C8	115.43 (19)	C5—C4—N2	118.5 (2)
O2—N1—O1	124.3 (2)	C3—C4—N2	118.8 (2)
O2—N1—C2	119.4 (2)	C4—C5—C6	119.1 (2)
O1—N1—C2	116.3 (2)	C4—C5—H5	120.4
O4—N2—O3	125.6 (2)	C6—C5—H5	120.4
O4—N2—C4	117.4 (2)	C5—C6—C1	120.3 (2)
O3—N2—C4	117.0 (2)	C5—C6—C7	118.5 (2)
C2—C1—C6	119.2 (2)	C1—C6—C7	121.1 (2)
C2—C1—H1	120.4	O6—C7—O5	125.7 (2)
C6—C1—H1	120.4	O6—C7—C6	122.5 (2)
C3—C2—C1	121.6 (2)	O5—C7—C6	111.8 (2)
C3—C2—N1	122.3 (2)	O5—C8—H8A	109.5
C1—C2—N1	116.1 (2)	O5—C8—H8B	109.5

C2—C3—C4	117.0 (2)	H8A—C8—H8B	109.5
C2—C3—C11	124.5 (2)	O5—C8—H8C	109.5
C4—C3—C11	118.43 (19)	H8A—C8—H8C	109.5
C5—C4—C3	122.7 (2)	H8B—C8—H8C	109.5
C6—C1—C2—C3	-0.5 (4)	O3—N2—C4—C5	-98.0 (3)
C6—C1—C2—N1	178.8 (2)	O4—N2—C4—C3	-97.7 (3)
O2—N1—C2—C3	-29.5 (4)	O3—N2—C4—C3	82.5 (3)
O1—N1—C2—C3	149.4 (3)	C3—C4—C5—C6	-1.7 (4)
O2—N1—C2—C1	151.2 (3)	N2—C4—C5—C6	178.8 (2)
O1—N1—C2—C1	-29.9 (3)	C4—C5—C6—C1	1.0 (4)
C1—C2—C3—C4	-0.1 (4)	C4—C5—C6—C7	178.7 (2)
N1—C2—C3—C4	-179.4 (2)	C2—C1—C6—C5	0.1 (4)
C1—C2—C3—C11	177.2 (2)	C2—C1—C6—C7	-177.6 (2)
N1—C2—C3—C11	-2.1 (4)	C8—O5—C7—O6	1.4 (4)
C2—C3—C4—C5	1.3 (4)	C8—O5—C7—C6	-179.4 (2)
C11—C3—C4—C5	-176.2 (2)	C5—C6—C7—O6	-12.6 (4)
C2—C3—C4—N2	-179.3 (2)	C1—C6—C7—O6	165.0 (2)
C11—C3—C4—N2	3.2 (3)	C5—C6—C7—O5	168.2 (2)
O4—N2—C4—C5	81.8 (3)	C1—C6—C7—O5	-14.2 (3)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C1—H1...O1 <sup>i</sup>	0.95	2.51	3.329 (3)	145
C5—H5...O6 <sup>ii</sup>	0.95	2.34	3.143 (3)	142

Symmetry codes: (i)  $-x+2, -y+1, -z+1$ ; (ii)  $-x+1, -y+2, -z$ .