Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

5-(2,6-Difluorophenyl)-1,3,4-thiadiazol-2-amine

Yao Wang, Rong Wan,* Feng Han and Peng Wang

Department of Applied Chemistry, College of Science, Nanjing University of Technology, No. 5 Xinmofan Road, Nanjing, Nanjing 210009, People's Republic of China
Correspondence e-mail: rwan@njut.edu.cn

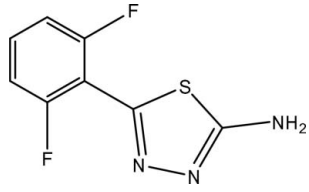
Received 6 May 2009; accepted 12 November 2009

Key indicators: single-crystal X-ray study; T = 293 K; mean σ(C–C) = 0.004 Å; R factor = 0.042; wR factor = 0.109; data-to-parameter ratio = 12.3.

The title compound, $C_8H_5F_2N_3S$, was synthesized by the reaction of 2,6-difluorobenzoic acid and thiosemicarbazide. The dihedral angle between the thiadiazole and phenyl ring is 35.19 (14)°. In the crystal structure, intermolecular N—H···N hydrogen bonds form chains along the b and c axes.

Related literature

For the biological activity of 1,3,4-thiadiazole derivatives, see: Nakagawa *et al.* (1996); Wang *et al.* (1999). For bond-length data see: Allen *et al.* (1987).



Experimental

Crystal data

$C_8H_5F_2N_3S$
M_r = 213.21
Monoclinic, $P2_1/c$
a = 9.0920 (18) Å
b = 8.7400 (17) Å
c = 10.936 (2) Å
β = 95.85 (3)°
V = 864.5 (3) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.37 mm^{-1}
T = 293 K
0.20 × 0.10 × 0.10 mm

Data collection

Enraf–Nonius CAD-4 diffractometer
Absorption correction: ψ scan (North *et al.*, 1968)
T_{min} = 0.931, T_{max} = 0.964
1670 measured reflections
1568 independent reflections
1189 reflections with $I > 2\sigma(I)$
R_{int} = 0.018
3 standard reflections
every 200 reflections
intensity decay: 1%

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.042
$wR(F^2)$ = 0.109
S = 1.01
1568 reflections
127 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.26 e Å^{-3}
$\Delta\rho_{min}$ = −0.28 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N3—H3A···N2^{i}	0.86	2.17	3.017 (4)	166
N3—H3B···N1^{ii}	0.86	2.30	3.088 (3)	152

Symmetry codes: (i) $-x+1, -y+1, -z$; (ii) $x, -y+\frac{1}{2}, z+\frac{1}{2}$.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1989); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

The authors gratefully acknowledge Professor Hua-Qin Wang of the Analysis Center, Nanjing University, for providing the Enraf–Nonius CAD-4 diffractometer for this research project.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RN2058).

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Enraf–Nonius (1989). *CAD-4 Software*. Enraf–Nonius, Delft, The Netherlands.
Harms, K. & Wocadlo, S. (1995). *XCAD4*. University of Marburg, Germany.
Nakagawa, Y., Nishimura, K., Izumi, K., Kinoshita, K., Kimura, T. & Kurihara, N. (1996). *J. Pesticide Sci*, **21**, 195–201.
North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* A**24**, 351–359.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Wang, Y. G., Cao, L., Yan, J., Ye, W. F., Zhou, Q. C. & Lu, B. X. (1999). *Chem. J. Chin. Univ.* **20**, 1903–1905.

supporting information

Acta Cryst. (2009). E**65**, o3108 [doi:10.1107/S1600536809047990]

5-(2,6-Difluorophenyl)-1,3,4-thiadiazol-2-amine

Yao Wang, Rong Wan, Feng Han and Peng Wang

S1. Comment

1,3,4-Thiadiazole derivatives represent a class of biologically important compounds, which often exhibit insecticidal, fungicidal and other biological activities (Nakagawa *et al.*, 1996; Wang *et al.*, 1999). We report here the crystal structure of the title compound, (I).

The molecular structure of (I) is shown in Fig.1, in which the bond lengths and angles are generally within normal ranges (Allen *et al.*, 1987). The dihedral angle between the thiadiazole and phenyl ring is 35.19 (14)°. In the crystal structure, intermolecular N—H···N hydrogen bonds (Fig. 2) form chains along the b and c axes. There are also intermolecular N-H···S contacts between the molecules, which may further stabilize the structure.

S2. Experimental

2,6-difluorobenzoic acid (2 mmol) and thiosemicarbazide (5 mmol) were mixed in a 25 ml flask, and kept in the oil bath at 90°C for 6 h. After cooling, the crude product (I) precipitated and was filtrated. Pure compound (I) was obtained by crystallization from ethanol (20 ml). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an acetone solution.

S3. Refinement

All H atoms bonded to the C atoms were placed geometrically at distances of 0.93–0.97 Å and included in the refinement in riding motion approximation with U_{iso}(H) = 1.2 or 1.5U_{eq} of the carrier atom.

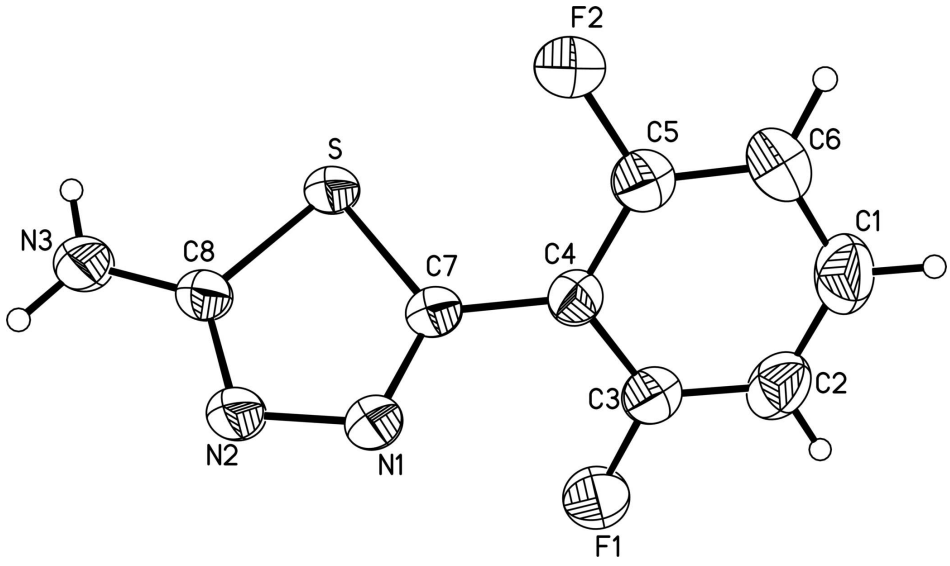


Figure 1

A view of the molecular structure of (I). Displacement ellipsoids are drawn at the 50% probability level.

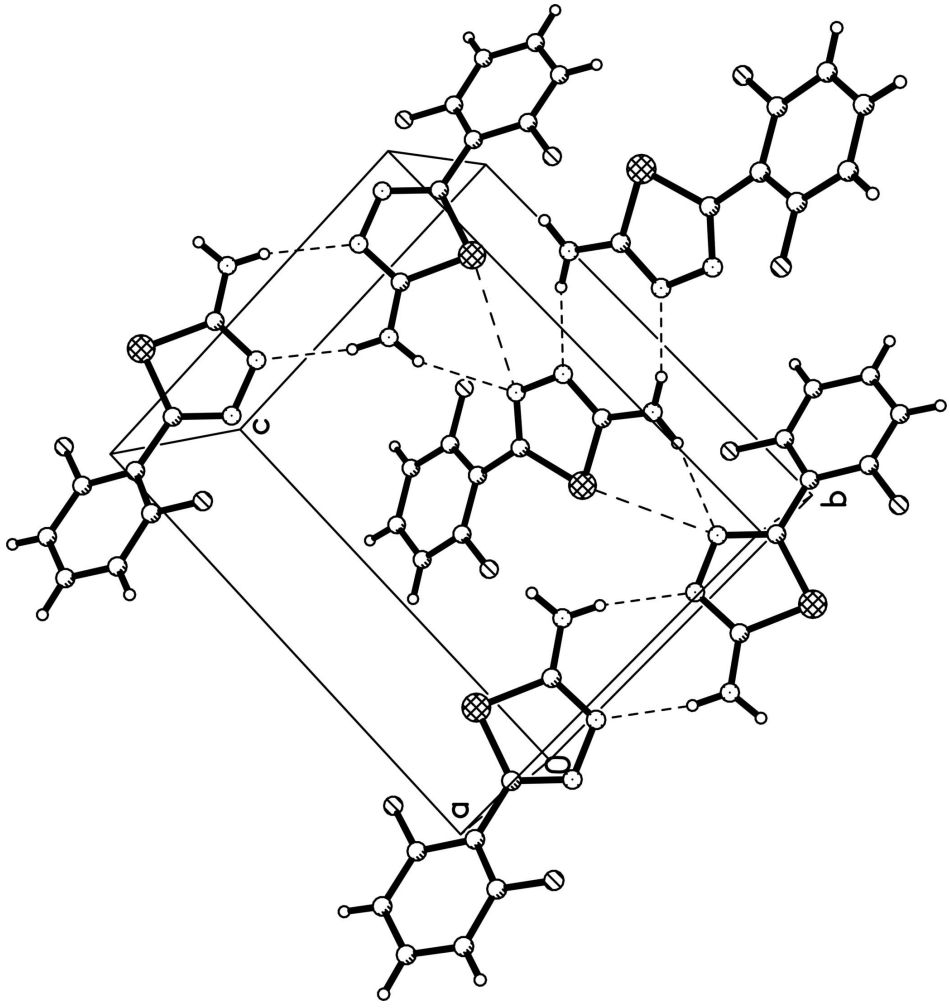


Figure 2

Partial packing view showing the hydrogen-bonded network. Dashed lines indicate intermolecular N—H···N hydrogen bonds and intermolecular N-H···S contacts between the molecules.

5-(2,6-Difluorophenyl)-1,3,4-thiadiazol-2-amine

Crystal data

$C_8H_5F_2N_3S$
$M_r = 213.21$
Monoclinic, $P2_1/c$
$a = 9.0920$ (18) Å
$b = 8.7400$ (17) Å
$c = 10.936$ (2) Å
$\beta = 95.85$ (3)°
$V = 864.5$ (3) Å^3
$Z = 4$
$F(000) = 432$
$D_x = 1.638$ Mg m^{-3}
Melting point: 533 K
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 25 reflections
$\theta = 10$–13°
$\mu = 0.37$ mm^{-1}
$T = 293$ K
Block, colorless
0.20 × 0.10 × 0.10 mm

Data collection

Enraf–Nonius CAD-4 diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
$\omega/2\theta$ scans
Absorption correction: ψ scan (North *et al.*, 1968)
T_{min} = 0.931, T_{max} = 0.964
1670 measured reflections
1568 independent reflections
1189 reflections with $I > 2\sigma(I)$
R_{int} = 0.018
θ_{max} = 25.3°, θ_{min} = 2.3°
h = 0→10
k = 0→10
l = −13→13
3 standard reflections every 200 reflections
intensity decay: 1%

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.042
$wR(F^2)$ = 0.109
S = 1.01
1568 reflections
127 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.060P)^2 + 0.150P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max}$ = 0.26 e Å^{-3}
$\Delta\rho_{min}$ = −0.28 e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	U_{iso}*/U_{eq}
S	0.68075 (8)	0.14488 (8)	0.16016 (6)	0.0423 (2)
F1	0.8904 (2)	0.0561 (2)	−0.18276 (16)	0.0637 (5)
N1	0.6608 (3)	0.1922 (3)	−0.07123 (19)	0.0475 (6)
C1	0.8942 (4)	−0.3350 (4)	−0.0851 (3)	0.0606 (9)
H1B	0.9319	−0.4285	−0.1078	0.073*
F2	0.6767 (2)	−0.1812 (2)	0.14183 (15)	0.0616 (5)
N2	0.6037 (3)	0.3224 (3)	−0.02139 (19)	0.0495 (6)
C2	0.9208 (3)	−0.2061 (4)	−0.1510 (3)	0.0531 (8)
H2B	0.9784	−0.2110	−0.2165	0.064*
N3	0.5574 (3)	0.4251 (3)	0.1678 (2)	0.0521 (7)
H3A	0.5209	0.5074	0.1338	0.062*
H3B	0.5620	0.4143	0.2462	0.062*
C3	0.8608 (3)	−0.0705 (3)	−0.1185 (2)	0.0442 (7)
C4	0.7736 (3)	−0.0549 (3)	−0.0209 (2)	0.0362 (6)
C5	0.7556 (3)	−0.1889 (3)	0.0433 (3)	0.0446 (7)

C6	0.8123 (4)	−0.3281 (3)	0.0144 (3)	0.0571 (8)
H6A	0.7961	−0.4149	0.0603	0.069*
C7	0.7062 (3)	0.0912 (3)	0.0101 (2)	0.0357 (6)
C8	0.6070 (3)	0.3142 (3)	0.0986 (2)	0.0376 (6)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S	0.0628 (5)	0.0382 (4)	0.0272 (3)	0.0092 (3)	0.0101 (3)	0.0052 (3)
F1	0.0799 (13)	0.0567 (11)	0.0601 (11)	0.0030 (10)	0.0351 (10)	0.0104 (9)
N1	0.0737 (17)	0.0411 (12)	0.0278 (11)	0.0138 (12)	0.0064 (11)	−0.0010 (9)
C1	0.069 (2)	0.0476 (18)	0.065 (2)	0.0180 (16)	0.0049 (17)	−0.0093 (16)
F2	0.0825 (13)	0.0493 (10)	0.0580 (11)	0.0069 (9)	0.0309 (10)	0.0111 (8)
N2	0.0803 (18)	0.0405 (13)	0.0280 (11)	0.0176 (12)	0.0069 (11)	0.0017 (10)
C2	0.0503 (18)	0.063 (2)	0.0471 (17)	0.0128 (16)	0.0116 (14)	−0.0065 (15)
N3	0.0824 (19)	0.0448 (13)	0.0304 (12)	0.0185 (13)	0.0128 (12)	0.0008 (10)
C3	0.0479 (16)	0.0468 (16)	0.0384 (14)	0.0020 (13)	0.0071 (13)	0.0000 (12)
C4	0.0389 (15)	0.0363 (14)	0.0333 (13)	0.0015 (11)	0.0035 (11)	−0.0016 (11)
C5	0.0468 (16)	0.0446 (15)	0.0430 (15)	0.0022 (13)	0.0077 (13)	0.0015 (13)
C6	0.069 (2)	0.0377 (16)	0.065 (2)	0.0041 (15)	0.0080 (17)	0.0039 (14)
C7	0.0432 (15)	0.0362 (13)	0.0279 (12)	0.0014 (12)	0.0051 (11)	0.0011 (11)
C8	0.0481 (16)	0.0348 (14)	0.0298 (13)	0.0042 (12)	0.0042 (11)	0.0032 (10)

Geometric parameters (Å, º)

S—C8	1.733 (3)	C2—C3	1.367 (4)
S—C7	1.745 (2)	C2—H2B	0.9300
F1—C3	1.352 (3)	N3—C8	1.336 (3)
N1—C7	1.291 (3)	N3—H3A	0.8600
N1—N2	1.385 (3)	N3—H3B	0.8600
C1—C2	1.372 (4)	C3—C4	1.400 (4)
C1—C6	1.382 (4)	C4—C5	1.384 (4)
C1—H1B	0.9300	C4—C7	1.471 (3)
F2—C5	1.356 (3)	C5—C6	1.370 (4)
N2—C8	1.311 (3)	C6—H6A	0.9300
C8—S—C7	86.98 (12)	C5—C4—C3	114.2 (2)
C7—N1—N2	113.4 (2)	C5—C4—C7	123.0 (2)
C2—C1—C6	121.1 (3)	C3—C4—C7	122.8 (2)
C2—C1—H1B	119.5	F2—C5—C6	118.0 (2)
C6—C1—H1B	119.5	F2—C5—C4	117.4 (2)
C8—N2—N1	112.2 (2)	C6—C5—C4	124.6 (3)
C3—C2—C1	118.6 (3)	C5—C6—C1	117.8 (3)
C3—C2—H2B	120.7	C5—C6—H6A	121.1
C1—C2—H2B	120.7	C1—C6—H6A	121.1
C8—N3—H3A	120.0	N1—C7—C4	123.1 (2)
C8—N3—H3B	120.0	N1—C7—S	113.60 (19)
H3A—N3—H3B	120.0	C4—C7—S	123.26 (18)

F1—C3—C2	117.9 (2)	N2—C8—N3	123.6 (2)
F1—C3—C4	118.4 (2)	N2—C8—S	113.82 (19)
C2—C3—C4	123.7 (3)	N3—C8—S	122.63 (19)
C7—N1—N2—C8	−0.5 (4)	C2—C1—C6—C5	1.5 (5)
C6—C1—C2—C3	−1.9 (5)	N2—N1—C7—C4	−178.8 (2)
C1—C2—C3—F1	178.7 (3)	N2—N1—C7—S	0.9 (3)
C1—C2—C3—C4	0.2 (5)	C5—C4—C7—N1	−146.4 (3)
F1—C3—C4—C5	−176.8 (2)	C3—C4—C7—N1	33.2 (4)
C2—C3—C4—C5	1.6 (4)	C5—C4—C7—S	33.9 (4)
F1—C3—C4—C7	3.6 (4)	C3—C4—C7—S	−146.5 (2)
C2—C3—C4—C7	−178.0 (3)	C8—S—C7—N1	−0.7 (2)
C3—C4—C5—F2	177.7 (2)	C8—S—C7—C4	178.9 (2)
C7—C4—C5—F2	−2.7 (4)	N1—N2—C8—N3	−179.6 (3)
C3—C4—C5—C6	−2.1 (4)	N1—N2—C8—S	−0.1 (3)
C7—C4—C5—C6	177.5 (3)	C7—S—C8—N2	0.4 (2)
F2—C5—C6—C1	−179.2 (3)	C7—S—C8—N3	−180.0 (3)
C4—C5—C6—C1	0.6 (5)		

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N3—H3*A*···N2^{i}	0.86	2.17	3.017 (4)	166
N3—H3*B*···N1^{ii}	0.86	2.30	3.088 (3)	152

Symmetry codes: (i) −*x*+1, −*y*+1, −*z*; (ii) *x*, −*y*+1/2, *z*+1/2.