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6-Methyl-2,7-diphenyl-1,4-diazepan-5-one

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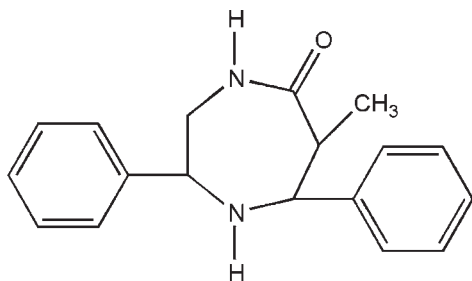
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 Key indicators: single-crystal X-ray study; $T = 295$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.049; wR factor = 0.150; data-to-parameter ratio = 17.6.

The title compound, $\text{C}_{18}\text{H}_{20}\text{N}_2\text{O}$, crystallizes with two molecules in the asymmetric unit. The seven-membered ring in both molecules adopts a distorted chair conformation. The dihedral angles between the phenyl rings are 43.2 (1) and 54.7 (1)° in the two molecules. The crystal packing features $\text{N}-\text{H}\cdots\text{O}$ and weak $\text{N}-\text{H}\cdots\pi$ and $\text{C}-\text{H}\cdots\pi$ interactions.

Related literature

For the biological activity of related compounds, see: Gopalakrishnan *et al.* (2007); Włodarczyk *et al.* (2006). For the synthetic procedure, see: Thennarasu & Perumal (2002). For hydrogen-bond motifs, see: Bernstein *et al.* (1995).



Experimental

Crystal data

 $\text{C}_{18}\text{H}_{20}\text{N}_2\text{O}$
 $M_r = 280.36$

 Monoclinic, $P2_1/c$
 $a = 10.8621$ (3) Å

 $b = 21.3210$ (7) Å
 $c = 13.3890$ (4) Å
 $\beta = 91.167$ (2)°
 $V = 3100.13$ (16) Å³
 $Z = 8$

 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 295$ K
 $0.26 \times 0.22 \times 0.18$ mm

Data collection

 Bruker Kappa APEXII
 diffractometer
 Absorption correction: multi-scan
 (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.981$, $T_{\max} = 0.987$

 32883 measured reflections
 7003 independent reflections
 4175 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.049$
 $wR(F^2) = 0.150$
 $S = 1.01$
 7003 reflections
 397 parameters
 1 restraint

 H atoms treated by a mixture of
 independent and constrained
 refinement
 $\Delta\rho_{\text{max}} = 0.26$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.18$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N2}-\text{H2A}\cdots\text{O2}^{\text{i}}$	0.90 (2)	2.02 (2)	2.911 (2)	170.2 (18)
$\text{N4}-\text{H4A}\cdots\text{O1}^{\text{i}}$	0.87 (2)	2.03 (2)	2.884 (2)	167 (2)
$\text{N1}-\text{H1A}\cdots\text{Cg1}^{\text{ii}}$	0.88 (2)	2.93 (2)	3.707 (2)	149.5 (2)
$\text{C18}-\text{H18}\cdots\text{Cg5}^{\text{iii}}$	0.93	2.95	3.872 (2)	171

Symmetry codes: (i) $-x, -y, -z + 1$; (ii) $-x, -y, -z$; (iii) $x + 1, y, z$. Cg1 and Cg5 are the centroids of the C1–C6 and C31–C36 rings, respectively.

Data collection: APEX2 (Bruker, 2004); cell refinement: SAINT (Bruker, 2004); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2009); software used to prepare material for publication: SHELXL97.

The authors wish to acknowledge SAIF, IIT, Madras for the data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: GK2234).

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supporting information

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6-Methyl-2,7-diphenyl-1,4-diazepan-5-one

S. Sathish Kumar, Helen P. Kavitha, Jasmine P. Vennila, G. Chakkaravarthi and V. Manivannan

S1. Comment

In view of the biological activities of the heterocyclic compounds with 1,4-diazepan-5-one fragment that have widespread applications in pharmaceuticals (Wlodarczyk *et al.*, 2006; Gopalakrishnan *et al.*, 2007), we report the crystal structure of the title compound.

In the molecule A atoms C7, C8 and C9 show S-configuration whereas in the enantiomeric molecule B atoms C25, C26 and C27 show R-configuration. The phenyl ring C1—C6 forms the dihedral angle of 43.2 (1)° with the phenyl ring C13—C18 in molecule A and the phenyl ring C19—C24 forms the dihedral angle of 54.7 (1)° with the phenyl ring C31—C36 in molecule B. Intermolecular N—H···O interactions between the two symmetry independent molecules generates an eight-membered ring with graph-set motif $R_2^2(8)$ (Bernstein *et al.*, 1995). The crystal packing is controlled by weak N—H···O, N—H··· π and C—H··· π interactions (see Table 1 & Fig. 2).

S2. Experimental

The title compound was prepared according to the general procedure reported by Thennarasu & Perumal (2002). 3-Methyl-2,6-diphenylpiperidin-4-one (1.82 g, 5 mmole) was added into the ice cold sulfuric acid and the mixture was allowed to reach the room temperature. Then, sodium azide (1.92 g, 30 mmole) was added in portions over a period of one hour. The solution was then poured into crushed ice. The pH of the solution was adjusted to approximately 8.0 using 2 N NaOH solution. The precipitated white solid was recrystallized from ethanol to yield colourless diffraction quality crystals.

S3. Refinement

The H atoms from the N—H groups were located in difference Fourier maps and their positions and isotropic displacement parameters were freely refined. All other H atoms were positioned geometrically and refined using riding model with C—H = 0.93 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic C—H, C—H = 0.98 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for methine C—H, C—H = 0.97 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for methylene C—H and C—H = 0.96 Å and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl. The components of the anisotropic displacement parameters in direction of the bond of C33 and C34 were restrained to be equal within an effective standard deviation of 0.001 using the DELU command in *SHELXL97* (Sheldrick, 2008).

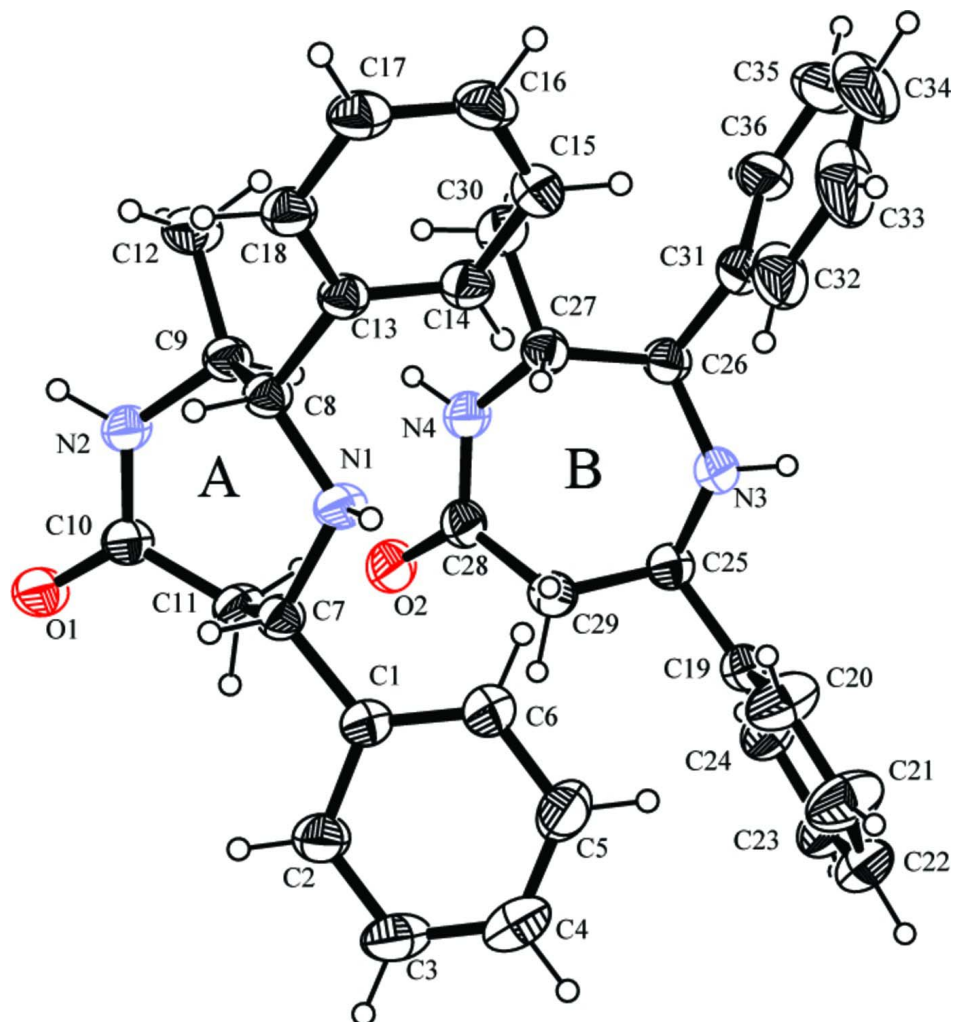


Figure 1

The molecular structure of the title compound, with atom labels and 30% probability displacement ellipsoids for non-H atoms. A and B are the two molecules in the asymmetric unit.

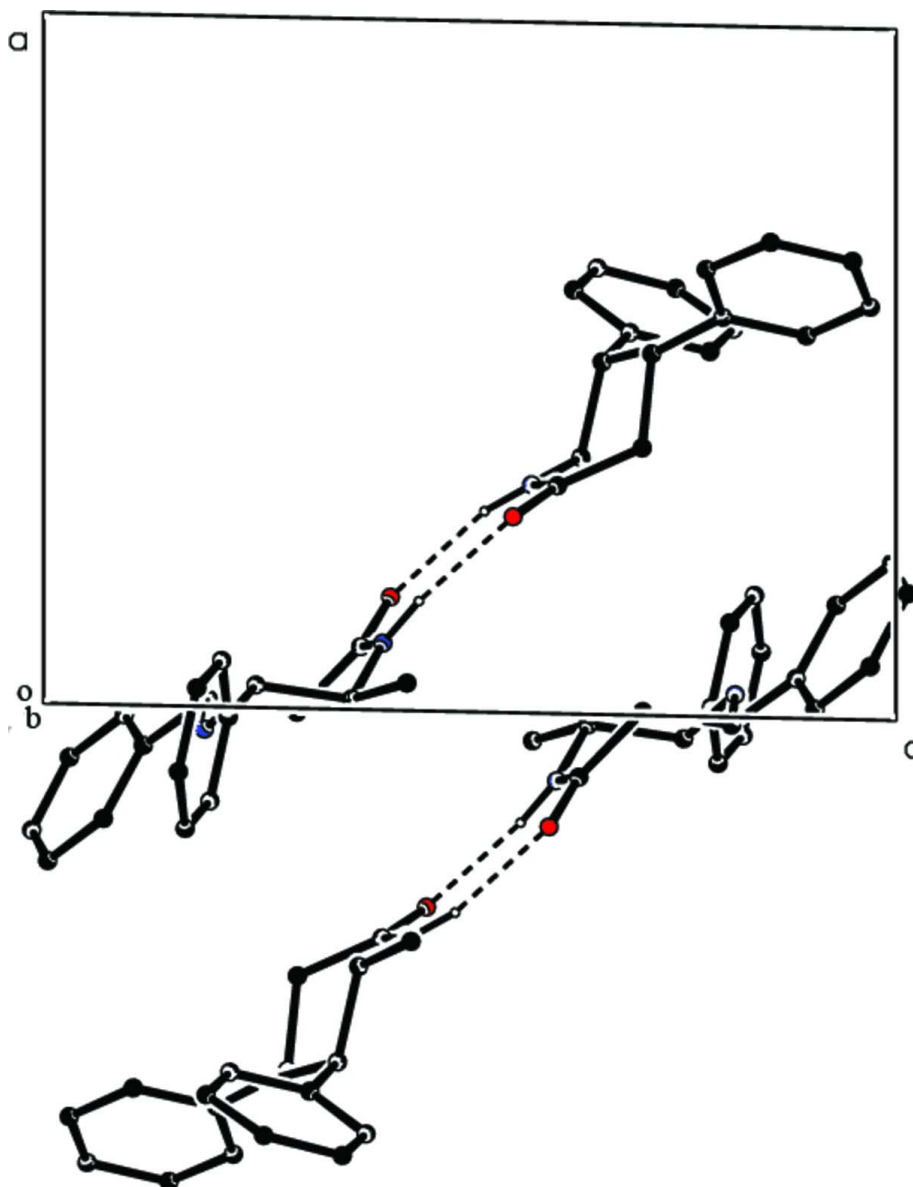


Figure 2

Intermolecular N—H···O interactions - view down the *b* axis. Hydrogen bonds are shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

6-Methyl-2,7-diphenyl-1,4-diazepan-5-one

Crystal data

$C_{18}H_{20}N_2O$

$M_r = 280.36$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 10.8621\ (3)\ \text{\AA}$

$b = 21.3210\ (7)\ \text{\AA}$

$c = 13.3890\ (4)\ \text{\AA}$

$\beta = 91.167\ (2)^\circ$

$V = 3100.13\ (16)\ \text{\AA}^3$

$Z = 8$

$F(000) = 1200$

$D_x = 1.201\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2786 reflections

$\theta = 1.9\text{--}27.4^\circ$

$\mu = 0.08\ \text{mm}^{-1}$

$T = 295$ K
Block, colourless

$0.26 \times 0.22 \times 0.18$ mm

Data collection

Bruker Kappa APEXII
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
 ω and φ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
 $T_{\min} = 0.981$, $T_{\max} = 0.987$

32883 measured reflections
7003 independent reflections
4175 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.034$
 $\theta_{\max} = 27.4^\circ$, $\theta_{\min} = 1.9^\circ$
 $h = -14 \rightarrow 13$
 $k = -27 \rightarrow 27$
 $l = -17 \rightarrow 17$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.049$
 $wR(F^2) = 0.150$
 $S = 1.01$
7003 reflections
397 parameters
1 restraint
Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map
Hydrogen site location: inferred from
neighbouring sites
H atoms treated by a mixture of independent
and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0613P)^2 + 0.757P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.26 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	-0.05795 (17)	-0.04801 (8)	0.11713 (12)	0.0485 (4)
C2	-0.0143 (2)	-0.10676 (9)	0.09516 (15)	0.0678 (6)
H2	0.0593	-0.1202	0.1245	0.081*
C3	-0.0773 (3)	-0.14624 (10)	0.03052 (16)	0.0795 (7)
H3	-0.0463	-0.1860	0.0173	0.095*
C4	-0.1837 (2)	-0.12746 (12)	-0.01365 (17)	0.0785 (7)
H4	-0.2269	-0.1543	-0.0565	0.094*
C5	-0.2275 (2)	-0.06876 (13)	0.00506 (18)	0.0847 (7)
H5	-0.2998	-0.0552	-0.0264	0.102*
C6	-0.16502 (19)	-0.02932 (11)	0.07031 (16)	0.0687 (6)
H6	-0.1961	0.0105	0.0826	0.082*
C7	0.01134 (17)	-0.00817 (8)	0.19307 (12)	0.0473 (4)
H7	0.0991	-0.0087	0.1775	0.057*
C8	0.03251 (17)	0.10337 (7)	0.24911 (12)	0.0463 (4)
H8	0.1206	0.1007	0.2351	0.056*
C9	0.01501 (17)	0.09158 (8)	0.36136 (12)	0.0493 (4)
H9	-0.0702	0.0782	0.3712	0.059*
C10	0.09125 (18)	-0.01783 (8)	0.37215 (13)	0.0526 (4)
C11	-0.00555 (19)	-0.03663 (8)	0.29722 (13)	0.0549 (5)
H11A	-0.0853	-0.0241	0.3216	0.066*
H11B	-0.0056	-0.0820	0.2914	0.066*
C12	0.0399 (2)	0.14835 (9)	0.42625 (14)	0.0662 (6)

H12A	0.0341	0.1368	0.4953	0.099*
H12B	-0.0197	0.1804	0.4108	0.099*
H12C	0.1210	0.1640	0.4139	0.099*
C13	-0.01396 (17)	0.16719 (8)	0.21681 (12)	0.0476 (4)
C14	-0.13663 (18)	0.17781 (9)	0.19508 (14)	0.0562 (5)
H14	-0.1928	0.1451	0.1998	0.067*
C15	-0.1770 (2)	0.23685 (9)	0.16622 (15)	0.0643 (5)
H15	-0.2601	0.2436	0.1522	0.077*
C16	-0.0951 (2)	0.28531 (9)	0.15823 (15)	0.0663 (6)
H16	-0.1222	0.3248	0.1383	0.080*
C17	0.0266 (2)	0.27512 (9)	0.17977 (15)	0.0637 (5)
H17	0.0825	0.3079	0.1747	0.076*
C18	0.06729 (18)	0.21665 (8)	0.20889 (14)	0.0556 (5)
H18	0.1505	0.2104	0.2234	0.067*
C19	-0.57859 (17)	-0.06524 (9)	0.20328 (13)	0.0528 (4)
C20	-0.5541 (3)	-0.05228 (12)	0.10529 (16)	0.0890 (8)
H20	-0.5076	-0.0171	0.0896	0.107*
C21	-0.5974 (3)	-0.09052 (13)	0.03036 (18)	0.1049 (9)
H21	-0.5814	-0.0806	-0.0357	0.126*
C22	-0.6634 (2)	-0.14265 (12)	0.05125 (19)	0.0848 (7)
H22	-0.6910	-0.1690	0.0001	0.102*
C23	-0.6884 (2)	-0.15582 (11)	0.14736 (19)	0.0822 (7)
H23	-0.7340	-0.1914	0.1625	0.099*
C24	-0.6472 (2)	-0.11726 (10)	0.22268 (16)	0.0685 (6)
H24	-0.6664	-0.1267	0.2884	0.082*
C25	-0.52358 (17)	-0.02607 (8)	0.28651 (13)	0.0519 (4)
H25	-0.5652	-0.0359	0.3489	0.062*
C26	-0.51055 (17)	0.08491 (8)	0.34472 (14)	0.0545 (5)
H26	-0.5551	0.0724	0.4044	0.065*
C27	-0.37168 (17)	0.08639 (8)	0.36986 (14)	0.0534 (4)
H27	-0.3269	0.0873	0.3071	0.064*
C28	-0.32986 (17)	-0.02759 (9)	0.39725 (14)	0.0557 (5)
C29	-0.38718 (18)	-0.04257 (9)	0.29788 (14)	0.0602 (5)
H29A	-0.3775	-0.0871	0.2855	0.072*
H29B	-0.3424	-0.0203	0.2469	0.072*
C30	-0.3340 (2)	0.14317 (10)	0.43108 (18)	0.0744 (6)
H30A	-0.3845	0.1460	0.4889	0.112*
H30B	-0.3442	0.1804	0.3914	0.112*
H30C	-0.2492	0.1392	0.4518	0.112*
C31	-0.55368 (18)	0.14851 (10)	0.31237 (19)	0.0691 (6)
C32	-0.5286 (2)	0.17105 (11)	0.2185 (2)	0.0944 (8)
H32	-0.4888	0.1454	0.1731	0.113*
C33	-0.5619 (3)	0.23130 (17)	0.1909 (4)	0.1510 (17)
H33	-0.5456	0.2459	0.1271	0.181*
C34	-0.6192 (4)	0.26937 (18)	0.2584 (6)	0.180 (3)
H34	-0.6395	0.3104	0.2413	0.216*
C35	-0.6460 (3)	0.24657 (18)	0.3505 (5)	0.161 (2)
H35	-0.6870	0.2720	0.3954	0.193*

C36	-0.6138 (2)	0.18703 (12)	0.3781 (3)	0.1021 (10)
H36	-0.6324	0.1724	0.4414	0.122*
N1	-0.03260 (15)	0.05628 (7)	0.18881 (11)	0.0518 (4)
N2	0.09714 (16)	0.04195 (7)	0.39814 (11)	0.0550 (4)
N3	-0.53901 (17)	0.04046 (7)	0.26410 (12)	0.0566 (4)
N4	-0.33046 (16)	0.03138 (8)	0.42746 (12)	0.0571 (4)
O1	0.16394 (14)	-0.05709 (6)	0.40701 (10)	0.0724 (4)
O2	-0.28387 (14)	-0.06962 (6)	0.44918 (11)	0.0751 (4)
H1A	-0.0273 (17)	0.0686 (9)	0.1266 (15)	0.060 (6)*
H2A	0.1588 (18)	0.0538 (9)	0.4403 (15)	0.065 (6)*
H3A	-0.616 (2)	0.0465 (10)	0.2477 (16)	0.077 (7)*
H4A	-0.291 (2)	0.0403 (10)	0.4827 (17)	0.073 (7)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0583 (11)	0.0441 (10)	0.0432 (9)	-0.0040 (8)	0.0021 (8)	0.0013 (7)
C2	0.0880 (16)	0.0517 (12)	0.0632 (12)	0.0099 (10)	-0.0133 (11)	-0.0067 (9)
C3	0.120 (2)	0.0524 (13)	0.0659 (13)	-0.0023 (13)	-0.0062 (13)	-0.0130 (10)
C4	0.0900 (17)	0.0830 (17)	0.0627 (13)	-0.0259 (14)	0.0033 (12)	-0.0208 (12)
C5	0.0633 (14)	0.111 (2)	0.0794 (15)	-0.0023 (13)	-0.0112 (11)	-0.0298 (14)
C6	0.0635 (13)	0.0708 (14)	0.0713 (13)	0.0067 (10)	-0.0083 (10)	-0.0169 (11)
C7	0.0540 (10)	0.0389 (9)	0.0487 (9)	-0.0007 (8)	-0.0040 (8)	0.0017 (7)
C8	0.0548 (11)	0.0380 (9)	0.0460 (9)	0.0003 (7)	-0.0025 (8)	0.0015 (7)
C9	0.0594 (11)	0.0429 (10)	0.0454 (9)	0.0030 (8)	-0.0038 (8)	0.0003 (7)
C10	0.0696 (12)	0.0448 (10)	0.0430 (9)	0.0034 (9)	-0.0063 (8)	0.0032 (8)
C11	0.0725 (13)	0.0401 (10)	0.0518 (10)	-0.0040 (9)	-0.0083 (9)	0.0048 (8)
C12	0.0933 (16)	0.0515 (11)	0.0533 (11)	0.0069 (10)	-0.0063 (10)	-0.0058 (9)
C13	0.0605 (11)	0.0397 (9)	0.0425 (9)	0.0014 (8)	-0.0036 (8)	0.0000 (7)
C14	0.0619 (12)	0.0448 (10)	0.0618 (11)	-0.0033 (9)	-0.0036 (9)	0.0008 (8)
C15	0.0677 (13)	0.0550 (12)	0.0697 (13)	0.0111 (10)	-0.0122 (10)	-0.0024 (10)
C16	0.0933 (17)	0.0392 (11)	0.0660 (12)	0.0083 (10)	-0.0054 (11)	0.0028 (9)
C17	0.0812 (15)	0.0404 (10)	0.0696 (12)	-0.0061 (10)	0.0014 (11)	-0.0004 (9)
C18	0.0632 (12)	0.0445 (10)	0.0588 (11)	-0.0032 (9)	-0.0033 (9)	-0.0008 (8)
C19	0.0559 (11)	0.0503 (11)	0.0522 (10)	0.0020 (8)	-0.0043 (8)	0.0017 (8)
C20	0.136 (2)	0.0767 (16)	0.0541 (12)	-0.0326 (15)	0.0007 (13)	0.0039 (11)
C21	0.165 (3)	0.096 (2)	0.0536 (13)	-0.027 (2)	-0.0054 (15)	-0.0060 (13)
C22	0.0978 (18)	0.0840 (17)	0.0718 (16)	-0.0069 (14)	-0.0201 (13)	-0.0205 (13)
C23	0.0805 (16)	0.0777 (16)	0.0879 (17)	-0.0227 (13)	-0.0094 (13)	-0.0082 (13)
C24	0.0713 (14)	0.0710 (14)	0.0632 (12)	-0.0150 (11)	-0.0001 (10)	0.0004 (11)
C25	0.0588 (11)	0.0487 (10)	0.0483 (9)	-0.0006 (8)	0.0012 (8)	0.0039 (8)
C26	0.0548 (11)	0.0483 (10)	0.0605 (11)	0.0006 (8)	0.0019 (9)	-0.0014 (9)
C27	0.0543 (11)	0.0501 (11)	0.0557 (10)	0.0002 (8)	0.0000 (8)	0.0034 (8)
C28	0.0569 (12)	0.0542 (11)	0.0556 (10)	0.0029 (9)	-0.0049 (9)	0.0044 (9)
C29	0.0682 (13)	0.0526 (11)	0.0596 (11)	0.0076 (9)	-0.0064 (9)	-0.0045 (9)
C30	0.0751 (15)	0.0560 (13)	0.0912 (16)	-0.0024 (10)	-0.0197 (12)	-0.0040 (11)
C31	0.0543 (12)	0.0509 (12)	0.1012 (17)	0.0008 (9)	-0.0175 (11)	-0.0044 (12)
C32	0.0843 (17)	0.0668 (15)	0.131 (2)	-0.0035 (12)	-0.0251 (16)	0.0311 (15)

C33	0.105 (3)	0.083 (2)	0.263 (5)	-0.0263 (18)	-0.072 (3)	0.079 (3)
C34	0.096 (3)	0.0458 (19)	0.395 (8)	-0.0086 (17)	-0.079 (4)	0.031 (3)
C35	0.084 (2)	0.067 (2)	0.331 (7)	0.0177 (19)	-0.044 (3)	-0.062 (3)
C36	0.0637 (15)	0.0716 (16)	0.170 (3)	0.0132 (12)	-0.0193 (16)	-0.0429 (17)
N1	0.0728 (11)	0.0384 (8)	0.0437 (8)	-0.0004 (7)	-0.0091 (7)	0.0037 (6)
N2	0.0701 (11)	0.0454 (9)	0.0487 (8)	0.0024 (7)	-0.0149 (8)	0.0007 (7)
N3	0.0583 (11)	0.0499 (9)	0.0610 (10)	0.0037 (8)	-0.0101 (8)	0.0004 (7)
N4	0.0679 (11)	0.0538 (10)	0.0491 (9)	0.0028 (8)	-0.0096 (8)	0.0018 (7)
O1	0.0996 (11)	0.0519 (8)	0.0645 (8)	0.0159 (8)	-0.0274 (8)	-0.0014 (7)
O2	0.0886 (11)	0.0569 (9)	0.0785 (10)	0.0038 (7)	-0.0264 (8)	0.0104 (7)

Geometric parameters (Å, °)

C1—C6	1.369 (3)	C19—C25	1.506 (2)
C1—C2	1.373 (3)	C20—C21	1.369 (3)
C1—C7	1.513 (2)	C20—H20	0.9300
C2—C3	1.379 (3)	C21—C22	1.355 (4)
C2—H2	0.9300	C21—H21	0.9300
C3—C4	1.349 (3)	C22—C23	1.350 (3)
C3—H3	0.9300	C22—H22	0.9300
C4—C5	1.364 (3)	C23—C24	1.369 (3)
C4—H4	0.9300	C23—H23	0.9300
C5—C6	1.381 (3)	C24—H24	0.9300
C5—H5	0.9300	C25—N3	1.459 (2)
C6—H6	0.9300	C25—C29	1.527 (3)
C7—N1	1.455 (2)	C25—H25	0.9800
C7—C11	1.535 (2)	C26—N3	1.465 (2)
C7—H7	0.9800	C26—C31	1.496 (3)
C8—N1	1.462 (2)	C26—C27	1.539 (3)
C8—C13	1.511 (2)	C26—H26	0.9800
C8—C9	1.539 (2)	C27—N4	1.469 (2)
C8—H8	0.9800	C27—C30	1.514 (3)
C9—N2	1.463 (2)	C27—H27	0.9800
C9—C12	1.511 (2)	C28—O2	1.234 (2)
C9—H9	0.9800	C28—N4	1.321 (2)
C10—O1	1.236 (2)	C28—C29	1.492 (3)
C10—N2	1.322 (2)	C29—H29A	0.9700
C10—C11	1.493 (2)	C29—H29B	0.9700
C11—H11A	0.9700	C30—H30A	0.9600
C11—H11B	0.9700	C30—H30B	0.9600
C12—H12A	0.9600	C30—H30C	0.9600
C12—H12B	0.9600	C31—C36	1.377 (3)
C12—H12C	0.9600	C31—C32	1.378 (4)
C13—C14	1.377 (3)	C32—C33	1.382 (4)
C13—C18	1.381 (2)	C32—H32	0.9300
C14—C15	1.385 (3)	C33—C34	1.373 (7)
C14—H14	0.9300	C33—H33	0.9300
C15—C16	1.369 (3)	C34—C35	1.363 (8)

C15—H15	0.9300	C34—H34	0.9300
C16—C17	1.365 (3)	C35—C36	1.366 (5)
C16—H16	0.9300	C35—H35	0.9300
C17—C18	1.376 (3)	C36—H36	0.9300
C17—H17	0.9300	N1—H1A	0.876 (19)
C18—H18	0.9300	N2—H2A	0.90 (2)
C19—C24	1.364 (3)	N3—H3A	0.87 (2)
C19—C20	1.372 (3)	N4—H4A	0.87 (2)
C6—C1—C2	117.44 (18)	C22—C21—C20	120.8 (2)
C6—C1—C7	123.39 (17)	C22—C21—H21	119.6
C2—C1—C7	119.14 (17)	C20—C21—H21	119.6
C1—C2—C3	121.5 (2)	C23—C22—C21	119.0 (2)
C1—C2—H2	119.2	C23—C22—H22	120.5
C3—C2—H2	119.2	C21—C22—H22	120.5
C4—C3—C2	120.2 (2)	C22—C23—C24	120.5 (2)
C4—C3—H3	119.9	C22—C23—H23	119.7
C2—C3—H3	119.9	C24—C23—H23	119.7
C3—C4—C5	119.4 (2)	C19—C24—C23	121.3 (2)
C3—C4—H4	120.3	C19—C24—H24	119.3
C5—C4—H4	120.3	C23—C24—H24	119.3
C4—C5—C6	120.5 (2)	N3—C25—C19	110.16 (14)
C4—C5—H5	119.8	N3—C25—C29	110.58 (16)
C6—C5—H5	119.8	C19—C25—C29	108.39 (15)
C1—C6—C5	120.9 (2)	N3—C25—H25	109.2
C1—C6—H6	119.5	C19—C25—H25	109.2
C5—C6—H6	119.5	C29—C25—H25	109.2
N1—C7—C1	110.20 (14)	N3—C26—C31	108.24 (16)
N1—C7—C11	111.32 (14)	N3—C26—C27	111.45 (15)
C1—C7—C11	108.75 (14)	C31—C26—C27	110.15 (15)
N1—C7—H7	108.8	N3—C26—H26	109.0
C1—C7—H7	108.8	C31—C26—H26	109.0
C11—C7—H7	108.8	C27—C26—H26	109.0
N1—C8—C13	107.78 (13)	N4—C27—C30	106.14 (15)
N1—C8—C9	111.05 (14)	N4—C27—C26	112.67 (15)
C13—C8—C9	112.28 (14)	C30—C27—C26	112.83 (16)
N1—C8—H8	108.5	N4—C27—H27	108.3
C13—C8—H8	108.5	C30—C27—H27	108.3
C9—C8—H8	108.5	C26—C27—H27	108.3
N2—C9—C12	106.57 (14)	O2—C28—N4	121.56 (17)
N2—C9—C8	111.11 (14)	O2—C28—C29	120.27 (18)
C12—C9—C8	113.98 (15)	N4—C28—C29	118.17 (17)
N2—C9—H9	108.3	C28—C29—C25	115.22 (16)
C12—C9—H9	108.3	C28—C29—H29A	108.5
C8—C9—H9	108.3	C25—C29—H29A	108.5
O1—C10—N2	121.76 (17)	C28—C29—H29B	108.5
O1—C10—C11	120.52 (16)	C25—C29—H29B	108.5
N2—C10—C11	117.72 (16)	H29A—C29—H29B	107.5

C10—C11—C7	114.13 (15)	C27—C30—H30A	109.5
C10—C11—H11A	108.7	C27—C30—H30B	109.5
C7—C11—H11A	108.7	H30A—C30—H30B	109.5
C10—C11—H11B	108.7	C27—C30—H30C	109.5
C7—C11—H11B	108.7	H30A—C30—H30C	109.5
H11A—C11—H11B	107.6	H30B—C30—H30C	109.5
C9—C12—H12A	109.5	C36—C31—C32	118.7 (2)
C9—C12—H12B	109.5	C36—C31—C26	120.3 (2)
H12A—C12—H12B	109.5	C32—C31—C26	120.9 (2)
C9—C12—H12C	109.5	C31—C32—C33	120.8 (4)
H12A—C12—H12C	109.5	C31—C32—H32	119.6
H12B—C12—H12C	109.5	C33—C32—H32	119.6
C14—C13—C18	118.40 (16)	C34—C33—C32	119.5 (5)
C14—C13—C8	121.63 (16)	C34—C33—H33	120.2
C18—C13—C8	119.96 (16)	C32—C33—H33	120.2
C13—C14—C15	120.50 (18)	C35—C34—C33	119.5 (4)
C13—C14—H14	119.8	C35—C34—H34	120.2
C15—C14—H14	119.8	C33—C34—H34	120.2
C16—C15—C14	120.38 (19)	C34—C35—C36	121.2 (5)
C16—C15—H15	119.8	C34—C35—H35	119.4
C14—C15—H15	119.8	C36—C35—H35	119.4
C17—C16—C15	119.41 (18)	C35—C36—C31	120.3 (4)
C17—C16—H16	120.3	C35—C36—H36	119.9
C15—C16—H16	120.3	C31—C36—H36	119.9
C16—C17—C18	120.56 (19)	C7—N1—C8	118.18 (14)
C16—C17—H17	119.7	C7—N1—H1A	107.0 (12)
C18—C17—H17	119.7	C8—N1—H1A	106.2 (12)
C17—C18—C13	120.75 (19)	C10—N2—C9	125.64 (16)
C17—C18—H18	119.6	C10—N2—H2A	117.7 (13)
C13—C18—H18	119.6	C9—N2—H2A	116.6 (13)
C24—C19—C20	117.58 (19)	C25—N3—C26	117.12 (15)
C24—C19—C25	121.33 (17)	C25—N3—H3A	107.6 (15)
C20—C19—C25	120.94 (18)	C26—N3—H3A	106.1 (15)
C21—C20—C19	120.7 (2)	C28—N4—C27	127.07 (16)
C21—C20—H20	119.6	C28—N4—H4A	117.6 (14)
C19—C20—H20	119.6	C27—N4—H4A	114.3 (14)
C6—C1—C2—C3	-1.7 (3)	C24—C19—C25—N3	-137.32 (19)
C7—C1—C2—C3	176.56 (19)	C20—C19—C25—N3	47.3 (3)
C1—C2—C3—C4	0.6 (4)	C24—C19—C25—C29	101.6 (2)
C2—C3—C4—C5	1.0 (4)	C20—C19—C25—C29	-73.8 (2)
C3—C4—C5—C6	-1.5 (4)	N3—C26—C27—N4	-75.63 (19)
C2—C1—C6—C5	1.2 (3)	C31—C26—C27—N4	164.21 (17)
C7—C1—C6—C5	-177.0 (2)	N3—C26—C27—C30	164.19 (16)
C4—C5—C6—C1	0.3 (4)	C31—C26—C27—C30	44.0 (2)
C6—C1—C7—N1	-14.8 (2)	O2—C28—C29—C25	119.9 (2)
C2—C1—C7—N1	167.09 (17)	N4—C28—C29—C25	-59.8 (2)
C6—C1—C7—C11	107.5 (2)	N3—C25—C29—C28	81.2 (2)

C2—C1—C7—C11	-70.6 (2)	C19—C25—C29—C28	-157.99 (16)
N1—C8—C9—N2	79.72 (18)	N3—C26—C31—C36	137.3 (2)
C13—C8—C9—N2	-159.53 (15)	C27—C26—C31—C36	-100.6 (2)
N1—C8—C9—C12	-159.86 (16)	N3—C26—C31—C32	-46.1 (3)
C13—C8—C9—C12	-39.1 (2)	C27—C26—C31—C32	75.9 (2)
O1—C10—C11—C7	-113.7 (2)	C36—C31—C32—C33	0.6 (4)
N2—C10—C11—C7	65.3 (2)	C26—C31—C32—C33	-176.0 (2)
N1—C7—C11—C10	-79.9 (2)	C31—C32—C33—C34	0.9 (5)
C1—C7—C11—C10	158.55 (16)	C32—C33—C34—C35	-2.1 (6)
N1—C8—C13—C14	40.2 (2)	C33—C34—C35—C36	1.9 (7)
C9—C8—C13—C14	-82.4 (2)	C34—C35—C36—C31	-0.4 (5)
N1—C8—C13—C18	-140.06 (17)	C32—C31—C36—C35	-0.9 (4)
C9—C8—C13—C18	97.33 (19)	C26—C31—C36—C35	175.7 (2)
C18—C13—C14—C15	-0.1 (3)	C1—C7—N1—C8	-173.45 (15)
C8—C13—C14—C15	179.61 (17)	C11—C7—N1—C8	65.8 (2)
C13—C14—C15—C16	0.5 (3)	C13—C8—N1—C7	169.15 (15)
C14—C15—C16—C17	-0.5 (3)	C9—C8—N1—C7	-67.5 (2)
C15—C16—C17—C18	0.3 (3)	O1—C10—N2—C9	-179.67 (18)
C16—C17—C18—C13	0.1 (3)	C11—C10—N2—C9	1.4 (3)
C14—C13—C18—C17	-0.1 (3)	C12—C9—N2—C10	168.85 (18)
C8—C13—C18—C17	-179.88 (17)	C8—C9—N2—C10	-66.4 (2)
C24—C19—C20—C21	0.0 (4)	C19—C25—N3—C26	169.44 (16)
C25—C19—C20—C21	175.5 (2)	C29—C25—N3—C26	-70.8 (2)
C19—C20—C21—C22	-1.3 (5)	C31—C26—N3—C25	-169.93 (17)
C20—C21—C22—C23	1.5 (5)	C27—C26—N3—C25	68.8 (2)
C21—C22—C23—C24	-0.4 (4)	O2—C28—N4—C27	174.06 (18)
C20—C19—C24—C23	1.2 (3)	C29—C28—N4—C27	-6.2 (3)
C25—C19—C24—C23	-174.4 (2)	C30—C27—N4—C28	-170.8 (2)
C22—C23—C24—C19	-1.0 (4)	C26—C27—N4—C28	65.3 (2)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C6—H6...N1	0.93	2.45	2.797 (2)	102
N2—H2 <i>A</i> ...O2 ⁱ	0.90 (2)	2.02 (2)	2.911 (2)	170.2 (18)
N4—H4 <i>A</i> ...O1 ⁱ	0.87 (2)	2.03 (2)	2.884 (2)	167 (2)
N1—H1 <i>A</i> ...Cg1 ⁱⁱ	0.88 (2)	2.93 (2)	3.707 (2)	149.5 (2)
C18—H18...Cg5 ⁱⁱⁱ	0.93	2.95	3.872 (2)	171

Symmetry codes: (i) $-x, -y, -z+1$; (ii) $-x, -y, -z$; (iii) $x+1, y, z$.