Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

# 1-Dodecyloxy-4-nitrobenzene

#### Xi-Gui Yue

Affiliation: Alan G. MacDiarmid Institute, Jilin University, Changchun 130012, People's Republic of China Correspondence e-mail: yuexigui@jlu.edu.cn

Received 12 October 2009; accepted 2 November 2009

Key indicators: single-crystal X-ray study; T = 291 K; mean  $\sigma$ (C–C) = 0.005 Å; R factor = 0.068; wR factor = 0.180; data-to-parameter ratio = 15.4.

The asymmetric unit of the title compound,  $C_{18}H_{29}NO_3$ , contains two independent molecules. The benzene ring and the mean plane of the alkyl unit form dihedral angles of 83.69 (12) and 77.14 (11)° in the two molecules. In the crystal structure, weak C-H···O hydrogen bonds link molecules into double-layer ribbons extending in [110].

#### **Related literature**

For the structure of a related nitrobenzene derivative, see Yue (2009).



**Experimental** 

Crystal data

b = 16.064 (7)  Å
c = 21.390 (12)  Å
$\alpha = 72.190 \ (15)^{\circ}$
$\beta = 87.290 \ (18)^{\circ}$

 $\gamma = 80.240 (16)^{\circ}$   $V = 1810.1 (15) \text{ Å}^3$  Z = 4Mo  $K\alpha$  radiation

#### Data collection

Rigaku R-AXIS RAPID diffractometer Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995)  $T_{\rm min} = 0.983, T_{\rm max} = 0.984$ 

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.068$  $wR(F^2) = 0.180$ S = 0.996158 reflections

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C3-H3\cdots O4^{i}$	0.93	2.69	3.329 (4)	127
C6−H6···O2 <sup>ii</sup>	0.93	2.60	3.375 (4)	141
C20−H20···O5 <sup>iii</sup>	0.93	2.52	3.372 (5)	152
		-		

Symmetry codes: (i) x - 1, y, z; (ii) -x + 3, -y, -z + 2; (iii) -x + 4, -y - 1, -z + 2.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSC, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV2628).

#### References

Higashi, T. (1995). ABSCOR. Rigaku Corporation, Tokyo, Japan.
Rigaku (1998). RAPID-AUTO. Rigaku Corporation, Tokyo, Japan.
Rigaku/MSC (2002). CrystalStructure. Rigaku/MSC Inc., The Woodlands, Texas, USA.
Sheldrick, G. M. (2008). Acta Cryst. A64, 112–122.
Spek, A. L. (2009). Acta Cryst. D65, 148–155.
Yue, X.-G. (2009). Acta Cryst. E65, o2627.

 $\mu = 0.08 \text{ mm}^{-1}$  T = 291 K $0.23 \times 0.23 \times 0.21 \text{ mm}$ 

13636 measured reflections 6158 independent reflections n 2548 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.064$ 

> 399 parameters H-atom parameters constrained  $\Delta \rho_{max} = 0.15$  e Å<sup>-3</sup>  $\Delta \rho_{min} = -0.16$  e Å<sup>-3</sup>

# supporting information

Acta Cryst. (2009). E65, o3034 [doi:10.1107/S1600536809045966]

# 1-Dodecyloxy-4-nitrobenzene

# Xi-Gui Yue

# S1. Comment

Nitrobenzene and its derivatives are of great interest for their various applications. Recently, we reported the crystal structure of 1-decyloxy-4-nitrobenzene (Yue, 2009). As an extension of our work on the structure characterizations of nitrobenzene derivatives, we report herein the crystal structure of the title compound.

The title compound, as shown in Fig. 1, comprises two independent molecules in the asymmetric unit. Two benzene rings of the two molecules form a dihedral angle of 48.12 (13) °. Weak C—H···O hydrogen bonds (Table 1) link molecules into ribbons extended in direction [110].

# S2. Experimental

4-Nitrophenol (0.14 g, 1 mmol) and dodecyl iodide (0.30 g, 1 mmol) were dissovled in 15 ml of acetone. The sodium hydroxide solution (10 ml, 8%) was added into the above solution. The resultant mixture was heated for 2 h under refluxing and then the solution was cooled to room temperaure in an ice bath with stirring. The colourless crystals suitable for single-crystal analysis were obtained by recrystallization from 95% ethanol.

## **S3. Refinement**

H atoms were placed in calculated positions (C—H 0.93-0.97 Å) and were included in the refinement in the riding model with  $U_{iso}(H) = 1.2$  or 1.5  $U_{eq}(C)$ .



#### Figure 1

Two independent molecules of the title compound showing the atom numbering and 30% probability displacement ellipsoids.

## 1-Dodecyloxy-4-nitrobenzene

Crystal data

 $C_{18}H_{29}NO_3$   $M_r = 307.42$ Triclinic, *P*1 Hall symbol: -P 1 a = 5.615 (3) Å b = 16.064 (7) Å c = 21.390 (12) Å  $a = 72.190 (15)^{\circ}$   $\beta = 87.290 (18)^{\circ}$   $\gamma = 80.240 (16)^{\circ}$  $V = 1810.1 (15) \text{ Å}^{3}$ 

#### Data collection

Rigaku R-AXIS RAPID	13636 measured reflections
diffractometer	6158 independent reflections
Radiation source: fine-focus sealed tube	2548 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\rm int} = 0.064$
$\omega$ scans	$\theta_{\rm max} = 25.0^{\circ}, \ \theta_{\rm min} = 3.0^{\circ}$
Absorption correction: multi-scan	$h = -6 \rightarrow 5$
(ABSCOR; Higashi, 1995)	$k = -18 \rightarrow 19$
$T_{\min} = 0.983, \ T_{\max} = 0.984$	$l = -25 \rightarrow 25$

Z = 4 F(000) = 672  $D_x = 1.128 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ Å}$ Cell parameters from 8462 reflections  $\theta = 3.0-25.0^{\circ}$   $\mu = 0.08 \text{ mm}^{-1}$  T = 291 KBlock, colourless  $0.23 \times 0.23 \times 0.21 \text{ mm}$  Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.068$	Hydrogen site location: inferred from
$wR(F^2) = 0.180$	neighbouring sites
S = 0.99	H-atom parameters constrained
6158 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0744P)^2]$
399 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{\rm max} < 0.001$
Primary atom site location: structure-invariant	$\Delta  ho_{ m max} = 0.15 \ { m e} \ { m \AA}^{-3}$
direct methods	$\Delta \rho_{\rm min} = -0.16 \text{ e} \text{ Å}^{-3}$

## Special details

Experimental. (See detailed section in the paper)

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes. **Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on  $F^2$ , the second se

conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$ are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(\hat{A}^2)$ 

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
C1	1.1633 (6)	-0.1114 (2)	0.98321 (16)	0.0464 (9)	
C2	0.9771 (7)	-0.1433 (2)	0.96353 (19)	0.0577 (11)	
H2	0.9392	-0.1982	0.9877	0.069*	
C3	0.8480 (7)	-0.0939 (2)	0.90832 (17)	0.0544 (10)	
H3	0.7243	-0.1159	0.8943	0.065*	
C4	0.9005 (6)	-0.0111 (2)	0.87311 (16)	0.0482 (9)	
C5	1.0903 (6)	0.0201 (2)	0.89214 (17)	0.0513 (10)	
Н5	1.1290	0.0748	0.8678	0.062*	
C6	1.2222 (7)	-0.0301 (2)	0.94732 (17)	0.0548 (10)	
H6	1.3504	-0.0094	0.9605	0.066*	
C7	0.7796 (7)	0.1225 (2)	0.78507 (17)	0.0593 (11)	
H7A	0.9356	0.1240	0.7636	0.071*	
H7B	0.7679	0.1582	0.8147	0.071*	
C8	0.5811 (7)	0.1577 (2)	0.73523 (17)	0.0594 (11)	
H8A	0.5787	0.2209	0.7158	0.071*	
H8B	0.4283	0.1498	0.7575	0.071*	
C9	0.5996 (7)	0.1151 (2)	0.68065 (17)	0.0576 (11)	
H9A	0.6023	0.0519	0.6998	0.069*	
H9B	0.7511	0.1236	0.6578	0.069*	
C10	0.3935 (7)	0.1523 (2)	0.63141 (17)	0.0582 (10)	
H10A	0.2426	0.1427	0.6543	0.070*	
H10B	0.3886	0.2158	0.6132	0.070*	
C11	0.4119 (7)	0.1121 (2)	0.57567 (18)	0.0622 (11)	
H11A	0.5619	0.1225	0.5524	0.075*	

H11B	0.4190	0.0486	0.5939	0.075*
C12	0.2058 (7)	0.1482 (2)	0.52735 (18)	0.0605 (11)
H12A	0.1992	0.2118	0.5092	0.073*
H12B	0.0561	0.1381	0.5508	0.073*
C13	0.2195 (7)	0.1091 (2)	0.47143 (19)	0.0650(11)
H13A	0.3689	0.1194	0.4479	0.078*
H13B	0.2270	0.0455	0.4896	0.078*
C14	0.0136(7)	0.1446 (3)	0.42319 (19)	0.0701 (12)
H14A	0.0065	0.2081	0.4048	0.084*
H14B	-0.1360	0.1344	0.4467	0.084*
C15	0.0286 (8)	0.1047 (3)	0.3678 (2)	0.0761 (13)
H15A	0.0399	0.0410	0.3863	0.091*
H15B	0.1767	0.1160	0.3438	0.091*
C16	-0.1797 (8)	0.1382 (3)	0.3197 (2)	0.0740 (13)
H16A	-0.3277	0.1273	0.3439	0.089*
H16B	-0.1902	0.2019	0.3011	0.089*
C17	-0.1683 (9)	0.0991 (4)	0.2650(2)	0.1038 (17)
H17A	-0.1457	0.0351	0.2835	0.125*
H17B	-0.0266	0.1138	0.2389	0.125*
C18	-0.3825 (8)	0.1275 (3)	0.2203 (2)	0.0944 (16)
H18A	-0.5184	0.1035	0.2434	0.142*
H18B	-0.3471	0.1062	0.1830	0.142*
H18C	-0.4200	0.1910	0.2058	0.142*
C19	1.5593 (7)	-0.3938(2)	0.92787 (16)	0.0513 (10)
C20	1.6746 (7)	-0.4774 (2)	0.93077 (16)	0.0531 (10)
H20	1.8218	-0.5002	0.9526	0.064*
C21	1.5700 (7)	-0.5274 (2)	0.90099 (17)	0.0531 (10)
H21	1.6470	-0.5839	0.9022	0.064*
C22	1.3504 (7)	-0.4930 (2)	0.86934 (16)	0.0513 (10)
C23	1.2366 (7)	-0.4080 (2)	0.86619 (18)	0.0584 (11)
H23	1.0905	-0.3846	0.8438	0.070*
C24	1.3406 (7)	-0.3587 (2)	0.89620 (17)	0.0561 (10)
H24	1.2642	-0.3022	0.8951	0.067*
C25	1.3257 (7)	-0.6272 (2)	0.84394 (18)	0.0613 (11)
H25A	1.3344	-0.6628	0.8897	0.074*
H25B	1.4870	-0.6319	0.8256	0.074*
C26	1.1596 (7)	-0.6589(2)	0.80708 (17)	0.0622 (11)
H26A	0.9977	-0.6498	0.8244	0.075*
H26B	1.2090	-0.7221	0.8150	0.075*
C27	1.1516(7)	-0.6140(2)	0.73391 (17)	0.0569 (11)
H27A	1.3134	-0.6227	0.7165	0.068*
H27B	1.1005	-0.5508	0.7258	0.068*
C28	0.9838 (7)	-0.6477 (2)	0.69766 (17)	0.0577 (11)
H28A	1.0324	-0.7112	0.7073	0.069*
H28B	0.8217	-0.6375	0.7147	0.069*
C29	0.9752 (7)	-0.6064 (2)	0.62407 (17)	0.0572 (11)
H29A	1.1368	-0.6165	0.6068	0.069*
H29B	0.9256	-0.5428	0.6142	0.069*

C30	0.8054 (7)	-0.6419 (2)	0.58938 (17)	0.0588 (11)
H30A	0.8532	-0.7056	0.6001	0.071*
H30B	0.6435	-0.6307	0.6062	0.071*
C31	0.7980 (7)	-0.6027 (2)	0.51597 (18)	0.0603 (11)
H31A	0.9597	-0.6138	0.4990	0.072*
H31B	0.7495	-0.5391	0.5051	0.072*
C32	0.6273 (7)	-0.6392 (2)	0.48198 (17)	0.0577 (11)
H32A	0.6749	-0.7030	0.4935	0.069*
H32B	0.4657	-0.6276	0.4988	0.069*
C33	0.6188 (7)	-0.6017 (2)	0.40820 (17)	0.0606 (11)
H33A	0.7798	-0.6141	0.3912	0.073*
H33B	0.5733	-0.5378	0.3966	0.073*
C34	0.4454 (7)	-0.6375 (2)	0.37487 (18)	0.0613 (11)
H34A	0.4905	-0.7014	0.3868	0.074*
H34B	0.2844	-0.6249	0.3918	0.074*
C35	0.4363 (8)	-0.6012 (3)	0.30168 (19)	0.0780 (13)
H35A	0.5972	-0.6137	0.2847	0.094*
H35B	0.3903	-0.5374	0.2896	0.094*
C36	0.2635 (8)	-0.6377 (3)	0.2690 (2)	0.0848 (14)
H36A	0.1016	-0.6221	0.2830	0.127*
H36B	0.3056	-0.7011	0.2810	0.127*
H36C	0.2735	-0.6132	0.2222	0.127*
N1	1.2958 (6)	-0.1617 (2)	1.04342 (15)	0.0629 (9)
N2	1.6664 (7)	-0.3418 (3)	0.96029 (16)	0.0653 (10)
01	1.2231 (5)	-0.22878 (19)	1.07737 (14)	0.0902 (10)
O2	1.4686 (5)	-0.13502 (17)	1.05870 (13)	0.0791 (9)
O3	0.7542 (5)	0.03322 (15)	0.82063 (12)	0.0614 (7)
O4	1.5594 (6)	-0.2675 (2)	0.95704 (17)	0.1001 (12)
05	1.8548 (6)	-0.37346 (19)	0.99037 (15)	0.0833 (10)
O6	1.2305 (4)	-0.53613 (15)	0.83828 (12)	0.0608 (7)

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.049 (2)	0.048 (2)	0.040 (2)	-0.0047 (18)	-0.0020 (18)	-0.0123 (16)
C2	0.058 (3)	0.054 (2)	0.063 (3)	-0.018 (2)	0.002 (2)	-0.0157 (19)
C3	0.056 (3)	0.056 (2)	0.052 (2)	-0.016 (2)	-0.012 (2)	-0.0131 (18)
C4	0.054 (2)	0.054 (2)	0.039 (2)	-0.0096 (19)	-0.0050 (19)	-0.0157 (17)
C5	0.057 (3)	0.054 (2)	0.045 (2)	-0.021 (2)	-0.003 (2)	-0.0119 (17)
C6	0.052 (2)	0.065 (2)	0.054 (3)	-0.015 (2)	-0.001 (2)	-0.0240 (19)
C7	0.079 (3)	0.054 (2)	0.045 (2)	-0.017 (2)	-0.007 (2)	-0.0109 (17)
C8	0.071 (3)	0.056 (2)	0.047 (2)	-0.008 (2)	-0.011 (2)	-0.0102 (18)
C9	0.066 (3)	0.064 (2)	0.045 (2)	-0.012 (2)	-0.005 (2)	-0.0184 (18)
C10	0.057 (3)	0.066 (2)	0.051 (2)	-0.005 (2)	-0.005 (2)	-0.0178 (19)
C11	0.068 (3)	0.070 (2)	0.051 (2)	-0.011 (2)	-0.007 (2)	-0.0208 (19)
C12	0.058 (3)	0.075 (2)	0.047 (2)	-0.004 (2)	-0.006 (2)	-0.0195 (19)
C13	0.058 (3)	0.082 (3)	0.059 (3)	0.000 (2)	-0.012 (2)	-0.032 (2)
C14	0.069 (3)	0.086 (3)	0.060 (3)	-0.007 (2)	-0.009 (2)	-0.032 (2)

C15	0.075 (3)	0.098 (3)	0.062 (3)	-0.010 (3)	-0.015 (2)	-0.035 (2)
C16	0.080 (3)	0.086 (3)	0.060 (3)	-0.009 (2)	-0.009 (2)	-0.029 (2)
C17	0.092 (4)	0.157 (4)	0.083 (4)	-0.005 (3)	-0.019 (3)	-0.071 (3)
C18	0.102 (4)	0.119 (4)	0.073 (3)	-0.023 (3)	-0.021 (3)	-0.039 (3)
C19	0.063 (3)	0.055 (2)	0.042 (2)	-0.022 (2)	0.003 (2)	-0.0170 (17)
C20	0.052 (2)	0.066 (2)	0.042 (2)	-0.012 (2)	-0.0028 (18)	-0.0159 (18)
C21	0.053 (3)	0.056 (2)	0.051 (2)	-0.0061 (19)	-0.003 (2)	-0.0186 (18)
C22	0.056 (3)	0.060 (2)	0.043 (2)	-0.016 (2)	0.0008 (19)	-0.0193 (18)
C23	0.055 (3)	0.060 (2)	0.061 (3)	-0.004 (2)	-0.006 (2)	-0.0212 (19)
C24	0.064 (3)	0.049 (2)	0.059 (3)	-0.009 (2)	0.004 (2)	-0.0223 (18)
C25	0.077 (3)	0.056 (2)	0.053 (3)	-0.007 (2)	-0.009 (2)	-0.0200 (18)
C26	0.078 (3)	0.063 (2)	0.053 (3)	-0.021 (2)	-0.003 (2)	-0.0233 (19)
C27	0.066 (3)	0.061 (2)	0.050 (2)	-0.019 (2)	-0.003 (2)	-0.0219 (18)
C28	0.061 (3)	0.062 (2)	0.055 (3)	-0.018 (2)	0.001 (2)	-0.0221 (18)
C29	0.063 (3)	0.065 (2)	0.049 (2)	-0.018 (2)	-0.004 (2)	-0.0201 (19)
C30	0.066 (3)	0.062 (2)	0.054 (3)	-0.020 (2)	-0.006 (2)	-0.0199 (18)
C31	0.060 (3)	0.065 (2)	0.056 (3)	-0.013 (2)	-0.009 (2)	-0.0158 (19)
C32	0.056 (3)	0.064 (2)	0.054 (3)	-0.012 (2)	-0.002 (2)	-0.0179 (18)
C33	0.065 (3)	0.071 (2)	0.046 (2)	-0.015 (2)	-0.005 (2)	-0.0155 (19)
C34	0.061 (3)	0.074 (2)	0.052 (3)	-0.013 (2)	-0.003 (2)	-0.0228 (19)
C35	0.076 (3)	0.110 (3)	0.051 (3)	-0.025 (3)	-0.008 (2)	-0.021 (2)
C36	0.074 (3)	0.115 (3)	0.067 (3)	-0.015 (3)	-0.019 (3)	-0.028 (3)
N1	0.066 (2)	0.070 (2)	0.051 (2)	-0.0071 (19)	-0.0105 (18)	-0.0163 (17)
N2	0.072 (3)	0.074 (2)	0.060 (2)	-0.031 (2)	0.004 (2)	-0.0266 (19)
01	0.093 (2)	0.0818 (19)	0.077 (2)	-0.0241 (18)	-0.0156 (18)	0.0116 (16)
O2	0.080 (2)	0.0864 (19)	0.069 (2)	-0.0167 (17)	-0.0267 (17)	-0.0151 (15)
O3	0.0731 (19)	0.0588 (15)	0.0521 (16)	-0.0209 (14)	-0.0134 (14)	-0.0092 (12)
04	0.107 (3)	0.076 (2)	0.136 (3)	-0.015 (2)	-0.021 (2)	-0.056 (2)
O5	0.077 (2)	0.098 (2)	0.090 (2)	-0.0213 (18)	-0.0191 (19)	-0.0434 (17)
O6	0.0667 (18)	0.0597 (15)	0.0628 (17)	-0.0066 (14)	-0.0164 (14)	-0.0282 (12)

Geometric parameters (Å, °)

C1—C2	1.373 (4)	C19—N2	1.451 (4)
C1—C6	1.384 (4)	C20—C21	1.380 (4)
C1—N1	1.454 (4)	C20—H20	0.9300
С2—С3	1.365 (4)	C21—C22	1.379 (5)
С2—Н2	0.9300	C21—H21	0.9300
C3—C4	1.387 (4)	C22—O6	1.362 (4)
С3—Н3	0.9300	C22—C23	1.389 (5)
C4—O3	1.357 (4)	C23—C24	1.373 (4)
C4—C5	1.378 (4)	С23—Н23	0.9300
C5—C6	1.374 (4)	C24—H24	0.9300
С5—Н5	0.9300	C25—O6	1.440 (4)
С6—Н6	0.9300	C25—C26	1.492 (4)
С7—О3	1.432 (3)	C25—H25A	0.9700
С7—С8	1.492 (4)	C25—H25B	0.9700
С7—Н7А	0.9700	C26—C27	1.509 (5)

С7—Н7В	0.9700	C26—H26A	0.9700
C8—C9	1.515 (5)	C26—H26B	0.9700
C8—H8A	0.9700	C27—C28	1.510 (4)
C8—H8B	0.9700	С27—Н27А	0.9700
C9—C10	1.517 (4)	С27—Н27В	0.9700
С9—Н9А	0.9700	C28—C29	1.509 (4)
С9—Н9В	0.9700	C28—H28A	0.9700
C10—C11	1.512 (5)	C28—H28B	0.9700
C10—H10A	0.9700	C29—C30	1.515 (4)
C10—H10B	0.9700	С29—Н29А	0.9700
C11—C12	1.504 (5)	C29—H29B	0.9700
C11—H11A	0.9700	C30—C31	1.502 (5)
C11—H11B	0.9700	C30—H30A	0.9700
C12—C13	1.506 (5)	C30—H30B	0.9700
C12—H12A	0.9700	C31—C32	1.519 (4)
C12—H12B	0.9700	C31—H31A	0.9700
C13—C14	1.502 (5)	C31—H31B	0.9700
C13—H13A	0.9700	C32—C33	1.507 (5)
C13—H13B	0.9700	С32—Н32А	0.9700
C14—C15	1.503 (5)	C32—H32B	0.9700
C14—H14A	0.9700	C33—C34	1.514 (4)
C14—H14B	0.9700	С33—Н33А	0.9700
C15—C16	1.508 (5)	С33—Н33В	0.9700
C15—H15A	0.9700	C34—C35	1.494 (5)
C15—H15B	0.9700	C34—H34A	0.9700
C16—C17	1.483 (6)	C34—H34B	0.9700
C16—H16A	0.9700	C35—C36	1.510 (5)
C16—H16B	0.9700	С35—Н35А	0.9700
C17—C18	1.491 (5)	С35—Н35В	0.9700
C17—H17A	0.9700	С36—Н36А	0.9600
С17—Н17В	0.9700	С36—Н36В	0.9600
C18—H18A	0.9600	С36—Н36С	0.9600
C18—H18B	0.9600	N1—O2	1.220 (3)
C18—H18C	0.9600	N101	1.224 (3)
C19—C20	1.373 (5)	N2—O5	1.214 (4)
C19—C24	1.377 (5)	N2—O4	1.225 (4)
C2—C1—C6	120.6 (3)	С19—С20—Н20	120.3
C2-C1-N1	120.2 (3)	С21—С20—Н20	120.3
C6-C1-N1	119.2 (3)	C22—C21—C20	119.6 (4)
C3—C2—C1	119.6 (3)	C22—C21—H21	120.2
С3—С2—Н2	120.2	C20—C21—H21	120.2
C1—C2—H2	120.2	O6—C22—C21	124.7 (3)
C2—C3—C4	120.2 (3)	O6—C22—C23	114.8 (3)
С2—С3—Н3	119.9	C21—C22—C23	120.5 (3)
C4—C3—H3	119.9	C24—C23—C22	119.7 (4)
O3—C4—C5	125.0 (3)	C24—C23—H23	120.1
O3—C4—C3	114.9 (3)	C22—C23—H23	120.1

C5—C4—C3	120.1 (3)	C23—C24—C19	119.3 (4)
C6—C5—C4	119.7 (3)	C23—C24—H24	120.4
С6—С5—Н5	120.2	C19—C24—H24	120.4
С4—С5—Н5	120.2	O6—C25—C26	107.4 (3)
C5—C6—C1	119.7 (3)	O6—C25—H25A	110.2
С5—С6—Н6	120.1	C26—C25—H25A	110.2
C1—C6—H6	120.1	06—C25—H25B	110.2
03 - 07 - 08	1074(3)	C26—C25—H25B	110.2
$O_3 - C_7 - H_7 A$	110.2	$H_{25}^{-}$ $H_{$	108.5
C8 - C7 - H7A	110.2	$C_{25}$ $C_{25}$ $C_{25}$ $C_{25}$ $C_{25}$ $C_{25}$ $C_{25}$ $C_{27}$	114.7(3)
$O_3 - C_7 - H_7B$	110.2	$C_{25} = C_{26} = H_{26A}$	108.6
$C_{8}$ $C_{7}$ $H_{7}$ $H_{7$	110.2	$C_{23} = C_{20} = H_{20} A$	108.6
	10.2	$C_{2}^{-}$ $C_{2$	108.0
$\Pi/A = C/ = \Pi/B$	100.3	$C_{23} = C_{20} = H_{20B}$	108.0
$C_{1} = C_{0} = C_{1}$	113.1 (3)	$C_2/-C_20-H_20B$	108.0
$C = C \delta = H \delta A$	108.5	H20A—C20—H20B	107.6
C9—C8—H8A	108.5	$C_{26} = C_{27} = C_{28}$	113.6 (3)
C/—C8—H8B	108.5	C26—C27—H27A	108.8
С9—С8—Н8В	108.5	C28—C27—H27A	108.8
H8A—C8—H8B	107.5	С26—С27—Н27В	108.8
C8—C9—C10	113.3 (3)	С28—С27—Н27В	108.8
С8—С9—Н9А	108.9	H27A—C27—H27B	107.7
С10—С9—Н9А	108.9	C29—C28—C27	115.5 (3)
С8—С9—Н9В	108.9	C29—C28—H28A	108.4
С10—С9—Н9В	108.9	C27—C28—H28A	108.4
H9A—C9—H9B	107.7	C29—C28—H28B	108.4
С11—С10—С9	114.1 (3)	C27—C28—H28B	108.4
C11-C10-H10A	108.7	H28A—C28—H28B	107.5
C9-C10-H10A	108.7	C28—C29—C30	114.0 (3)
C11—C10—H10B	108.7	C28—C29—H29A	108.8
C9-C10-H10B	108.7	C30—C29—H29A	108.8
H10A—C10—H10B	107.6	C28—C29—H29B	108.8
C12—C11—C10	114.1 (3)	C30—C29—H29B	108.8
C12—C11—H11A	108.7	H29A—C29—H29B	107.7
C10-C11-H11A	108.7	C31—C30—C29	114.7 (3)
C12—C11—H11B	108.7	C31—C30—H30A	108.6
C10-C11-H11B	108.7	C29—C30—H30A	108.6
H11A-C11-H11B	107.6	$C_{31}$ $C_{30}$ $H_{30B}$	108.6
$C_{11}$ $C_{12}$ $C_{13}$	115 1 (3)	$C_{29}$ $C_{30}$ $H_{30B}$	108.6
$C_{11} - C_{12} - H_{12A}$	108.5	H30A_C30_H30B	103.6
C13 $C12$ $H12A$	108.5	$C_{30}$ $C_{31}$ $C_{32}$	107.0 114 0 (3)
$C_{11}$ $C_{12}$ $H_{12R}$	108.5	$C_{30} = C_{31} = C_{32}$	108.7
$C_{12}$ $C_{12}$ $H_{12}$ $H_{12}$	108.5	$C_{30} = C_{31} = H_{31A}$	108.7
$H_{12A} = C_{12} = H_{12B}$	108.5	$C_{20} = C_{21} = H_{21} R$	108.7
$\frac{1112}{112} = \frac{112}{112} =$	107.3	$C_{20} = C_{21} = C_{21} = C_{21}$	100.7
C14 - C13 - C12	113.4 (4)		108.7
C12 C12 H12A	108.4	$H_{21} = H_{21} = H$	10/.0
U12— $U13$ — $H13A$	108.4	$C_{22} = C_{22} = U_{22}$	115.1 (3)
CI4—CI3—HI3B	108.4	$C_{33}$ — $C_{32}$ — $H_{32A}$	108.5
C12-C13-H13B	108.4	C31—C32—H32A	108.5

H13A—C13—H13B	107.5	C33—C32—H32B	108.5
C13—C14—C15	114.9 (4)	С31—С32—Н32В	108.5
C13—C14—H14A	108.5	H32A—C32—H32B	107.5
C15—C14—H14A	108.5	C32—C33—C34	114.7 (3)
C13—C14—H14B	108.5	С32—С33—Н33А	108.6
C15—C14—H14B	108.5	С34—С33—Н33А	108.6
H14A—C14—H14B	107.5	С32—С33—Н33В	108.6
C14—C15—C16	115.5 (4)	С34—С33—Н33В	108.6
C14—C15—H15A	108.4	H33A—C33—H33B	107.6
C16—C15—H15A	108.4	C35—C34—C33	115.2 (3)
C14—C15—H15B	108.4	С35—С34—Н34А	108.5
C16—C15—H15B	108.4	С33—С34—Н34А	108.5
H15A—C15—H15B	107.5	С35—С34—Н34В	108.5
C17—C16—C15	116.2 (4)	C33—C34—H34B	108.5
C17—C16—H16A	108.2	H34A—C34—H34B	107.5
C15—C16—H16A	108.2	C34—C35—C36	114.7 (3)
C17—C16—H16B	108.2	С34—С35—Н35А	108.6
C15—C16—H16B	108.2	С36—С35—Н35А	108.6
H16A—C16—H16B	107.4	С34—С35—Н35В	108.6
C16—C17—C18	116.1 (4)	С36—С35—Н35В	108.6
C16—C17—H17A	108.3	H35A—C35—H35B	107.6
C18—C17—H17A	108.3	С35—С36—Н36А	109.5
C16—C17—H17B	108.3	С35—С36—Н36В	109.5
C18—C17—H17B	108.3	H36A—C36—H36B	109.5
H17A—C17—H17B	107.4	С35—С36—Н36С	109.5
C17—C18—H18A	109.5	H36A—C36—H36C	109.5
C17—C18—H18B	109.5	H36B—C36—H36C	109.5
H18A—C18—H18B	109.5	O2—N1—O1	123.0 (3)
C17—C18—H18C	109.5	O2—N1—C1	119.6 (3)
H18A—C18—H18C	109.5	O1—N1—C1	117.4 (3)
H18B—C18—H18C	109.5	O5—N2—O4	122.5 (3)
C20—C19—C24	121.5 (3)	O5—N2—C19	119.4 (4)
C20—C19—N2	119.5 (4)	O4—N2—C19	118.1 (4)
C24—C19—N2	118.9 (4)	C4—O3—C7	118.9 (2)
C19—C20—C21	119.4 (3)	C22—O6—C25	118.2 (3)

# Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D···A	<i>D</i> —H··· <i>A</i>
C3—H3…O4 <sup>i</sup>	0.93	2.69	3.329 (4)	127
С6—Н6…О2 <sup>іі</sup>	0.93	2.60	3.375 (4)	141
C20—H20…O5 <sup>iii</sup>	0.93	2.52	3.372 (5)	152

Symmetry codes: (i) *x*-1, *y*, *z*; (ii) -*x*+3, -*y*, -*z*+2; (iii) -*x*+4, -*y*-1, -*z*+2.