

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

N-Butyladamantane-1-carboxamide

Weiwei SiMa

Ordered Matter Science Research Center, College of Chemistry and Chemical Engineering, Southeast University, Nanjing 210096, People's Republic of China
Correspondence e-mail: nysima@126.com

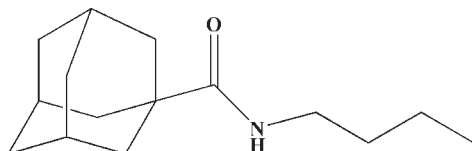
Received 25 August 2009; accepted 12 September 2009

Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; R factor = 0.086; wR factor = 0.246; data-to-parameter ratio = 21.0.

In the crystal of the title compound, $\text{C}_{15}\text{H}_{25}\text{NO}$, the molecules are linked into chains propagating in $[001]$ by intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds.

Related literature

For a related structure, see: SiMa (2009). For further synthetic details, see: Tadashi & Sasaki (1969).



Experimental

Crystal data

$\text{C}_{15}\text{H}_{25}\text{NO}$
 $M_r = 235.36$
Monoclinic, $C2/c$
 $a = 32.257$ (7) Å
 $b = 9.4353$ (19) Å

$c = 9.5328$ (19) Å
 $\beta = 101.69$ (3)°
 $V = 2841.1$ (10) Å³
 $Z = 8$
Mo $K\alpha$ radiation

$\mu = 0.07$ mm⁻¹
 $T = 298$ K

0.20 × 0.20 × 0.20 mm

Data collection

Rigaku SCXmini diffractometer
Absorption correction: none
14458 measured reflections

3260 independent reflections
1544 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.077$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.086$
 $wR(F^2) = 0.246$
 $S = 1.04$
3260 reflections

155 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.36$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.32$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1C}\cdots\text{O1}^i$	0.99	1.96	2.896 (3)	158

Symmetry code: (i) $x, -y, z - \frac{1}{2}$.

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

The author is grateful to the starter fund of Southeast University for financial support to buy the X-ray diffractometer.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5069).

References

- Rigaku (2005). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
SiMa, W. (2009). *Acta Cryst.* **E65**, o2452.
Tadashi, X. & Sasaki, Y. (1969). *Bull. Chem. Soc. Jpn.* **42**, 1617–1621.

supporting information

Acta Cryst. (2009). E65, o2492 [doi:10.1107/S1600536809036897]

N-Butyladamantane-1-carboxamide

Weiwei SiMa

S1. Experimental

A solution of freshly prepared 1-adamantane carbonyl chloride (1 mmol, prepared by refluxing 1-adamantane carboxylic acid with 3M excess of SOCl_2) in dry CH_2Cl_2 was added dropwise to a well stirred and ice-cooled solution of butanamine (1 mmol) and triethylamine (2 mmol) in the same solvent. After 24 h of stirring at room temperature, the solvents were removed *in vacuo* and the residue was recrystallized from methanol. Colourless prisms of (I) were obtained in 80% yield (Tadashi Sasaki *et al.*, 1969).

S2. Refinement

Positional parameters of all the H atoms were calculated geometrically and were allowed to ride on the N or C atoms to which they are bonded, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N,C})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

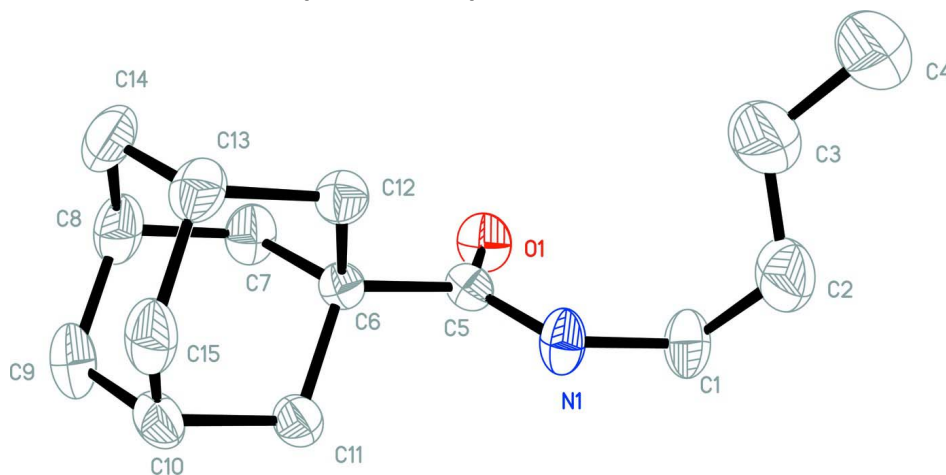


Figure 1

The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level, and all H atoms have been omitted for clarity.

N-Butyladamantane-1-carboxamide

Crystal data

$\text{C}_{15}\text{H}_{25}\text{NO}$

$M_r = 235.36$

Monoclinic, $C2/c$

Hall symbol: $-C 2yc$

$a = 32.257 (7) \text{ \AA}$

$b = 9.4353 (19) \text{ \AA}$

$c = 9.5328 (19) \text{ \AA}$

$\beta = 101.69 (3)^\circ$

$V = 2841.1 (10) \text{ \AA}^3$

$Z = 8$

$F(000) = 1040$

$D_x = 1.100 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 4730 reflections

$\theta = 3.0\text{--}27.8^\circ$
 $\mu = 0.07\text{ mm}^{-1}$
 $T = 298\text{ K}$

Prism, colourless
 $0.20 \times 0.20 \times 0.20\text{ mm}$

Data collection

Rigaku SCXmini
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 Detector resolution: $13.6612\text{ pixels mm}^{-1}$
 ω scans
 14458 measured reflections

3260 independent reflections
 1544 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.077$
 $\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 3.0^\circ$
 $h = -41 \rightarrow 41$
 $k = -12 \rightarrow 12$
 $l = -12 \rightarrow 12$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.086$
 $wR(F^2) = 0.246$
 $S = 1.04$
 3260 reflections
 155 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0977P)^2 + 2.030P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.36\text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.32\text{ e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.30657 (10)	-0.1173 (4)	0.4805 (4)	0.0671 (10)
H1A	0.3025	-0.1014	0.5774	0.080*
H1B	0.3146	-0.2157	0.4733	0.080*
C2	0.26562 (13)	-0.0911 (5)	0.3781 (5)	0.0935 (13)
H2A	0.2702	-0.1038	0.2813	0.112*
H2B	0.2453	-0.1622	0.3941	0.112*
C3	0.24721 (15)	0.0484 (6)	0.3883 (7)	0.143 (2)
H3A	0.2643	0.1164	0.3488	0.172*
H3B	0.2502	0.0705	0.4892	0.172*
C4	0.20314 (15)	0.0729 (6)	0.3200 (7)	0.152 (2)
H4A	0.1997	0.0609	0.2183	0.229*
H4B	0.1952	0.1677	0.3403	0.229*
H4C	0.1854	0.0065	0.3564	0.229*
C5	0.36027 (8)	0.0693 (3)	0.5470 (3)	0.0431 (7)

C6	0.39355 (8)	0.1604 (3)	0.4964 (3)	0.0411 (7)
C7	0.41628 (10)	0.2560 (4)	0.6189 (3)	0.0615 (9)
H7A	0.3957	0.3154	0.6524	0.074*
H7B	0.4303	0.1979	0.6984	0.074*
C8	0.44885 (11)	0.3491 (4)	0.5667 (4)	0.0715 (10)
H8	0.4630	0.4095	0.6456	0.086*
C9	0.48163 (10)	0.2564 (4)	0.5166 (4)	0.0795 (11)
H9A	0.5025	0.3155	0.4844	0.095*
H9B	0.4961	0.1981	0.5952	0.095*
C10	0.45958 (10)	0.1630 (4)	0.3948 (4)	0.0614 (9)
H10	0.4806	0.1031	0.3624	0.074*
C11	0.42699 (9)	0.0688 (3)	0.4458 (3)	0.0525 (8)
H11A	0.4411	0.0091	0.5238	0.063*
H11B	0.4134	0.0078	0.3680	0.063*
C12	0.37184 (9)	0.2551 (3)	0.3720 (3)	0.0513 (8)
H12A	0.3576	0.1965	0.2932	0.062*
H12B	0.3507	0.3140	0.4030	0.062*
C13	0.40482 (11)	0.3492 (3)	0.3218 (3)	0.0619 (9)
H13	0.3907	0.4089	0.2422	0.074*
C14	0.42639 (12)	0.4422 (4)	0.4441 (4)	0.0773 (11)
H14A	0.4056	0.5012	0.4765	0.093*
H14B	0.4467	0.5036	0.4121	0.093*
C15	0.43746 (11)	0.2543 (4)	0.2710 (3)	0.0674 (10)
H15A	0.4234	0.1945	0.1929	0.081*
H15B	0.4581	0.3125	0.2364	0.081*
N1	0.34081 (7)	-0.0273 (3)	0.4547 (2)	0.0582 (7)
H1C	0.3501	-0.0278	0.3622	0.070*
O1	0.35154 (7)	0.0859 (2)	0.6660 (2)	0.0619 (7)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.062 (2)	0.075 (2)	0.065 (2)	-0.0255 (18)	0.0153 (18)	0.0110 (18)
C2	0.074 (3)	0.087 (3)	0.123 (4)	-0.018 (2)	0.028 (3)	0.022 (3)
C3	0.086 (3)	0.105 (4)	0.244 (7)	0.001 (3)	0.044 (4)	0.024 (4)
C4	0.085 (4)	0.126 (5)	0.245 (8)	0.003 (3)	0.030 (4)	0.008 (5)
C5	0.0435 (16)	0.0514 (18)	0.0365 (14)	0.0034 (13)	0.0128 (13)	0.0079 (13)
C6	0.0425 (15)	0.0466 (16)	0.0357 (14)	-0.0005 (12)	0.0111 (12)	-0.0016 (12)
C7	0.069 (2)	0.072 (2)	0.0451 (17)	-0.0157 (18)	0.0151 (15)	-0.0170 (15)
C8	0.076 (2)	0.078 (2)	0.061 (2)	-0.032 (2)	0.0136 (19)	-0.0210 (19)
C9	0.055 (2)	0.098 (3)	0.085 (3)	-0.029 (2)	0.0109 (19)	-0.003 (2)
C10	0.0479 (17)	0.071 (2)	0.072 (2)	-0.0038 (16)	0.0278 (17)	-0.0084 (18)
C11	0.0473 (16)	0.0553 (19)	0.0565 (18)	0.0001 (14)	0.0142 (14)	0.0015 (14)
C12	0.0528 (17)	0.0508 (18)	0.0497 (17)	-0.0002 (14)	0.0090 (14)	0.0037 (14)
C13	0.070 (2)	0.056 (2)	0.060 (2)	-0.0064 (16)	0.0150 (17)	0.0139 (16)
C14	0.093 (3)	0.050 (2)	0.094 (3)	-0.0227 (19)	0.032 (2)	-0.0086 (19)
C15	0.070 (2)	0.081 (2)	0.059 (2)	-0.0260 (19)	0.0307 (17)	-0.0057 (18)
N1	0.0609 (16)	0.0749 (18)	0.0441 (13)	-0.0258 (14)	0.0234 (12)	-0.0055 (13)

O1	0.0715 (15)	0.0796 (16)	0.0407 (11)	-0.0063 (12)	0.0263 (11)	0.0002 (11)
----	-------------	-------------	-------------	--------------	-------------	-------------

Geometric parameters (Å, °)

C1—N1	1.453 (4)	C8—C9	1.522 (5)
C1—C2	1.496 (5)	C8—C14	1.523 (5)
C1—H1A	0.9700	C8—H8	0.9800
C1—H1B	0.9700	C9—C10	1.516 (5)
C2—C3	1.456 (6)	C9—H9A	0.9700
C2—H2A	0.9700	C9—H9B	0.9700
C2—H2B	0.9700	C10—C15	1.517 (5)
C3—C4	1.457 (6)	C10—C11	1.529 (4)
C3—H3A	0.9700	C10—H10	0.9800
C3—H3B	0.9700	C11—H11A	0.9700
C4—H4A	0.9600	C11—H11B	0.9700
C4—H4B	0.9600	C12—C13	1.535 (4)
C4—H4C	0.9600	C12—H12A	0.9700
C5—O1	1.232 (3)	C12—H12B	0.9700
C5—N1	1.333 (3)	C13—C14	1.512 (4)
C5—C6	1.527 (4)	C13—C15	1.533 (5)
C6—C11	1.535 (4)	C13—H13	0.9800
C6—C12	1.536 (4)	C14—H14A	0.9700
C6—C7	1.539 (4)	C14—H14B	0.9700
C7—C8	1.528 (4)	C15—H15A	0.9700
C7—H7A	0.9700	C15—H15B	0.9700
C7—H7B	0.9700	N1—H1C	0.9861
N1—C1—C2	113.2 (3)	C10—C9—C8	109.1 (3)
N1—C1—H1A	108.9	C10—C9—H9A	109.9
C2—C1—H1A	108.9	C8—C9—H9A	109.9
N1—C1—H1B	108.9	C10—C9—H9B	109.9
C2—C1—H1B	108.9	C8—C9—H9B	109.9
H1A—C1—H1B	107.8	H9A—C9—H9B	108.3
C3—C2—C1	115.1 (4)	C9—C10—C15	109.8 (3)
C3—C2—H2A	108.5	C9—C10—C11	109.9 (3)
C1—C2—H2A	108.5	C15—C10—C11	109.4 (3)
C3—C2—H2B	108.5	C9—C10—H10	109.2
C1—C2—H2B	108.5	C15—C10—H10	109.2
H2A—C2—H2B	107.5	C11—C10—H10	109.2
C2—C3—C4	119.2 (5)	C10—C11—C6	110.2 (2)
C2—C3—H3A	107.5	C10—C11—H11A	109.6
C4—C3—H3A	107.5	C6—C11—H11A	109.6
C2—C3—H3B	107.5	C10—C11—H11B	109.6
C4—C3—H3B	107.5	C6—C11—H11B	109.6
H3A—C3—H3B	107.0	H11A—C11—H11B	108.1
C3—C4—H4A	109.5	C13—C12—C6	110.0 (2)
C3—C4—H4B	109.5	C13—C12—H12A	109.7
H4A—C4—H4B	109.5	C6—C12—H12A	109.7

C3—C4—H4C	109.5	C13—C12—H12B	109.7
H4A—C4—H4C	109.5	C6—C12—H12B	109.7
H4B—C4—H4C	109.5	H12A—C12—H12B	108.2
O1—C5—N1	122.0 (2)	C14—C13—C15	110.1 (3)
O1—C5—C6	121.7 (3)	C14—C13—C12	109.6 (3)
N1—C5—C6	116.3 (2)	C15—C13—C12	108.9 (3)
C5—C6—C11	111.5 (2)	C14—C13—H13	109.4
C5—C6—C12	109.4 (2)	C15—C13—H13	109.4
C11—C6—C12	108.9 (2)	C12—C13—H13	109.4
C5—C6—C7	110.3 (2)	C13—C14—C8	109.3 (3)
C11—C6—C7	108.2 (2)	C13—C14—H14A	109.8
C12—C6—C7	108.5 (2)	C8—C14—H14A	109.8
C8—C7—C6	110.1 (2)	C13—C14—H14B	109.8
C8—C7—H7A	109.6	C8—C14—H14B	109.8
C6—C7—H7A	109.6	H14A—C14—H14B	108.3
C8—C7—H7B	109.6	C10—C15—C13	109.3 (3)
C6—C7—H7B	109.6	C10—C15—H15A	109.8
H7A—C7—H7B	108.2	C13—C15—H15A	109.8
C9—C8—C14	110.1 (3)	C10—C15—H15B	109.8
C9—C8—C7	109.8 (3)	C13—C15—H15B	109.8
C14—C8—C7	109.3 (3)	H15A—C15—H15B	108.3
C9—C8—H8	109.2	C5—N1—C1	124.1 (2)
C14—C8—H8	109.2	C5—N1—H1C	113.9
C7—C8—H8	109.2	C1—N1—H1C	121.8
N1—C1—C2—C3	-65.2 (5)	C5—C6—C11—C10	179.5 (2)
C1—C2—C3—C4	-165.0 (5)	C12—C6—C11—C10	58.8 (3)
O1—C5—C6—C11	126.9 (3)	C7—C6—C11—C10	-59.0 (3)
N1—C5—C6—C11	-54.1 (3)	C5—C6—C12—C13	179.0 (2)
O1—C5—C6—C12	-112.6 (3)	C11—C6—C12—C13	-58.9 (3)
N1—C5—C6—C12	66.4 (3)	C7—C6—C12—C13	58.6 (3)
O1—C5—C6—C7	6.6 (4)	C6—C12—C13—C14	-60.3 (3)
N1—C5—C6—C7	-174.4 (2)	C6—C12—C13—C15	60.2 (3)
C5—C6—C7—C8	-178.8 (3)	C15—C13—C14—C8	-58.9 (4)
C11—C6—C7—C8	59.0 (3)	C12—C13—C14—C8	61.0 (4)
C12—C6—C7—C8	-59.0 (3)	C9—C8—C14—C13	59.6 (4)
C6—C7—C8—C9	-60.3 (4)	C7—C8—C14—C13	-61.1 (4)
C6—C7—C8—C14	60.5 (4)	C9—C10—C15—C13	-59.7 (3)
C14—C8—C9—C10	-60.2 (4)	C11—C10—C15—C13	61.0 (3)
C7—C8—C9—C10	60.1 (4)	C14—C13—C15—C10	59.2 (3)
C8—C9—C10—C15	60.3 (3)	C12—C13—C15—C10	-61.0 (3)
C8—C9—C10—C11	-60.1 (4)	O1—C5—N1—C1	2.9 (5)
C9—C10—C11—C6	60.4 (3)	C6—C5—N1—C1	-176.1 (3)
C15—C10—C11—C6	-60.2 (3)	C2—C1—N1—C5	115.9 (4)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N1—H1C \cdots O1 ⁱ	0.99	1.96	2.896 (3)	158

Symmetry code: (i) $x, -y, z-1/2$.