

Poly[[diaqua(3-carboxy-5-nitrobenzoato)(μ -5-nitrobenzene-1,3-dicarboxylato)neodymium(III)] 2.5-hydrate]

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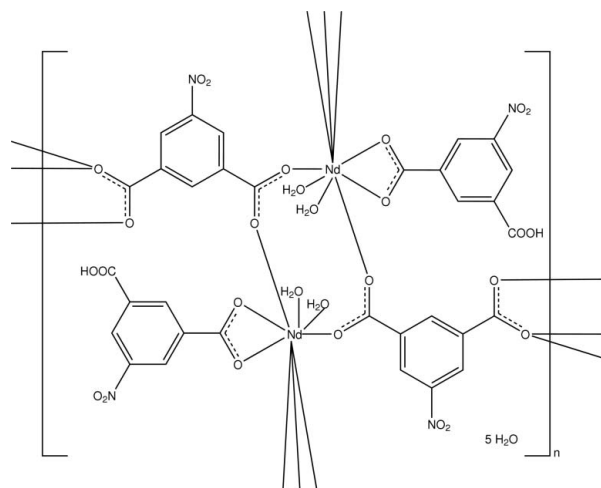
Received 27 July 2009; accepted 15 August 2009

Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; disorder in main residue; R factor = 0.018; wR factor = 0.057; data-to-parameter ratio = 14.8.

In the title compound, $\{[\text{Nd}(\text{C}_8\text{H}_3\text{NO}_6)(\text{C}_8\text{H}_4\text{NO}_6)(\text{H}_2\text{O})_2] \cdot 2.5\text{H}_2\text{O}\}_n$, the Nd^{III} ion is nine-coordinated by seven O atoms from five carboxylate groups and two water molecules. The $[\text{Nd}(\text{C}_8\text{H}_3\text{NO}_6)(\text{H}_2\text{O})_2]^{2+}$ units are bridged by 5-nitroisophthalate dianions, forming polymeric sheets parallel to the ab plane. The polymeric sheets are linked into a three-dimensional network by $\text{O}-\text{H} \cdots \text{O}$ and $\text{C}-\text{H} \cdots \text{O}$ hydrogen bonds, and $\pi-\pi$ interactions [centroid-centroid distance = $3.5533(11)$ Å]. The 5-nitroisophthalate(1⁻) anion is disordered over three positions with an occupancy ratio of 0.68:0.23:0.09. Two of the uncoordinated water molecules are disordered over two positions, with occupancy ratios of 0.722 (15):0.278 (15) and 0.279 (6):0.221 (6), respectively.

Related literature

For related structures, see: Ye *et al.* (2008); Eddoudi *et al.* (2001); Bünzli & Choppin (1989); Huang *et al.* (2008); Cui *et al.* (2002); Yan *et al.* (2005); Ren *et al.* (2006); Li *et al.* (2005). For the stability of the temperature controller used for the data collection, see: Cosier & Glazer (1986).



Experimental

Crystal data

$[\text{Nd}(\text{C}_8\text{H}_3\text{NO}_6)(\text{C}_8\text{H}_4\text{NO}_6)(\text{H}_2\text{O})_2] \cdot 2.5\text{H}_2\text{O}$

$M_r = 644.55$

Triclinic, $P\bar{1}$

$a = 9.5748(1)$ Å

$b = 10.4634(1)$ Å

$c = 13.3285(2)$ Å

$\alpha = 69.279(1)^\circ$

$\beta = 71.753(1)^\circ$

$\gamma = 66.046(1)^\circ$

$V = 1118.71(2)$ Å³

$Z = 2$

Mo $K\alpha$ radiation

$\mu = 2.41$ mm⁻¹

$T = 100$ K

$0.35 \times 0.24 \times 0.13$ mm

Data collection

Bruker SMART APEXII CCD

area-detector diffractometer

Absorption correction: multi-scan

(SADABS; Bruker, 2005)

$T_{\text{min}} = 0.484$, $T_{\text{max}} = 0.742$

22527 measured reflections

5879 independent reflections

5725 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.019$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.018$

$wR(F^2) = 0.057$

$S = 1.21$

5879 reflections

398 parameters

115 restraints

H-atom parameters constrained

$\Delta\rho_{\text{max}} = 1.18$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.77$ e Å⁻³

Table 1

Selected bond lengths (Å).

Nd1—O7	2.3302 (13)	Nd1—O2W	2.4775 (15)
Nd1—O8 ⁱ	2.4214 (14)	Nd1—O2	2.5223 (13)
Nd1—O10 ⁱⁱ	2.4481 (14)	Nd1—O1	2.5743 (14)
Nd1—O1W	2.4612 (14)	Nd1—O10 ⁱⁱⁱ	2.9332 (14)
Nd1—O9 ⁱⁱⁱ	2.4708 (13)		

Symmetry codes: (i) $-x, -y + 1, -z + 1$; (ii) $-x, -y + 2, -z + 1$; (iii) $x + 1, y - 1, z$.

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Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O5A—H5A \cdots O3W ^{iv}	0.84	1.76	2.596 (5)	178
O1W—H1W1 \cdots O2 ^v	0.74	1.99	2.731 (3)	175
O1W—H2W1 \cdots O4WA	0.94	2.00	2.849 (4)	149
O2W—H1W2 \cdots O4WA	0.84	1.86	2.667 (3)	159
O2W—H2W2 \cdots O1 ⁱ	0.82	2.06	2.874 (3)	172
O3W—H1W3 \cdots O11 ^{vi}	0.91	2.56	3.246 (3)	132
O3W—H1W3 \cdots O12 ^{vi}	0.91	2.05	2.902 (3)	156
O3W—H2W3 \cdots O9 ⁱⁱⁱ	0.96	1.76	2.688 (2)	162
O4WA—H2W4 \cdots O3W	0.85	2.51	3.118 (6)	130
C12—H12A \cdots O1 ⁱⁱ	0.95	2.41	3.351 (2)	169

Symmetry codes: (i) $-x, -y + 1, -z + 1$; (ii) $-x, -y + 2, -z + 1$; (iii) $x + 1, y - 1, z$; (iv) $x - 1, y, z + 1$; (v) $-x + 1, -y + 1, -z + 1$; (vi) $x, y - 1, z$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINTE* (Bruker, 2005); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

The authors thank the Malaysian Government, the Ministry of Science, Technology and Innovation (MOSTI) and Universiti Sains Malaysia (USM) for a Research University grant (No. 1001/PKIMIA/811120). HKF thanks USM for a

Research University Golden Goose grant (No. 1001/PFIZIK/811012). MMR thanks the Malaysian Government and USM for the award of the post of Research Officer under the Science Fund (grant No. 305/PFIZIK/613312).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI2868).

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supporting information

Acta Cryst. (2009). E65, m1108–m1109 [doi:10.1107/S1600536809032486]

Poly[[diaqua(3-carboxy-5-nitrobenzoato)(μ -5-nitrobenzene-1,3-dicarboxylato)neodymium(III)] 2.5-hydrate]

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S1. Comment

The research on the assembly of lanthanide coordination networks has increased over the years (Ye *et al.*, 2008; Eddaoudi *et al.*, 2001). Lanthanide provide opportunities for the discovery of unusual network topologies (Bünzli & Choppin, 1989; Huang *et al.*, 2008; Cui *et al.*, 2002) due to its high and variable coordination numbers and flexible coordination environments (Yan *et al.*, 2005).

5-Nitroisophthalic acid (**nia**) has two carboxylic groups which may be completely or partially deprotonated and thus produces versatile coordination modes with lanthanide ions. Moreover, the existence of nitro group as an electron-withdrawing group has profound impacts on the electron density of the whole ligand, and thereby different physical phenomena can be produced (Ren *et al.*, 2006; Li *et al.*, 2005). In this paper, we report the crystal structure of a polymeric coordination complex formed from hydrothermal reaction between trivalent neodymium ion and nitroisophthalic acid.

The asymmetric unit of the title polymeric compound is shown in Fig. 1. In the crystal structure, each Nd^{III} ion adopts a nine-coordination environment being coordinated by two O atoms from two water molecules and seven O atoms from five carboxylate groups. The Nd—O distances range from 2.3302 (13) to 2.9332 (14) Å for carboxylate groups and 2.4612 (14) and 2.4775 (15) Å, respectively for O1W and O2W (Table 1). The nitro group attached to the C1-C6 benzene ring is slightly twisted, with a dihedral angle of 12.2 (3)° for the major disorder component A [9(1)° for B and 3(1)° for C]. The nitro group attached to the C9-C14 benzene ring is twisted by a dihedral angle of 21.2 (1)°. The dihedral angles between C1-C6 benzene rings and planes of carboxyl groups in disorder components A, B and C are 5.8 (5)°, 9(1)° and 29 (1)°, respectively. The adjacent [Nd(C₈H₃NO₆)(H₂O)₂]²⁺ units are bridged by 5-nitroisophthalate dianions to form polymeric sheets parallel to the *ab* plane (Fig. 2).

The polymeric sheets are linked into a three-dimensional network (Fig. 3) by O—H \cdots O and C—H \cdots O hydrogen bonds (Table 2). There also exist a π – π interaction between the C9–C14 benzene rings at (x, y, z) and (-x, 2-y, 1-z), with a centroid-to-centroid distance of 3.5533 (11) Å.

S2. Experimental

A mixture of 5-nitroisophthalic acid (0.4243 g, 2.0 mmol), sodium hydroxide (0.140 g, 3.5 mmol) and distilled water (30 ml) was heated till boiling. The solution was left to cool to room temperature before neodymium(III) nitrate hexahydrate salt, Nd(NO₃)₆·6H₂O (0.4380 g, 1.0 mmol) was added to it. The pH of the mixture was controlled to be between 3–4 by adding 1 M NaOH or 1 M HNO₃. Subsequently the solution was poured into a 40 ml Teflon tube, which was then sealed and heated at 403 K for 3 d. Upon cooling to room temperature, light purple crystals of the title compound were obtained which were filtered, washed with distilled water and left to dry in air.

S3. Refinement

The 5-nitroisophthalate(1-) anion is disordered over three positions with an occupancy ratio of 0.683 (7):0.234 (7):0.087 (7) which was fixed at 0.68:0.23:0.09 for the final refinement. The two minor disorder components B and C were refined isotropically; a common U_{iso} was used for the disorder component C. In all disorder components, the C1—C6 benzene ring is constrained to a regular hexagon with $d = 1.39$ Å. Atom O4W and hemihydrate O5W are also disordered over two positions, with occupancies of 0.722 (15) and 0.278 (15), and 0.279 (6) and 0.221 (6) respectively. For the disordered 5-nitroisophthalate(1-) anion, similarity restraints were applied. The O atoms of the uncoordinated water molecules are restrained so that their U_{ij} components approximate to isotropic behavior. H atoms for O1W, O2W, O3W and O4WB were located in a difference Fourier map and refined as riding with the parent atom with $U_{iso}(H) = 1.5U_{eq}(O)$. The rest of H atoms were positioned geometrically and refined using a riding model, with C-H = 0.95 Å, O-H = 0.84 Å and $U_{iso} = 1.2U_{eq}(C)$ and $1.5U_{eq}(O)$.

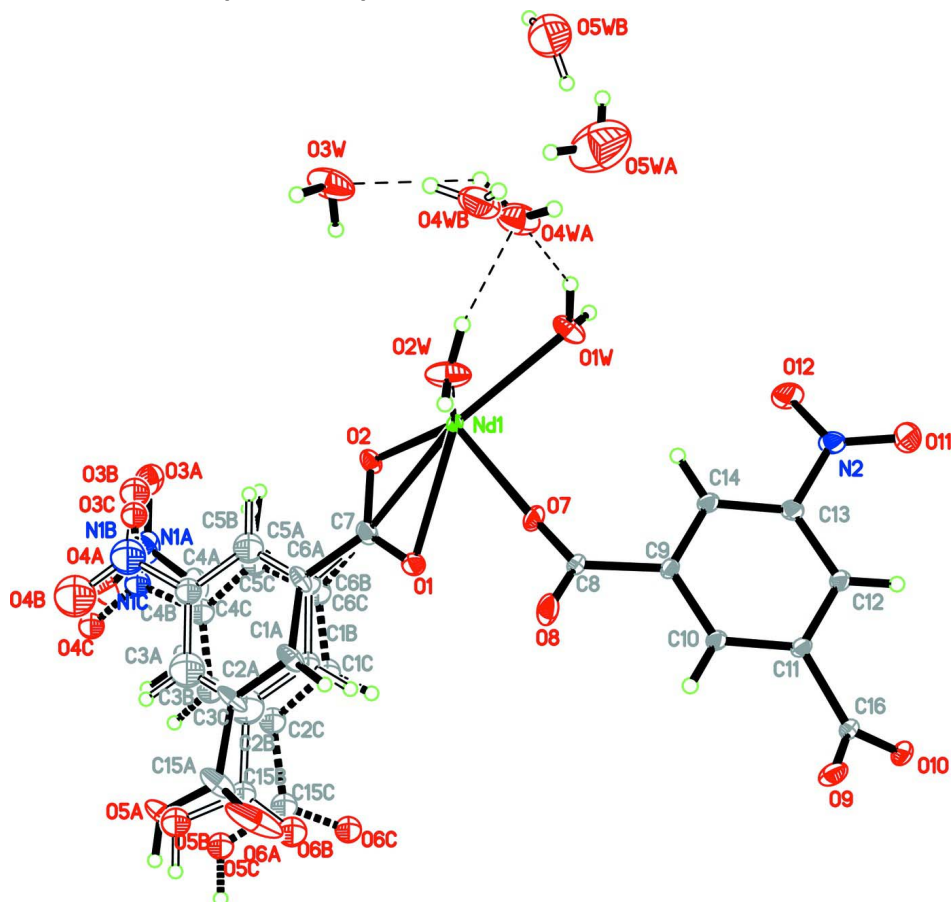
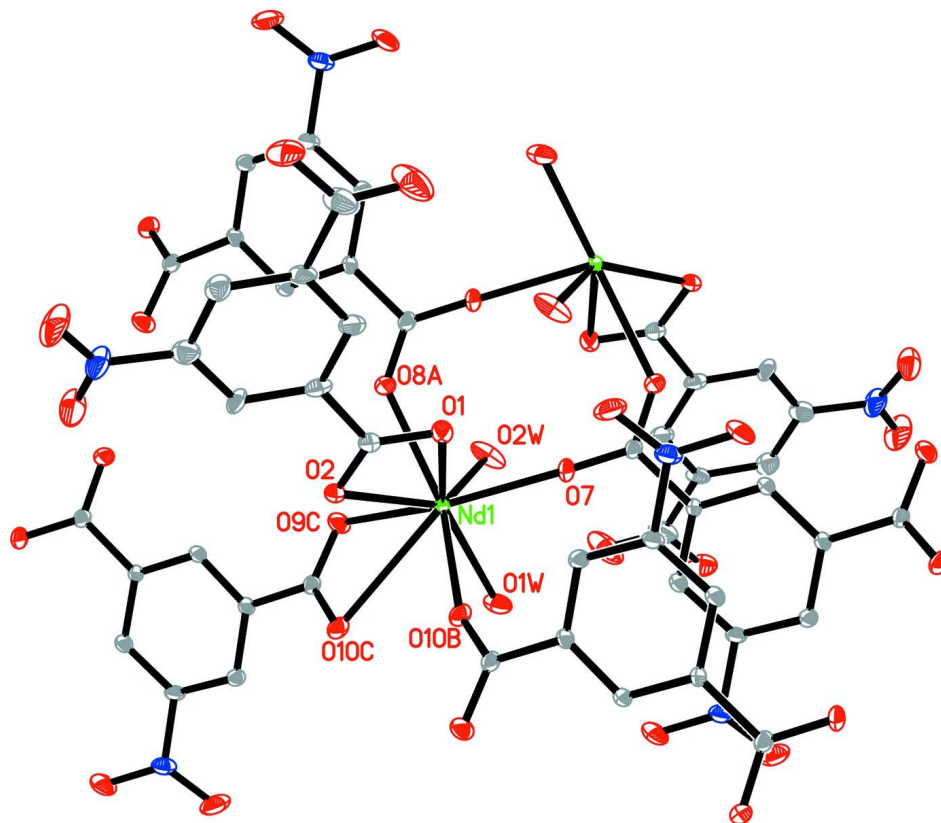
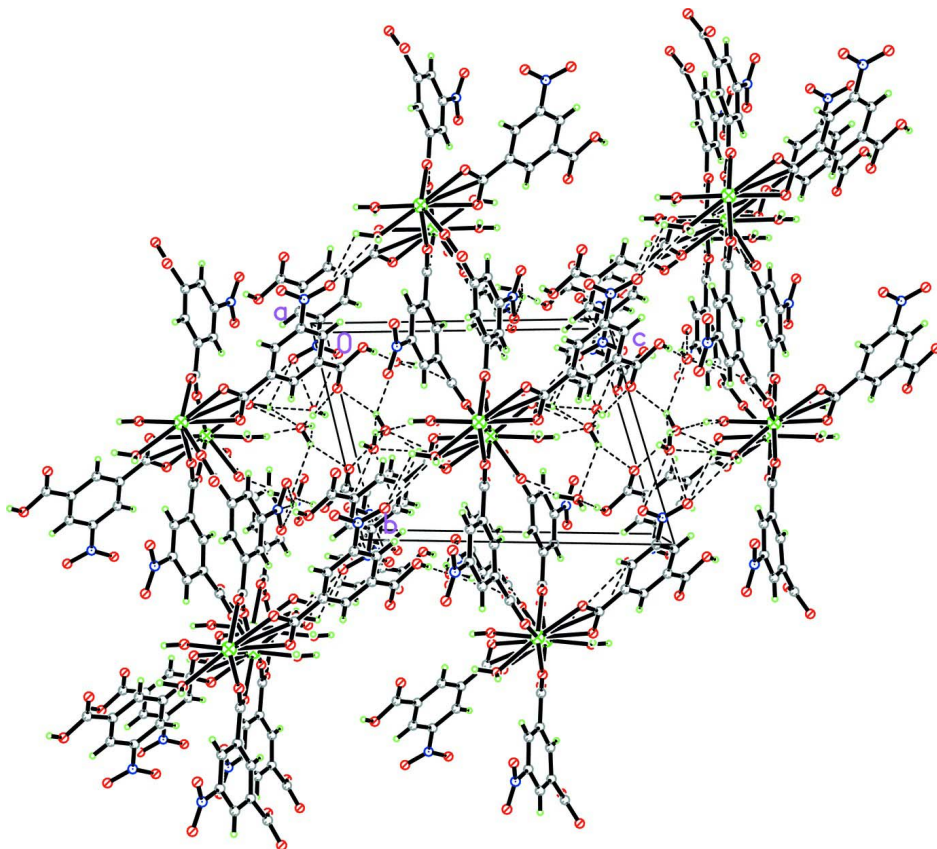


Figure 1

The asymmetric unit of the title compound. Displacement ellipsoids are drawn at the 50% probability level. All disorder components are shown. Hydrogen bonds are shown as dashed lines.

**Figure 2**

Part of the polymeric sheet, showing the coordination environment of Nd^{III} ion. Displacement ellipsoids are drawn at the 50% probability level. Only major disorder components are shown. Uncoordinated water molecules and H atoms have been omitted for clarity. Symmetry codes: (A) $-x, 1-y, 1-z$; (B) $-x, 2-y, 1-z$; (c) $1+x, -1+y, z$.

**Figure 3**

The crystal packing of the title compound, viewed down the *a* axis. Hydrogen bonds are shown as dashed lines. Only major disorder components are shown.

Poly[[diaqua(3-carboxy-5-nitrobenzoato)(μ -5-nitrobenzene-1,3-dicarboxylato)neodymium(III)] 2.5-hydrate]

Crystal data

[Nd(C₈H₃NO₆)(C₈H₄NO₆)(H₂O)₂] \cdot 2.5H₂O

$M_r = 644.55$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 9.5748$ (1) Å

$b = 10.4634$ (1) Å

$c = 13.3285$ (2) Å

$\alpha = 69.279$ (1) $^\circ$

$\beta = 71.753$ (1) $^\circ$

$\gamma = 66.046$ (1) $^\circ$

$V = 1118.71$ (2) Å³

$Z = 2$

$F(000) = 636$

$D_x = 1.913$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 9930 reflections

$\theta = 2.7$ – 29.0 $^\circ$

$\mu = 2.41$ mm⁻¹

$T = 100$ K

Plate, purple

$0.35 \times 0.24 \times 0.13$ mm

Data collection

Bruker SMART APEXII CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2005)

$T_{\min} = 0.484$, $T_{\max} = 0.742$

22527 measured reflections

5879 independent reflections

5725 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.019$
 $\theta_{\text{max}} = 29.0^\circ$, $\theta_{\text{min}} = 1.7^\circ$
 $h = -13 \rightarrow 13$

$k = -14 \rightarrow 14$
 $l = -18 \rightarrow 18$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.018$
 $wR(F^2) = 0.057$
 $S = 1.21$
 5879 reflections
 398 parameters
 115 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0337P)^2 + 0.5091P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 1.18 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.77 \text{ e } \text{\AA}^{-3}$

Special details

Experimental. The crystal was placed in the cold stream of an Oxford Cyrosystems Cobra open-flow nitrogen cryostat (Cosier & Glazer, 1986) operating at 100.0 (1) K.

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
Nd1	0.285572 (9)	0.468414 (9)	0.486911 (7)	0.01037 (4)	
O3A	0.5272 (5)	-0.1271 (4)	0.9533 (3)	0.0330 (9)	0.68
O4A	0.3769 (4)	-0.1558 (3)	1.1135 (2)	0.0463 (8)	0.68
O5A	-0.1663 (5)	0.1114 (5)	1.1553 (3)	0.0263 (8)	0.68
H5A	-0.2601	0.1301	1.1875	0.039*	0.68
O6A	-0.2644 (3)	0.3037 (5)	1.0252 (3)	0.0605 (13)	0.68
N1A	0.4039 (6)	-0.0919 (4)	1.0164 (3)	0.0297 (7)	0.68
C1A	0.0383 (3)	0.2754 (4)	0.8934 (3)	0.0205 (8)	0.68
H1AA	-0.0439	0.3572	0.8654	0.025*	0.68
C2A	0.0078 (3)	0.1835 (5)	0.9962 (3)	0.0198 (8)	0.68
C3A	0.1281 (4)	0.0639 (4)	1.0372 (2)	0.0238 (9)	0.68
H3AA	0.1072	0.0011	1.1075	0.029*	0.68
C4A	0.2789 (4)	0.0361 (3)	0.9754 (3)	0.0245 (8)	0.68
C5A	0.3094 (3)	0.1279 (4)	0.8726 (3)	0.0189 (9)	0.68
H5AA	0.4125	0.1089	0.8304	0.023*	0.68
C6A	0.1891 (4)	0.2476 (4)	0.8316 (3)	0.0165 (13)	0.68
C15A	-0.1548 (5)	0.2079 (7)	1.0602 (3)	0.0268 (9)	0.68
O3B	0.4896 (15)	-0.1413 (17)	0.9604 (13)	0.027 (3)*	0.23
O4B	0.3076 (14)	-0.1937 (12)	1.0907 (9)	0.056 (3)*	0.23
O5B	-0.1948 (16)	0.1468 (13)	1.1445 (11)	0.025 (3)*	0.23

H5B	-0.2852	0.1742	1.1814	0.037*	0.23
O6B	-0.2641 (10)	0.3699 (8)	1.0410 (7)	0.0257 (18)*	0.23
N1B	0.3557 (15)	-0.1115 (12)	1.0079 (10)	0.040 (3)*	0.23
C1B	0.0367 (14)	0.3066 (10)	0.8974 (11)	0.016 (3)*	0.23
H1BA	-0.0349	0.4004	0.8737	0.019*	0.23
C2B	-0.0066 (12)	0.2128 (14)	0.9957 (11)	0.038 (5)*	0.23
C3B	0.0982 (14)	0.0756 (13)	1.0304 (10)	0.039 (5)*	0.23
H3BA	0.0686	0.0115	1.0976	0.046*	0.23
C4B	0.2463 (12)	0.0322 (10)	0.9668 (10)	0.021 (3)*	0.23
C5B	0.2896 (12)	0.1260 (15)	0.8684 (10)	0.028 (4)*	0.23
H5BA	0.3907	0.0963	0.8249	0.033*	0.23
C6B	0.1848 (16)	0.2632 (14)	0.8337 (10)	0.016 (4)*	0.23
C15B	-0.1695 (13)	0.2567 (12)	1.0630 (10)	0.019 (3)*	0.23
O3C	0.449 (2)	-0.130 (3)	0.9824 (19)	0.0208 (13)*	0.09
O4C	0.272 (2)	-0.1538 (18)	1.1276 (12)	0.0208 (13)*	0.09
O5C	-0.2380 (19)	0.2386 (18)	1.1423 (12)	0.0208 (13)*	0.09
H5C	-0.3302	0.2675	1.1873	0.025*	0.09
O6C	-0.2201 (18)	0.4585 (14)	1.0575 (13)	0.0208 (13)*	0.09
N1C	0.318 (2)	-0.0826 (18)	1.0374 (14)	0.0208 (13)*	0.09
C1C	0.0294 (19)	0.3429 (15)	0.9028 (13)	0.0208 (13)*	0.09
H1CA	-0.0355	0.4382	0.8740	0.025*	0.09
C2C	-0.0198 (15)	0.2637 (16)	1.0071 (12)	0.0208 (13)*	0.09
C3C	0.0750 (18)	0.1244 (16)	1.0493 (11)	0.0208 (13)*	0.09
H3CA	0.0414	0.0702	1.1206	0.025*	0.09
C4C	0.2191 (19)	0.0642 (16)	0.9873 (14)	0.0208 (13)*	0.09
C5C	0.2683 (19)	0.143 (2)	0.8830 (15)	0.0208 (13)*	0.09
H5CA	0.3668	0.1023	0.8406	0.025*	0.09
C6C	0.173 (2)	0.283 (2)	0.8408 (12)	0.0208 (13)*	0.09
C15C	-0.174 (2)	0.3312 (17)	1.0721 (15)	0.0208 (13)*	0.09
O1	0.11719 (16)	0.45727 (15)	0.68090 (11)	0.0170 (3)	
O2	0.36376 (16)	0.31566 (15)	0.66785 (11)	0.0165 (3)	
O7	0.05148 (16)	0.65931 (14)	0.46945 (12)	0.0176 (3)	
O8	-0.19586 (17)	0.73727 (15)	0.45331 (13)	0.0209 (3)	
O9	-0.49580 (16)	1.26834 (15)	0.42188 (12)	0.0188 (3)	
O10	-0.37511 (16)	1.38214 (14)	0.45852 (11)	0.0157 (3)	
O11	0.1860 (2)	1.24682 (18)	0.22910 (16)	0.0325 (4)	
O12	0.32156 (17)	1.02380 (18)	0.29539 (14)	0.0253 (3)	
N2	0.19668 (19)	1.12233 (19)	0.28241 (14)	0.0178 (3)	
C7	0.2251 (2)	0.3495 (2)	0.72073 (15)	0.0155 (3)	
C8	-0.0736 (2)	0.75718 (19)	0.44717 (14)	0.0122 (3)	
C9	-0.0762 (2)	0.91156 (18)	0.41124 (14)	0.0107 (3)	
C10	-0.2178 (2)	1.02243 (19)	0.42723 (14)	0.0115 (3)	
H10A	-0.3122	1.0009	0.4554	0.014*	
C11	-0.2213 (2)	1.16466 (19)	0.40205 (14)	0.0113 (3)	
C12	-0.0849 (2)	1.19940 (19)	0.35355 (14)	0.0124 (3)	
H12A	-0.0861	1.2963	0.3343	0.015*	
C13	0.0528 (2)	1.0867 (2)	0.33446 (14)	0.0129 (3)	
C14	0.0619 (2)	0.94284 (19)	0.36509 (14)	0.0124 (3)	

H14A	0.1593	0.8681	0.3549	0.015*	
C16	−0.3728 (2)	1.27944 (19)	0.42875 (14)	0.0127 (3)	
O1W	0.39421 (17)	0.62127 (17)	0.31765 (11)	0.0214 (3)	
H1W1	0.4595	0.6422	0.3181	0.032*	
H2W1	0.4207	0.6070	0.2475	0.032*	
O2W	0.20344 (18)	0.4620 (2)	0.33051 (13)	0.0285 (4)	
H1W2	0.2599	0.4648	0.2674	0.043*	
H2W2	0.1146	0.4785	0.3245	0.043*	
O3W	0.5412 (2)	0.1731 (2)	0.24946 (15)	0.0377 (4)	
H1W3	0.4742	0.1364	0.2425	0.057*	
H2W3	0.5221	0.1903	0.3193	0.057*	
O4WA	0.4139 (3)	0.5063 (6)	0.1473 (2)	0.0342 (12)	0.722 (15)
H1W4	0.3780	0.5660	0.0913	0.051*	0.722 (15)
H2W4	0.4716	0.4275	0.1301	0.051*	0.722 (15)
O4WB	0.4055 (9)	0.4415 (15)	0.1312 (8)	0.037 (3)	0.278 (15)
H3W4	0.3704	0.4757	0.0809	0.056*	0.278 (15)
H4W4	0.4353	0.3561	0.1412	0.056*	0.278 (15)
O5WA	0.6107 (14)	0.5930 (14)	0.0951 (10)	0.082 (4)	0.279 (6)
H1W5	0.7059	0.5714	0.0630	0.123*	0.279 (6)
H2W5	0.5861	0.5169	0.1155	0.123*	0.279 (6)
O5WB	0.9195 (14)	0.4364 (12)	0.0923 (8)	0.051 (3)	0.221 (6)
H3W5	0.8615	0.4864	0.1375	0.077*	0.221 (6)
H4W5	1.0037	0.3835	0.1152	0.077*	0.221 (6)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Nd1	0.00707 (6)	0.00691 (6)	0.01558 (6)	−0.00155 (4)	−0.00089 (3)	−0.00326 (4)
O3A	0.038 (2)	0.0212 (16)	0.0264 (15)	−0.0010 (16)	−0.0073 (15)	−0.0004 (11)
O4A	0.064 (2)	0.0291 (15)	0.0201 (12)	−0.0069 (14)	−0.0055 (13)	0.0107 (10)
O5A	0.0263 (18)	0.037 (2)	0.0144 (13)	−0.0199 (18)	0.0051 (12)	−0.0027 (14)
O6A	0.0223 (14)	0.103 (3)	0.0277 (14)	−0.0250 (18)	−0.0064 (11)	0.0225 (18)
N1A	0.043 (2)	0.0183 (15)	0.0194 (14)	−0.0077 (16)	−0.0092 (15)	0.0041 (11)
C1A	0.0177 (17)	0.027 (2)	0.0192 (16)	−0.0148 (15)	−0.0041 (11)	−0.0003 (16)
C2A	0.0189 (16)	0.034 (2)	0.0095 (14)	−0.0193 (14)	−0.0011 (10)	0.0016 (13)
C3A	0.036 (2)	0.0240 (19)	0.0128 (14)	−0.0182 (16)	−0.0061 (15)	0.0037 (11)
C4A	0.035 (2)	0.0231 (19)	0.0186 (16)	−0.0142 (16)	−0.0103 (16)	0.0006 (12)
C5A	0.0233 (17)	0.0150 (17)	0.0155 (15)	−0.0079 (14)	−0.0039 (13)	0.0010 (11)
C6A	0.018 (2)	0.019 (2)	0.0147 (17)	−0.0128 (14)	−0.0026 (10)	0.0009 (12)
C15A	0.0245 (18)	0.044 (3)	0.0150 (16)	−0.0223 (19)	−0.0037 (12)	0.0009 (17)
O1	0.0127 (6)	0.0151 (6)	0.0183 (6)	−0.0033 (5)	−0.0015 (5)	−0.0018 (5)
O2	0.0112 (6)	0.0155 (6)	0.0192 (6)	−0.0062 (5)	−0.0027 (5)	0.0014 (5)
O7	0.0123 (6)	0.0091 (6)	0.0256 (7)	0.0005 (5)	−0.0037 (5)	−0.0026 (5)
O8	0.0148 (7)	0.0110 (6)	0.0374 (8)	−0.0051 (5)	−0.0068 (6)	−0.0047 (6)
O9	0.0096 (6)	0.0161 (7)	0.0311 (7)	−0.0019 (5)	−0.0015 (5)	−0.0117 (6)
O10	0.0170 (6)	0.0099 (6)	0.0203 (6)	−0.0039 (5)	−0.0009 (5)	−0.0072 (5)
O11	0.0217 (8)	0.0195 (8)	0.0512 (10)	−0.0135 (6)	0.0080 (7)	−0.0089 (7)
O12	0.0092 (6)	0.0269 (8)	0.0378 (8)	−0.0053 (6)	0.0001 (6)	−0.0111 (7)

N2	0.0120 (7)	0.0190 (8)	0.0244 (8)	-0.0080 (6)	0.0022 (6)	-0.0099 (6)
C7	0.0139 (8)	0.0152 (9)	0.0172 (8)	-0.0072 (7)	-0.0030 (6)	-0.0014 (6)
C8	0.0118 (8)	0.0086 (7)	0.0145 (7)	-0.0024 (6)	-0.0007 (6)	-0.0040 (6)
C9	0.0091 (7)	0.0079 (7)	0.0143 (7)	-0.0014 (6)	-0.0028 (6)	-0.0030 (6)
C10	0.0081 (7)	0.0107 (8)	0.0150 (7)	-0.0027 (6)	-0.0014 (6)	-0.0039 (6)
C11	0.0085 (7)	0.0097 (8)	0.0148 (7)	-0.0014 (6)	-0.0018 (6)	-0.0048 (6)
C12	0.0111 (8)	0.0109 (8)	0.0155 (7)	-0.0041 (6)	-0.0008 (6)	-0.0048 (6)
C13	0.0094 (8)	0.0139 (8)	0.0168 (7)	-0.0057 (6)	0.0003 (6)	-0.0059 (6)
C14	0.0086 (7)	0.0121 (8)	0.0155 (7)	-0.0023 (6)	-0.0013 (6)	-0.0048 (6)
C16	0.0113 (8)	0.0089 (8)	0.0155 (7)	-0.0024 (6)	-0.0001 (6)	-0.0038 (6)
O1W	0.0170 (7)	0.0319 (8)	0.0164 (6)	-0.0145 (6)	-0.0027 (5)	-0.0008 (6)
O2W	0.0138 (7)	0.0522 (11)	0.0270 (7)	-0.0106 (7)	-0.0007 (6)	-0.0222 (7)
O3W	0.0305 (9)	0.0669 (13)	0.0304 (8)	-0.0311 (9)	0.0039 (7)	-0.0205 (9)
O4WA	0.0301 (14)	0.046 (3)	0.0198 (11)	-0.0101 (13)	-0.0026 (9)	-0.0065 (11)
O4WB	0.027 (3)	0.045 (5)	0.030 (3)	-0.013 (3)	0.005 (3)	-0.005 (4)
O5WA	0.072 (6)	0.091 (7)	0.103 (7)	-0.024 (5)	-0.012 (5)	-0.058 (6)
O5WB	0.063 (6)	0.052 (6)	0.045 (5)	-0.022 (5)	-0.011 (4)	-0.017 (4)

Geometric parameters (Å, °)

Nd1—O7	2.3302 (13)	C1C—C2C	1.39
Nd1—O8 ⁱ	2.4214 (14)	C1C—C6C	1.39
Nd1—O10 ⁱⁱⁱ	2.4481 (14)	C1C—H1CA	0.95
Nd1—O1W	2.4612 (14)	C2C—C3C	1.39
Nd1—O9 ⁱⁱⁱ	2.4708 (13)	C2C—C15C	1.486 (13)
Nd1—O2W	2.4775 (15)	C3C—C4C	1.39
Nd1—O2	2.5223 (13)	C3C—H3CA	0.95
Nd1—O1	2.5743 (14)	C4C—C5C	1.39
Nd1—C7	2.8822 (18)	C5C—C6C	1.39
Nd1—O10 ⁱⁱⁱ	2.9332 (14)	C5C—H5CA	0.95
Nd1—C16 ⁱⁱⁱ	3.0749 (18)	C6C—C7	1.512 (12)
O3A—N1A	1.218 (6)	O1—C7	1.263 (2)
O4A—N1A	1.228 (4)	O2—C7	1.263 (2)
O5A—C15A	1.317 (5)	O7—C8	1.258 (2)
O5A—H5A	0.84	O8—C8	1.243 (2)
O6A—C15A	1.202 (6)	O8—Nd1 ⁱ	2.4214 (14)
N1A—C4A	1.454 (4)	O9—C16	1.263 (2)
C1A—C2A	1.39	O9—Nd1 ^{iv}	2.4708 (13)
C1A—C6A	1.39	O10—C16	1.261 (2)
C1A—H1AA	0.95	O10—Nd1 ⁱⁱ	2.4482 (14)
C2A—C3A	1.39	O10—Nd1 ^{iv}	2.9332 (14)
C2A—C15A	1.489 (4)	O11—N2	1.222 (2)
C3A—C4A	1.39	O12—N2	1.235 (2)
C3A—H3AA	0.95	N2—C13	1.468 (2)
C4A—C5A	1.39	C8—C9	1.505 (2)
C5A—C6A	1.39	C9—C14	1.391 (2)
C5A—H5AA	0.95	C9—C10	1.395 (2)
C6A—C7	1.518 (3)	C10—C11	1.393 (2)

O3B—N1B	1.203 (14)	C10—H10A	0.95
O4B—N1B	1.225 (12)	C11—C12	1.395 (2)
O5B—C15B	1.319 (13)	C11—C16	1.496 (2)
O5B—H5B	0.84	C12—C13	1.390 (2)
O6B—C15B	1.166 (12)	C12—H12A	0.95
N1B—C4B	1.469 (11)	C13—C14	1.386 (3)
C1B—C2B	1.39	C14—H14A	0.95
C1B—C6B	1.39	C16—Nd1 ^{iv}	3.0749 (18)
C1B—H1BA	0.95	O1W—H1W1	0.74
C2B—C3B	1.39	O1W—H2W1	0.94
C2B—C15B	1.511 (11)	O2W—H1W2	0.84
C3B—C4B	1.39	O2W—H2W2	0.82
C3B—H3BA	0.95	O3W—H1W3	0.91
C4B—C5B	1.39	O3W—H2W3	0.96
C5B—C6B	1.39	O4WA—H1W4	0.85
C5B—H5BA	0.95	O4WA—H2W4	0.85
C6B—C7	1.480 (9)	O4WB—H3W4	0.75
O3C—N1C	1.239 (16)	O4WB—H4W4	0.79
O4C—N1C	1.223 (15)	O5WA—H1W5	0.85
O5C—C15C	1.294 (15)	O5WA—H2W5	0.85
O5C—H5C	0.90	O5WB—H3W5	0.85
O6C—C15C	1.184 (16)	O5WB—H4W5	0.85
N1C—C4C	1.481 (13)		
O7—Nd1—O8 ⁱ	100.76 (5)	C5B—C6B—C1B	120.0
O7—Nd1—O10 ⁱⁱ	88.82 (5)	C5B—C6B—C7	118.1 (8)
O8 ⁱ —Nd1—O10 ⁱⁱⁱ	145.95 (5)	C1B—C6B—C7	121.1 (8)
O7—Nd1—O1W	84.42 (5)	O6B—C15B—O5B	124.3 (11)
O8 ⁱ —Nd1—O1W	139.99 (5)	O6B—C15B—C2B	124.3 (10)
O10 ⁱⁱ —Nd1—O1W	72.98 (5)	O5B—C15B—C2B	111.2 (10)
O7—Nd1—O9 ⁱⁱⁱ	151.36 (5)	C15C—O5C—H5C	121.2
O8 ⁱ —Nd1—O9 ⁱⁱⁱ	73.06 (5)	O4C—N1C—O3C	121.9 (17)
O10 ⁱⁱ —Nd1—O9 ⁱⁱⁱ	112.18 (5)	O4C—N1C—C4C	121.7 (16)
O1W—Nd1—O9 ⁱⁱⁱ	83.59 (5)	O3C—N1C—C4C	116.4 (15)
O7—Nd1—O2W	74.02 (5)	C2C—C1C—C6C	120.0
O8 ⁱ —Nd1—O2W	72.51 (6)	C2C—C1C—H1CA	120.0
O10 ⁱⁱ —Nd1—O2W	141.22 (5)	C6C—C1C—H1CA	120.0
O1W—Nd1—O2W	70.97 (5)	C3C—C2C—C1C	120.0
O9 ⁱⁱⁱ —Nd1—O2W	77.50 (5)	C3C—C2C—C15C	121.1 (10)
O7—Nd1—O2	123.94 (5)	C1C—C2C—C15C	118.9 (10)
O8 ⁱ —Nd1—O2	72.14 (5)	C4C—C3C—C2C	120.0
O10 ⁱⁱ —Nd1—O2	75.40 (5)	C4C—C3C—H3CA	120.0
O1W—Nd1—O2	136.59 (5)	C2C—C3C—H3CA	120.0
O9 ⁱⁱⁱ —Nd1—O2	81.61 (5)	C3C—C4C—C5C	120.0
O2W—Nd1—O2	142.83 (6)	C3C—C4C—N1C	117.9 (11)
O7—Nd1—O1	73.37 (5)	C5C—C4C—N1C	122.1 (11)
O8 ⁱ —Nd1—O1	70.84 (5)	C6C—C5C—C4C	120.0
O10 ⁱⁱ —Nd1—O1	81.06 (5)	C6C—C5C—H5CA	120.0

O1W—Nd1—O1	146.05 (5)	C4C—C5C—H5CA	120.0
O9 ⁱⁱⁱ —Nd1—O1	127.22 (5)	C5C—C6C—C1C	120.0
O2W—Nd1—O1	124.29 (5)	C5C—C6C—C7	115.6 (10)
O2—Nd1—O1	51.38 (4)	C1C—C6C—C7	123.6 (10)
O7—Nd1—C7	99.34 (5)	O6C—C15C—O5C	128.0 (16)
O8 ⁱ —Nd1—C7	65.51 (5)	O6C—C15C—C2C	118.1 (14)
O10 ⁱⁱ —Nd1—C7	80.77 (5)	O5C—C15C—C2C	113.8 (14)
O1W—Nd1—C7	153.43 (5)	C7—O1—Nd1	90.75 (11)
O9 ⁱⁱⁱ —Nd1—C7	102.98 (5)	C7—O2—Nd1	93.16 (11)
O2W—Nd1—C7	135.46 (6)	C8—O7—Nd1	171.21 (13)
O2—Nd1—C7	25.94 (5)	C8—O8—Nd1 ⁱ	136.37 (12)
O1—Nd1—C7	25.99 (5)	C16—O9—Nd1 ^{iv}	106.33 (11)
O7—Nd1—O10 ⁱⁱⁱ	146.58 (4)	C16—O10—Nd1 ⁱⁱ	160.39 (12)
O8 ⁱ —Nd1—O10 ⁱⁱⁱ	112.58 (4)	C16—O10—Nd1 ^{iv}	84.27 (11)
O10 ⁱⁱ —Nd1—O10 ⁱⁱⁱ	64.92 (5)	Nd1 ⁱⁱ —O10—Nd1 ^{iv}	115.08 (5)
O1W—Nd1—O10 ⁱⁱⁱ	68.84 (4)	O11—N2—O12	124.08 (17)
O9 ⁱⁱⁱ —Nd1—O10 ⁱⁱⁱ	47.27 (4)	O11—N2—C13	118.38 (17)
O2W—Nd1—O10 ⁱⁱⁱ	113.16 (4)	O12—N2—C13	117.54 (16)
O2—Nd1—O10 ⁱⁱⁱ	71.17 (4)	O2—C7—O1	122.11 (17)
O1—Nd1—O10 ⁱⁱⁱ	119.02 (4)	O2—C7—C6B	120.1 (5)
C7—Nd1—O10 ⁱⁱⁱ	96.50 (5)	O1—C7—C6B	117.7 (5)
O7—Nd1—C16 ⁱⁱⁱ	159.88 (5)	O2—C7—C6C	125.0 (7)
O8 ⁱ —Nd1—C16 ⁱⁱⁱ	92.18 (5)	O1—C7—C6C	112.3 (7)
O10 ⁱⁱ —Nd1—C16 ⁱⁱⁱ	88.96 (5)	O2—C7—C6A	117.2 (2)
O1W—Nd1—C16 ⁱⁱⁱ	75.82 (5)	O1—C7—C6A	120.5 (2)
O9 ⁱⁱⁱ —Nd1—C16 ⁱⁱⁱ	23.21 (5)	O2—C7—Nd1	60.91 (9)
O2W—Nd1—C16 ⁱⁱⁱ	95.64 (5)	O1—C7—Nd1	63.26 (10)
O2—Nd1—C16 ⁱⁱⁱ	74.63 (5)	C6B—C7—Nd1	166.1 (8)
O1—Nd1—C16 ⁱⁱⁱ	125.93 (4)	C6C—C7—Nd1	171.4 (11)
C7—Nd1—C16 ⁱⁱⁱ	100.03 (5)	C6A—C7—Nd1	160.6 (3)
O10 ⁱⁱⁱ —Nd1—C16 ⁱⁱⁱ	24.08 (4)	O8—C8—O7	125.51 (17)
C15A—O5A—H5A	109.5	O8—C8—C9	117.18 (16)
O3A—N1A—O4A	124.8 (4)	O7—C8—C9	117.30 (16)
O3A—N1A—C4A	118.3 (3)	C14—C9—C10	120.32 (16)
O4A—N1A—C4A	116.9 (4)	C14—C9—C8	120.22 (15)
C2A—C1A—C6A	120.0	C10—C9—C8	119.44 (16)
C2A—C1A—H1AA	120.0	C11—C10—C9	120.31 (16)
C6A—C1A—H1AA	120.0	C11—C10—H10A	119.8
C3A—C2A—C1A	120.0	C9—C10—H10A	119.8
C3A—C2A—C15A	119.6 (3)	C10—C11—C12	120.42 (16)
C1A—C2A—C15A	120.3 (3)	C10—C11—C16	119.21 (16)
C4A—C3A—C2A	120.0	C12—C11—C16	120.37 (16)
C4A—C3A—H3AA	120.0	C13—C12—C11	117.41 (16)
C2A—C3A—H3AA	120.0	C13—C12—H12A	121.3
C3A—C4A—C5A	120.0	C11—C12—H12A	121.3
C3A—C4A—N1A	120.1 (3)	C14—C13—C12	123.58 (16)
C5A—C4A—N1A	119.9 (3)	C14—C13—N2	118.54 (16)
C4A—C5A—C6A	120.0	C12—C13—N2	117.84 (16)

C4A—C5A—H5AA	120.0	C13—C14—C9	117.75 (16)
C6A—C5A—H5AA	120.0	C13—C14—H14A	121.1
C5A—C6A—C1A	120.0	C9—C14—H14A	121.1
C5A—C6A—C7	119.4 (2)	O10—C16—O9	122.04 (17)
C1A—C6A—C7	120.6 (2)	O10—C16—C11	119.94 (16)
O6A—C15A—O5A	124.1 (4)	O9—C16—C11	118.02 (16)
O6A—C15A—C2A	122.3 (3)	O10—C16—Nd1 ^{iv}	71.65 (10)
O5A—C15A—C2A	113.5 (4)	O9—C16—Nd1 ^{iv}	50.45 (9)
C15B—O5B—H5B	109.5	C11—C16—Nd1 ^{iv}	167.91 (13)
O3B—N1B—O4B	123.1 (13)	Nd1—O1W—H1W1	118.6
O3B—N1B—C4B	118.3 (12)	Nd1—O1W—H2W1	125.6
O4B—N1B—C4B	118.5 (12)	H1W1—O1W—H2W1	102.4
C2B—C1B—C6B	120.0	Nd1—O2W—H1W2	123.6
C2B—C1B—H1BA	120.0	Nd1—O2W—H2W2	127.6
C6B—C1B—H1BA	120.0	H1W2—O2W—H2W2	106.9
C3B—C2B—C1B	120.0	H1W3—O3W—H2W3	115.1
C3B—C2B—C15B	119.6 (8)	H1W4—O4WA—H2W4	107.4
C1B—C2B—C15B	120.3 (8)	H1W4—O4WA—H3W4	54.4
C2B—C3B—C4B	120.0	H2W4—O4WA—H3W4	64.4
C2B—C3B—H3BA	120.0	H1W4—O4WB—H2W4	97.0
C4B—C3B—H3BA	120.0	H1W4—O4WB—H3W4	60.6
C3B—C4B—C5B	120.0	H2W4—O4WB—H3W4	123.4
C3B—C4B—N1B	118.5 (9)	H1W4—O4WB—H4W4	164.0
C5B—C4B—N1B	121.5 (9)	H2W4—O4WB—H4W4	79.4
C6B—C5B—C4B	120.0	H3W4—O4WB—H4W4	108.3
C6B—C5B—H5BA	120.0	H1W5—O5WA—H2W5	107.4
C4B—C5B—H5BA	120.0	H3W5—O5WB—H4W5	107.7
C6A—C1A—C2A—C3A	0.0	C5B—C6B—C7—Nd1	73 (2)
C6A—C1A—C2A—C15A	-176.5 (5)	C1B—C6B—C7—Nd1	-96 (2)
C1A—C2A—C3A—C4A	0.0	C5C—C6C—C7—O2	-25.6 (17)
C15A—C2A—C3A—C4A	176.6 (5)	C1C—C6C—C7—O2	164.7 (10)
C2A—C3A—C4A—C5A	0.0	C5C—C6C—C7—O1	163.1 (9)
C2A—C3A—C4A—N1A	-177.8 (4)	C1C—C6C—C7—O1	-6.7 (19)
O3A—N1A—C4A—C3A	167.1 (4)	C5C—C6C—C7—C6B	33 (8)
O4A—N1A—C4A—C3A	-13.6 (6)	C1C—C6C—C7—C6B	-137 (10)
O3A—N1A—C4A—C5A	-10.7 (6)	C5C—C6C—C7—C6A	36 (3)
O4A—N1A—C4A—C5A	168.6 (4)	C1C—C6C—C7—C6A	-134 (5)
C3A—C4A—C5A—C6A	0.0	C5A—C6A—C7—O2	3.6 (4)
N1A—C4A—C5A—C6A	177.8 (4)	C1A—C6A—C7—O2	-178.3 (3)
C4A—C5A—C6A—C1A	0.0	C5A—C6A—C7—O1	178.6 (2)
C4A—C5A—C6A—C7	178.2 (5)	C1A—C6A—C7—O1	-3.2 (5)
C2A—C1A—C6A—C5A	0.0	C5A—C6A—C7—C6B	-117 (7)
C2A—C1A—C6A—C7	-178.2 (5)	C1A—C6A—C7—C6B	61 (7)
C3A—C2A—C15A—O6A	-174.7 (5)	C5A—C6A—C7—C6C	-122 (4)
C1A—C2A—C15A—O6A	1.8 (8)	C1A—C6A—C7—C6C	56 (4)
C3A—C2A—C15A—O5A	2.6 (6)	C5A—C6A—C7—Nd1	82.8 (6)
C1A—C2A—C15A—O5A	179.2 (4)	C1A—C6A—C7—Nd1	-99.0 (5)

C6B—C1B—C2B—C3B	0.0	O7—Nd1—C7—O2	-162.80 (11)
C6B—C1B—C2B—C15B	-177.4 (17)	O8 ⁱ —Nd1—C7—O2	99.56 (12)
C1B—C2B—C3B—C4B	0.0	O10 ⁱⁱ —Nd1—C7—O2	-75.54 (11)
C15B—C2B—C3B—C4B	177.4 (17)	O1W—Nd1—C7—O2	-66.56 (17)
C2B—C3B—C4B—C5B	0.0	O9 ⁱⁱⁱ —Nd1—C7—O2	35.26 (12)
C2B—C3B—C4B—N1B	177.8 (12)	O2W—Nd1—C7—O2	120.53 (12)
O3B—N1B—C4B—C3B	-170.2 (14)	O1—Nd1—C7—O2	-164.04 (19)
O4B—N1B—C4B—C3B	7.7 (19)	O10 ⁱⁱⁱ —Nd1—C7—O2	-12.32 (12)
O3B—N1B—C4B—C5B	8 (2)	C16 ⁱⁱⁱ —Nd1—C7—O2	11.75 (12)
O4B—N1B—C4B—C5B	-174.5 (13)	O7—Nd1—C7—O1	1.25 (12)
C3B—C4B—C5B—C6B	0.0	O8 ⁱ —Nd1—C7—O1	-96.39 (12)
N1B—C4B—C5B—C6B	-177.8 (12)	O10 ⁱⁱ —Nd1—C7—O1	88.51 (11)
C4B—C5B—C6B—C1B	0.0	O1W—Nd1—C7—O1	97.48 (15)
C4B—C5B—C6B—C7	-169.7 (17)	O9 ⁱⁱⁱ —Nd1—C7—O1	-160.70 (11)
C2B—C1B—C6B—C5B	0.0	O2W—Nd1—C7—O1	-75.42 (13)
C2B—C1B—C6B—C7	169.3 (17)	O2—Nd1—C7—O1	164.04 (19)
C3B—C2B—C15B—O6B	178.9 (11)	O10 ⁱⁱⁱ —Nd1—C7—O1	151.72 (11)
C1B—C2B—C15B—O6B	-4 (2)	C16 ⁱⁱⁱ —Nd1—C7—O1	175.79 (11)
C3B—C2B—C15B—O5B	-6.3 (17)	O7—Nd1—C7—C6B	99 (2)
C1B—C2B—C15B—O5B	171.1 (13)	O8 ⁱ —Nd1—C7—C6B	1 (2)
C6C—C1C—C2C—C3C	0.0	O10 ⁱⁱ —Nd1—C7—C6B	-174 (2)
C6C—C1C—C2C—C15C	179.4 (19)	O1W—Nd1—C7—C6B	-165 (2)
C1C—C2C—C3C—C4C	0.0	O9 ⁱⁱⁱ —Nd1—C7—C6B	-63 (2)
C15C—C2C—C3C—C4C	-179 (2)	O2W—Nd1—C7—C6B	22 (2)
C2C—C3C—C4C—C5C	0.0	O2—Nd1—C7—C6B	-98 (2)
C2C—C3C—C4C—N1C	178 (2)	O1—Nd1—C7—C6B	98 (2)
O4C—N1C—C4C—C3C	5 (3)	O10 ⁱⁱⁱ —Nd1—C7—C6B	-111 (2)
O3C—N1C—C4C—C3C	-176 (2)	C16 ⁱⁱⁱ —Nd1—C7—C6B	-86 (2)
O4C—N1C—C4C—C5C	-177 (2)	O7—Nd1—C7—C6A	107.5 (5)
O3C—N1C—C4C—C5C	2 (3)	O8 ⁱ —Nd1—C7—C6A	9.9 (5)
C3C—C4C—C5C—C6C	0.0	O10 ⁱⁱ —Nd1—C7—C6A	-165.2 (5)
N1C—C4C—C5C—C6C	-178 (2)	O1W—Nd1—C7—C6A	-156.2 (5)
C4C—C5C—C6C—C1C	0.0	O9 ⁱⁱⁱ —Nd1—C7—C6A	-54.4 (5)
C4C—C5C—C6C—C7	-170 (2)	O2W—Nd1—C7—C6A	30.9 (5)
C2C—C1C—C6C—C5C	0.0	O2—Nd1—C7—C6A	-89.7 (5)
C2C—C1C—C6C—C7	169 (2)	O1—Nd1—C7—C6A	106.3 (5)
C3C—C2C—C15C—O6C	149.3 (18)	O10 ⁱⁱⁱ —Nd1—C7—C6A	-102.0 (5)
C1C—C2C—C15C—O6C	-30 (3)	C16 ⁱⁱⁱ —Nd1—C7—C6A	-77.9 (5)
C3C—C2C—C15C—O5C	-28 (3)	Nd1 ⁱ —O8—C8—O7	4.0 (3)
C1C—C2C—C15C—O5C	152.6 (16)	Nd1 ⁱ —O8—C8—C9	-175.30 (12)
O7—Nd1—O1—C7	-178.71 (12)	O8—C8—C9—C14	-156.86 (17)
O8 ⁱ —Nd1—O1—C7	73.22 (11)	O7—C8—C9—C14	23.8 (2)
O10 ⁱⁱ —Nd1—O1—C7	-87.26 (11)	O8—C8—C9—C10	24.8 (2)
O1W—Nd1—O1—C7	-127.43 (12)	O7—C8—C9—C10	-154.54 (17)
O9 ⁱⁱⁱ —Nd1—O1—C7	23.86 (13)	C14—C9—C10—C11	-3.1 (3)
O2W—Nd1—O1—C7	124.75 (12)	C8—C9—C10—C11	175.22 (15)
O2—Nd1—O1—C7	-8.85 (11)	C9—C10—C11—C12	4.5 (3)
O10 ⁱⁱⁱ —Nd1—O1—C7	-32.57 (12)	C9—C10—C11—C16	-174.87 (16)

C16 ⁱⁱⁱ —Nd1—O1—C7	-5.12 (13)	C10—C11—C12—C13	-1.6 (3)
O7—Nd1—O2—C7	20.60 (13)	C16—C11—C12—C13	177.77 (16)
O8 ⁱ —Nd1—O2—C7	-70.53 (12)	C11—C12—C13—C14	-2.8 (3)
O10 ⁱⁱ —Nd1—O2—C7	99.01 (12)	C11—C12—C13—N2	179.38 (16)
O1W—Nd1—O2—C7	143.33 (11)	O11—N2—C13—C14	160.22 (19)
O9 ⁱⁱⁱ —Nd1—O2—C7	-145.35 (12)	O12—N2—C13—C14	-19.7 (3)
O2W—Nd1—O2—C7	-89.17 (13)	O11—N2—C13—C12	-21.9 (3)
O1—Nd1—O2—C7	8.87 (11)	O12—N2—C13—C12	158.19 (17)
O10 ⁱⁱⁱ —Nd1—O2—C7	167.05 (12)	C12—C13—C14—C9	4.1 (3)
C16 ⁱⁱⁱ —Nd1—O2—C7	-168.00 (12)	N2—C13—C14—C9	-178.07 (16)
Nd1—O2—C7—O1	-16.8 (2)	C10—C9—C14—C13	-1.1 (3)
Nd1—O2—C7—C6B	164.1 (9)	C8—C9—C14—C13	-179.40 (16)
Nd1—O2—C7—C6C	172.6 (13)	Nd1 ⁱⁱ —O10—C16—O9	173.8 (3)
Nd1—O2—C7—C6A	158.1 (3)	Nd1 ^{iv} —O10—C16—O9	2.76 (17)
Nd1—O1—C7—O2	16.47 (19)	Nd1 ⁱⁱ —O10—C16—C11	-5.2 (5)
Nd1—O1—C7—C6B	-164.4 (9)	Nd1 ^{iv} —O10—C16—C11	-176.24 (15)
Nd1—O1—C7—C6C	-171.9 (12)	Nd1 ⁱⁱ —O10—C16—Nd1 ^{iv}	171.0 (4)
Nd1—O1—C7—C6A	-158.3 (3)	Nd1 ^{iv} —O9—C16—O10	-3.4 (2)
C5B—C6B—C7—O2	-17.0 (11)	Nd1 ^{iv} —O9—C16—C11	175.62 (12)
C1B—C6B—C7—O2	173.5 (8)	C10—C11—C16—O10	145.14 (17)
C5B—C6B—C7—O1	163.9 (5)	C12—C11—C16—O10	-34.2 (2)
C1B—C6B—C7—O1	-5.6 (16)	C10—C11—C16—O9	-33.9 (2)
C5B—C6B—C7—C6C	-143 (9)	C12—C11—C16—O9	146.75 (17)
C1B—C6B—C7—C6C	47 (8)	C10—C11—C16—Nd1 ^{iv}	-17.6 (7)
C5B—C6B—C7—C6A	45 (7)	C12—C11—C16—Nd1 ^{iv}	163.1 (5)
C1B—C6B—C7—C6A	-124 (8)		

Symmetry codes: (i) $-x, -y+1, -z+1$; (ii) $-x, -y+2, -z+1$; (iii) $x+1, y-1, z$; (iv) $x-1, y+1, z$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O5A—H5A \cdots O3W ^v	0.84	1.76	2.596 (5)	178
O1W—H1W1 \cdots O2 ^{vi}	0.74	1.99	2.731 (3)	175
O1W—H2W1 \cdots O4WA	0.94	2.00	2.849 (4)	149
O2W—H1W2 \cdots O4WA	0.84	1.86	2.667 (3)	159
O2W—H2W2 \cdots O1 ⁱ	0.82	2.06	2.874 (3)	172
O3W—H1W3 \cdots O11 ^{vii}	0.91	2.56	3.246 (3)	132
O3W—H1W3 \cdots O12 ^{vii}	0.91	2.05	2.902 (3)	156
O3W—H2W3 \cdots O9 ⁱⁱⁱ	0.96	1.76	2.688 (2)	162
O4WA—H2W4 \cdots O3W	0.85	2.51	3.118 (6)	130
C12—H12A \cdots O1 ⁱⁱ	0.95	2.41	3.351 (2)	169

Symmetry codes: (i) $-x, -y+1, -z+1$; (ii) $-x, -y+2, -z+1$; (iii) $x+1, y-1, z$; (v) $x-1, y, z+1$; (vi) $-x+1, -y+1, -z+1$; (vii) $x, y-1, z$.