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## Structure Reports

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**(E)-2-[(6-Ethoxybenzothiazol-2-yl)imino-methyl]-6-methoxyphenol**

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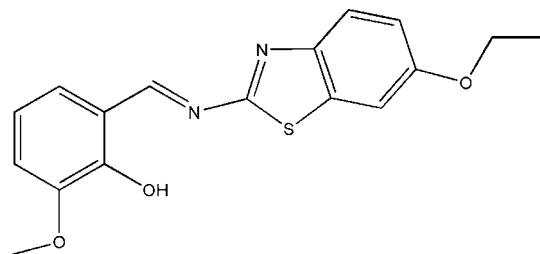
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Key indicators: single-crystal X-ray study;  $T = 298$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.044;  $wR$  factor = 0.114; data-to-parameter ratio = 13.0.

In the title molecule,  $\text{C}_{17}\text{H}_{16}\text{N}_2\text{O}_3\text{S}$ , the benzothiazole fragment and the benzene ring form a dihedral angle of  $13.8$  (4)°, and an intramolecular  $\text{O}-\text{H}\cdots\text{N}$  hydrogen bond occurs. In the crystal structure, pairs of weak intermolecular  $\text{O}-\text{H}\cdots\text{S}$  and  $\text{C}-\text{H}\cdots(\text{O},\text{O})$  hydrogen bonds link molecules into centrosymmetric dimers. These dimers are related by translation along the  $a$  axis and form stacks *via*  $\pi-\pi$  interactions, with a short intermolecular distance of  $3.766$  (5) Å between the centroids of the benzene and thiazole rings.

## Related literature

For a related crystal structure, see: Zhao *et al.* (2008). For details of the crystallography and coordination chemistry of Schiff base compounds, see: Garnovski *et al.* (1993).



## Experimental

## Crystal data

$\text{C}_{17}\text{H}_{16}\text{N}_2\text{O}_3\text{S}$   
 $M_r = 328.38$   
 Triclinic,  $P\bar{1}$   
 $a = 6.0178$  (14) Å  
 $b = 10.941$  (3) Å  
 $c = 12.164$  (3) Å  
 $\alpha = 85.479$  (4)°  
 $\beta = 83.693$  (5)°  
 $\gamma = 76.486$  (3)°  
 $V = 772.9$  (3) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.23$  mm<sup>-1</sup>  
 $T = 298$  K  
 $0.12 \times 0.08 \times 0.06$  mm

## Data collection

Bruker SMART APEX diffractometer  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.973$ ,  $T_{\max} = 0.987$   
 4102 measured reflections  
 2720 independent reflections  
 1911 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$   
 $wR(F^2) = 0.114$   
 $S = 1.03$   
 2720 reflections  
 209 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.18$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.21$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O1}-\text{H1}\cdots\text{N1}$	0.82	1.88	2.606 (3)	147
$\text{O1}-\text{H1}\cdots\text{S1}^{\dagger}$	0.82	2.92	3.1746 (18)	100
$\text{C12}-\text{H12}\cdots\text{O1}^{\dagger}$	0.93	2.59	3.328 (3)	136
$\text{C12}-\text{H12}\cdots\text{O2}^{\dagger}$	0.93	2.60	3.491 (3)	160

Symmetry code: (i)  $-x + 1, -y + 1, -z + 1$ .

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV2526).

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## supporting information

*Acta Cryst.* (2009). E65, o832 [doi:10.1107/S1600536809009337]

**(*E*)-2-[(6-Ethoxybenzothiazol-2-yl)iminomethyl]-6-methoxyphenol****Ling-Qian Kong****S1. Comment**

Recently, a number of Schiff base compounds have been investigated in terms of their crystallography and coordination chemistry (Garnovski *et al.*, 1993). In order to continue our studies on Schiff bases, we now report the synthesis and crystal structure of the title compound, (I).

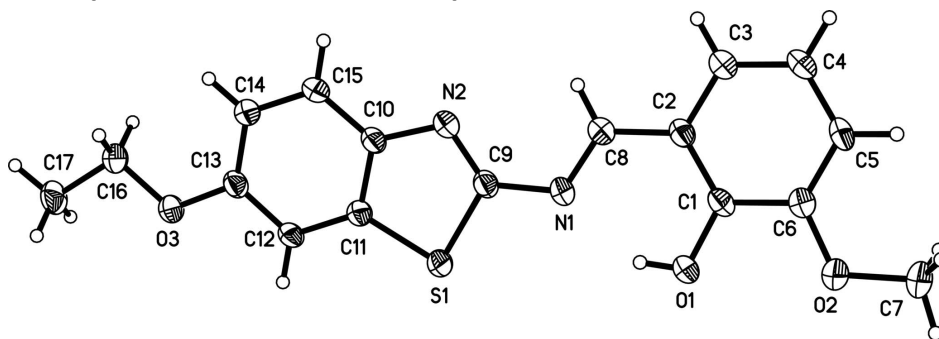
In (I) (Fig. 1), all the geometric parameters are in a good agreement with those found in (*E*)-2-methoxy-6-[(5-methylisoxazol-3-yl)-iminomethyl] phenol (Zhao *et al.*, 2008). The benzene and the benzothiazole rings make a dihedral angle of 13.8 (4)° showing that the Schiff base ligand adopts a non-planar conformation in the case. Moreover, weak intermolecular O—H···S and C—H···O hydrogen bonds (Table 1) link the molecules into centrosymmetric dimers. These dimers related by translation along axis *a* form stacks *via*  $\pi$ - $\pi$  interactions proved by short intermolecular distance of 3.766 (5) Å between the centroids of benzene and thiazole rings.

**S2. Experimental**

The title compound was synthesized by the reaction of 2-hydroxy-3-methoxybenzaldehyde (0.152 g, 1 mmol) and 6-ethoxybenzothiazol-2-amine (0.194 g, 1 mmol) in ethanol solution and stirred under reflux conditions (353 K) for 5 h. When cooled to room temperature the solution was filtered and after a week yellow crystals suitable for X-ray diffraction study were obtained. Yield, 0.283 g, 86%. m.p. 342–344 K.

**S3. Refinement**

The H atoms were included in the riding-model approximation with C—H = 0.93 Å, C—H = 0.96 Å and O—H = 0.82 Å, and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C-aromatic})$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C-methyl, methylene and O})$ .

**Figure 1**

The molecular structure of (I) showing 30% probability displacement ellipsoids and the atom-numbering scheme.

**(E)-2-[(6-Ethoxybenzothiazol-2-yl)iminomethyl]-6-methoxyphenol***Crystal data*C<sub>17</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>S $M_r = 328.38$ Triclinic,  $P\bar{1}$ 

Hall symbol: -P 1

 $a = 6.0178$  (14) Å $b = 10.941$  (3) Å $c = 12.164$  (3) Å $\alpha = 85.479$  (4)° $\beta = 83.693$  (5)° $\gamma = 76.486$  (3)° $V = 772.9$  (3) Å<sup>3</sup> $Z = 2$  $F(000) = 344$  $D_x = 1.411$  Mg m<sup>-3</sup>Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 982 reflections

 $\theta = 2.6$ – $22.3$ ° $\mu = 0.23$  mm<sup>-1</sup> $T = 298$  K

Block, yellow

 $0.12 \times 0.08 \times 0.06$  mm*Data collection*

Bruker SMART APEX

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 $\varphi$  and  $\omega$  scans

Absorption correction: multi-scan

(SADABS; Sheldrick, 1996)

 $T_{\min} = 0.973$ ,  $T_{\max} = 0.987$ 

4102 measured reflections

2720 independent reflections

1911 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.020$  $\theta_{\text{max}} = 25.1$ °,  $\theta_{\text{min}} = 1.7$ ° $h = -7 \rightarrow 7$  $k = -12 \rightarrow 12$  $l = -14 \rightarrow 9$ *Refinement*Refinement on  $F^2$ 

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.044$  $wR(F^2) = 0.114$  $S = 1.03$ 

2720 reflections

209 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0548P)^2 + 0.0301P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\text{max}} < 0.001$  $\Delta\rho_{\text{max}} = 0.18$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.21$  e Å<sup>-3</sup>*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>)*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.46082 (11)	0.40652 (6)	0.37670 (5)	0.0489 (2)
O1	0.8898 (3)	0.68081 (19)	0.43648 (14)	0.0622 (5)
H1	0.8120	0.6363	0.4157	0.093*

O2	1.1451 (3)	0.83142 (17)	0.47463 (15)	0.0609 (5)
O3	-0.0448 (3)	0.15300 (16)	0.22235 (13)	0.0528 (5)
N1	0.7555 (3)	0.54862 (17)	0.29849 (16)	0.0439 (5)
N2	0.5822 (3)	0.45416 (18)	0.16878 (16)	0.0447 (5)
C1	1.0271 (4)	0.7137 (2)	0.3503 (2)	0.0437 (6)
C2	1.0358 (4)	0.6701 (2)	0.2444 (2)	0.0413 (6)
C3	1.1832 (4)	0.7085 (2)	0.1588 (2)	0.0498 (7)
H3	1.1915	0.6787	0.0886	0.060*
C4	1.3155 (4)	0.7898 (2)	0.1772 (2)	0.0539 (7)
H4	1.4111	0.8161	0.1194	0.065*
C5	1.3068 (4)	0.8329 (2)	0.2824 (2)	0.0518 (7)
H5	1.3978	0.8877	0.2945	0.062*
C6	1.1657 (4)	0.7957 (2)	0.3686 (2)	0.0458 (6)
C7	1.2911 (5)	0.9082 (3)	0.5008 (2)	0.0687 (9)
H7A	1.4484	0.8668	0.4822	0.103*
H7B	1.2668	0.9221	0.5786	0.103*
H7C	1.2566	0.9876	0.4593	0.103*
C8	0.8943 (4)	0.5867 (2)	0.2223 (2)	0.0436 (6)
H8	0.9036	0.5596	0.1511	0.052*
C9	0.6154 (4)	0.4739 (2)	0.2685 (2)	0.0413 (6)
C10	0.4252 (4)	0.3792 (2)	0.17276 (19)	0.0406 (6)
C11	0.3392 (4)	0.3421 (2)	0.27927 (19)	0.0397 (6)
C12	0.1811 (4)	0.2670 (2)	0.2947 (2)	0.0418 (6)
H12	0.1246	0.2435	0.3655	0.050*
C13	0.1098 (4)	0.2280 (2)	0.2018 (2)	0.0423 (6)
C14	0.1905 (4)	0.2659 (2)	0.0958 (2)	0.0473 (6)
H14	0.1394	0.2394	0.0343	0.057*
C15	0.3456 (4)	0.3422 (2)	0.0812 (2)	0.0477 (6)
H15	0.3963	0.3686	0.0103	0.057*
C16	-0.1092 (4)	0.1015 (2)	0.1306 (2)	0.0520 (7)
H16A	0.0253	0.0516	0.0905	0.062*
H16B	-0.1827	0.1684	0.0805	0.062*
C17	-0.2723 (4)	0.0206 (2)	0.1744 (2)	0.0595 (8)
H17A	-0.1959	-0.0472	0.2216	0.089*
H17B	-0.3230	-0.0133	0.1137	0.089*
H17C	-0.4023	0.0703	0.2160	0.089*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0572 (4)	0.0570 (4)	0.0411 (4)	-0.0323 (3)	-0.0019 (3)	0.0001 (3)
O1	0.0725 (13)	0.0819 (14)	0.0470 (11)	-0.0529 (11)	0.0118 (9)	-0.0104 (10)
O2	0.0722 (13)	0.0712 (12)	0.0521 (11)	-0.0419 (10)	-0.0002 (9)	-0.0115 (10)
O3	0.0598 (11)	0.0620 (11)	0.0476 (10)	-0.0381 (9)	0.0015 (8)	-0.0062 (9)
N1	0.0431 (11)	0.0474 (12)	0.0461 (12)	-0.0220 (10)	-0.0008 (10)	-0.0009 (10)
N2	0.0443 (12)	0.0506 (12)	0.0429 (12)	-0.0211 (10)	0.0007 (9)	-0.0006 (10)
C1	0.0410 (13)	0.0452 (14)	0.0464 (15)	-0.0183 (12)	0.0033 (11)	0.0027 (12)
C2	0.0393 (13)	0.0419 (14)	0.0443 (14)	-0.0154 (11)	-0.0021 (11)	0.0033 (11)

C3	0.0490 (15)	0.0566 (16)	0.0474 (15)	-0.0225 (13)	0.0015 (12)	-0.0016 (13)
C4	0.0519 (16)	0.0616 (17)	0.0519 (16)	-0.0289 (14)	0.0097 (13)	0.0014 (14)
C5	0.0498 (15)	0.0530 (16)	0.0602 (17)	-0.0306 (13)	0.0020 (13)	-0.0012 (14)
C6	0.0473 (14)	0.0462 (14)	0.0473 (15)	-0.0183 (12)	-0.0024 (12)	-0.0025 (12)
C7	0.081 (2)	0.0717 (19)	0.0689 (19)	-0.0453 (17)	-0.0054 (16)	-0.0165 (16)
C8	0.0402 (13)	0.0486 (15)	0.0432 (14)	-0.0146 (12)	0.0006 (11)	-0.0032 (12)
C9	0.0391 (13)	0.0418 (14)	0.0453 (15)	-0.0158 (11)	-0.0010 (11)	-0.0010 (12)
C10	0.0390 (13)	0.0435 (14)	0.0418 (14)	-0.0177 (11)	0.0031 (11)	-0.0028 (11)
C11	0.0414 (13)	0.0401 (13)	0.0394 (13)	-0.0140 (11)	-0.0008 (11)	-0.0035 (11)
C12	0.0450 (14)	0.0446 (14)	0.0390 (13)	-0.0208 (12)	0.0046 (11)	-0.0012 (11)
C13	0.0403 (13)	0.0423 (14)	0.0475 (15)	-0.0181 (11)	0.0011 (11)	-0.0024 (12)
C14	0.0487 (15)	0.0580 (16)	0.0409 (14)	-0.0245 (13)	0.0004 (11)	-0.0070 (12)
C15	0.0490 (15)	0.0600 (16)	0.0381 (14)	-0.0248 (13)	0.0042 (11)	-0.0014 (12)
C16	0.0538 (16)	0.0576 (16)	0.0527 (16)	-0.0279 (13)	-0.0038 (13)	-0.0075 (13)
C17	0.0587 (17)	0.0564 (17)	0.0731 (19)	-0.0333 (14)	-0.0030 (15)	-0.0058 (15)

*Geometric parameters (Å, °)*

S1—C11	1.732 (2)	C5—H5	0.9300
S1—C9	1.743 (2)	C7—H7A	0.9600
O1—C1	1.342 (3)	C7—H7B	0.9600
O1—H1	0.8200	C7—H7C	0.9600
O2—C6	1.361 (3)	C8—H8	0.9300
O2—C7	1.425 (3)	C10—C15	1.383 (3)
O3—C13	1.370 (3)	C10—C11	1.407 (3)
O3—C16	1.417 (3)	C11—C12	1.385 (3)
N1—C8	1.289 (3)	C12—C13	1.382 (3)
N1—C9	1.396 (3)	C12—H12	0.9300
N2—C9	1.293 (3)	C13—C14	1.394 (3)
N2—C10	1.383 (3)	C14—C15	1.380 (3)
C1—C2	1.399 (3)	C14—H14	0.9300
C1—C6	1.405 (3)	C15—H15	0.9300
C2—C3	1.397 (3)	C16—C17	1.500 (3)
C2—C8	1.443 (3)	C16—H16A	0.9700
C3—C4	1.370 (3)	C16—H16B	0.9700
C3—H3	0.9300	C17—H17A	0.9600
C4—C5	1.390 (4)	C17—H17B	0.9600
C4—H4	0.9300	C17—H17C	0.9600
C5—C6	1.373 (3)		
C11—S1—C9	88.69 (11)	N2—C9—N1	126.4 (2)
C1—O1—H1	109.5	N2—C9—S1	117.12 (17)
C6—O2—C7	117.7 (2)	N1—C9—S1	116.38 (18)
C13—O3—C16	117.82 (18)	C15—C10—N2	124.9 (2)
C8—N1—C9	118.2 (2)	C15—C10—C11	119.1 (2)
C9—N2—C10	109.4 (2)	N2—C10—C11	115.9 (2)
O1—C1—C2	122.7 (2)	C12—C11—C10	121.7 (2)
O1—C1—C6	117.7 (2)	C12—C11—S1	129.48 (19)

C2—C1—C6	119.6 (2)	C10—C11—S1	108.82 (16)
C3—C2—C1	119.4 (2)	C13—C12—C11	118.0 (2)
C3—C2—C8	119.6 (2)	C13—C12—H12	121.0
C1—C2—C8	121.1 (2)	C11—C12—H12	121.0
C4—C3—C2	120.6 (2)	O3—C13—C12	115.4 (2)
C4—C3—H3	119.7	O3—C13—C14	123.7 (2)
C2—C3—H3	119.7	C12—C13—C14	120.9 (2)
C3—C4—C5	120.0 (2)	C15—C14—C13	120.6 (2)
C3—C4—H4	120.0	C15—C14—H14	119.7
C5—C4—H4	120.0	C13—C14—H14	119.7
C6—C5—C4	120.8 (2)	C14—C15—C10	119.6 (2)
C6—C5—H5	119.6	C14—C15—H15	120.2
C4—C5—H5	119.6	C10—C15—H15	120.2
O2—C6—C5	125.7 (2)	O3—C16—C17	107.7 (2)
O2—C6—C1	114.7 (2)	O3—C16—H16A	110.2
C5—C6—C1	119.7 (2)	C17—C16—H16A	110.2
O2—C7—H7A	109.5	O3—C16—H16B	110.2
O2—C7—H7B	109.5	C17—C16—H16B	110.2
H7A—C7—H7B	109.5	H16A—C16—H16B	108.5
O2—C7—H7C	109.5	C16—C17—H17A	109.5
H7A—C7—H7C	109.5	C16—C17—H17B	109.5
H7B—C7—H7C	109.5	H17A—C17—H17B	109.5
N1—C8—C2	121.9 (2)	C16—C17—H17C	109.5
N1—C8—H8	119.0	H17A—C17—H17C	109.5
C2—C8—H8	119.0	H17B—C17—H17C	109.5

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O1—H1 $\cdots$ N1	0.82	1.88	2.606 (3)	147
O1—H1 $\cdots$ S1 <sup>i</sup>	0.82	2.92	3.1746 (18)	100
C12—H12 $\cdots$ O1 <sup>i</sup>	0.93	2.59	3.328 (3)	136
C12—H12 $\cdots$ O2 <sup>i</sup>	0.93	2.60	3.491 (3)	160

Symmetry code: (i)  $-x+1, -y+1, -z+1$ .