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## Structure Reports

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# *N,N'*-Bis(4-methoxybenzylidene)-4,4'-(*m*-phenylenedioxy)dianiline

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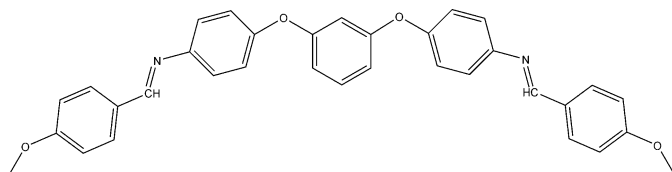
Received 21 February 2009; accepted 24 March 2009

Key indicators: single-crystal X-ray study;  $T = 163$  K; mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å;  $R$  factor = 0.052;  $wR$  factor = 0.142; data-to-parameter ratio = 7.3.

Molecules of the title compound,  $\text{C}_{34}\text{H}_{28}\text{N}_2\text{O}_4$ , a Schiff base precursor for macrocycles, are located on a mirror plane. The  $\text{C}=\text{N}$  double bond is *trans* configured. Intermolecular  $\text{C}-\text{H}\cdots\text{O}$  interactions stabilize the crystal packing.

## Related literature

For the importance of Schiff base macrocycles in macrocyclic and supramolecular chemistry, see: Ali *et al.* (2008).



## Experimental

### Crystal data

$\text{C}_{34}\text{H}_{28}\text{N}_2\text{O}_4$   
 $M_r = 528.58$

Orthorhombic,  $Cmc2_1$   
 $a = 59.344$  (13) Å

$b = 7.484$  (3) Å  
 $c = 5.988$  (2) Å  
 $V = 2659.4$  (15) Å<sup>3</sup>  
 $Z = 4$

Mo  $K\alpha$  radiation  
 $\mu = 0.09$  mm<sup>-1</sup>  
 $T = 163$  K  
 $0.60 \times 0.41 \times 0.02$  mm

### Data collection

Rigaku AFC8S Mercury CCD diffractometer  
Absorption correction: multi-scan (Jacobson, 1998)  
 $T_{\min} = 0.837$ ,  $T_{\max} = 1.000$   
(expected range = 0.836–0.998)

7742 measured reflections  
1351 independent reflections  
1164 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.046$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.052$   
 $wR(F^2) = 0.142$   
 $S = 1.09$   
1351 reflections  
185 parameters

1 restraint  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.29$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.23$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C8}-\text{H8B}\cdots\text{O7}^i$	0.96	2.57	3.411 (5)	147
$\text{C13}-\text{H13}\cdots\text{O17}^{ii}$	0.96	2.52	3.405 (4)	154

Symmetry codes: (i)  $-x + \frac{1}{2}, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (ii)  $x, -y, z - \frac{1}{2}$ .

Data collection: *CrystalClear* (Rigaku, 2001); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT2885).

## References

- Ali, Q., Hussain, Z., Shah, M. R. & VanDerveer, D. (2008). *Acta Cryst.* **E64**, o1377.  
Jacobson, R. (1998). Private communication to the Rigaku Corporation, Tokyo, Japan.  
Rigaku (2001). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

## supporting information

*Acta Cryst.* (2009). E65, o897 [doi:10.1107/S1600536809010939]

## *N,N'*-Bis(4-methoxybenzylidene)-4,4'-(*m*-phenylenedioxy)dianiline

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### S1. Comment

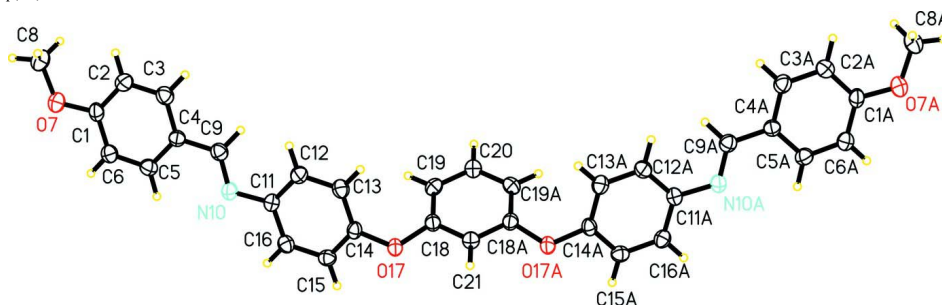
Schiff base macrocycles have been of great importance in macrocyclic and supramolecular chemistry (Ali *et al.*, 2008). Here we report the synthesis and crystal structure of the title compound which is a precursor for the synthesis of a Schiff base macrocycle. The crystal packing is stabilized by some short C—H···O contacts and van der Waals interactions. The methoxy group is coplanar with the benzene ring as revealed by the C6—C1—O7—C8 torsion of 175.5 (3)°. The methoxy oxygen atom acts as an acceptor for a weak C—H···O hydrogen bond from a neighboring molecule. C—H···O interactions stabilize the crystal packing.

### S2. Experimental

100 mg (0.34 mmol) of 4,4'-(1,3-phenylenebis(oxy))dianiline was dissolved in 2 ml of dichloromethane and then a solution of 4-methoxybenzaldehyde (0.1 ml, 0.85 mmol) in 2 ml of dichloromethane was added dropwise with stirring. The reaction was stirred at 330 K for 30 min and cooled to room temperature. The solvent was then removed using a rotary evaporator to give a crude solid. The solid was dissolved in dichloromethane and slow evaporation of the dichloromethane afforded needle like crystals in 80% yield.

### S3. Refinement

In the absence of anomalous scatterers Friedel pairs had been merged. All H atoms were geometrically fixed and allowed to ride on the corresponding non-H atom with C—H = 0.96 Å,  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  of the attached C atom for methyl H atoms and  $1.2U_{\text{eq}}(\text{C})$  for other H atoms.



**Figure 1**

Crystal Structure of the title compound; anisotropic displacement ellipsoid plot, 50% probability; symmetry operator for generating equivalent atoms  $-x, y, z$ ).

***N,N'*-Bis(4-methoxybenzylidene)-4,4'-(*m*-phenylenedioxy)dianiline***Crystal data*C<sub>34</sub>H<sub>28</sub>N<sub>2</sub>O<sub>4</sub> $M_r = 528.58$ Orthorhombic, *Cmc*2<sub>1</sub> $a = 59.344$  (13) Å $b = 7.484$  (3) Å $c = 5.988$  (2) Å $V = 2659.4$  (15) Å<sup>3</sup> $Z = 4$  $F(000) = 1112$  $D_x = 1.320$  Mg m<sup>-3</sup>Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 3132 reflections

 $\theta = 2.7$ – $26.4^\circ$  $\mu = 0.09$  mm<sup>-1</sup> $T = 163$  K

Plate, colorless

 $0.60 \times 0.41 \times 0.02$  mm*Data collection*Rigaku AFC8S Mercury CCD  
diffractometer

Radiation source: Sealed Tube

Graphite Monochromator monochromator

Detector resolution: 14.6306 pixels mm<sup>-1</sup> $\omega$  scansAbsorption correction: multi-scan  
(Jacobson, 1998) $T_{\min} = 0.837$ ,  $T_{\max} = 1.000$ 

7742 measured reflections

1351 independent reflections

1164 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.046$  $\theta_{\max} = 25.4^\circ$ ,  $\theta_{\min} = 2.7^\circ$  $h = -71 \rightarrow 53$  $k = -8 \rightarrow 9$  $l = -7 \rightarrow 7$ *Refinement*Refinement on  $F^2$ 

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.052$  $wR(F^2) = 0.142$  $S = 1.09$ 

1351 reflections

185 parameters

1 restraint

Primary atom site location: structure-invariant  
direct methodsSecondary atom site location: difference Fourier  
mapHydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0795P)^2 + 2.3925P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\max} < 0.001$  $\Delta\rho_{\max} = 0.29$  e Å<sup>-3</sup> $\Delta\rho_{\min} = -0.23$  e Å<sup>-3</sup>*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>)*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.20032 (5)	0.2529 (4)	0.2108 (8)	0.0298 (9)
C2	0.18360 (6)	0.1751 (5)	0.0811 (7)	0.0336 (9)
H2	0.1873	0.1211	-0.0596	0.040*
C3	0.16140 (6)	0.1762 (5)	0.1569 (7)	0.0322 (9)
H3	0.1497	0.1270	0.0649	0.039*

C4	0.15600 (6)	0.2481 (4)	0.3652 (7)	0.0266 (8)
C5	0.17305 (6)	0.3287 (4)	0.4920 (7)	0.0309 (8)
H5	0.1695	0.3815	0.6339	0.037*
C6	0.19483 (6)	0.3321 (5)	0.4130 (7)	0.0319 (9)
H6	0.2064	0.3901	0.4990	0.038*
O7	0.22265 (4)	0.2623 (4)	0.1501 (5)	0.0417 (7)
C8	0.22934 (6)	0.1709 (7)	-0.0480 (9)	0.0566 (13)
H8A	0.2260	0.0460	-0.0333	0.085*
H8B	0.2452	0.1864	-0.0706	0.085*
H8C	0.2213	0.2189	-0.1734	0.085*
C9	0.13286 (6)	0.2372 (4)	0.4458 (7)	0.0294 (9)
H9	0.1212	0.2054	0.3420	0.035*
N10	0.12749 (5)	0.2679 (4)	0.6473 (6)	0.0291 (7)
C11	0.10465 (5)	0.2522 (4)	0.7171 (7)	0.0262 (8)
C12	0.08845 (6)	0.1478 (4)	0.6058 (7)	0.0299 (8)
H12	0.0926	0.0816	0.4748	0.036*
C13	0.06651 (6)	0.1403 (4)	0.6851 (7)	0.0298 (8)
H13	0.0555	0.0692	0.6089	0.036*
C14	0.06059 (5)	0.2357 (4)	0.8746 (6)	0.0247 (8)
C15	0.07629 (5)	0.3366 (4)	0.9897 (7)	0.0277 (8)
H15	0.0720	0.4020	1.1210	0.033*
C16	0.09834 (5)	0.3414 (4)	0.9119 (7)	0.0275 (8)
H16	0.1095	0.4075	0.9939	0.033*
O17	0.03923 (4)	0.2212 (3)	0.9705 (4)	0.0297 (6)
C18	0.02009 (5)	0.2435 (4)	0.8384 (7)	0.0265 (8)
C19	0.02040 (6)	0.3185 (4)	0.6262 (7)	0.0282 (8)
H19	0.0344	0.3466	0.5542	0.034*
C20	0.0000	0.3517 (6)	0.5207 (9)	0.0282 (11)
H20	0.0000	0.3988	0.3715	0.034*
C21	0.0000	0.2007 (6)	0.9448 (10)	0.0267 (11)
H21	0.0000	0.1432	1.0881	0.032*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0204 (16)	0.037 (2)	0.032 (2)	-0.0001 (13)	0.0009 (16)	0.0060 (16)
C2	0.0318 (18)	0.039 (2)	0.030 (2)	-0.0004 (15)	0.0030 (16)	-0.0016 (16)
C3	0.0286 (18)	0.0354 (19)	0.033 (2)	-0.0043 (13)	-0.0012 (17)	0.0008 (17)
C4	0.0277 (17)	0.0271 (17)	0.025 (2)	0.0007 (12)	0.0006 (15)	0.0032 (14)
C5	0.0312 (17)	0.0318 (18)	0.030 (2)	-0.0006 (13)	-0.0033 (16)	-0.0025 (16)
C6	0.0264 (16)	0.0343 (18)	0.035 (2)	-0.0021 (13)	-0.0005 (17)	-0.0011 (17)
O7	0.0250 (13)	0.0602 (17)	0.0398 (17)	-0.0013 (11)	0.0049 (13)	-0.0029 (14)
C8	0.031 (2)	0.092 (4)	0.046 (3)	0.004 (2)	0.010 (2)	-0.009 (3)
C9	0.0294 (17)	0.0305 (18)	0.028 (2)	-0.0009 (13)	-0.0015 (16)	0.0030 (17)
N10	0.0229 (13)	0.0323 (15)	0.0321 (19)	-0.0009 (11)	0.0007 (14)	-0.0020 (14)
C11	0.0230 (16)	0.0260 (17)	0.030 (2)	0.0013 (12)	0.0016 (16)	0.0015 (15)
C12	0.0281 (17)	0.0298 (17)	0.032 (2)	0.0029 (13)	0.0002 (16)	-0.0040 (17)
C13	0.0255 (16)	0.0296 (16)	0.034 (2)	-0.0031 (13)	-0.0010 (16)	-0.0008 (16)

C14	0.0205 (15)	0.0254 (16)	0.028 (2)	0.0018 (12)	0.0004 (15)	0.0057 (14)
C15	0.0318 (17)	0.0255 (16)	0.0257 (19)	0.0025 (12)	-0.0012 (15)	-0.0004 (15)
C16	0.0265 (15)	0.0254 (16)	0.031 (2)	-0.0001 (12)	-0.0031 (15)	-0.0005 (15)
O17	0.0207 (11)	0.0401 (13)	0.0282 (16)	0.0032 (9)	-0.0002 (11)	0.0020 (12)
C18	0.0223 (17)	0.0264 (15)	0.031 (2)	0.0007 (12)	-0.0016 (14)	-0.0034 (16)
C19	0.0284 (17)	0.0260 (15)	0.030 (2)	-0.0004 (12)	0.0051 (15)	0.0006 (16)
C20	0.028 (2)	0.026 (2)	0.030 (3)	0.000	0.000	0.001 (2)
C21	0.028 (2)	0.025 (2)	0.027 (3)	0.000	0.000	0.002 (2)

*Geometric parameters (Å, °)*

C1—O7	1.376 (4)	C11—C12	1.407 (5)
C1—C6	1.387 (6)	C12—C13	1.387 (4)
C1—C2	1.388 (5)	C12—H12	0.9600
C2—C3	1.394 (5)	C13—C14	1.386 (5)
C2—H2	0.9600	C13—H13	0.9600
C3—C4	1.395 (6)	C14—C15	1.384 (5)
C3—H3	0.9600	C14—O17	1.396 (4)
C4—C5	1.402 (5)	C15—C16	1.390 (5)
C4—C9	1.458 (5)	C15—H15	0.9600
C5—C6	1.377 (5)	C16—H16	0.9600
C5—H5	0.9600	O17—C18	1.394 (4)
C6—H6	0.9600	C18—C19	1.389 (6)
O7—C8	1.425 (6)	C18—C21	1.389 (4)
C8—H8A	0.9599	C19—C20	1.388 (4)
C8—H8B	0.9599	C19—H19	0.9600
C8—H8C	0.9599	C20—C19 <sup>i</sup>	1.388 (4)
C9—N10	1.269 (5)	C20—H20	0.9600
C9—H9	0.9600	C21—C18 <sup>i</sup>	1.389 (4)
N10—C11	1.423 (4)	C21—H21	0.9600
C11—C16	1.395 (5)		
O7—C1—C6	115.8 (3)	C16—C11—N10	117.5 (3)
O7—C1—C2	124.2 (4)	C12—C11—N10	123.9 (3)
C6—C1—C2	120.0 (3)	C13—C12—C11	120.1 (3)
C1—C2—C3	119.4 (4)	C13—C12—H12	119.9
C1—C2—H2	120.3	C11—C12—H12	119.9
C3—C2—H2	120.3	C14—C13—C12	119.8 (3)
C2—C3—C4	120.7 (3)	C14—C13—H13	120.1
C2—C3—H3	119.6	C12—C13—H13	120.1
C4—C3—H3	119.6	C15—C14—C13	121.2 (3)
C3—C4—C5	119.0 (3)	C15—C14—O17	116.7 (3)
C3—C4—C9	119.4 (3)	C13—C14—O17	121.8 (3)
C5—C4—C9	121.7 (3)	C14—C15—C16	118.8 (3)
C6—C5—C4	120.0 (4)	C14—C15—H15	120.6
C6—C5—H5	120.0	C16—C15—H15	120.6
C4—C5—H5	120.0	C15—C16—C11	121.4 (3)
C5—C6—C1	120.8 (3)	C15—C16—H16	119.3

C5—C6—H6	119.6	C11—C16—H16	119.3
C1—C6—H6	119.6	C18—O17—C14	119.8 (3)
C1—O7—C8	117.6 (3)	C19—C18—C21	121.6 (3)
O7—C8—H8A	109.5	C19—C18—O17	123.8 (3)
O7—C8—H8B	109.5	C21—C18—O17	114.3 (4)
H8A—C8—H8B	109.5	C20—C19—C18	118.5 (3)
O7—C8—H8C	109.5	C20—C19—H19	120.7
H8A—C8—H8C	109.5	C18—C19—H19	120.7
H8B—C8—H8C	109.5	C19—C20—C19 <sup>i</sup>	121.4 (5)
N10—C9—C4	122.8 (3)	C19—C20—H20	119.3
N10—C9—H9	118.6	C19 <sup>i</sup> —C20—H20	119.3
C4—C9—H9	118.6	C18 <sup>i</sup> —C21—C18	118.3 (5)
C9—N10—C11	120.3 (3)	C18 <sup>i</sup> —C21—H21	120.9
C16—C11—C12	118.6 (3)	C18—C21—H21	120.9
O7—C1—C2—C3	-178.7 (3)	N10—C11—C12—C13	-179.8 (3)
C6—C1—C2—C3	-0.6 (5)	C11—C12—C13—C14	-0.2 (5)
C1—C2—C3—C4	-2.6 (5)	C12—C13—C14—C15	-1.1 (5)
C2—C3—C4—C5	3.6 (5)	C12—C13—C14—O17	-174.9 (3)
C2—C3—C4—C9	-176.2 (3)	C13—C14—C15—C16	0.1 (5)
C3—C4—C5—C6	-1.6 (5)	O17—C14—C15—C16	174.1 (3)
C9—C4—C5—C6	178.2 (3)	C14—C15—C16—C11	2.3 (5)
C4—C5—C6—C1	-1.5 (5)	C12—C11—C16—C15	-3.5 (5)
O7—C1—C6—C5	-179.1 (3)	N10—C11—C16—C15	178.6 (3)
C2—C1—C6—C5	2.6 (5)	C15—C14—O17—C18	134.4 (3)
C6—C1—O7—C8	175.3 (4)	C13—C14—O17—C18	-51.6 (4)
C2—C1—O7—C8	-6.4 (5)	C14—O17—C18—C19	-15.0 (4)
C3—C4—C9—N10	166.3 (3)	C14—O17—C18—C21	171.2 (3)
C5—C4—C9—N10	-13.6 (5)	C21—C18—C19—C20	0.4 (5)
C4—C9—N10—C11	-179.0 (3)	O17—C18—C19—C20	-173.0 (3)
C9—N10—C11—C16	-159.2 (3)	C18—C19—C20—C19 <sup>i</sup>	2.6 (6)
C9—N10—C11—C12	23.0 (5)	C19—C18—C21—C18 <sup>i</sup>	-3.3 (6)
C16—C11—C12—C13	2.4 (5)	O17—C18—C21—C18 <sup>i</sup>	170.7 (3)

Symmetry code: (i)  $-x, y, z$ .

*Hydrogen-bond geometry* ( $\text{\AA}, ^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C8—H8B $\cdots$ O7 <sup>ii</sup>	0.96	2.57	3.411 (5)	147
C13—H13 $\cdots$ O17 <sup>iii</sup>	0.96	2.52	3.405 (4)	154

Symmetry codes: (ii)  $-x+1/2, -y+1/2, z-1/2$ ; (iii)  $x, -y, z-1/2$ .