# organic compounds

Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

## 2-(1-Adamantyl)-1-{4-[(2-chloro-9isopropyl-9H-purin-6-yl)aminomethyl]phenyl}ethanone

### Michal Rouchal,<sup>a</sup> Marek Nečas,<sup>b</sup> Fabiana Pires de Carvalho<sup>a</sup> and Robert Vícha<sup>a</sup>\*

<sup>a</sup>Department of Chemistry, Faculty of Technology, Tomas Bata University in Zlin, Nám. T. G. Masaryka 275, Zlín, 762 72, Czech Republic, and <sup>b</sup>Department of Chemistry, Faculty of Science, Masaryk University in Brno, Kamenice 5, Brno-Bohunice, 625 00, Czech Republic Correspondence e-mail: rvicha@ft.utb.cz

Received 10 December 2008; accepted 6 January 2009

Key indicators: single-crystal X-ray study; T = 120 K; mean  $\sigma$ (C–C) = 0.004 Å; R factor = 0.042; wR factor = 0.150; data-to-parameter ratio = 13.5.

The structure of the title compound, C<sub>27</sub>H<sub>32</sub>ClN<sub>5</sub>O, consists of two crystallographically independent conformers differing slightly in all geometric parameters. Both contain nearly planar purine and benzene ring systems [maximum deviations of 0.046 (3) and 0.005 (2) Å, respectively], the dihedral angles between them being 76.44 (6) and 82.39 (6) $^{\circ}$ , and an adamantane cage consisting of three fused cyclohexane rings in almost ideal chair conformations, with C-C-C angles in the range 108.7 (2)–110.6 (2) $^{\circ}$ . The carbonyl plane and the benzene ring are almost coplanar [dihedral angles of 6.43 (9) and  $0.64 (8)^{\circ}$  in the two conformers]. The crystal structure is stabilized by intermolecular N-H···N interactions that link adjacent molecules into dimers and by some non-bonding contacts of the  $C-H \cdot \cdot \cdot Cl$  type.

### **Related literature**

The title compound was prepared according to a modified procedure published by Fiorini & Abel (1989). For the synthesis and/or biological activity of related compounds, see: Veselý et al. (1994); Havlíček et al. (1997); de Azevedo et al. (1997); Kryštof et al. (2002); Kryštof et al. (2005); Legraverend & Grierson (2006). For some important properties of adamantane-bearing compounds, see: van Bommel et al. (2001); Cromwell et al. (1985). For related structures, see: Wang et al. (2001); Trávníček & Zatloukal (2004); Trávníček & Popa (2007*a*,*b*).



V = 4750.7 (3) Å<sup>3</sup>

Mo Ka radiation

47831 measured reflections

8353 independent reflections

5567 reflections with  $I > 2\sigma(I)$ 

H-atom parameters constrained

 $\mu = 0.19 \text{ mm}^{-3}$ 

T = 120 (2) K  $0.50 \times 0.40 \times 0.30 \text{ mm}$ 

 $R_{\rm int} = 0.026$ 

617 parameters

 $\Delta \rho_{\rm max} = 0.35 \ {\rm e} \ {\rm \AA}^{-3}$ 

 $\Delta \rho_{\rm min} = -0.29 \text{ e} \text{ Å}^{-3}$ 

Z = 8

### **Experimental**

Crystal data
C <sub>27</sub> H <sub>32</sub> ClN <sub>5</sub> O
$M_r = 478.03$
Monoclinic, $P2_1/c$
$a = 15.8778 (5) \text{\AA}$
b = 20.2779 (5) Å
c = 15.2225 (5) Å
$\beta = 104.233 \ (3)^{\circ}$

### Data collection

Kuma KM-4 CCD diffractometer Absorption correction: multi-scan (Xcalibur; Oxford Diffraction, 2006) $T_{\min} = 0.872, T_{\max} = 0.944$ 

### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.042$  $wR(F^2) = 0.150$ S = 1.098353 reflections

### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N5-H5A\cdots N53^{i}$	0.88	2.20	2.997 (3)	150
$N55 - H55A \cdots N3^{ii}$	0.88	2.18	2.946 (3)	145
$C27 - H27A \cdots Cl1^{iii}$	0.98	2.86	3.732 (3)	149
$C54-H54B\cdots Cl51^{iv}$	0.99	2.76	3.698 (3)	158
Summation and an (i)	x 1 2 1	3. (::)		- + 3. (:::)

-x + 3, -y, -z + 2; (iv)  $x - 1, -y + \frac{1}{2}, z - \frac{1}{2}$ .

#### Table 2

Comparative torsion angles (°) for selected 2,6,9-trisubstituted purines containing the 2-chloro 6-benzylamino and 9-isopropyl unit.

Compound	langle	value	angle	value
NG38 <sup>a</sup>	C6-N6-C9-C10	115.22 (13	)H17-C17-N9-C4	-13.38(18)
CIBAP1 <sup>b</sup>	C6-N6-C9-C10	178.97 (15	)H16-C16-N9-C4	-63.03(2)
CIBAP2 <sup>c</sup>	C6-N6-C9-C10	-117.35(2)	H16-C16-N9-C4	30.35 (3)
$CIABAP^d$	C20-N5-C19-C16	100.28 (3)	H25-C25-N4-C22	45.01 (3)
$CIABAP^d$	C70-N55-C69-C66	-99.62 (3)	H75-C75-N54-C72	2 - 40.79(3)

Notes: (a) Trávníček & Zatloukal (2004), where NG38 is N-[(2-azepan-1-yl)-9-isopropyl-9H-purin-6-yl]-4-methoxybenzylamine; (b) Trávníček & Popa (2007a), where CIBAP1 is 2-chloro-6-[(2,6-dimethoxybenzyl)amino]-9-isopropylpurine; (c) Trávníček & Popa (2007b), where CIBAP2 is 2-chloro-6-[(4-hydroxy-3,5-dimethoxybenzyl)amino]-9-isopropylpurine; (d) this work, where CIABAP is the title compound (the structure consists of two crystallographically independent molecules).

Data collection: *Xcalibur* (Oxford Diffraction, 2006); cell refinement: *Xcalibur*; data reduction: *Xcalibur*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

The financial support of this work by the Science Foundation of the Czech Republic (grant No. 203/06/P362) and by the Czech Ministry of Education (project No. MSM 7088352101) is gratefully acknowledged.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PK2141).

### References

- Azevedo, W. F. de, Leclerc, S., Meijer, L., Havlíček, L., Strnad, M. & Kim, S.-H. (1997). *Eur. J. Biochem.* **243**, 518–526.
- Bommel, K. J. C. van, Metselaar, G. A., Verboom, W. & Reinhoudt, D. N. (2001). J. Org. Chem. 66, 5405–5412.

- Cromwell, W. C., Bystrom, K. & Eftink, M. R. (1985). J. Phys. Chem. 89, 326–332.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Fiorini, M. T. & Abel, Ch. (1989). Tetrahedron Lett. 39, 1827–1830.
- Havlíček, L., Hanuš, J., Veselý, J., Leclerc, S., Meijer, L., Shaw, G. & Strnad, M. (1997). *J. Med. Chem.* **40**, 408–412.
- Kryštof, V., Lenobel, R., Havlíček, L., Kuzma, M. & Strnad, M. (2002). *Bioorg. Med. Chem. Lett.* 12, 3283–3286.
- Kryštof, V., McNae, I. W., Walkinshaw, M. D., Fischer, P. M., Muller, P., Vojtěšek, B., Orság, M., Havlíček, L. & Strnad, M. (2005). *Cell. Mol. Life Sci.* 62, 1763–1771.
- Legraverend, M. & Grierson, D. S. (2006). *Bioorg. Med. Chem.* 14, 3987–4006. Oxford Diffraction (2006). *Xcalibur*. Oxford Diffraction Ltd, Abingdon,
- England.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Trávníček, Z. & Popa, I. (2007a). Acta Cryst. E63, 0629-0631.
- Trávníček, Z. & Popa, I. (2007b). Acta Cryst. E63, 0728-0730.
- Trávníček, Z. & Zatloukal, M. (2004). Acta Cryst. E60, 0924-0926.
- Veselý, J., Havlíček, L., Strnad, M., Blow, J. J., Donnella-Deana, A., Pinna, L., Letham, D. S., Kato, J., Detivaud, L., Leclerc, S. & Meijer, L. (1994). *Eur. J. Biochem.* 224, 771–786.
- Wang, S., McClue, S. J., Ferguson, J. R., Hull, J. D., Stokes, S., Parsons, S., Westwood, R. & Fischer, P. M. (2001). *Tetrahedron Asymmetry*, **12**, 2891– 2894.

# supporting information

Acta Cryst. (2009). E65, o298-o299 [doi:10.1107/S160053680900052X]

# 2-(1-Adamantyl)-1-{4-[(2-chloro-9-isopropyl-9*H*-purin-6-yl)aminomethyl]-phenyl}ethanone

### Michal Rouchal, Marek Nečas, Fabiana Pires de Carvalho and Robert Vícha

### S1. Comment

The title molecule is related to the family of 2,6,9-trisubstituted purines that behave as potent inhibitors of cyclindependent kinases and show anticancer activity. The antiproliferative and proapoptotic effects of these drugs have been studied extensively with both an important and a promising result (Veselý *et al.*, 1994; Havlíček *et al.*, 1997; de Azevedo *et al.*, 1997; Kryštof *et al.*, 2002; Kryštof *et al.*, 2005; Legraverend & Grierson, 2006). The adamantane group is frequently used to improve the pharmacological properties of potential drugs. Whereas the lipophilic adamantane cage itself may increase solubility in non-polar systems (*e.g.* cell membranes), the non-covalent complex of adamantane with cyclodextrins can enhance solubility in water based media (Cromwell *et al.*, 1985; van Bommel *et al.*, 2001). Both these facilities have considerable importance in drug design and formulation. To the best of our knowledge, the title compound is the first described derivative of a 2,6,9-trisubstituted purine with an adamantyl group linked to 6-benzylamino substituent.

The structure consists of two crystallographically independent molecules slightly variant in geometry. Each ring is essentially planar, the maximum deviations from the best planes being 0.026 (2) Å for atom C23 (pyrimidine rings), 0.0050 (10) Å for atoms C72 and C73 (imidazole rings) and 0.005 (2) Å for atoms C13, C16 and C68 (benzene rings). The dihedral angles between purine and benzene rings are 76.44 (6)° and 82.39 (6)° respectively. The torsion angles C19–N5–C20–C23, C20–N5–C19–C16, N5–C19–C16–C17, C18–C13–C12–C11 and C13–C12–C11–C1 are 172.35 (2), 100.28 (3), 146.27 (2), -6.92 (4) and -100.18 (3)° respectively. The corresponding values of torsion angles for the second distinct conformer are -176.98 (2), -99.62 (3), -168.68 (2), 0.16 (4) and 95.57 (3) respectively. Comparative torsion angles for selected related molecules are shown in Table 2. The crystal structure is stabilized by intermolecular N–H…N interactions that link the molecules into pairs (Fig. 2 and Table 1), the N…N distances being 2.997 (3) and 2.946 (3) Å. respectively. There are also some additional intermolecular non-bonding contacts of the type C–H…Cl (Table 1). One from the two conformers is linked by C27–H27A…Cl1 interaction into pairs and the second conformer is linked by C54–H54B…Cl51 into linear chains. No other short intermolecular interactions were found.

### **S2.** Experimental

The title compound was prepared according to a slightly modified literature procedure (Fiorini & Abel, 1989). 2,6-Dichloro-9-(propan-2-yl)-9*H*-purine (0.65 mmol, 150 mg) and 1-[4-(aminoethyl)phenyl)]-2-(1-adamantyl)ethanone hydrochloride (0.68 mmol, 218 mg) were dissolved in the mixture of DMF (2 cm<sup>3</sup>) and triethylamine (1.30 mmol, 0.18 cm<sup>3</sup>). The resulting solution was stirred and refluxed for 2.5 h. After the starting material had all reacted (according to TLC), the mixture was diluted with water and extracted five times with 15 cm<sup>3</sup> of diethyl ether. The combined organic layers were washed twice with brine, dried over sodium sulfate and evaporated in vacuum. The crude product was purified by column chromatography (silica gel; petroleum ether/ethyl acetate, v/v, 1/1). The desired product was obtained as pale yellow crystalline powder (228 mg, 74%, mp 146–150°C). The crystal used for data collection was grown by liquid diffusion (acetone/hexane, v/v, 1/3) at -18°C within 48 h.



### Figure 1

*ORTEP* of the asymmetric unit with atoms represented as 50% probability ellipsoids. H-atoms have been omitted to enhance clarity.



### Figure 2

Part of the crystal structure of the title compound, showing the hydrogen bonding (dashed lines).

### 2-(1-Adamantyl)-1-{4-[(2-chloro-9-isopropyl-9H-purin-6-yl)aminomethyl]phenyl}ethanone

Crystal data	
C <sub>27</sub> H <sub>32</sub> ClN <sub>5</sub> O	<i>c</i> = 15.2225 (5) Å
$M_r = 478.03$	$\beta = 104.233 \ (3)^{\circ}$
Monoclinic, $P2_1/c$	$V = 4750.7 (3) Å^3$
Hall symbol: -P 2ybc	Z = 8
a = 15.8778 (5)  Å	F(000) = 2032
b = 20.2779 (5) Å	$D_{\rm x} = 1.337 {\rm ~Mg} {\rm ~m}^{-3}$

Melting point: 148 K Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å Cell parameters from 8353 reflections  $\theta = 2.8-25.0^{\circ}$ 

### Data collection

Kuma KM-4 CCD	47831 measured reflections
diffractometer	8353 independent reflections
Radiation source: fine-focus sealed tube	5567 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\rm int} = 0.026$
Detector resolution: 0.06 pixels mm <sup>-1</sup>	$\theta_{\rm max} = 25.0^\circ, \ \theta_{\rm min} = 2.8^\circ$
$\omega$ scans	$h = -16 \rightarrow 18$
Absorption correction: multi-scan	$k = -24 \rightarrow 23$
(Xcalibur; Oxford Diffraction, 2006)	$l = -18 \rightarrow 18$
$T_{\min} = 0.872, \ T_{\max} = 0.944$	
Refinement	

 $\mu = 0.19 \text{ mm}^{-1}$ 

Block, yellow

 $0.50 \times 0.40 \times 0.30 \text{ mm}$ 

T = 120 K

Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained  $w = 1/[\sigma^2(F_o^2) + (0.0761P)^2 + 2.6597P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{max} = 0.005$  $\Delta\rho_{max} = 0.35$  e Å<sup>-3</sup>  $\Delta\rho_{min} = -0.29$  e Å<sup>-3</sup>

### Special details

direct methods

Refinement on  $F^2$ 

 $wR(F^2) = 0.150$ 

8353 reflections

617 parameters 0 restraints

S = 1.09

Least-squares matrix: full

Primary atom site location: structure-invariant

 $R[F^2 > 2\sigma(F^2)] = 0.042$ 

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on  $F^2$ , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates an	nd isotropic or e	equivalent isotropic a	lisplacement	parameters	$(Å^2$	?)
				P	1	/

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
Cl1	1.54371 (4)	0.16850 (4)	1.01111 (5)	0.0416 (2)
01	1.15991 (12)	0.38506 (9)	0.65047 (14)	0.0418 (5)
N1	1.39808 (13)	0.11570 (10)	0.92710 (14)	0.0265 (5)
N2	1.53027 (13)	0.05913 (10)	0.92127 (14)	0.0278 (5)
N4	1.50180 (14)	-0.04121 (10)	0.83228 (14)	0.0289 (5)
N3	1.35644 (14)	-0.03462 (10)	0.78459 (15)	0.0306 (5)
N5	1.26108 (13)	0.07666 (10)	0.85571 (14)	0.0261 (5)
H5A	1.2293	0.0495	0.8159	0.031*
C1	0.95337 (16)	0.38700 (11)	0.59602 (15)	0.0222 (5)
C2	0.98831 (16)	0.39531 (12)	0.51181 (17)	0.0269 (6)
H2B	1.0340	0.3620	0.5123	0.032*
H2C	1.0147	0.4396	0.5122	0.032*

C3	0.91501 (17)	0.38722 (12)	0.42610 (17)	0.0281 (6)
H3A	0.9391	0.3929	0.3717	0.034*
C4	0.84508 (17)	0.43886 (13)	0.42383 (18)	0.0321 (6)
H4B	0.7976	0.4333	0.3684	0.038*
H4C	0.8699	0.4836	0.4228	0.038*
C5	0.80936 (17)	0.43108 (13)	0.50779 (17)	0.0285 (6)
H5B	0.7636	0.4651	0.5070	0.034*
C6	0.77037 (17)	0.36252 (13)	0.50670 (17)	0.0302 (6)
H6A	0.7225	0.3571	0.4516	0.036*
H6B	0.7461	0.3569	0.5602	0.036*
C7	0.83958 (17)	0.31037 (13)	0.50795 (17)	0.0282 (6)
H7A	0.8132	0.2656	0.5074	0.034*
C8	0.91320 (16)	0.31812 (12)	0.59364 (16)	0.0254 (6)
H8A	0.9583	0.2842	0.5944	0.030*
H8B	0.8902	0.3117	0.6479	0.030*
C9	0.87618 (17)	0.31819 (13)	0.42455 (17)	0.0308 (6)
H9A	0.9215	0.2845	0.4253	0.037*
H9B	0.8294	0.3121	0.3687	0.037*
C10	0.88214 (17)	0.43880 (12)	0.59306 (17)	0.0284 (6)
H10A	0.8584	0.4338	0.6470	0.034*
H10B	0.9075	0.4835	0.5946	0.034*
C11	1.02429 (17)	0.39841 (12)	0.68438 (17)	0.0291 (6)
H11A	1.0369	0.4462	0.6908	0.035*
H11B	1.0011	0.3847	0.7363	0.035*
C12	1.10835 (17)	0.36200 (13)	0.68945 (17)	0.0287 (6)
C13	1.12982 (16)	0.29932 (12)	0.74137 (16)	0.0258 (6)
C14	1.20554 (18)	0.26697 (13)	0.73665 (18)	0.0331 (6)
H14A	1.2396	0.2843	0.6987	0.040*
C15	1.23249 (17)	0.21078 (13)	0.78510(18)	0.0327 (6)
H15A	1.2847	0.1898	0.7804	0.039*
C16	1.18432 (16)	0.18431 (12)	0.84080 (16)	0.0240 (5)
C17	1.10862 (16)	0.21535 (13)	0.84536 (17)	0.0279 (6)
H17A	1.0745	0.1974	0.8828	0.033*
C18	1.08111 (16)	0.27234 (12)	0.79635 (17)	0.0280 (6)
H18A	1.0285	0.2930	0.8005	0.034*
C19	1.21638 (16)	0.12463 (12)	0.89844 (17)	0.0263 (6)
H19A	1.2563	0.1394	0.9557	0.032*
H19B	1.1662	0.1029	0.9140	0.032*
C20	1.34751 (16)	0.07103 (12)	0.87280 (15)	0.0228 (5)
C21	1.48320 (16)	0.10626 (13)	0.94474 (16)	0.0275 (6)
C22	1.47769 (16)	0.01555 (12)	0.86771 (16)	0.0246 (6)
C23	1.38849 (16)	0.01919 (12)	0.83828 (17)	0.0255 (6)
C24	1.42650 (18)	-0.06881 (13)	0.78385 (19)	0.0333 (6)
H24A	1.4248	-0.1094	0.7522	0.040*
C25	1.59119 (17)	-0.06731 (13)	0.84933 (18)	0.0326 (6)
H25A	1.6181	-0.0638	0.9158	0.039*
C26	1.64435 (18)	-0.02650 (14)	0.8010 (2)	0.0389 (7)
H26A	1.6197	-0.0295	0.7355	0.058*

H26B	1.7043	-0.0428	0.8157	0.058*
H26C	1.6438	0.0196	0.8202	0.058*
C27	1.5911 (2)	-0.13913 (15)	0.8235 (3)	0.0566 (10)
H27A	1.5524	-0.1637	0.8528	0.085*
H27B	1.6502	-0.1568	0.8433	0.085*
H27C	1.5706	-0.1435	0.7575	0.085*
C151	1.06711 (4)	0.32556 (3)	1.02370 (4)	0.03138 (18)
O51	0.65672 (14)	0.16100 (11)	0.55786 (14)	0.0508 (6)
N51	0.92396 (13)	0.37756 (10)	0.93336 (13)	0.0251 (5)
N52	1.05595 (13)	0.43726 (10)	0.93901 (13)	0.0231 (5)
N53	0.88368 (13)	0.53117 (10)	0.79834 (13)	0.0237 (5)
N54	1.02812 (13)	0.53960 (10)	0.85313 (13)	0.0223 (5)
N55	0.78821 (13)	0.41714 (10)	0.86172 (13)	0.0264 (5)
H55A	0.7566	0.4474	0.8274	0.032*
C51	0.47184 (16)	0.11567 (12)	0.58865 (16)	0.0259 (6)
C52	0.43209 (18)	0.16706 (13)	0.63953 (18)	0.0326 (6)
H52B	0.4403	0.1533	0.7035	0.039*
H52C	0.4618	0.2099	0.6388	0.039*
C53	0.33568 (18)	0.17478 (14)	0.5956 (2)	0.0370 (7)
H53B	0.3103	0.2083	0.6298	0.044*
C54	0.2906 (2)	0.10865 (16)	0.5982 (2)	0.0478 (8)
H54A	0.2991	0.0938	0.6617	0.057*
H54B	0.2274	0.1133	0.5712	0.057*
C55	0.3289 (2)	0.05839 (15)	0.5449 (2)	0.0496 (8)
H55B	0.2988	0.0151	0.5453	0.060*
C56	0.3168 (2)	0.08135 (18)	0.4476 (2)	0.0605 (10)
H56A	0.3407	0.0479	0.4129	0.073*
H56B	0.2541	0.0867	0.4188	0.073*
C57	0.3632 (2)	0.14650 (17)	0.44614 (19)	0.0457 (8)
H57A	0.3556	0.1613	0.3820	0.055*
C58	0.45957 (18)	0.13811 (15)	0.49055 (18)	0.0373 (7)
H58A	0.4901	0.1805	0.4890	0.045*
H58B	0.4849	0.1050	0.4567	0.045*
C59	0.3243 (2)	0.19791 (16)	0.49812 (19)	0.0437 (8)
H59A	0.3539	0.2408	0.4973	0.052*
H59B	0.2618	0.2039	0.4690	0.052*
C60	0.42482 (18)	0.05001 (13)	0.5893 (2)	0.0380 (7)
H60A	0.4500	0.0164	0.5560	0.046*
H60B	0.4324	0.0347	0.6525	0.046*
C61	0.56868 (17)	0.10361 (13)	0.63550 (19)	0.0329 (6)
H61A	0.5739	0.0958	0.7008	0.039*
H61B	0.5877	0.0629	0.6101	0.039*
C62	0.62922 (17)	0.15816 (14)	0.62633 (19)	0.0344 (7)
C63	0.65636 (16)	0.20984 (13)	0.69797 (17)	0.0279 (6)
C68	0.62858 (17)	0.21168 (13)	0.77775 (17)	0.0311 (6)
H68A	0.5905	0.1786	0.7895	0.037*
C67	0.65643 (17)	0.26183 (13)	0.84013 (17)	0.0294 (6)
H67A	0.6364	0.2632	0.8940	0.035*

C66	0.71300 (16)	0.30974 (12)	0.82460 (16)	0.0247 (6)	
C65	0.74072 (16)	0.30712 (13)	0.74527 (17)	0.0281 (6)	
H65A	0.7795	0.3399	0.7339	0.034*	
C64	0.71310 (17)	0.25799 (13)	0.68273 (17)	0.0295 (6)	
H64A	0.7330	0.2570	0.6288	0.035*	
C69	0.74333 (17)	0.36378 (13)	0.89382 (16)	0.0276 (6)	
H69A	0.6923	0.3819	0.9120	0.033*	
H69B	0.7826	0.3443	0.9484	0.033*	
C70	0.87432 (15)	0.42347 (11)	0.88074 (15)	0.0218 (5)	
C71	1.00903 (16)	0.38826 (12)	0.95686 (15)	0.0230 (5)	
C72	1.00400 (16)	0.48160 (11)	0.88524 (15)	0.0216 (5)	
C73	0.91542 (16)	0.47738 (11)	0.85193 (15)	0.0214 (5)	
C74	0.95342 (16)	0.56661 (12)	0.80151 (16)	0.0259 (6)	
H74A	0.9521	0.6075	0.7706	0.031*	
C75	1.11732 (16)	0.56615 (12)	0.87100 (16)	0.0250 (6)	
H75A	1.1462	0.5577	0.9362	0.030*	
C76	1.11638 (17)	0.63973 (12)	0.85592 (18)	0.0311 (6)	
H76A	1.0919	0.6492	0.7916	0.047*	
H76B	1.1759	0.6568	0.8743	0.047*	
H76C	1.0808	0.6609	0.8921	0.047*	
C77	1.16747 (17)	0.52945 (13)	0.81434 (18)	0.0308 (6)	
H77A	1.1671	0.4822	0.8277	0.046*	
H77B	1.2275	0.5455	0.8285	0.046*	
H77C	1.1404	0.5368	0.7500	0.046*	

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0251 (4)	0.0477 (4)	0.0511 (4)	-0.0055 (3)	0.0075 (3)	-0.0211 (3)
01	0.0265 (11)	0.0396 (12)	0.0568 (13)	-0.0032 (9)	0.0051 (10)	0.0171 (10)
N1	0.0216 (12)	0.0287 (12)	0.0294 (11)	0.0004 (9)	0.0067 (9)	0.0004 (9)
N2	0.0225 (12)	0.0299 (12)	0.0305 (11)	0.0001 (9)	0.0057 (9)	0.0000 (9)
N4	0.0235 (12)	0.0237 (11)	0.0394 (12)	0.0046 (9)	0.0079 (10)	0.0022 (9)
N3	0.0285 (13)	0.0233 (12)	0.0387 (12)	0.0021 (9)	0.0055 (10)	0.0002 (10)
N5	0.0201 (12)	0.0246 (11)	0.0319 (11)	0.0002 (9)	0.0030 (9)	-0.0023 (9)
C1	0.0228 (13)	0.0199 (12)	0.0232 (12)	0.0009 (10)	0.0045 (10)	-0.0027 (10)
C2	0.0238 (14)	0.0245 (14)	0.0326 (14)	-0.0020 (11)	0.0074 (11)	0.0031 (11)
C3	0.0293 (15)	0.0328 (15)	0.0232 (12)	-0.0005 (11)	0.0081 (11)	0.0013 (11)
C4	0.0291 (15)	0.0322 (15)	0.0337 (14)	-0.0007 (12)	0.0055 (12)	0.0043 (12)
C5	0.0252 (14)	0.0265 (14)	0.0324 (14)	0.0075 (11)	0.0043 (11)	-0.0013 (11)
C6	0.0226 (14)	0.0416 (16)	0.0271 (13)	-0.0001 (12)	0.0072 (11)	0.0015 (12)
C7	0.0285 (15)	0.0241 (13)	0.0314 (14)	-0.0054 (11)	0.0061 (11)	-0.0007 (11)
C8	0.0265 (14)	0.0230 (13)	0.0277 (13)	0.0006 (11)	0.0088 (11)	0.0028 (10)
C9	0.0271 (15)	0.0347 (15)	0.0276 (13)	0.0001 (12)	0.0010 (11)	-0.0084 (11)
C10	0.0310 (15)	0.0253 (14)	0.0281 (13)	0.0049 (11)	0.0060 (11)	-0.0057 (11)
C11	0.0306 (15)	0.0243 (13)	0.0285 (13)	0.0017 (11)	0.0000 (11)	-0.0031 (11)
C12	0.0242 (14)	0.0283 (14)	0.0297 (13)	-0.0047 (11)	-0.0007 (11)	-0.0025 (11)
C13	0.0252 (14)	0.0241 (13)	0.0263 (13)	-0.0019 (11)	0.0027 (11)	-0.0048 (10)

C14	0.0313 (16)	0.0327 (15)	0.0393 (15)	-0.0010 (12)	0.0165 (12)	0.0056 (12)
C15	0.0236 (14)	0.0309 (15)	0.0462 (16)	0.0043 (11)	0.0137 (12)	0.0078 (12)
C16	0.0202 (13)	0.0271 (13)	0.0237 (12)	-0.0045 (10)	0.0034 (10)	-0.0063 (10)
C17	0.0248 (14)	0.0332 (15)	0.0286 (13)	0.0004 (11)	0.0122 (11)	0.0025 (11)
C18	0.0206 (13)	0.0294 (14)	0.0336 (14)	0.0041 (11)	0.0061 (11)	-0.0032 (11)
C19	0.0218 (14)	0.0282 (14)	0.0296 (13)	0.0033 (11)	0.0074 (11)	0.0003 (11)
C20	0.0247 (14)	0.0213 (13)	0.0220 (12)	0.0000 (10)	0.0051 (10)	0.0034 (10)
C21	0.0226 (14)	0.0333 (15)	0.0256 (13)	-0.0025 (11)	0.0039 (11)	-0.0014 (11)
C22	0.0268 (15)	0.0218 (13)	0.0254 (12)	0.0033 (11)	0.0066 (11)	0.0032 (10)
C23	0.0219 (14)	0.0245 (13)	0.0307 (13)	0.0008 (10)	0.0073 (11)	0.0058 (11)
C24	0.0314 (16)	0.0229 (14)	0.0441 (16)	0.0010 (12)	0.0063 (13)	-0.0036(12)
C25	0.0267 (15)	0.0363(15)	0.0357 (14)	0.0108 (12)	0.0097(12)	0.0078(12)
C26	0.0286(16)	0.0364(16)	0.0519(18)	0.0054(12)	0.0101(13)	0.0125(13)
C27	0.047(2)	0.0323(17)	0.104 (3)	0.0129(15)	0.044 (2)	0.0069(17)
C151	0.0289(4)	0.0222(1)	0.0354(4)	0.0007(3)	0.0048(3)	0.0088(3)
051	0.0203(1)	0.0690(15)	0.0482(13)	-0.0183(11)	0.0224(11)	-0.0241(11)
N51	0.0265(12)	0.0000(10)	0.0229(10)	-0.0025(9)	0.0055 (9)	0.0000 (9)
N52	0.0203(12) 0.0223(11)	0.0235(11)	0.0227(10)	-0.0015(9)	0.0055(9) 0.0042(9)	-0.0000(9)
N53	0.0223(11) 0.0227(12)	0.0232(11)	0.0227(10) 0.0243(10)	-0.0013(9)	0.0012(9)	0.0010 (8)
N54	0.0227(12)	0.0232(11) 0.0224(11)	0.0240(10)	-0.0027(8)	0.0041(9)	0.0010 (8)
N55	0.0200(11) 0.0247(12)	0.0224(11) 0.0270(12)	0.0240(10) 0.0273(11)	-0.0024(9)	0.0060 (9)	0.0007(0)
C51	0.0247(12) 0.0247(14)	0.0270(12) 0.0229(13)	0.0275(11) 0.0295(13)	-0.0024(9)	0.0000(9)	-0.0010(10)
C52	0.0247(14) 0.0335(16)	0.0229(15)	0.0295(13) 0.0335(14)	0.0003(10)	0.0055(11) 0.0057(12)	-0.0010(10)
C52	0.0333(10)	0.0277(16)	0.0333(14)	0.0033(12) 0.0082(12)	0.0057(12)	-0.0017(13)
C54	0.0293(10) 0.0282(17)	0.0577(10)	0.0448(10)	0.0082(12) 0.0011(14)	0.0100(15)	0.0017(13)
C55	0.0202(17) 0.0321(18)	0.030(2)	0.001(2)	-0.0115(14)	0.0141(15) 0.0124(16)	-0.0005(16)
C56	0.0321(10)	0.0370(18)	0.079(2)	0.0113(14) 0.0038(17)	-0.0024(10)	-0.0293(18)
C57	0.039(2)	0.074(3)	0.037(2)	0.0038(17)	0.0099(10)	0.0293(18)
C58	0.0411(18) 0.0352(17)	0.000(2)	0.0203(14) 0.0331(15)	0.0110(10)	0.0002(13)	0.0028(14)
C50	0.0332(17)	0.0457(18)	0.0331(13)	0.0000(13)	0.0125(13)	0.0003(13)
C59	0.0400(18) 0.0375(17)	0.0409(18)	0.0419(17) 0.0537(18)	-0.0033(12)	0.0030(14)	-0.0094(14)
C00	0.0373(17)	0.0240(14)	0.0337(18)	0.0033(12)	0.0132(14) 0.0054(12)	-0.0003(13)
C61	0.0313(10)	0.0273(14)	0.0383(13)	0.0033(12)	0.0034(12)	-0.0024(12)
C62	0.0217(13)	0.0423(17) 0.0213(14)	0.0372(10)	0.0028(12)	0.0038(12) 0.0037(11)	-0.0049(12)
C03	0.0238(14)	0.0313(14)	0.0272(13)	0.0070(11)	0.0037(11)	0.0027(11)
C08	0.0257(14)	0.0309(15)	0.0338(14)	-0.0035(11)	0.0018(11)	0.0078(12)
	0.0237(14)	0.0333(13)	0.0200(13)	-0.0043(11)	0.0031(11)	0.0038(11)
C00	0.0206(13)	0.0250(13)	0.0255(12)	0.0034(10)	-0.0008(10)	0.0060(10)
C65	0.0245(14)	0.0282(14)	0.0330(14)	0.0004 (11)	0.0108(11)	0.0031(11)
C64	0.0280(15)	0.0327(15)	0.0302(14)	0.0014(11)	0.0110(12)	0.0020(11)
C69	0.0245 (14)	0.0345 (15)	0.0238 (12)	-0.0065 (11)	0.0058 (11)	0.0051 (11)
C/0	0.0215 (14)	0.0234 (13)	0.0212 (12)	-0.0024 (10)	0.0066 (10)	-0.0054 (10)
C/I	0.0253(14)	0.0236(13)	0.0200 (12)	-0.0011(10)	0.0051(10)	-0.0001(10)
C72	0.0254 (14)	0.0221(13)	0.0184 (11)	-0.000/(10)	0.00/3(10)	-0.0032(10)
C73	0.0242 (14)	0.0222 (13)	0.0188 (11)	-0.0017 (10)	0.0069 (10)	-0.0023 (10)
C/4	0.0261 (14)	0.0246 (13)	0.0266 (13)	-0.0002 (11)	0.0056 (11)	0.0002 (10)
C75	0.0211 (14)	0.0286 (14)	0.0245 (12)	-0.0043 (10)	0.0038 (10)	0.0029 (10)
C76	0.0277 (15)	0.0275 (14)	0.0384 (15)	-0.0088 (11)	0.0087 (12)	-0.0009 (12)
C77	0.0246 (15)	0.0313 (15)	0.0363 (15)	-0.0017 (11)	0.0069 (12)	0.0018 (11)

Geometric parameters (Å, °)

Cl1—C21	1.749 (3)	Cl51—C71	1.744 (2)
O1—C12	1.217 (3)	O51—C62	1.226 (3)
N1-C21	1.326 (3)	N51—C71	1.327 (3)
N1-C20	1.350 (3)	N51—C70	1.349 (3)
N2-C21	1.316 (3)	N52—C71	1.310 (3)
N2-C22	1.345 (3)	N52—C72	1.351 (3)
N4—C24	1.361 (3)	N53—C74	1.311 (3)
N4—C22	1.365 (3)	N53—C73	1.381 (3)
N4—C25	1.477 (3)	N54—C72	1.364 (3)
N3—C24	1.313 (3)	N54—C74	1.366 (3)
N3—C23	1.383 (3)	N54—C75	1.477 (3)
N5-C20	1.337 (3)	N55—C70	1.332 (3)
N5-C19	1.449 (3)	N55—C69	1.445 (3)
N5—H5A	0.8800	N55—H55A	0.8800
C1—C2	1.526 (3)	C51—C52	1.525 (4)
C1—C8	1.532 (3)	C51—C58	1.527 (4)
C1—C10	1.536 (3)	C51—C60	1.528 (4)
C1C11	1.545 (3)	C51—C61	1.547 (4)
С2—С3	1.529 (3)	C52—C53	1.521 (4)
C2—H2B	0.9900	C52—H52B	0.9900
C2—H2C	0.9900	C52—H52C	0.9900
C3—C4	1.520 (4)	C53—C59	1.524 (4)
С3—С9	1.527 (4)	C53—C54	1.525 (4)
С3—НЗА	1.0000	С53—Н53В	1.0000
C4—C5	1.528 (4)	C54—C55	1.520 (4)
C4—H4B	0.9900	C54—H54A	0.9900
C4—H4C	0.9900	C54—H54B	0.9900
C5—C10	1.519 (3)	C55—C60	1.516 (4)
C5—C6	1.520 (4)	C55—C56	1.518 (5)
С5—Н5В	1.0000	C55—H55B	1.0000
C6—C7	1.522 (4)	C56—C57	1.515 (5)
С6—Н6А	0.9900	C56—H56A	0.9900
С6—Н6В	0.9900	C56—H56B	0.9900
С7—С9	1.529 (4)	C57—C58	1.523 (4)
С7—С8	1.531 (3)	C57—C59	1.528 (4)
С7—Н7А	1.0000	С57—Н57А	1.0000
C8—H8A	0.9900	C58—H58A	0.9900
C8—H8B	0.9900	C58—H58B	0.9900
С9—Н9А	0.9900	С59—Н59А	0.9900
С9—Н9В	0.9900	С59—Н59В	0.9900
C10—H10A	0.9900	C60—H60A	0.9900
C10—H10B	0.9900	C60—H60B	0.9900
C11—C12	1.511 (4)	C61—C62	1.494 (4)
C11—H11A	0.9900	C61—H61A	0.9900
C11—H11B	0.9900	C61—H61B	0.9900
C12—C13	1.491 (4)	C62—C63	1.497 (4)

C13—C18	1.384 (4)	C63—C64	1.386 (4)
C13—C14	1.387 (4)	C63—C68	1.391 (4)
C14—C15	1.368 (4)	C68—C67	1.387 (4)
C14—H14A	0.9500	C68—H68A	0.9500
C15—C16	1.382 (4)	C67—C66	1.383 (3)
C15—H15A	0.9500	С67—Н67А	0.9500
C16—C17	1.374 (3)	C66—C65	1.384 (4)
C16—C19	1.508 (3)	C66—C69	1.514 (3)
C17—C18	1.386 (4)	C65—C64	1.373 (4)
C17—H17A	0.9500	С65—Н65А	0.9500
C18—H18A	0.9500	C64—H64A	0.9500
C19—H19A	0.9900	С69—Н69А	0.9900
C19—H19B	0.9900	С69—Н69В	0.9900
C20—C23	1.405 (3)	С70—С73	1.398 (3)
C22—C23	1.378 (4)	С72—С73	1.375 (3)
C24—H24A	0.9500	С74—Н74А	0.9500
C25—C26	1.498 (4)	C75—C77	1.507 (4)
C25—C27	1.508 (4)	C75—C76	1.509 (3)
C25—H25A	1.0000	С75—Н75А	1.0000
C26—H26A	0.9800	С76—Н76А	0.9800
C26—H26B	0.9800	С76—Н76В	0.9800
C26—H26C	0.9800	С76—Н76С	0.9800
C27—H27A	0.9800	С77—Н77А	0.9800
С27—Н27В	0.9800	С77—Н77В	0.9800
C27—H27C	0.9800	С77—Н77С	0.9800
C21—N1—C20	116.8 (2)	C71—N51—C70	117.0 (2)
C21—N2—C22	109.4 (2)	C71—N52—C72	109.7 (2)
C24—N4—C22	105.5 (2)	C74—N53—C73	103.2 (2)
C24—N4—C25	128.8 (2)	C72—N54—C74	105.7 (2)
C22—N4—C25	125.7 (2)	C72—N54—C75	126.1 (2)
C24—N3—C23	103.4 (2)	C74—N54—C75	128.1 (2)
C20—N5—C19	124.2 (2)	C70—N55—C69	124.3 (2)
C20—N5—H5A	117.9	C70—N55—H55A	117.8
C19—N5—H5A	117.9	C69—N55—H55A	117.8
C2—C1—C8	108.42 (19)	C52—C51—C58	108.9 (2)
C2-C1-C10	108.5 (2)	C52—C51—C60	109.2 (2)
C8—C1—C10	108.9 (2)	C58—C51—C60	108.7 (2)
C2—C1—C11	112.2 (2)	C52—C51—C61	111.3 (2)
C8—C1—C11	111.55 (19)	C58—C51—C61	111.8 (2)
C10—C1—C11	107.27 (19)	C60—C51—C61	106.9 (2)
C1—C2—C3	110.3 (2)	C53—C52—C51	110.2 (2)
C1—C2—H2B	109.6	С53—С52—Н52В	109.6
C3—C2—H2B	109.6	С51—С52—Н52В	109.6
C1—C2—H2C	109.6	С53—С52—Н52С	109.6
C3—C2—H2C	109.6	С51—С52—Н52С	109.6
H2B—C2—H2C	108.1	H52B—C52—H52C	108.1
$C_1$ $C_2$ $C_0$	100.0(2)	C52 C53 C50	100.0(2)

$C_{4}$ $C_{2}$ $C_{2}$	110.1.(2)	C52 C52 C54	100.1.(2)
$C_4 - C_3 - C_2$	110.1(2) 100.2(2)	$C_{32} = C_{33} = C_{34}$	109.1(2) 110.6(2)
$C_{2}$	109.5 (2)	$C_{59} = C_{53} = C_{54}$	110.0 (5)
C4 - C3 - H3A	109.1	С52—С53—П55В	109.4
$C_{2}$ $C_{2}$ $H_{2}$	109.1	С54 С52 Ц52Р	109.4
$C_2 = C_3 = H_3 A$	109.1	С54—С53—Н53В	109.4
$C_3 - C_4 - C_5$	109.2 (2)	055-054-053	108.9 (2)
C3—C4—H4B	109.8	С55—С54—Н54А	109.9
C5—C4—H4B	109.8	С53—С54—Н54А	109.9
C3—C4—H4C	109.8	С55—С54—Н54В	109.9
C5—C4—H4C	109.8	С53—С54—Н54В	109.9
H4B—C4—H4C	108.3	H54A—C54—H54B	108.3
C10—C5—C6	109.0 (2)	C60—C55—C56	110.0 (3)
C10—C5—C4	110.1 (2)	C60—C55—C54	109.1 (3)
C6—C5—C4	108.6 (2)	C56—C55—C54	110.1 (3)
C10—C5—H5B	109.7	С60—С55—Н55В	109.2
С6—С5—Н5В	109.7	С56—С55—Н55В	109.2
С4—С5—Н5В	109.7	С54—С55—Н55В	109.2
C5—C6—C7	110.1 (2)	C57—C56—C55	109.7 (2)
С5—С6—Н6А	109.6	С57—С56—Н56А	109.7
С7—С6—Н6А	109.6	С55—С56—Н56А	109.7
С5—С6—Н6В	109.6	С57—С56—Н56В	109.7
С7—С6—Н6В	109.6	С55—С56—Н56В	109.7
H6A—C6—H6B	108.1	H56A—C56—H56B	108.2
C6-C7-C9	109.8 (2)	$C_{56} - C_{57} - C_{58}$	100.2 109.6 (3)
C6-C7-C8	109.3(2) 109.7(2)	$C_{56} - C_{57} - C_{59}$	109.0(3) 109.1(3)
C9-C7-C8	109.7(2) 109.3(2)	$C_{58} - C_{57} - C_{59}$	109.1(3) 109.6(2)
C6 C7 H7A	109.5 (2)	C56 C57 H57A	109.6 (2)
$C_0 = C_7 = H_7 \Lambda$	109.4	$C_{50} = C_{57} = H_{57A}$	109.5
$C_{2} = C_{1} = H_{1}^{2} A$	109.4	$C_{50} = C_{57} = H_{57A}$	109.5
$C_{0} = C_{0} = C_{1}$	109.4	C57 - C57 - C51	109.3 100.0(2)
$C^{-}$	109.88 (19)	$C_{57} = C_{58} = U_{58}$	109.9 (2)
$C_{-}C_{0}$ HeA	109.7	$C_{51} = C_{50} = H_{50A}$	109.7
$C1 - C\delta - H\delta A$	109.7	C51—C58—H58A	109.7
C/-C8-H8B	109.7	C57-C58-H58B	109.7
CI-C8-H8B	109.7	С51—С58—Н58В	109.7
H8A—C8—H8B	108.2	H58A—C58—H58B	108.2
С3—С9—С7	108.5 (2)	C53—C59—C57	109.1 (2)
С3—С9—Н9А	110.0	С53—С59—Н59А	109.9
С7—С9—Н9А	110.0	С57—С59—Н59А	109.9
С3—С9—Н9В	110.0	С53—С59—Н59В	109.9
С7—С9—Н9В	110.0	С57—С59—Н59В	109.9
H9A—C9—H9B	108.4	H59A—C59—H59B	108.3
C5-C10-C1	110.76 (19)	C55—C60—C51	109.8 (2)
C5-C10-H10A	109.5	С55—С60—Н60А	109.7
C1C10H10A	109.5	С51—С60—Н60А	109.7
C5-C10-H10B	109.5	С55—С60—Н60В	109.7
C1C10H10B	109.5	C51—C60—H60B	109.7
H10A—C10—H10B	108.1	H60A—C60—H60B	108.2
C12—C11—C1	114.6 (2)	C62—C61—C51	115.4 (2)

C12—C11—H11A	108.6	C62—C61—H61A	108.4
C1-C11-H11A	108.6	C51—C61—H61A	108.4
C12—C11—H11B	108.6	C62—C61—H61B	108.4
C1—C11—H11B	108.6	C51—C61—H61B	108.4
H11A—C11—H11B	107.6	H61A—C61—H61B	107.5
O1—C12—C13	119.5 (2)	O51—C62—C61	119.1 (2)
O1—C12—C11	118.6 (2)	O51—C62—C63	119.2 (3)
C13—C12—C11	121.8 (2)	C61—C62—C63	121.8 (2)
C18—C13—C14	117.7 (2)	C64—C63—C68	119.1 (2)
C18—C13—C12	124.6 (2)	C64—C63—C62	117.0 (2)
C14—C13—C12	117.6 (2)	C68—C63—C62	124.0 (2)
C15—C14—C13	121.8 (2)	C67—C68—C63	120.1 (2)
C15—C14—H14A	119.1	C67—C68—H68A	120.0
C13—C14—H14A	119.1	C63—C68—H68A	120.0
C14-C15-C16	120.5(2)	C66—C67—C68	120.6(2)
C14—C15—H15A	119.8	C66—C67—H67A	119.7
C16-C15-H15A	119.8	C68 - C67 - H67A	119.7
$C_{17}$ $C_{16}$ $C_{15}$ $C_{15}$	119.0 118.4(2)	C67 - C66 - C65	119.7 118.8(2)
$C_{17}$ $C_{16}$ $C_{19}$	120.8(2)	C67 - C66 - C69	110.0(2) 119.9(2)
$C_{1}^{-1} = C_{1}^{-1} = C_{1}^{-1}$	120.3(2) 120.7(2)	$C_{0} = C_{0} = C_{0}$	119.9(2) 121.3(2)
$C_{15} = C_{10} = C_{17}$	120.7(2)	$C_{00} = C_{00} = C_{00}$	121.3(2) 121.1(2)
$C_{10} - C_{17} - C_{18}$	121.2(2)	C64 - C65 - H65A	121.1(2)
C10 - C17 - H17A	119.4	C04 - C05 - H05A	119.4
C13 - C17 - H1/A	119.4	C00—C05—H05A	119.4
C13 - C18 - C17	120.4 (2)	C65 - C64 - C63	120.3 (2)
C13—C18—H18A	119.8	C65—C64—H64A	119.8
C17—C18—H18A	119.8	C63—C64—H64A	119.8
N5—C19—C16	114.2 (2)	N55—C69—C66	113.9 (2)
N5—C19—H19A	108.7	N55—C69—H69A	108.8
С16—С19—Н19А	108.7	С66—С69—Н69А	108.8
N5—C19—H19B	108.7	N55—C69—H69B	108.8
C16—C19—H19B	108.7	C66—C69—H69B	108.8
H19A—C19—H19B	107.6	H69A—C69—H69B	107.7
N5C20N1	119.3 (2)	N55—C70—N51	118.8 (2)
N5—C20—C23	122.6 (2)	N55—C70—C73	122.7 (2)
N1—C20—C23	118.1 (2)	N51—C70—C73	118.4 (2)
N2—C21—N1	132.0 (2)	N52—C71—N51	131.4 (2)
N2—C21—Cl1	114.40 (19)	N52—C71—Cl51	115.34 (18)
N1—C21—Cl1	113.63 (19)	N51—C71—Cl51	113.28 (18)
N2—C22—N4	126.9 (2)	N52—C72—N54	127.4 (2)
N2—C22—C23	126.7 (2)	N52—C72—C73	126.8 (2)
N4—C22—C23	106.4 (2)	N54—C72—C73	105.9 (2)
C22—C23—N3	110.4 (2)	C72—C73—N53	111.1 (2)
C22—C23—C20	116.9 (2)	C72—C73—C70	116.5 (2)
N3—C23—C20	132.4 (2)	N53—C73—C70	132.3 (2)
N3—C24—N4	114.3 (2)	N53—C74—N54	114.0 (2)
N3—C24—H24A	122.9	N53—C74—H74A	123.0
N4—C24—H24A	122.9	N54—C74—H74A	123.0
N4—C25—C26	110.4 (2)	N54—C75—C77	109.4 (2)

N4—C25—C27	111.2 (2)	N54—C75—C76	111.1 (2)
C26—C25—C27	111.7 (2)	C77—C75—C76	112.8 (2)
N4—C25—H25A	107.8	N54—C75—H75A	107.8
С26—С25—Н25А	107.8	С77—С75—Н75А	107.8
С27—С25—Н25А	107.8	С76—С75—Н75А	107.8
C25—C26—H26A	109.5	С75—С76—Н76А	109.5
C25—C26—H26B	109.5	С75—С76—Н76В	109.5
H26A—C26—H26B	109.5	H76A—C76—H76B	109.5
C25—C26—H26C	109.5	С75—С76—Н76С	109.5
H26A—C26—H26C	109.5	H76A—C76—H76C	109.5
H26B—C26—H26C	109.5	H76B—C76—H76C	109.5
С25—С27—Н27А	109.5	С75—С77—Н77А	109.5
С25—С27—Н27В	109.5	С75—С77—Н77В	109.5
H27A—C27—H27B	109.5	H77A—C77—H77B	109.5
C25—C27—H27C	109.5	C75—C77—H77C	109.5
H27A - C27 - H27C	109.5	H77A—C77—H77C	109.5
H27B-C27-H27C	109.5	H77B - C77 - H77C	109.5
	109.0		109.0
C8—C1—C2—C3	59.4 (3)	C58—C51—C52—C53	59.9 (3)
C10-C1-C2-C3	-58.7 (3)	C60—C51—C52—C53	-58.6(3)
C11—C1—C2—C3	-177.0(2)	C61—C51—C52—C53	-176.4(2)
C1—C2—C3—C4	60.1 (3)	C51—C52—C53—C59	-60.9 (3)
C1—C2—C3—C9	-60.8 (3)	C51—C52—C53—C54	59.9 (3)
C9—C3—C4—C5	61.4 (3)	C52—C53—C54—C55	-61.1 (3)
C2—C3—C4—C5	-59.2 (3)	C59—C53—C54—C55	58.7 (3)
C3—C4—C5—C10	58.8 (3)	C53—C54—C55—C60	61.8 (3)
C3—C4—C5—C6	-60.5 (3)	C53—C54—C55—C56	-59.0 (3)
C10—C5—C6—C7	-59.8 (3)	C60—C55—C56—C57	-59.5 (3)
C4—C5—C6—C7	60.1 (3)	C54—C55—C56—C57	60.7 (3)
С5—С6—С7—С9	-60.2 (3)	C55—C56—C57—C58	59.4 (3)
C5—C6—C7—C8	59.9 (3)	C55—C56—C57—C59	-60.7 (3)
C6—C7—C8—C1	-59.3 (3)	C56—C57—C58—C51	-60.0 (3)
C9—C7—C8—C1	61.1 (3)	C59—C57—C58—C51	59.7 (3)
C2—C1—C8—C7	-59.5 (3)	C52—C51—C58—C57	-59.1 (3)
C10-C1-C8-C7	58.3 (3)	C60—C51—C58—C57	59.8 (3)
C11—C1—C8—C7	176.5 (2)	C61—C51—C58—C57	177.6 (2)
C4—C3—C9—C7	-60.3 (3)	C52—C53—C59—C57	60.5 (3)
C2—C3—C9—C7	60.7 (3)	C54—C53—C59—C57	-59.4 (3)
C6—C7—C9—C3	59.3 (3)	C56—C57—C59—C53	59.8 (3)
C8—C7—C9—C3	-61.1 (3)	C58—C57—C59—C53	-60.2 (3)
C6C10C1	59.7 (3)	C56—C55—C60—C51	59.9 (3)
C4C5C10C1	-59.3 (3)	C54—C55—C60—C51	-61.0 (3)
C2-C1-C10-C5	58.6 (3)	C52—C51—C60—C55	59.0 (3)
C8-C1-C10-C5	-59.2 (3)	C58—C51—C60—C55	-59.6 (3)
C11—C1—C10—C5	180.0 (2)	C61—C51—C60—C55	179.5 (2)
C2-C1-C11-C12	-48.3 (3)	C52—C51—C61—C62	-72.2 (3)
C8—C1—C11—C12	73.5 (3)	C58—C51—C61—C62	49.8 (3)
C10-C1-C11-C12	-167.4 (2)	C60—C51—C61—C62	168.7 (2)

C1-C11-C12-O1	80.4 (3)	C51—C61—C62—O51	-83.7 (3)
C1-C11-C12-C13	-100.2 (3)	C51—C61—C62—C63	95.6 (3)
O1—C12—C13—C18	172.5 (2)	O51—C62—C63—C64	-1.0 (4)
C11—C12—C13—C18	-6.9 (4)	C61—C62—C63—C64	179.7 (2)
O1—C12—C13—C14	-5.2 (4)	O51—C62—C63—C68	179.4 (3)
C11—C12—C13—C14	175.4 (2)	C61—C62—C63—C68	0.2 (4)
C18—C13—C14—C15	-0.8(4)	C64—C63—C68—C67	1.1 (4)
C12—C13—C14—C15	177.1 (2)	C62—C63—C68—C67	-179.4(2)
C13 - C14 - C15 - C16	0.0 (4)	C63—C68—C67—C66	-1.1 (4)
C14-C15-C16-C17	0.7 (4)	C68—C67—C66—C65	0.6 (4)
C14-C15-C16-C19	-1765(2)	C68—C67—C66—C69	-1795(2)
$C_{15}$ $C_{16}$ $C_{17}$ $C_{18}$	-0.7(4)	C67 - C66 - C65 - C64	-0.1(4)
C19 - C16 - C17 - C18	1764(2)	C69 - C66 - C65 - C64	1799(2)
$C_{14}$ $C_{13}$ $C_{18}$ $C_{17}$	0.8(4)	C66-C65-C64-C63	0.1(4)
$C_{12}$ $C_{13}$ $C_{18}$ $C_{17}$	-1769(2)	C68 - C63 - C64 - C65	-0.6(4)
$C_{16}$ $C_{17}$ $C_{18}$ $C_{13}$	0.0(4)	C62 - C63 - C64 - C65	179.8(2)
$C_{10} = 0.05 = 0.05 = 0.05$	1003(3)	C70-N55-C69-C66	-99.6(3)
$C_{17}$ $C_{16}$ $C_{19}$ $N_{5}$	100.3(3) 146.3(2)	C67 - C66 - C69 - N55	-1687(2)
$C_{17} = C_{10} = C_{19} = N_5$	-36.6(2)	C65 C66 C69 N55	108.7(2)
$C_{10} = 0.0 = 0$	-60(3)	C60 N55 C70 N51	11.3(3)
C19 - N5 - C20 - C23	-0.0(3)	C60 N55 C70 C73	0.0(3)
C19 - N3 - C20 - C23	172.3(2)	$C_{09}$ N51 C70 N55	-177.0(2)
$C_{21} = N_{1} = C_{20} = N_{3}$	1/7.5(2)	C/1 = N51 = C70 = N53	-1/4.7(2)
$C_{21} = N_{1} = C_{20} = C_{23}$	-1.1(3)	C/1 = N51 = C/0 = C/3	2.3 (3)
$C_{22}$ N2 $C_{21}$ $C_{11}$	2.1(4)	C/2 = N52 = C/1 = N51	-1.8(4)
$C_{22}$ N1 $C_{21}$ N2	-1//.54(1/)	C/2 = N52 = C/1 = C151	1/8.53 (16)
C20—N1—C21—N2	-2.5 (4)	C/0—N51—C/1—N52	1.0 (4)
C20—N1—C21—C11	177.14 (17)	C/0—N51—C/1—Cl51	-179.33 (16)
C21—N2—C22—N4	-177.5 (2)	C71—N52—C72—N54	178.9 (2)
C21—N2—C22—C23	1.9 (3)	C71—N52—C72—C73	-0.8 (3)
C24—N4—C22—N2	179.4 (2)	C74—N54—C72—N52	179.4 (2)
C25—N4—C22—N2	2.7 (4)	C75—N54—C72—N52	-0.1 (4)
C24—N4—C22—C23	-0.2 (3)	C74—N54—C72—C73	-0.8(2)
C25—N4—C22—C23	-176.9 (2)	C75—N54—C72—C73	179.7 (2)
N2—C22—C23—N3	-179.7 (2)	N52—C72—C73—N53	-179.3 (2)
N4—C22—C23—N3	-0.1 (3)	N54—C72—C73—N53	1.0 (3)
N2—C22—C23—C20	-5.0 (4)	N52—C72—C73—C70	3.8 (3)
N4—C22—C23—C20	174.6 (2)	N54—C72—C73—C70	-176.0 (2)
C24—N3—C23—C22	0.4 (3)	C74—N53—C73—C72	-0.7 (3)
C24—N3—C23—C20	-173.2 (3)	C74—N53—C73—C70	175.6 (2)
N5—C20—C23—C22	-174.1 (2)	N55—C70—C73—C72	172.5 (2)
N1-C20-C23-C22	4.2 (3)	N51-C70-C73-C72	-4.4 (3)
N5-C20-C23-N3	-0.8 (4)	N55—C70—C73—N53	-3.6 (4)
N1-C20-C23-N3	177.5 (2)	N51—C70—C73—N53	179.5 (2)
C23—N3—C24—N4	-0.5 (3)	C73—N53—C74—N54	0.1 (3)
C22—N4—C24—N3	0.4 (3)	C72—N54—C74—N53	0.4 (3)
C25—N4—C24—N3	177.0 (2)	C75—N54—C74—N53	179.9 (2)
C24—N4—C25—C26	111.6 (3)	C72—N54—C75—C77	76.2 (3)
C22—N4—C25—C26	-72.5 (3)	C74—N54—C75—C77	-103.2 (3)
	× /		~ /

C24—N4—C25—C27	-13.0 (4)	C72—N54—C75—C76	-158.6 (2)
C22—N4—C25—C27	162.9 (3)	C74—N54—C75—C76	22.0 (3)

Hydrogen-bond geometry (Å, °)

<i>D</i> —Н	H···A	$D^{\dots}A$	D—H···A
0.88	2.20	2.997 (3)	150
0.88	2.18	2.946 (3)	145
0.98	2.86	3.732 (3)	149
0.99	2.76	3.698 (3)	158
	<i>D</i> —H 0.88 0.88 0.98 0.99	D—H         H…A           0.88         2.20           0.88         2.18           0.98         2.86           0.99         2.76	D—H         H···A         D···A           0.88         2.20         2.997 (3)           0.88         2.18         2.946 (3)           0.98         2.86         3.732 (3)           0.99         2.76         3.698 (3)

Symmetry codes: (i) -x+2, y-1/2, -z+3/2; (ii) -x+2, y+1/2, -z+3/2; (iii) -x+3, -y, -z+2; (iv) x-1, -y+1/2, z-1/2.