

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

# *N*-(Quinoxalin-2-yl)-4-toluidine

## Wan Ainna Mardhiah Wan Saffiee, Azila Idris, Zaharah Aiyub, Zanariah Abdullah and Seik Weng Ng\*

Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia Correspondence e-mail: seikweng@um.edu.my

Received 4 December 2008; accepted 5 December 2008

Key indicators: single-crystal X-ray study; T = 100 K; mean  $\sigma$ (C–C) = 0.002 Å; *R* factor = 0.041; w*R* factor = 0.112; data-to-parameter ratio = 16.7.

The aromatic and the aromatic fused-rings in the title compound,  $C_{15}H_{13}N_3$ , open the angle at the planar N atom to 130.07 (13) and 129.98 (13)° in the two independent molecules in the asymmetric unit. The amino N atom of one molecule forms a hydrogen bond to the 4-N atom of an adjacent quinoxalinyl ring, generating a supramolecular chain.

#### **Related literature**

For the structure of *N*-(2-pyridyl)-4-toluidine, see: Fairuz *et al.* (2008); for that of *N*-(pyrazin-2-yl)-4-toluidine, see: Wan Saffiee *et al.* (2008). The title compound is isostructural with *N*-(quinoxalin-2-yl)-4-chloroaniline; see: Idris *et al.* (2008).



#### **Experimental**

Crystal data

 $\begin{array}{l} C_{15}H_{13}N_3 \\ M_r = 235.28 \\ \text{Orthorhombic, } Pbca \\ a = 12.2081 \ (9) \ \text{\AA} \\ b = 11.3720 \ (9) \ \text{\AA} \\ c = 35.097 \ (3) \ \text{\AA} \end{array}$ 

 $V = 4872.5 \text{ (6) } \text{\AA}^{3}$  Z = 16Mo K\alpha radiation  $\mu = 0.08 \text{ mm}^{-1}$  T = 100 (2) K $0.40 \times 0.15 \times 0.05 \text{ mm}$  5592 independent reflections

 $R_{\rm int} = 0.051$ 

4089 reflections with  $I > 2\sigma(I)$ 

Data collection

Bruker SMART APEX diffractometer Absorption correction: none 26747 measured reflections

Refinement

$$\begin{split} R[F^2 > 2\sigma(F^2)] &= 0.041 & \text{H atoms treated by a mixture of} \\ wR(F^2) &= 0.112 & \text{independent and constrained} \\ S &= 1.03 & \text{refinement} \\ 5592 \text{ reflections} & \Delta\rho_{\text{max}} &= 0.25 \text{ e } \text{ Å}^{-3} \\ 335 \text{ parameters} & \Delta\rho_{\text{min}} &= -0.25 \text{ e } \text{ Å}^{-3} \end{split}$$

 Table 1

 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N1-H1\cdots N5$ $N4-H4\cdots N2^{i}$	0.89 (1) 0.87 (1)	2.26 (1) 2.19 (1)	3.114 (2) 3.017 (2)	163 (2) 157 (2)
	3 1			

Symmetry code: (i)  $-x + \frac{3}{2}, y + \frac{1}{2}, z$ .

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2009).

We thank the University of Malaya for supporting this study (FS358/2008 A).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK2339).

#### References

Barbour, L. J. (2001). J. Supramol. Chem. 1, 189-191.

Bruker (2007). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.

- Fairuz, Z. A., Aiyub, Z., Abdullah, Z. & Ng, S. W. (2008). Acta Cryst. E64, 02441.
- Idris, A., Wan Saffiee, W. A. M., Abdullah, Z., Ariffin, A. & Ng, S. W. (2008). Acta Cryst. E64, 02443.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Wan Saffiee, W. A. M., Idris, A., Aiyub, Z., Abdullah, Z. & Ng, S. W. (2008). Acta Cryst. E64, o2440.
- Westrip, S. P. (2009). publCIF. In preparation.

# supporting information

Acta Cryst. (2009). E65, o93 [doi:10.1107/S1600536808041160]

# N-(Quinoxalin-2-yl)-4-toluidine

# Wan Ainna Mardhiah Wan Saffiee, Azila Idris, Zaharah Aiyub, Zanariah Abdullah and Seik Weng Ng

# S1. Experimental

2-Chloroquinoxaline (1.64 g, 10 mmol) and 4-toluidine (1.07 g, 10 mmol) were mixed with ethanol (2 ml) and the mixture was heated at 423–433 K for 3 h. The product was dissolved in water and the solution extracted with ether. The ether phase was dried over sodium sulfate; the evaporation of the solvent gave well shaped crystals along with some unidentified brown material.

# S2. Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95 to 0.98 Å) and were included in the refinement in the riding model approximation, with U(H) set to 1.2 to  $1.5_{eq}U(C)$ . The amino H-atoms were located in a difference Fourier map, and were refined with a distance restraint of N–H 0.88±0.01 Å; their temperature factors were freely refined.



## Figure 1

Thermal ellipsoid plot (Barbour, 2001) for the two independent molecules of  $C_{15}H_{13}N_3$  at the 70% probability level. Hydrogen atoms are drawn as spheres of arbitrary radius.

## N-(Quinoxalin-2-yl)-4-toluidine

## Crystal data

 $C_{15}H_{13}N_3$   $M_r = 235.28$ Orthorhombic, *Pbca* Hall symbol: -P 2ac 2ab a = 12.2081 (9) Å b = 11.3720 (9) Å c = 35.097 (3) Å V = 4872.5 (6) Å<sup>3</sup> Z = 16

# Data collection

Bruker SMART APEX	4089 reflections with $I > 2\sigma(I)$
diffractometer	$R_{\rm int} = 0.051$
Radiation source: fine-focus sealed tube	$\theta_{\rm max} = 27.5^{\circ},  \theta_{\rm min} = 1.2^{\circ}$
Graphite monochromator	$h = -15 \rightarrow 15$
$\omega$ scans	$k = -14 \rightarrow 9$
26747 measured reflections	$l = -45 \rightarrow 45$
5592 independent reflections	

F(000) = 1984

 $\theta = 2.5 - 27.6^{\circ}$ 

 $\mu = 0.08 \text{ mm}^{-1}$ 

Block, yellow

 $0.40 \times 0.15 \times 0.05 \text{ mm}$ 

T = 100 K

 $D_{\rm x} = 1.283 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å Cell parameters from 3585 reflections

### Refinement

-j	
Refinement on $F^2$ Least-squares matrix: full	Secondary atom site location: difference Fourier
$P[E^2 > 2\sigma(E^2)] = 0.041$	Hydrogen site location: inferred from
$wR(F^2) = 0.112$	neighbouring sites
S = 1.03	H atoms treated by a mixture of independent
5592 reflections	and constrained refinement
335 parameters	$w = 1/[\sigma^2(F_o^2) + (0.0453P)^2 + 2.027P]$
2 restraints	where $P = (F_o^2 + 2F_c^2)/3$
Primary atom site location: structure-invariant	$(\Delta/\sigma)_{\rm max} = 0.001$
direct methods	$\Delta  ho_{ m max} = 0.25 \ { m e} \ { m \AA}^{-3}$
	$\Delta \rho_{\rm min} = -0.25 \text{ e } \text{\AA}^{-3}$

	x	у	Ζ	$U_{\rm iso}$ */ $U_{\rm eq}$
N1	0.48774 (11)	0.63997 (12)	0.55869 (4)	0.0200 (3)
N2	0.54428 (10)	0.38803 (11)	0.61730 (3)	0.0183 (3)
N3	0.41112 (10)	0.45078 (11)	0.55410 (3)	0.0169 (3)
N4	0.78134 (10)	0.95678 (12)	0.67401 (4)	0.0194 (3)
N5	0.57078 (11)	0.78948 (12)	0.62719 (4)	0.0207 (3)
N6	0.59133 (10)	0.96931 (11)	0.68290 (3)	0.0183 (3)
C1	0.47718 (12)	0.52591 (13)	0.57059 (4)	0.0166 (3)
C2	0.54419 (12)	0.49311 (14)	0.60265 (4)	0.0188 (3)
H2	0.5906	0.5510	0.6136	0.023*
C3	0.47637 (12)	0.30629 (13)	0.60042 (4)	0.0164 (3)
C4	0.47172 (12)	0.19136 (14)	0.61498 (4)	0.0193 (3)
H4A	0.5162	0.1697	0.6361	0.023*
C5	0.40277 (13)	0.11048 (14)	0.59870 (4)	0.0217 (3)
Н5	0.3993	0.0328	0.6086	0.026*

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(Å^2)$ 

04	0.0070((10)	0.1.4050 (1.4)		0.0000 (0)
C6	0.33726 (13)	0.14250 (14)	0.56743 (4)	0.0220 (3)
H6	0.2897	0.0859	0.5563	0.026*
C7	0.34082 (12)	0.25413 (14)	0.55266 (4)	0.0204 (3)
H7	0.2964	0.2741	0.5314	0.024*
C8	0.41034 (12)	0.33894 (13)	0.56900 (4)	0.0166 (3)
C9	0.43642 (12)	0.69945 (14)	0.52834 (4)	0.0178 (3)
C10	0.38087 (12)	0.64377 (14)	0.49874 (4)	0.0182 (3)
H10	0.3741	0.5606	0.4983	0.022*
C11	0.33549 (13)	0.71156 (14)	0.46985 (4)	0.0205 (3)
H11	0.2981	0.6731	0.4497	0.025*
C12	0.34273 (12)	0.83326 (14)	0.46935 (4)	0.0194 (3)
C13	0.39785 (13)	0.88706 (14)	0.49930 (5)	0.0222 (3)
H13	0.4036	0.9704	0.4999	0.027*
C14	0.44450 (13)	0.82189 (14)	0.52828 (5)	0.0222 (3)
H14	0.4823	0.8607	0.5483	0.027*
C15	0.29166 (14)	0.90520 (15)	0.43798(5)	0.0269 (4)
H15A	0 3407	0.9702	0.4312	0.040*
H15R	0.2213	0.9369	0.4467	0.040*
H15C	0.2797	0.8553	0.4156	0.040*
C16	0.2797	0.0000	0.66615 (4)	0.040
C10	0.66505 (13)	0.92171(13) 0.83080(14)	0.00013(4) 0.63794(4)	0.0174(3)
U17	0.00505 (15)	0.85080 (14)	0.6266	0.0198(3)
П1/ С19	0.7294 0.47052 (12)	0.7994 0.82502 (14)	0.0200	$0.024^{\circ}$
C18	0.47932(13)	0.85502(14)	0.04323(4) 0.02(17(5))	0.0190(3)
U19	0.37401 (14)	0.79237 (15)	0.03017 (3)	0.0247 (4)
HI9	0.3666	0.7318	0.61//	0.030*
C20	0.28394 (14)	0.837/1 (16)	0.65375 (5)	0.0274 (4)
H20	0.2133	0.8084	0.6475	0.033*
C21	0.29513 (13)	0.92756 (16)	0.68096 (5)	0.0265 (4)
H21	0.2317	0.9591	0.6929	0.032*
C22	0.39656 (13)	0.97021 (15)	0.69051 (4)	0.0226 (4)
H22	0.4032	1.0306	0.7091	0.027*
C23	0.49085 (12)	0.92445 (14)	0.67284 (4)	0.0178 (3)
C24	0.81936 (13)	1.04794 (13)	0.69774 (4)	0.0177 (3)
C25	0.75245 (13)	1.11936 (15)	0.72024 (4)	0.0233 (4)
H25	0.6753	1.1083	0.7203	0.028*
C26	0.79967 (13)	1.20656 (15)	0.74242 (5)	0.0242 (4)
H26	0.7535	1.2550	0.7575	0.029*
C27	0.91210 (13)	1.22577 (14)	0.74339 (4)	0.0202 (3)
C28	0.97681 (12)	1.15459 (14)	0.72058 (4)	0.0191 (3)
H28	1.0539	1.1662	0.7204	0.023*
C29	0.93207 (12)	1.06744 (14)	0.69819 (4)	0.0180 (3)
H29	0.9786	1.0200	0.6829	0.022*
C30	0.96087 (14)	1.31961 (15)	0.76827 (5)	0.0263 (4)
H30A	1.0407	1.3193	0.7654	0.040*
H30B	0.9417	1.3042	0.7949	0.040*
H30C	0.9320	1 3965	0 7607	0.040*
H1	0 5235 (14)	0.6854 (15)	0 5750 (5)	0.038 (6)*
H4	0.8327 (12)	0.9193(14)	0.6617 (5)	0.030(0) 0.027(5)*
117	0.0527 (12)	(17)	0.0017 (3)	0.027(3)

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0240 (7)	0.0156 (7)	0.0204 (6)	-0.0025 (6)	-0.0051 (5)	0.0004 (5)
N2	0.0179 (6)	0.0195 (7)	0.0176 (6)	-0.0004 (5)	-0.0005 (5)	-0.0004 (5)
N3	0.0164 (6)	0.0162 (7)	0.0181 (6)	0.0003 (5)	0.0000 (5)	0.0003 (5)
N4	0.0166 (6)	0.0193 (7)	0.0223 (7)	-0.0015 (6)	0.0030 (5)	-0.0053 (5)
N5	0.0234 (7)	0.0180 (7)	0.0208 (6)	-0.0017 (6)	-0.0018 (5)	-0.0008 (5)
N6	0.0181 (6)	0.0189 (7)	0.0179 (6)	-0.0022 (5)	-0.0003 (5)	-0.0008 (5)
C1	0.0162 (7)	0.0163 (8)	0.0174 (7)	0.0000 (6)	0.0021 (6)	-0.0006 (6)
C2	0.0190 (7)	0.0195 (8)	0.0180 (7)	-0.0029 (6)	0.0000 (6)	-0.0020 (6)
C3	0.0147 (7)	0.0180 (8)	0.0164 (7)	0.0002 (6)	0.0024 (6)	-0.0010 (6)
C4	0.0198 (7)	0.0204 (8)	0.0177 (7)	0.0030 (6)	-0.0010 (6)	0.0023 (6)
C5	0.0244 (8)	0.0166 (8)	0.0240 (8)	0.0013 (7)	0.0004 (7)	0.0029 (7)
C6	0.0214 (8)	0.0187 (8)	0.0259 (8)	-0.0028 (7)	-0.0023 (6)	-0.0023 (7)
C7	0.0190 (8)	0.0210 (8)	0.0211 (7)	-0.0004 (7)	-0.0038 (6)	0.0001 (6)
C8	0.0152 (7)	0.0172 (8)	0.0174 (7)	0.0014 (6)	0.0015 (6)	0.0002 (6)
C9	0.0167 (7)	0.0184 (8)	0.0182 (7)	0.0003 (6)	0.0028 (6)	0.0011 (6)
C10	0.0205 (8)	0.0146 (8)	0.0194 (7)	0.0006 (6)	0.0032 (6)	-0.0017 (6)
C11	0.0219 (8)	0.0229 (9)	0.0167 (7)	0.0023 (7)	0.0009 (6)	-0.0016 (6)
C12	0.0183 (7)	0.0193 (8)	0.0206 (7)	0.0029 (6)	0.0045 (6)	0.0016 (6)
C13	0.0242 (8)	0.0151 (8)	0.0273 (8)	-0.0022 (6)	0.0021 (7)	0.0019 (7)
C14	0.0233 (8)	0.0197 (8)	0.0235 (8)	-0.0045 (7)	-0.0021 (6)	-0.0007 (7)
C15	0.0292 (9)	0.0241 (9)	0.0273 (9)	0.0034 (7)	-0.0026 (7)	0.0043 (7)
C16	0.0198 (7)	0.0155 (8)	0.0169 (7)	-0.0025 (6)	0.0003 (6)	0.0019 (6)
C17	0.0223 (8)	0.0177 (8)	0.0195 (7)	-0.0003 (7)	0.0012 (6)	-0.0019 (6)
C18	0.0225 (8)	0.0172 (8)	0.0175 (7)	-0.0023 (7)	-0.0026 (6)	0.0030 (6)
C19	0.0277 (9)	0.0235 (9)	0.0228 (8)	-0.0052 (7)	-0.0056 (7)	-0.0009 (7)
C20	0.0196 (8)	0.0334 (10)	0.0292 (9)	-0.0080 (7)	-0.0048 (7)	0.0012 (8)
C21	0.0199 (8)	0.0326 (10)	0.0270 (8)	-0.0012 (7)	0.0014 (7)	0.0003 (7)
C22	0.0218 (8)	0.0241 (9)	0.0219 (8)	-0.0012 (7)	0.0002 (6)	-0.0020(7)
C23	0.0205 (8)	0.0171 (8)	0.0159 (7)	-0.0023 (6)	-0.0020 (6)	0.0027 (6)
C24	0.0211 (7)	0.0161 (8)	0.0159 (7)	-0.0018 (6)	-0.0009 (6)	0.0008 (6)
C25	0.0167 (7)	0.0251 (9)	0.0280 (8)	-0.0019 (7)	0.0019 (7)	-0.0059 (7)
C26	0.0220 (8)	0.0236 (9)	0.0269 (8)	-0.0008 (7)	0.0032 (7)	-0.0091 (7)
C27	0.0228 (8)	0.0184 (8)	0.0195 (7)	-0.0041 (6)	-0.0028 (6)	0.0009 (6)
C28	0.0169 (7)	0.0220 (8)	0.0184 (7)	-0.0028 (6)	-0.0018 (6)	0.0038 (6)
C29	0.0189 (7)	0.0183 (8)	0.0167 (7)	0.0029 (6)	0.0001 (6)	0.0011 (6)
C30	0.0252 (8)	0.0270 (9)	0.0269 (8)	-0.0060 (7)	-0.0014 (7)	-0.0060 (7)

Geometric parameters (Å, °)

N1-C1	1.369 (2)	C12—C15	1.507 (2)	
N1—C9	1.4090 (19)	C13—C14	1.381 (2)	
N1—H1	0.886 (9)	C13—H13	0.9500	
N2—C2	1.301 (2)	C14—H14	0.9500	
N2—C3	1.3793 (19)	C15—H15A	0.9800	
N3—C1	1.3096 (19)	C15—H15B	0.9800	

N3—C8	1.3751 (19)	C15—H15C	0.9800
N4—C16	1.365 (2)	C16—C17	1.439 (2)
N4—C24	1.4084 (19)	С17—Н17	0.9500
N4—H4	0.872 (9)	C18—C19	1.406 (2)
N5—C17	1.299 (2)	C18—C23	1.411 (2)
N5—C18	1.382 (2)	C19—C20	1.368 (2)
N6—C16	1.314 (2)	С19—Н19	0.9500
N6—C23	1.3745 (19)	C20—C21	1.405 (2)
C1—C2	1.440 (2)	C20—H20	0.9500
C2—H2	0.9500	C21—C22	1.371 (2)
C3—C4	1.405 (2)	C21—H21	0.9500
C3—C8	1.415 (2)	C22—C23	1.407 (2)
C4—C5	1.372 (2)	C22—H22	0.9500
C4—H4A	0.9500	C24—C29	1.394 (2)
C5—C6	1.406 (2)	$C_{24}$ $C_{25}$	1.397 (2)
C5—H5	0.9500	$C_{25} = C_{26}$	1 386 (2)
C6—C7	1 372 (2)	C25—H25	0.9500
C6—H6	0.9500	$C_{26} - C_{27}$	1.390(2)
C7 - C8	1407(2)	C26—H26	0.9500
C7—H7	0.9500	$C_{27}$ $C_{28}$	1.386(2)
C9-C10	1 393 (2)	$C_{27} = C_{20}$	1.500(2) 1.502(2)
C9-C14	1 396 (2)	$C_{28}$ $C_{29}$	1.302(2) 1.378(2)
C10-C11	1 389 (2)	C28—H28	0.9500
C10_H10	0.9500	C20—H20	0.9500
$C_{11} - C_{12}$	1.387(2)	C30_H30A	0.9500
C11_H11	0.9500	C30—H30R	0.9800
C12-C13	1 390 (2)	C30—H30D	0.9800
012-015	1.590 (2)	050-11500	0.9000
C1—N1—C9	130.07 (13)	C12—C15—H15B	109.5
C1—N1—H1	113.7 (13)	H15A—C15—H15B	109.5
C9—N1—H1	115.3 (13)	C12—C15—H15C	109.5
C2—N2—C3	116.66 (13)	H15A—C15—H15C	109.5
C1—N3—C8	116.08 (13)	H15B—C15—H15C	109.5
C16—N4—C24	129.98 (13)	N6-C16-N4	122.10 (14)
C16—N4—H4	115.4 (12)	N6—C16—C17	121.65 (14)
C24—N4—H4	114.5 (12)	N4—C16—C17	116.24 (14)
C17—N5—C18	116.44 (13)	N5—C17—C16	123.24 (14)
C16—N6—C23	116.15 (13)	N5—C17—H17	118.4
N3—C1—N1	122.78 (14)	C16—C17—H17	118.4
N3—C1—C2	121.74 (14)	N5-C18-C19	120.06 (14)
N1-C1-C2	115.48 (13)	N5-C18-C23	120.37 (14)
N2-C2-C1	123.17 (14)	C19—C18—C23	119.57 (15)
N2-C2-H2	118.4	$C_{20}$ $C_{19}$ $C_{18}$	120.31 (15)
C1—C2—H2	118.4	C20—C19—H19	119.8
N2-C3-C4	119.67 (13)	C18—C19—H19	119.8
N2-C3-C8	120.02 (14)	C19-C20-C21	120 13 (15)
C4 - C3 - C8	120.31 (14)	C19—C20—H20	119.9
$C_{5} - C_{4} - C_{3}$	119 80 (14)	$C_{21} - C_{20} - H_{20}$	119.9
	117.00 (17)	021 020 1120	11/./

C5—C4—H4A	120.1	C22—C21—C20	120.72 (16)
C3—C4—H4A	120.1	C22—C21—H21	119.6
C4—C5—C6	120.05 (15)	C20—C21—H21	119.6
С4—С5—Н5	120.0	C21—C22—C23	120.01 (15)
С6—С5—Н5	120.0	C21—C22—H22	120.0
C7 - C6 - C5	121.10(15)	$C^{23}$ $C^{22}$ $H^{22}$	120.0
C7_C6_H6	110 4	N6_C23_C22	120.0 118.66 (14)
$C_{2} = C_{0} = H_{0}$	110.4	N6 C23 C18	110.00(14) 122.00(14)
$C_{5}$	117.4	10 - 23 - 218	122.09(14)
$C_{0} - C_{1} - C_{8}$	119.90 (14)	$C_{22} = C_{23} = C_{18}$	119.23 (14)
	120.0	$C_{29} = C_{24} = C_{25}$	118.58 (14)
C8—C/—H/	120.0	C29—C24—N4	116.69 (14)
N3—C8—C7	118.90 (13)	C25—C24—N4	124.72 (14)
N3—C8—C3	122.33 (14)	C26—C25—C24	119.36 (15)
C7—C8—C3	118.77 (14)	С26—С25—Н25	120.3
C10—C9—C14	119.13 (14)	С24—С25—Н25	120.3
C10—C9—N1	124.21 (14)	C25—C26—C27	122.46 (15)
C14—C9—N1	116.65 (14)	С25—С26—Н26	118.8
C11—C10—C9	119.11 (14)	С27—С26—Н26	118.8
C11—C10—H10	120.4	C28—C27—C26	117.18 (14)
С9—С10—Н10	120.4	C28—C27—C30	121.66 (14)
C12—C11—C10	122.51 (15)	$C_{26} - C_{27} - C_{30}$	121.15 (15)
C12—C11—H11	118 7	$C_{29} C_{28} C_{27}$	121.58 (14)
C10_C11_H11	118.7	$C_{29} = C_{28} = H_{28}$	119.2
$C_{11}$ $C_{12}$ $C_{13}$	117.42(14)	$C_{23} = C_{23} = H_{23}$	119.2
$C_{11} = C_{12} = C_{13}$	117.42(14) 121.65(15)	$C_2 = C_2 $	119.2
	121.03 (13)	$C_{28} = C_{29} = C_{24}$	120.82 (14)
	120.93 (15)	C28—C29—H29	119.6
C14—C13—C12	121.35 (15)	С24—С29—Н29	119.6
C14—C13—H13	119.3	С27—С30—Н30А	109.5
C12—C13—H13	119.3	С27—С30—Н30В	109.5
C13—C14—C9	120.47 (15)	H30A—C30—H30B	109.5
C13—C14—H14	119.8	С27—С30—Н30С	109.5
C9—C14—H14	119.8	H30A—C30—H30C	109.5
C12—C15—H15A	109.5	H30B—C30—H30C	109.5
C8—N3—C1—N1	179.27 (13)	C23—N6—C16—N4	-178.95 (14)
C8—N3—C1—C2	-0.5 (2)	C23—N6—C16—C17	2.1 (2)
C9—N1—C1—N3	-0.9(2)	C24—N4—C16—N6	-4.0(3)
C9—N1—C1—C2	178.94 (14)	C24—N4—C16—C17	174.97 (14)
$C_3 - N_2 - C_2 - C_1$	04(2)	C18 - N5 - C17 - C16	-1.9(2)
$N_{3}$ $C_{1}$ $C_{2}$ $N_{2}$	0.1(2) 0.2(2)	N6-C16-C17-N5	-0.2(2)
$N_1 = C_1 = C_2 = N_2$	-170.62(14)	N4 C16 C17 N5	-170 18 (14)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	179.02(14) 170.47(14)	$N_{-} = C_{10} = C_{17} = N_{5}$	177.10(14)
$C_2 = N_2 = C_2 = C_2$	1/3.4/(14)	C17 = N5 = C19 = C12	(177.93(13))
12 - 102 - 103 - 103	-0.0(2)	$U_1 / - N_3 - U_1 \delta - U_2 \delta$	2.1 (2)
$N_2 - C_3 - C_4 - C_5$	1/8.80 (14)	$N_{0} - C_{10} - C_{10} - C_{20}$	-1/9.52(15)
C8—C3—C4—C5	-0.1 (2)	C23—C18—C19—C20	0.5 (2)
C3—C4—C5—C6	0.2 (2)	C18—C19—C20—C21	0.1 (3)
C4—C5—C6—C7	0.0 (2)	C19—C20—C21—C22	-0.6(3)
C5—C6—C7—C8	-0.4 (2)	C20—C21—C22—C23	0.4 (3)

C1—N3—C8—C7	179.76 (13)	C16—N6—C23—C22	178.58 (14)
C1—N3—C8—C3	0.3 (2)	C16—N6—C23—C18	-1.9 (2)
C6—C7—C8—N3	-178.90 (14)	C21—C22—C23—N6	179.75 (15)
C6—C7—C8—C3	0.6 (2)	C21—C22—C23—C18	0.3 (2)
N2—C3—C8—N3	0.3 (2)	N5-C18-C23-N6	-0.2 (2)
C4—C3—C8—N3	179.13 (14)	C19—C18—C23—N6	179.87 (14)
N2—C3—C8—C7	-179.19 (13)	N5-C18-C23-C22	179.32 (14)
C4—C3—C8—C7	-0.3 (2)	C19—C18—C23—C22	-0.7 (2)
C1-N1-C9-C10	-15.1 (2)	C16—N4—C24—C29	-176.07 (15)
C1-N1-C9-C14	166.00 (15)	C16—N4—C24—C25	3.4 (3)
C14—C9—C10—C11	0.3 (2)	C29—C24—C25—C26	-0.4 (2)
N1-C9-C10-C11	-178.57 (14)	N4-C24-C25-C26	-179.90 (15)
C9-C10-C11-C12	-0.4 (2)	C24—C25—C26—C27	-0.3 (3)
C10-C11-C12-C13	-0.1 (2)	C25—C26—C27—C28	0.9 (2)
C10-C11-C12-C15	-179.21 (14)	C25—C26—C27—C30	-178.90 (15)
C11—C12—C13—C14	0.6 (2)	C26—C27—C28—C29	-0.8 (2)
C15—C12—C13—C14	179.69 (15)	C30—C27—C28—C29	179.05 (14)
C12—C13—C14—C9	-0.6 (2)	C27—C28—C29—C24	0.1 (2)
C10-C9-C14-C13	0.1 (2)	C25—C24—C29—C28	0.5 (2)
N1-C9-C14-C13	179.11 (14)	N4-C24-C29-C28	-179.91 (13)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D····A	D—H···A
N1—H1…N5	0.89(1)	2.26(1)	3.114 (2)	163 (2)
$N4$ — $H4$ ··· $N2^{i}$	0.87 (1)	2.19(1)	3.017 (2)	157 (2)

Symmetry code: (i) -x+3/2, y+1/2, z.