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3-Fluoro-*N*-(*p*-tolyl)benzamideAamer Saeed,^{a*} Rasheed Ahmad Khera,^a Kazuma Gotoh^b and Hiroyuki Ishida^b^aDepartment of Chemistry, Quaid-i-Azam University, Islamabad 45320, Pakistan, and ^bDepartment of Chemistry, Faculty of Science, Okayama University, Okayama 700-8530, Japan

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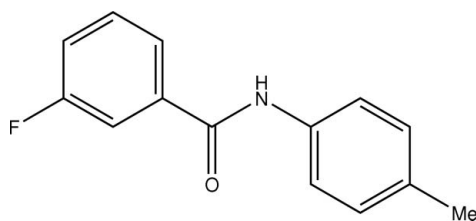
Received 17 September 2008; accepted 7 October 2008

Key indicators: single-crystal X-ray study; $T = 223$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.073; wR factor = 0.240; data-to-parameter ratio = 21.2.

In the crystal structure of the title compound, $\text{C}_{14}\text{H}_{12}\text{FNO}$, the amide $-\text{NHCO}-$ mean plane makes dihedral angles of 28.6 (2) and 37.5 (2)° with the mean planes through the fluorobenzene and methylbenzene units, respectively. The dihedral angle between the two benzene ring mean planes is 65.69 (10)°. In the crystal structure, molecules are linked through $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds and stack along the b axis.

Related literature

For related structures, see: Chopra & Row (2005); Saeed *et al.* (2008).



Experimental

Crystal data

 $\text{C}_{14}\text{H}_{12}\text{FNO}$ $M_r = 229.25$

Monoclinic, $C2/c$
 $a = 27.645$ (3) Å
 $b = 5.2618$ (6) Å
 $c = 15.892$ (2) Å
 $\beta = 93.519$ (3)°
 $V = 2307.3$ (5) Å³

$Z = 8$
 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 223$ (1) K
 $0.40 \times 0.35 \times 0.18$ mm

Data collection

Rigaku R-Axis RAPIDII diffractometer
 Absorption correction: numerical (*ABSCOR*; Higashi, 1999)
 $T_{\min} = 0.968$, $T_{\max} = 0.983$

13860 measured reflections
 3357 independent reflections
 1779 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.055$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.073$
 $wR(F^2) = 0.240$
 $S = 1.01$
 3357 reflections
 158 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.32$ e Å⁻³
 $\Delta\rho_{\min} = -0.21$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O1}^i$	0.77 (2)	2.35 (2)	3.087 (3)	161 (2)

Symmetry code: (i) $x, y - 1, z$.

Data collection: *PROCESS-AUTO* (Rigaku/MSK, 2004); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku/MSK, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *CrystalStructure* and *PLATON* (Spek, 2003).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2064).

References

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supporting information

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3-Fluoro-*N*-(*p*-tolyl)benzamide

Aamer Saeed, Rasheed Ahmad Khera, Kazuma Gotoh and Hiroyuki Ishida

S1. Comment

The background to this study has been described in our earlier paper on 4-chloro-*N*-(2-chlorophenyl)-benzamide (Saeed *et al.*, 2008).

In the crystal structure of the title compound the two benzene rings are considerably twisted with respect to one another, with a dihedral angle of 65.69 (10)°. The amide –NHCO– mean plane makes dihedral angles of 28.6 (2) and 37.5 (2)° with the best mean planes through the fluorobenzene and methylbenzene units, respectively. In the crystal the molecules are linked through N—H···O hydrogen bonds and stack up the *b* axis.

No C—H···F hydrogen bonds were observed here, in contrast to the situation in 4-fluoro-*N*-(2-fluorophenyl)-benzamide (Chopra & Row, 2005).

S2. Experimental

4-Fluorobenzoyl chloride (5.4 mmol) in CHCl₃ was treated with 4-methylaniline (21.6 mmol) under a nitrogen atmosphere at reflux for 4 h. Upon cooling the reaction mixture was diluted with CHCl₃ and washed consecutively with aq 1 M HCl and saturated aq NaHCO₃. The organic layer was dried over anhydrous sodium sulfate and concentrated under reduced pressure. Crystallization of the residue in CHCl₃ afforded the title compound (84%) as white needles: Anal. calcd. for C₁₄H₁₂FNO: C 73.35, H 5.28, N 6.11%; found: C 73.30, H 5.32, N 6.09%.

S3. Refinement

The N-bound H atom was located in a difference Fourier map and was freely refined. The other H atoms were positioned geometrically (C—H = 0.94 and 0.97 Å) and treated as riding atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

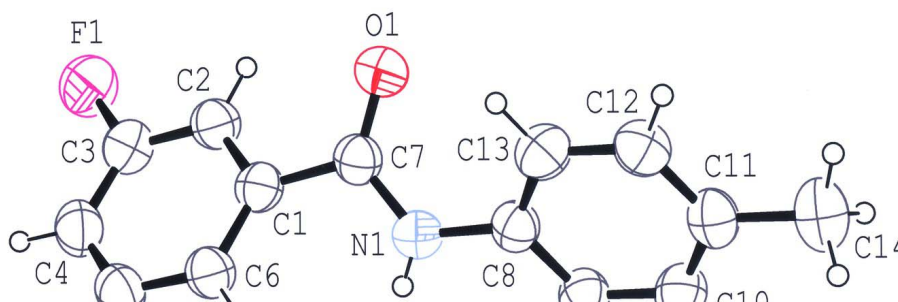


Figure 1

A view of the molecular structure of the title compound. The displacement ellipsoids are drawn at the 40% probability level.

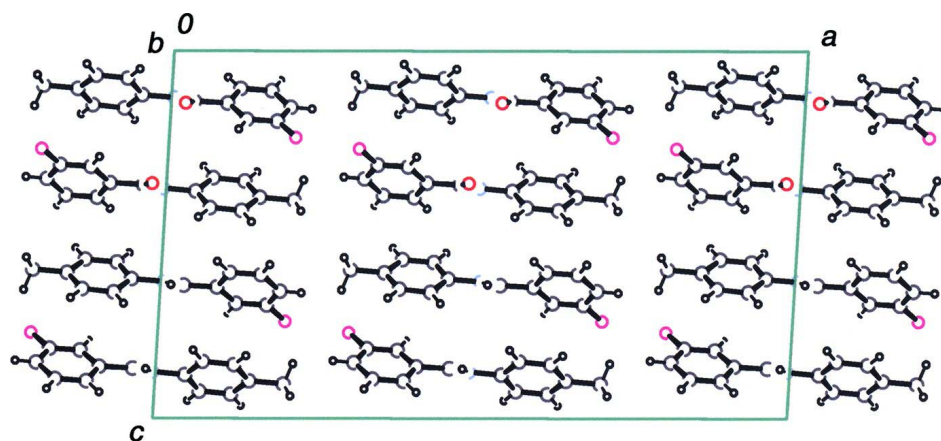


Figure 2

A view along the b axis of the crystal packing of the title compound.

3-Fluoro-*N*-(*p*-tolyl)benzamide

Crystal data

$C_{14}H_{12}FNO$

$M_r = 229.25$

Monoclinic, $C2/c$

Hall symbol: $-C\ 2yc$

$a = 27.645\ (3)\ \text{\AA}$

$b = 5.2618\ (6)\ \text{\AA}$

$c = 15.892\ (2)\ \text{\AA}$

$\beta = 93.519\ (3)^\circ$

$V = 2307.3\ (5)\ \text{\AA}^3$

$Z = 8$

$F(000) = 960.00$

$D_x = 1.320\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71075\ \text{\AA}$

Cell parameters from 7127 reflections

$\theta = 3.0\text{--}30.0^\circ$

$\mu = 0.09\ \text{mm}^{-1}$

$T = 223\ \text{K}$

Block, colorless

$0.40 \times 0.35 \times 0.18\ \text{mm}$

Data collection

Rigaku R-AXIS RAPIDII
diffractometer

Detector resolution: $10.00\ \text{pixels mm}^{-1}$

ω scans

Absorption correction: numerical
(*ABSCOR*; Higashi, 1999)

$T_{\min} = 0.968$, $T_{\max} = 0.983$

13860 measured reflections

3357 independent reflections

1779 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.055$

$\theta_{\max} = 30.0^\circ$

$h = -38 \rightarrow 38$

$k = -6 \rightarrow 7$

$l = -22 \rightarrow 22$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.073$

$wR(F^2) = 0.240$

$S = 1.01$

3357 reflections

158 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H atoms treated by a mixture of independent
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.1335P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.32\ \text{e \AA}^{-3}$

$\Delta\rho_{\min} = -0.21\ \text{e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.69906 (5)	0.5995 (3)	0.23465 (10)	0.0965 (5)
O1	0.52251 (5)	0.6096 (3)	0.14071 (11)	0.0781 (5)
N1	0.50521 (6)	0.1882 (4)	0.12722 (11)	0.0639 (5)
C1	0.58887 (6)	0.3226 (4)	0.14164 (11)	0.0594 (5)
C2	0.62009 (7)	0.4878 (4)	0.18591 (12)	0.0656 (5)
H2	0.6081	0.6327	0.2122	0.079*
C3	0.66868 (8)	0.4361 (4)	0.19066 (14)	0.0703 (5)
C4	0.68819 (7)	0.2292 (5)	0.15293 (14)	0.0745 (6)
H4	0.7217	0.1984	0.1576	0.089*
C5	0.65712 (7)	0.0677 (4)	0.10794 (14)	0.0732 (6)
H5	0.6697	-0.0741	0.0807	0.088*
C6	0.60772 (7)	0.1105 (4)	0.10214 (12)	0.0657 (5)
H6	0.5869	-0.0026	0.0718	0.079*
C7	0.53603 (7)	0.3860 (4)	0.13667 (11)	0.0608 (5)
C8	0.45358 (7)	0.2026 (4)	0.11934 (11)	0.0599 (5)
C9	0.42706 (7)	0.0146 (4)	0.15561 (13)	0.0675 (5)
H9	0.4431	-0.1164	0.1865	0.081*
C10	0.37703 (7)	0.0181 (4)	0.14673 (13)	0.0729 (6)
H10	0.3594	-0.1125	0.1712	0.087*
C11	0.35222 (7)	0.2093 (4)	0.10264 (11)	0.0675 (5)
C12	0.37940 (7)	0.3956 (4)	0.06655 (13)	0.0705 (6)
H12	0.3634	0.5269	0.0359	0.085*
C13	0.42967 (8)	0.3940 (4)	0.07434 (13)	0.0704 (5)
H13	0.4474	0.5228	0.0491	0.085*
C14	0.29774 (8)	0.2111 (6)	0.09397 (16)	0.0916 (8)
H14A	0.2850	0.2043	0.1495	0.137*
H14B	0.2864	0.0647	0.0613	0.137*
H14C	0.2867	0.3655	0.0655	0.137*
H1	0.5158 (7)	0.055 (4)	0.1353 (12)	0.058 (6)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0752 (8)	0.0995 (10)	0.1131 (11)	-0.0114 (7)	-0.0082 (7)	-0.0173 (8)
O1	0.0711 (9)	0.0616 (9)	0.1023 (12)	0.0027 (7)	0.0107 (7)	-0.0059 (7)
N1	0.0633 (9)	0.0566 (9)	0.0719 (10)	0.0009 (9)	0.0063 (7)	0.0030 (8)

C1	0.0655 (11)	0.0619 (10)	0.0518 (9)	-0.0016 (8)	0.0112 (7)	0.0049 (7)
C2	0.0692 (12)	0.0646 (11)	0.0636 (11)	-0.0002 (10)	0.0087 (8)	-0.0013 (9)
C3	0.0681 (12)	0.0758 (13)	0.0669 (11)	-0.0049 (10)	0.0035 (9)	0.0027 (9)
C4	0.0648 (11)	0.0822 (14)	0.0779 (13)	0.0043 (11)	0.0148 (9)	0.0089 (11)
C5	0.0746 (13)	0.0759 (13)	0.0709 (12)	0.0088 (11)	0.0190 (9)	-0.0006 (10)
C6	0.0705 (11)	0.0665 (11)	0.0610 (10)	0.0004 (9)	0.0127 (8)	-0.0028 (8)
C7	0.0625 (10)	0.0636 (11)	0.0570 (10)	0.0008 (9)	0.0088 (8)	-0.0002 (8)
C8	0.0605 (10)	0.0645 (10)	0.0551 (9)	-0.0013 (8)	0.0076 (7)	-0.0050 (8)
C9	0.0657 (11)	0.0717 (12)	0.0659 (11)	0.0011 (9)	0.0100 (8)	0.0067 (9)
C10	0.0717 (12)	0.0787 (13)	0.0694 (12)	-0.0049 (11)	0.0141 (9)	0.0063 (10)
C11	0.0635 (11)	0.0852 (14)	0.0540 (10)	0.0019 (10)	0.0058 (8)	-0.0085 (9)
C12	0.0712 (12)	0.0729 (13)	0.0663 (12)	0.0063 (10)	-0.0044 (9)	0.0028 (9)
C13	0.0763 (12)	0.0725 (12)	0.0626 (11)	-0.0045 (10)	0.0041 (9)	0.0088 (9)
C14	0.0683 (13)	0.126 (2)	0.0807 (15)	0.0047 (14)	0.0050 (11)	-0.0041 (14)

Geometric parameters (Å, °)

F1—C3	1.364 (2)	C6—H6	0.9400
O1—C7	1.238 (2)	C8—C9	1.378 (3)
N1—C7	1.347 (3)	C8—C13	1.380 (3)
N1—C8	1.427 (2)	C9—C10	1.382 (3)
N1—H1	0.77 (2)	C9—H9	0.9400
C1—C2	1.386 (3)	C10—C11	1.384 (3)
C1—C6	1.397 (3)	C10—H10	0.9400
C1—C7	1.496 (3)	C11—C12	1.381 (3)
C2—C3	1.368 (3)	C11—C14	1.504 (3)
C2—H2	0.9400	C12—C13	1.388 (3)
C3—C4	1.370 (3)	C12—H12	0.9400
C4—C5	1.377 (3)	C13—H13	0.9400
C4—H4	0.9400	C14—H14A	0.9700
C5—C6	1.382 (3)	C14—H14B	0.9700
C5—H5	0.9400	C14—H14C	0.9700
C7—N1—C8	126.25 (17)	C9—C8—C13	119.37 (18)
C7—N1—H1	117.0 (15)	C9—C8—N1	118.68 (17)
C8—N1—H1	115.6 (15)	C13—C8—N1	121.91 (17)
C2—C1—C6	119.43 (17)	C8—C9—C10	120.14 (19)
C2—C1—C7	117.56 (17)	C8—C9—H9	119.9
C6—C1—C7	122.99 (17)	C10—C9—H9	119.9
C3—C2—C1	118.83 (19)	C9—C10—C11	121.58 (19)
C3—C2—H2	120.6	C9—C10—H10	119.2
C1—C2—H2	120.6	C11—C10—H10	119.2
F1—C3—C2	118.34 (19)	C12—C11—C10	117.44 (17)
F1—C3—C4	118.61 (19)	C12—C11—C14	121.6 (2)
C2—C3—C4	123.05 (19)	C10—C11—C14	120.9 (2)
C3—C4—C5	117.92 (19)	C11—C12—C13	121.75 (18)
C3—C4—H4	121.0	C11—C12—H12	119.1
C5—C4—H4	121.0	C13—C12—H12	119.1

C4—C5—C6	121.1 (2)	C8—C13—C12	119.71 (18)
C4—C5—H5	119.4	C8—C13—H13	120.1
C6—C5—H5	119.4	C12—C13—H13	120.1
C5—C6—C1	119.65 (19)	C11—C14—H14A	109.5
C5—C6—H6	120.2	C11—C14—H14B	109.5
C1—C6—H6	120.2	H14A—C14—H14B	109.5
O1—C7—N1	123.30 (18)	C11—C14—H14C	109.5
O1—C7—C1	120.43 (17)	H14A—C14—H14C	109.5
N1—C7—C1	116.27 (17)	H14B—C14—H14C	109.5
C6—C1—C2—C3	-0.9 (3)	C2—C1—C7—N1	-152.62 (17)
C7—C1—C2—C3	-179.24 (17)	C6—C1—C7—N1	29.1 (3)
C1—C2—C3—F1	-179.77 (18)	C7—N1—C8—C9	-144.0 (2)
C1—C2—C3—C4	0.6 (3)	C7—N1—C8—C13	38.1 (3)
F1—C3—C4—C5	-179.27 (19)	C13—C8—C9—C10	0.1 (3)
C2—C3—C4—C5	0.3 (3)	N1—C8—C9—C10	-177.82 (17)
C3—C4—C5—C6	-1.0 (3)	C8—C9—C10—C11	-0.8 (3)
C4—C5—C6—C1	0.8 (3)	C9—C10—C11—C12	0.9 (3)
C2—C1—C6—C5	0.2 (3)	C9—C10—C11—C14	-179.65 (19)
C7—C1—C6—C5	178.47 (18)	C10—C11—C12—C13	-0.5 (3)
C8—N1—C7—O1	0.9 (3)	C14—C11—C12—C13	-179.9 (2)
C8—N1—C7—C1	-178.66 (15)	C9—C8—C13—C12	0.3 (3)
C2—C1—C7—O1	27.8 (3)	N1—C8—C13—C12	178.19 (17)
C6—C1—C7—O1	-150.5 (2)	C11—C12—C13—C8	-0.1 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots O1 ⁱ	0.77 (2)	2.35 (2)	3.087 (3)	161 (2)

Symmetry code: (i) $x, y-1, z$.