

Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

1-{Phenyl[1-(p-tolyl)ethylamino]methyl}-2-naphthol

Yong Hua Li,* Min Min Zhao and Yuan Zhang

Ordered Matter Science Research Center, College of Chemistry and Chemical Engineering, Southeast University, Nanjing 211189, People's Republic of China Correspondence e-mail: liyhnju@hotmail.com

Received 26 August 2008; accepted 22 September 2008

Key indicators: single-crystal X-ray study; T = 292 K; mean σ (C–C) = 0.007 Å; R factor = 0.055; wR factor = 0.136; data-to-parameter ratio = 9.5.

The title compound, C₂₆H₂₅NO, was obtained via a one-pot synthesis from the reaction of 2-naphthol, 1-(p-tolyl)ethylamine, p-toluenesulfonic acid and benzaldehyde. There are three molecules per asymmetric unit, all having similar conformations. There are intramolecular $O-H \cdots N$ and C- $H \cdots O$ hydrogen bonds, with only van der Waals forces found between molecules.

Related literature

For background, see: Devi & Bhuyan (2004); Domling & Ugi (2000); Hulme & Gore (2003); Ugi (1962). For related literature, see: Liu et al. (2001).



Experimental

Crystal data

β

C ₂₆ H ₂₅ NO	$\gamma = 72.39 \ (3)^{\circ}$
$M_r = 367.47$	V = 1579.0 (5) Å ³
Triclinic, P1	Z = 3
a = 9.3046 (19) Å	Mo $K\alpha$ radiation
b = 13.126 (3) Å	$\mu = 0.07 \text{ mm}^{-1}$
c = 13.572 (3) Å	T = 292 (2) K
$\alpha = 88.14 \ (3)^{\circ}$	$0.50 \times 0.40 \times 0.30 \text{ mm}$
$\beta = 89.99 \ (2)^{\circ}$	

Data collection

Rigaku SCXmini diffractometer Absorption correction: multi-scan (CrystalClear; Rigaku, 2005) $T_{\min} = 0.950, T_{\max} = 0.980$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.055$	
$wR(F^2) = 0.136$	
S = 1.05	
7208 reflections	
760 parameters	

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$01 - H1 \cdots N1$ $02 - H8 \cdots N2$ $03 - H2 \cdots N3$ $C6 - H53A \cdots O2$	0.82	1.86	2.571 (4)	144
	0.82	1.92	2.629 (5)	144
	0.82	1.87	2.597 (5)	147
	0.93	2.58	3.327 (6)	138

Data collection: CrystalClear (Rigaku, 2005); cell refinement: CrystalClear; data reduction: CrystalClear; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL/PC (Sheldrick, 2008); software used to prepare material for publication: SHELXTL/PC.

This work was supported by a start-up grant from Southeast University to Professor Yong-Hua Li.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FL2220).

References

Devi, I. & Bhuyan, P. J. (2004). Tetrahedron Lett. 45, 8625-8627. Domling, A. & Ugi, I. (2000). Angew. Chem. Int. Ed. 39, 3168-3210. Hulme, C. & Gore, V. (2003). Curr. Med. Chem. 10, 51-8. Liu, D.-X., Zhang, L.-C., Wang, Q., Da, C.-S., Xin, Z.-Q., Wang, R., Choi, M. C. K. & Chan, A. S. C. (2001). Org. Lett. 3, 2733-2735. Rigaku (2005). CrystalClear. Rigaku Corporation, Tokyo, Japan. Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122. Ugi, I. (1962). Angew. Chem. Int. Ed. Engl. 1, 8-21.

16610 measured reflections

 $R_{\rm int} = 0.048$

3 restraints

 $\Delta \rho_{\rm max} = 0.15 \text{ e} \text{ Å}^{-3}$ $\Delta \rho_{\rm min}$ = -0.16 e Å⁻³

7208 independent reflections

4449 reflections with $I > 2\sigma(I)$

H-atom parameters constrained

supporting information

Acta Cryst. (2008). E64, o2005 [doi:10.1107/S1600536808030365]

1-{Phenyl[1-(p-tolyl)ethylamino]methyl}-2-naphthol

Yong Hua Li, Min Min Zhao and Yuan Zhang

S1. Comment

Multi-component reactions (MCRs) (Hulme & Gore, 2003; Ugi,1962) involving at least three starting materials in a onepot reaction have attracted considerable attention in terms of saving both energy and raw materials (Devi & Bhuyan, 2004). Compared to conventional multi-step organic syntheses, MCRs have merits over multi-step reactions that include the simplicity of a one-pot procedure and the buildup of complex molecules (Domling & Ugi, 2000). Here we report the synthesis and crystal structure of the title compound (I, Fig. 1), obtained by a four-component condensation reaction as described in the experimental section.

I is an optically active derivative that crystallized with three molecules per asymmetric unit which is very rare. A few examples of similar compounds which contain two molecules per asymmetric unit have been reported (Liu *et al.*, 2001).

All three molecules in the asymmetric unit have the same relative conformation for both chiral carbon atoms. The dihedral angles between the rings A (C1–C6), B (C8–C17) and C (C20–C25) are A/B =71.90°, B/C =20.28°. The dihedral angles between the rings D (C38–C43), E (C27–C36) and F (C46–C51) are D/E =73.62°, E/F = 45.16° and the dihedral angles between the rings G (C64–C69), H (C53–C62) and I (C72–C77) are G/H =77.41°, H/I =62.04°. The three molecules are all stabilized by intramolecular O—H…N hydrogen bonding however, only one is involved in intramolecular C—H…O hydrogen bonds, no similar C—H…O distances are found in the other two molecules (Table 1). Intermolecular attractions are only on the order of Van der Waals forces.

S2. Experimental

Benzaldehyde (1.59 g, 0.015 mol) and *p*-toluenesulfonic acid (0.1 g) was added to 2-naphthol (2.16 g, 0.015 mol) without solvent. 1-(*p*-tolyl)ethylamine (2.025 g, 0.015 mol) was added dropwise with cooling to 0°C to the above mixture under nitrogen. The temperature was then gradually raised to 120°C over a period of hour and the mixture was stirred at this temperature for 24 h, then 20 ml of ethanol 95% was added, the precipitate was filtered and washed with a small amount of ethanol 95%. The title compound was isolated using column chromatography (Petroleum ether: ethyl acetate-5:1). Single crystals suitable for X-ray diffraction analysis were obtained from slow evaporation of an ethyl acetate solution.

S3. Refinement

H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93–0.98 Å and $U_{iso}(H) = 1.2-1.5U_{eq}(C)$. In the absence of significant anomalous scattering effects, 6529 Friedel pairs were merged.



Figure 1

The molecular structure of the title compound, showing the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

1-{Phenyl[1-(p-tolyl)ethylamino]methyl}-2-naphthol

Crystal data

C₂₆H₂₅NO $M_r = 367.47$ Triclinic, P1 Hall symbol: P1 a = 9.3046 (19) Å b = 13.126 (3) Å c = 13.572 (3) Å $a = 88.14 (3)^{\circ}$ $\beta = 89.99 (2)^{\circ}$ $\gamma = 72.39 (3)^{\circ}$ $V = 1579.0 (5) \text{ Å}^{3}$

Data collection

Rigaku SCXmini	16610 measur
diffractometer	7208 independ
Radiation source: fine-focus sealed tube	4449 reflection
Graphite monochromator	$R_{\rm int} = 0.048$
Detector resolution: 13.6612 pixels mm ⁻¹	$\theta_{\rm max} = 27.5^{\circ}, \theta$
CCD p rofilefitting scans	$h = -12 \rightarrow 12$
Absorption correction: multi-scan	$k = -17 \rightarrow 17$
(CrystalClear; Rigaku, 2005)	$l = -17 \rightarrow 17$
$T_{\min} = 0.950, \ T_{\max} = 0.980$	
P. 4	

Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.055$ $wR(F^2) = 0.136$ S = 1.057208 reflections 760 parameters 3 restraints Z = 3 F(000) = 588 $D_x = 1.159 \text{ Mg m}^{-3}$ Mo K α radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 7208 reflections $\theta = 3.0-27.5^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$ T = 292 K Prism, colorless $0.50 \times 0.40 \times 0.30 \text{ mm}$

16610 measured reflections 7208 independent reflections 4449 reflections with $I > 2\sigma(I)$ $R_{int} = 0.048$ $\theta_{max} = 27.5^{\circ}, \ \theta_{min} = 3.0^{\circ}$ $h = -12 \rightarrow 12$ $k = -17 \rightarrow 17$ $l = -17 \rightarrow 17$

Primary atom site location: structure-invariant direct methods Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0531P)^{2} + 0.0528P] \qquad \Delta \rho_{\max} = 0.15 \text{ e } \text{\AA}^{-3}$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3 \qquad \Delta \rho_{\min} = -0.16 \text{ e } \text{\AA}^{-3}$ $(\Delta/\sigma)_{\max} < 0.001$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes. **Refinement**. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 ,

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates	and isotropic of	or equivalent isotro	pic displacement	parameters	$(Å^2)$	ļ
-------------------------------	------------------	----------------------	------------------	------------	---------	---

	x	у	Z	$U_{ m iso}$ */ $U_{ m eq}$	
C1	0.3155 (4)	0.7172 (3)	0.8864 (3)	0.0628 (9)	
C2	0.2500 (5)	0.6843 (4)	0.9671 (3)	0.0804 (12)	
H50A	0.1711	0.6557	0.9583	0.096*	
C3	0.3005 (7)	0.6934 (4)	1.0618 (4)	0.0998 (15)	
H75A	0.2539	0.6720	1.1158	0.120*	
C4	0.4167 (7)	0.7332 (4)	1.0759 (4)	0.0918 (15)	
H67A	0.4523	0.7367	1.1394	0.110*	
C5	0.4819 (6)	0.7682 (4)	0.9965 (4)	0.0867 (13)	
H61A	0.5593	0.7982	1.0059	0.104*	
C6	0.4315 (5)	0.7588 (3)	0.9024 (3)	0.0759 (11)	
H53A	0.4775	0.7812	0.8486	0.091*	
C7	0.2582 (4)	0.7047 (3)	0.7836 (3)	0.0640 (10)	
H30A	0.1699	0.6793	0.7913	0.077*	
C8	0.2098 (5)	0.8090 (3)	0.7229 (3)	0.0695 (10)	
C9	0.3141 (7)	0.8430 (4)	0.6695 (3)	0.0906 (14)	
C10	0.2735 (11)	0.9451 (6)	0.6209 (4)	0.116 (2)	
H73A	0.3460	0.9675	0.5871	0.140*	
C11	0.1332 (13)	1.0083 (6)	0.6237 (5)	0.131 (3)	
H76A	0.1100	1.0751	0.5917	0.157*	
C12	0.0190 (8)	0.9803 (4)	0.6716 (4)	0.1030 (18)	
C13	-0.1273 (12)	1.0453 (6)	0.6728 (6)	0.158 (4)	
H81A	-0.1519	1.1115	0.6395	0.190*	
C14	-0.2387 (11)	1.0159 (9)	0.7215 (6)	0.182 (5)	
H83A	-0.3369	1.0618	0.7215	0.218*	
C15	-0.2030 (8)	0.9166 (8)	0.7711 (5)	0.156 (3)	
H84A	-0.2781	0.8957	0.8035	0.187*	
C16	-0.0564 (6)	0.8488 (5)	0.7722 (4)	0.1069 (17)	
H72A	-0.0346	0.7832	0.8062	0.128*	
C17	0.0586 (6)	0.8762 (4)	0.7240 (3)	0.0815 (13)	
C18	0.3134 (4)	0.5779 (3)	0.6466 (3)	0.0664 (10)	
H99A	0.2650	0.6378	0.6009	0.080*	
C19	0.4459 (5)	0.5019 (4)	0.5929 (4)	0.0968 (15)	

H79A	0.5165	0.5391	0.5739	0.145*
H79B	0.4093	0.4770	0.5351	0.145*
H79C	0.4951	0.4420	0.6359	0.145*
C20	0.1955 (4)	0.5256 (3)	0.6774 (2)	0.0573 (9)
C21	0.0467 (5)	0.5712 (3)	0.6513 (3)	0.0695 (10)
H54A	0.0181	0.6346	0.6135	0.083*
C22	-0.0609(5)	0.5239 (5)	0.6809 (3)	0.0864 (13)
H51A	-0.1607	0.5561	0.6615	0.104*
C23	-0.0267 (7)	0.4327 (5)	0.7369 (3)	0.0868 (14)
C24	0.1216 (7)	0.3864 (4)	0.7642 (3)	0.0882 (14)
H52A	0.1485	0.3237	0.8031	0.106*
C25	0.2306 (5)	0.4318 (3)	0.7348 (3)	0.0700 (10)
H27A	0.3303	0.3987	0.7539	0.084*
C26	-0.1445 (9)	0.3811 (7)	0.7711 (5)	0.151 (3)
H10D	-0.2419	0.4228	0.7463	0.227*
H10E	-0.1461	0.3778	0.8418	0.227*
H10F	-0.1202	0.3101	0.7468	0.227*
C27	0.0607 (4)	0.6526 (3)	0.3775 (2)	0.0514 (8)
C28	0.2086 (4)	0.6212 (3)	0.3414 (3)	0.0612 (9)
H3A	0.2409	0.5591	0.3058	0.073*
C29	0.3079 (5)	0.6790 (3)	0.3568 (3)	0.0753 (11)
H47A	0.4052	0.6557	0.3320	0.090*
C30	0.2615 (6)	0.7725 (4)	0.4098 (3)	0.0811 (12)
H44A	0.3285	0.8110	0.4214	0.097*
C31	0.1192 (5)	0.8072 (3)	0.4444 (3)	0.0746 (11)
H20A	0.0893	0.8699	0.4792	0.090*
C32	0.0139 (4)	0.7496 (3)	0.4285 (3)	0.0594 (9)
C33	-0.1334 (5)	0.7844 (3)	0.4658 (3)	0.0693 (10)
H18A	-0.1637	0.8467	0.5011	0.083*
C34	-0.2299 (5)	0.7295 (3)	0.4514 (3)	0.0723 (11)
H35A	-0.3274	0.7552	0.4754	0.087*
C35	-0.1874 (4)	0.6327 (3)	0.4001 (3)	0.0603 (9)
C36	-0.0418 (4)	0.5921 (3)	0.3643 (2)	0.0516 (8)
C37	0.0086 (4)	0.4862 (3)	0.3122 (3)	0.0530 (8)
H6A	0.1141	0.4508	0.3309	0.064*
C38	0.0010 (4)	0.4984 (3)	0.2004 (3)	0.0514 (8)
C39	0.1208 (4)	0.4455 (3)	0.1437 (3)	0.0696 (10)
H36A	0.2089	0.4031	0.1745	0.084*
C40	0.1140 (5)	0.4536 (4)	0.0429 (3)	0.0782 (11)
H80A	0.1972	0.4178	0.0061	0.094*
C41	-0.0152 (5)	0.5143 (3)	-0.0036 (3)	0.0721 (11)
H38A	-0.0201	0.5200	-0.0720	0.086*
C42	-0.1368 (5)	0.5663 (3)	0.0509 (3)	0.0740 (11)
H46A	-0.2253	0.6070	0.0194	0.089*
C43	-0.1292 (4)	0.5590 (3)	0.1527 (3)	0.0652 (10)
H57A	-0.2125	0.5952	0.1893	0.078*
C44	-0.0337 (4)	0.3634 (3)	0.4442 (3)	0.0628 (9)
H32A	-0.0308	0.4192	0.4896	0.075*

C45	-0.1560(5)	0.3160 (4)	0.4792 (4)	0.0927 (14)
H56A	-0.2519	0.3706	0.4768	0.139*
H56B	-0.1343	0.2883	0.5457	0.139*
H56C	-0.1591	0.2592	0.4373	0.139*
C46	0.1198 (4)	0.2804 (3)	0.4471 (2)	0.0535 (8)
C47	0.2259 (5)	0.2823 (3)	0.5164 (3)	0.0752 (11)
H64A	0.2042	0.3380	0.5601	0.090*
C48	0.3640 (5)	0.2044 (4)	0.5235 (4)	0.0903 (13)
H71A	0.4328	0.2081	0.5719	0.108*
C49	0.4011 (5)	0.1209 (4)	0.4594 (3)	0.0783 (12)
C50	0.2946 (5)	0.1187 (3)	0.3913 (3)	0.0719 (11)
H59A	0.3156	0.0621	0.3487	0.086*
C51	0.1580 (4)	0.1963 (3)	0.3830 (3)	0.0660 (10)
H22A	0.0901	0.1926	0.3340	0.079*
C52	0.5525 (6)	0.0356 (5)	0.4674 (5)	0.129 (2)
H82A	0.5590	-0.0161	0.4179	0.193*
H82B	0.5632	0.0008	0.5315	0.193*
H82C	0.6315	0.0680	0.4580	0.193*
C53	0.5031 (3)	0.2743 (3)	0.1125 (3)	0.0528 (8)
C54	0.4856 (4)	0.2479 (3)	0.2123 (3)	0.0571 (8)
H4A	0.4625	0.1852	0.2281	0.069*
C55	0.5015 (4)	0.3117 (3)	0.2868 (3)	0.0658 (10)
H15A	0.4881	0.2923	0.3518	0.079*
C56	0.5374 (4)	0.4049 (3)	0.2657 (4)	0.0761 (12)
H41A	0.5478	0.4481	0.3165	0.091*
C57	0.5572 (4)	0.4327 (3)	0.1722 (4)	0.0755 (12)
H21A	0.5826	0.4950	0.1590	0.091*
C58	0.5403 (4)	0.3698 (3)	0.0928 (3)	0.0637 (9)
C59	0.5630 (5)	0.3966 (4)	-0.0060 (4)	0.0822 (13)
H26A	0.5890	0.4585	-0.0203	0.099*
C60	0.5482 (5)	0.3352 (4)	-0.0801 (4)	0.0822 (13)
H43A	0.5657	0.3545	-0.1445	0.099*
C61	0.5063 (4)	0.2417 (4)	-0.0614 (3)	0.0675 (10)
C62	0.4835 (4)	0.2101 (3)	0.0338 (3)	0.0557 (8)
C63	0.4476 (4)	0.1062 (3)	0.0570 (3)	0.0576 (9)
H25A	0.3793	0.1178	0.1133	0.069*
C64	0.5875 (4)	0.0136 (3)	0.0847 (3)	0.0622 (9)
C65	0.5855 (6)	-0.0533 (3)	0.1655 (3)	0.0812 (12)
H55A	0.4999	-0.0408	0.2042	0.097*
C66	0.7132 (8)	-0.1398 (4)	0.1884 (4)	0.0960 (16)
H70A	0.7131	-0.1833	0.2439	0.115*
C67	0.8359 (7)	-0.1609 (4)	0.1313 (5)	0.1028 (17)
H74A	0.9186	-0.2201	0.1462	0.123*
C68	0.8397 (5)	-0.0961 (4)	0.0522 (5)	0.0941 (15)
H68A	0.9257	-0.1106	0.0136	0.113*
C69	0.7166 (5)	-0.0084 (3)	0.0284 (3)	0.0761 (11)
H63A	0.7208	0.0358	-0.0258	0.091*
C70	0.2047 (4)	0.1251 (3)	-0.0320 (3)	0.0685 (10)

H37A	0.1860	0.2023	-0.0423	0.082*
C71	0.1472 (6)	0.0848 (5)	-0.1227 (4)	0.1079 (17)
H77A	0.2022	0.0973	-0.1794	0.162*
H77B	0.0419	0.1221	-0.1323	0.162*
H77C	0.1611	0.0096	-0.1141	0.162*
C72	0.1207 (4)	0.1087 (3)	0.0598 (3)	0.0632 (9)
C73	0.1590 (5)	0.0127 (3)	0.1132 (4)	0.0746 (11)
H49A	0.2421	-0.0425	0.0939	0.090*
C74	0.0789 (5)	-0.0041 (3)	0.1937 (4)	0.0751 (11)
H48A	0.1100	-0.0693	0.2287	0.090*
C75	-0.0481 (4)	0.0753 (3)	0.2232 (3)	0.0671 (10)
C76	-0.0865 (4)	0.1706 (3)	0.1707 (3)	0.0699 (10)
H19A	-0.1707	0.2252	0.1896	0.084*
C77	-0.0053 (4)	0.1885 (3)	0.0910 (3)	0.0711 (11)
H42A	-0.0348	0.2547	0.0576	0.085*
C78	-0.1366 (5)	0.0563 (4)	0.3107 (4)	0.0927 (14)
H10A	-0.0909	-0.0146	0.3379	0.139*
H10B	-0.2383	0.0639	0.2905	0.139*
H10C	-0.1371	0.1076	0.3598	0.139*
N2	0.3758 (4)	0.6216 (3)	0.7309 (3)	0.0727 (8)
H1A	0.4194	0.5675	0.7739	0.087*
N3	0.3698 (4)	0.0750 (3)	-0.0274 (2)	0.0682 (8)
H3C	0.3882	0.0036	-0.0233	0.082*
N1	-0.0798 (3)	0.4155 (2)	0.3457 (2)	0.0610(7)
H4B	-0.0692	0.3642	0.3015	0.073*
01	-0.2945 (3)	0.5840 (2)	0.3874 (3)	0.0827 (8)
H1	-0.2573	0.5266	0.3613	0.124*
O3	0.4909 (4)	0.1879 (3)	-0.14207 (19)	0.0893 (9)
H2	0.4449	0.1453	-0.1274	0.134*
O2	0.4601 (4)	0.7828 (3)	0.6589 (3)	0.1158 (12)
H8	0.4740	0.7246	0.6876	0.174*

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.069 (2)	0.050 (2)	0.068 (2)	-0.0162 (18)	-0.0044 (19)	0.0020 (18)
C2	0.088 (3)	0.083 (3)	0.074 (3)	-0.032 (2)	-0.005(2)	0.002 (2)
C3	0.121 (4)	0.097 (4)	0.073 (3)	-0.022 (3)	-0.001 (3)	0.007 (3)
C4	0.107 (4)	0.069 (3)	0.085 (3)	-0.005 (3)	-0.031 (3)	-0.009 (3)
C5	0.087 (3)	0.063 (3)	0.109 (4)	-0.018 (2)	-0.020 (3)	-0.017 (3)
C6	0.079 (3)	0.065 (2)	0.087 (3)	-0.027 (2)	0.001 (2)	-0.009(2)
C7	0.063 (2)	0.050 (2)	0.082 (3)	-0.0216 (18)	0.007 (2)	-0.0088 (18)
C8	0.089 (3)	0.069 (2)	0.058 (2)	-0.034 (2)	0.007 (2)	-0.0069 (19)
C9	0.127 (4)	0.085 (3)	0.077 (3)	-0.058 (3)	-0.002 (3)	-0.004 (2)
C10	0.201 (7)	0.095 (4)	0.080 (4)	-0.084 (5)	0.002 (4)	0.003 (3)
C11	0.237 (10)	0.088 (5)	0.079 (4)	-0.068 (6)	-0.021 (5)	-0.004 (3)
C12	0.155 (5)	0.075 (3)	0.062 (3)	-0.007 (4)	-0.026 (3)	-0.014 (3)
C13	0.219 (9)	0.090 (4)	0.098 (5)	0.057 (6)	-0.038 (5)	-0.027 (4)

C14	0.156 (7)	0.199 (10)	0.098 (5)	0.087 (7)	-0.025 (5)	-0.034 (6)
C15	0.104 (4)	0.209 (9)	0.100 (5)	0.037 (5)	-0.001 (4)	-0.018 (5)
C16	0.083 (3)	0.121 (4)	0.089 (3)	0.009 (3)	0.000 (3)	-0.008(3)
C17	0.101 (3)	0.074 (3)	0.057 (2)	-0.007(3)	-0.009(2)	-0.013(2)
C18	0.067 (2)	0.062 (2)	0.072 (2)	-0.0216 (19)	0.0000 (19)	-0.0044 (19)
C19	0.089 (3)	0.094 (3)	0.106 (4)	-0.024 (3)	0.034 (3)	-0.027 (3)
C20	0.070 (2)	0.056 (2)	0.0469 (19)	-0.0196 (18)	-0.0037 (17)	-0.0068 (16)
C21	0.071 (3)	0.071 (2)	0.062 (2)	-0.016(2)	-0.0101 (19)	0.0071 (19)
C22	0.074(3)	0.124(4)	0.067(3)	-0.038(3)	-0.001(2)	-0.009(3)
C23	0.117 (4)	0.116 (4)	0.050(2)	-0.069(3)	0.006 (3)	-0.007(3)
C24	0.137 (5)	0.083 (3)	0.056(2)	-0.051(3)	-0.002(3)	0.009 (2)
C25	0.087(3)	0.064(2)	0.056(2)	-0.019(2)	-0.013(2)	0.0050(19)
C26	0.191(7)	0.221(8)	0.095(4)	-0.144(7)	0.015(4)	-0.004(4)
C27	0.058(2)	0.0221(0) 0.0477(18)	0.093(1)	-0.0085(15)	-0.0042(15)	-0.0026(14)
C28	0.050(2)	0.057(2)	0.066(2)	-0.0184(18)	0.0022(18)	-0.0063(17)
C29	0.001(2)	0.083(3)	0.000(2) 0.078(3)	-0.029(2)	-0.001(2)	-0.004(2)
C30	0.009(2)	0.003(3)	0.070(3)	-0.037(3)	-0.016(3)	-0.004(2)
C31	0.091(3)	0.077(3)	0.009(3)	-0.015(2)	-0.017(2)	-0.0100(19)
C32	0.070(2)	0.052(2)	0.009(2)	-0.0059(18)	-0.0112(17)	-0.0006(15)
C33	0.070(2)	0.051(2)	0.0497(19)	0.005(2)	-0.005(2)	-0.0066(18)
C34	0.070(3)	0.050(2) 0.068(3)	0.050(2) 0.071(2)	0.005(2)	0.005(2)	-0.005(2)
C35	0.002(2)	0.000(3)	0.071(2)	-0.0025(17)	-0.0003(17)	0.000(2)
C36	0.030(2) 0.0481(18)	0.030(2)	0.003(2)	-0.0023(17)	-0.0015(15)	-0.0005(15)
C37	0.0461(18)	0.0430(19)	0.0525(17)	-0.0031(13)	-0.0061(15)	-0.0003(15)
C38	0.0464(10) 0.0465(17)	0.0431(18) 0.0447(18)	0.004(2)	-0.0048(14)	-0.0026(15)	-0.0025(15)
C39	0.0405(17)	0.070(3)	0.000(2)	-0.0005(18)	0.0020(13)	-0.0033(13)
C40	0.035(2)	0.070(3)	0.075(3)	-0.009(2)	0.0010(1))	-0.015(2)
C40	0.070(3)	0.061(3)	0.000(3)	-0.027(2)	0.013(2)	-0.007(2)
C42	0.039(3)	0.009(3)	0.001(2)	-0.027(2)	-0.010(2)	-0.007(2)
C42	0.075(3)	0.077(3)	0.000(3)	-0.0070(18)	-0.0020(18)	-0.0086(10)
C43	0.034(2)	0.007(2)	0.007(2)	-0.0070(18)	0.0020(18)	-0.0036(17)
C44 C45	0.073(2)	0.048(2)	0.001(2)	-0.0082(18)	0.0039(19)	-0.0020(17)
C45	0.064(3)	0.081(3)	0.101(3)	-0.0144(16)	0.024(3)	-0.0020(15)
C40	0.008(2)	0.0431(19)	0.0433(18)	-0.0144(10)	0.0030(10)	-0.0029(13)
C47	0.088(3)	0.072(3)	0.001(2)	-0.010(2)	-0.010(2) -0.020(2)	-0.013(2)
C40	0.077(3)	0.103(4)	0.081(3)	-0.010(3)	-0.020(2)	0.003(3)
C49	0.000(3)	0.081(3)	0.070(3)	-0.006(2)	0.014(2)	0.017(2)
C50	0.080(3)	0.059(2)	0.067(2)	-0.005(2)	0.011(2)	-0.0061(19)
C51	0.074(2)	0.002(2)	0.038(2)	-0.014(2)	-0.0038(19)	-0.0087(18)
C52	0.072(3)	0.130(3)	0.144(5)	0.015(3)	0.022(3)	0.034(4)
C53	0.0380(10)	0.0464(18)	0.069(2)	-0.0063(14)	-0.0032(15)	-0.0003(16)
C54	0.055(2)	0.055(2)	0.063(2)	-0.0121(10)	0.0084 (16)	-0.0035(17)
C55	0.065(2)	0.059 (2)	0.0/1(2)	-0.0135(18)	0.0023(19)	-0.0160(19)
C56	0.063(2)	0.059 (3)	0.103 (4)	-0.0125 (19)	-0.009(2)	-0.019(2)
057	0.063 (2)	0.052 (2)	0.113 (4)	-0.0199 (19)	-0.014 (2)	0.001 (2)
058	0.0474 (19)	0.058 (2)	0.085 (3)	-0.0155 (17)	-0.0019 (18)	0.0094 (19)
C59	0.069 (3)	0.076 (3)	0.106 (4)	-0.032 (2)	-0.007(2)	0.028 (3)
C60	0.068 (3)	0.102 (3)	0.078 (3)	-0.030 (2)	0.001 (2)	0.028 (3)
C61	0.053 (2)	0.088 (3)	0.057 (2)	-0.0143 (19)	0.0048 (17)	0.005 (2)

supporting information

C62	0.0451 (18)	0.058 (2)	0.060 (2)	-0.0095 (15)	0.0018 (15)	0.0013 (17)
C63	0.065 (2)	0.059 (2)	0.055 (2)	-0.0262 (18)	0.0045 (17)	-0.0107 (17)
C64	0.079 (3)	0.052 (2)	0.058 (2)	-0.0221 (19)	-0.0068 (19)	-0.0111 (17)
C65	0.118 (4)	0.061 (3)	0.065 (3)	-0.027 (3)	-0.009 (2)	-0.009 (2)
C66	0.150 (5)	0.057 (3)	0.079 (3)	-0.029 (3)	-0.038 (3)	0.004 (2)
C67	0.109 (4)	0.068 (3)	0.122 (5)	-0.013 (3)	-0.046 (4)	-0.005 (3)
C68	0.072 (3)	0.080 (3)	0.126 (4)	-0.014 (3)	-0.010 (3)	-0.019 (3)
C69	0.070 (3)	0.068 (3)	0.085 (3)	-0.014 (2)	-0.006 (2)	-0.003 (2)
C70	0.073 (3)	0.062 (2)	0.070 (2)	-0.020 (2)	-0.013 (2)	-0.0050 (19)
C71	0.108 (4)	0.127 (4)	0.090 (3)	-0.036 (3)	-0.030 (3)	-0.017 (3)
C72	0.060 (2)	0.053 (2)	0.077 (2)	-0.0167 (18)	-0.0134 (19)	-0.0044 (18)
C73	0.062 (2)	0.052 (2)	0.103 (3)	-0.0084 (18)	-0.001 (2)	-0.002 (2)
C74	0.068 (2)	0.055 (2)	0.100 (3)	-0.016 (2)	-0.006 (2)	0.005 (2)
C75	0.061 (2)	0.060 (2)	0.082 (3)	-0.0188 (19)	-0.012 (2)	-0.006 (2)
C76	0.051 (2)	0.069 (3)	0.087 (3)	-0.0127 (19)	-0.004 (2)	-0.009 (2)
C77	0.062 (2)	0.050(2)	0.096 (3)	-0.0090 (18)	-0.018 (2)	-0.002 (2)
C78	0.081 (3)	0.091 (3)	0.108 (4)	-0.029 (3)	0.013 (3)	-0.007 (3)
N2	0.0649 (19)	0.071 (2)	0.083 (2)	-0.0212 (16)	-0.0017 (16)	-0.0115 (17)
N3	0.073 (2)	0.0654 (19)	0.0655 (19)	-0.0184 (16)	-0.0036 (15)	-0.0163 (15)
N1	0.0672 (18)	0.0509 (17)	0.0645 (18)	-0.0179 (14)	-0.0019 (15)	0.0030 (14)
01	0.0465 (14)	0.0728 (18)	0.123 (3)	-0.0096 (13)	0.0032 (15)	-0.0069 (17)
03	0.098 (2)	0.116 (3)	0.0550 (16)	-0.0339 (19)	0.0088 (15)	-0.0049 (17)
O2	0.105 (3)	0.140 (3)	0.128 (3)	-0.075 (3)	0.025 (2)	-0.007 (3)

Geometric parameters (Å, °)

C1—C6	1.369 (5)	C42—C43	1.382 (5)
C1—C2	1.374 (6)	C42—H46A	0.9300
C1—C7	1.525 (5)	C43—H57A	0.9300
С2—С3	1.389 (7)	C44—N1	1.484 (5)
С2—Н50А	0.9300	C44—C46	1.511 (5)
C3—C4	1.352 (7)	C44—C45	1.523 (6)
С3—Н75А	0.9300	C44—H32A	0.9800
C4—C5	1.370 (7)	C45—H56A	0.9600
С4—Н67А	0.9300	C45—H56B	0.9600
С5—С6	1.384 (6)	C45—H56C	0.9600
C5—H61A	0.9300	C46—C47	1.371 (5)
С6—Н53А	0.9300	C46—C51	1.387 (5)
C7—N2	1.492 (5)	C47—C48	1.380 (6)
С7—С8	1.520 (6)	C47—H64A	0.9300
С7—Н30А	0.9800	C48—C49	1.382 (7)
С8—С9	1.381 (6)	C48—H71A	0.9300
C8—C17	1.416 (6)	C49—C50	1.363 (6)
С9—О2	1.359 (6)	C49—C52	1.513 (6)
C9—C10	1.417 (8)	C50—C51	1.370 (5)
C10-C11	1.320 (9)	С50—Н59А	0.9300
С10—Н73А	0.9300	C51—H22A	0.9300
C11—C12	1.381 (9)	C52—H82A	0.9600

С11—Н76А	0.9300	С52—Н82В	0.9600
C12—C13	1.371 (10)	C52—H82C	0.9600
C12—C17	1.464 (7)	C53—C54	1.409 (5)
C13—C14	1.374 (13)	C53—C58	1.415 (5)
C13—H81A	0.9300	C53—C62	1 426 (5)
C14-C15	1 394 (13)	C54—C55	1.120(5) 1.369(5)
C14—H83A	0.9300	C54—H4A	0.9300
C15-C16	1 385 (8)	C55_C56	1 383 (6)
C15 H84A	0.0300	C55 H15A	0.0300
C16 C17	1 385 (7)	C56 C57	1 337 (6)
C16 H72A	0.0300	C56 H41A	0.0300
C18 N2	1 400 (5)	C57 C58	1.416 (6)
C_{10} C_{10} C_{20}	1.490(5)	C57_H21A	1.410(0)
$C_{10} = C_{20}$	1.515(5) 1.523(6)	C_{5}^{5} C_{5}^{50}	0.9300
C_{10} U_{100A}	1.555 (0)	$C_{50} = C_{59}$	1.408(0) 1.242(7)
C10_H99A	0.9800	C59—C60	1.342 (7)
C19—H/9A	0.9600	C59—H26A	0.9300
С19—Н/9В	0.9600		1.412 (6)
С19—Н79С	0.9600	C60—H43A	0.9300
C20—C21	1.372 (5)	C61—O3	1.353 (5)
C20—C25	1.387 (5)	C61—C62	1.380 (5)
C21—C22	1.382 (6)	C62—C63	1.523 (5)
C21—H54A	0.9300	C63—N3	1.488 (4)
C22—C23	1.350 (7)	C63—C64	1.525 (5)
C22—H51A	0.9300	C63—H25A	0.9800
C23—C24	1.374 (7)	C64—C69	1.383 (6)
C23—C26	1.519 (7)	C64—C65	1.387 (6)
C24—C25	1.376 (6)	C65—C66	1.399 (7)
C24—H52A	0.9300	C65—H55A	0.9300
С25—Н27А	0.9300	C66—C67	1.342 (8)
C26—H10D	0.9600	C66—H70A	0.9300
C26—H10E	0.9600	C67—C68	1.355 (8)
C26—H10F	0.9600	С67—Н74А	0.9300
C27—C28	1.405 (5)	C68—C69	1.385 (6)
C27—C32	1.419 (5)	C68—H68A	0.9300
C27—C36	1.428 (5)	С69—Н63А	0.9300
C28—C29	1.382 (5)	C70—N3	1.477 (5)
С28—НЗА	0.9300	C70—C72	1.514 (6)
C29—C30	1.395 (6)	C70—C71	1.516 (6)
С29—Н47А	0.9300	C70—H37A	0.9800
C30—C31	1.353 (6)	C71—H77A	0.9600
C30—H44A	0.9300	C71—H77B	0.9600
$C_{31} - C_{32}$	1 427 (6)	C71 - H77C	0.9600
C31—H20A	0.9300	C72-C73	1.382(5)
C_{32} — C_{33}	1 408 (6)	C72—C77	1 392 (5)
$C_{33} - C_{34}$	1 328 (6)	C73—C74	1 371 (6)
C33—H18A	0.9300	C73—H49A	0.9300
C_{34} C_{35}	1 418 (6)	C74 - C75	1 388 (5)
C34—H35A	0.9300	C74—H48A	0.9300
CO 1 1100/1	0.2000		0.7500

~~~			
C35—01	1.350 (5)	C/5-C/6	1.368 (6)
C35—C36	1.391 (5)	C75—C78	1.503 (6)
C36—C37	1.523 (5)	C76—C77	1.373 (6)
C37—N1	1.476 (4)	С76—Н19А	0.9300
C37—C38	1.520 (5)	С77—Н42А	0.9300
С37—Н6А	0.9800	C78—H10A	0.9600
C38—C39	1.373 (5)	C78—H10B	0.9600
C38—C43	1 379 (5)	C78—H10C	0 9600
$C_{39}$ $-C_{40}$	1 368 (6)	N2—H1A	0.9000
$C_{30}$ $H_{36A}$	0.0300	N3 H3C	0.9000
C40 C41	1 266 (6)	NJ U4D	0.9000
$C_{40}$	1.300 (0)		0.9001
C40—H80A	0.9300		0.8200
C41—C42	1.363 (6)	03—H2	0.8200
C41—H38A	0.9300	O2—H8	0.8200
C6—C1—C2	117.9 (4)	C38—C43—C42	120.5 (3)
C6—C1—C7	123.0 (4)	С38—С43—Н57А	119.8
$C_{2}$ — $C_{1}$ — $C_{7}$	119.1 (3)	C42—C43—H57A	119.8
C1 - C2 - C3	120.7(4)	N1-C44-C46	1144(3)
C1 - C2 - H50A	110.7	N1 - C44 - C45	107.8(3)
$C_1 = C_2 = H50\Lambda$	110.7	$C_{46} C_{44} C_{45}$	107.0(3)
$C_4 = C_2 = C_2$	120.5 (5)	N1  C44  H324	107.7
C4 = C3 = C2	120.3 (3)	NI = C44 = II32A	107.7
C4 - C3 - H75A	119.7	C40—C44—H32A	107.7
C2—C3—H/5A	119.7	C45—C44—H32A	107.7
C3—C4—C5	119.8 (5)	С44—С45—Н56А	109.5
С3—С4—Н67А	120.1	C44—C45—H56B	109.5
С5—С4—Н67А	120.1	H56A—C45—H56B	109.5
C4—C5—C6	119.5 (4)	C44—C45—H56C	109.5
C4—C5—H61A	120.3	H56A—C45—H56C	109.5
C6—C5—H61A	120.3	H56B—C45—H56C	109.5
C1—C6—C5	121.7 (4)	C47—C46—C51	116.8 (3)
C1—C6—H53A	119.2	C47—C46—C44	121.2 (3)
С5—С6—Н53А	119.2	C51—C46—C44	122.0 (3)
N2-C7-C8	110.6 (3)	C46—C47—C48	122.1 (4)
$N_2 - C_7 - C_1$	109.7 (3)	C46—C47—H64A	118.9
C8-C7-C1	109.1(3)	C48 - C47 - H64A	118.9
$N_2 C_7 H_{30A}$	107.8	C47 $C48$ $C40$	120.7(4)
$N_2 = C_7 = H_2 O_A$	107.0	C47 = C48 = U71 A	120.7 (4)
$C_{0}$ $C_{1}$ $C_{2}$ $H_{20A}$	107.8	C47 - C48 - H71A	119.7
C1 - C / - H30A	107.8	C49—C48—H/IA	119.7
C9—C8—C17	118.6 (4)	C50—C49—C48	117.0 (4)
C9—C8—C7	120.5 (4)	C50—C49—C52	122.4 (5)
C17—C8—C7	120.8 (4)	C48—C49—C52	120.5 (5)
O2—C9—C8	123.6 (4)	C49—C50—C51	122.7 (4)
O2—C9—C10	115.4 (6)	С49—С50—Н59А	118.7
C8—C9—C10	121.0 (6)	С51—С50—Н59А	118.7
C11—C10—C9	120.2 (7)	C50—C51—C46	120.7 (4)
С11—С10—Н73А	119.9	C50—C51—H22A	119.7
С9—С10—Н73А	119.9	C46—C51—H22A	119.7

C10—C11—C12	123.3 (6)	C49—C52—H82A	109.5
С10—С11—Н76А	118.3	С49—С52—Н82В	109.5
С12—С11—Н76А	118.3	H82A—C52—H82B	109.5
C13—C12—C11	123.2 (8)	С49—С52—Н82С	109.5
C13—C12—C17	119.2 (7)	H82A—C52—H82C	109.5
C11—C12—C17	117.5 (6)	H82B—C52—H82C	109.5
C12—C13—C14	122.1 (8)	C54—C53—C58	116.5 (3)
С12—С13—Н81А	118.9	C54—C53—C62	122.9 (3)
C14—C13—H81A	118.9	C58—C53—C62	120.6 (3)
C13—C14—C15	119.3 (7)	C55—C54—C53	122.1 (4)
C13—C14—H83A	120.3	С55—С54—Н4А	119.0
C15—C14—H83A	120.3	С53—С54—Н4А	119.0
C16—C15—C14	120.3 (8)	C54—C55—C56	120.3 (4)
C16—C15—H84A	119.8	С54—С55—Н15А	119.8
C14—C15—H84A	119.8	С56—С55—Н15А	119.8
C15—C16—C17	121.8 (7)	C57—C56—C55	119.9 (4)
C15—C16—H72A	119.1	С57—С56—Н41А	120.0
C17—C16—H72A	119.1	C55—C56—H41A	120.0
C16—C17—C8	123.6 (4)	C56—C57—C58	121.7 (4)
C16—C17—C12	117.2 (5)	С56—С57—Н21А	119.2
C8—C17—C12	119.2 (5)	C58—C57—H21A	119.2
N2—C18—C20	113.1 (3)	C59—C58—C53	117.9 (4)
N2—C18—C19	107.9 (3)	C59—C58—C57	122.6 (4)
C20—C18—C19	112.7 (3)	C53—C58—C57	119.5 (4)
N2—C18—H99A	107.6	C60—C59—C58	121.7 (4)
С20—С18—Н99А	107.6	С60—С59—Н26А	119.1
С19—С18—Н99А	107.6	С58—С59—Н26А	119.1
С18—С19—Н79А	109.5	C59—C60—C61	120.7 (4)
C18—C19—H79B	109.5	С59—С60—Н43А	119.7
H79A—C19—H79B	109.5	C61—C60—H43A	119.7
С18—С19—Н79С	109.5	O3—C61—C62	124.0 (4)
Н79А—С19—Н79С	109.5	O3—C61—C60	115.4 (4)
H79B—C19—H79C	109.5	C62—C61—C60	120.6 (4)
C21—C20—C25	116.9 (4)	C61—C62—C53	118.4 (3)
C21—C20—C18	120.8 (3)	C61—C62—C63	121.9 (3)
C25—C20—C18	122.2 (3)	C53—C62—C63	119.6 (3)
C20—C21—C22	120.6 (4)	N3—C63—C62	111.4 (3)
C20—C21—H54A	119.7	N3—C63—C64	108.8 (3)
С22—С21—Н54А	119.7	C62—C63—C64	112.7 (3)
C23—C22—C21	122.4 (5)	N3—C63—H25A	107.9
C23—C22—H51A	118.8	C62—C63—H25A	107.9
С21—С22—Н51А	118.8	C64—C63—H25A	107.9
C22—C23—C24	117.7 (4)	C69—C64—C65	118.4 (4)
C22—C23—C26	122.8 (6)	C69—C64—C63	121.4 (3)
C24—C23—C26	119.5 (5)	C65—C64—C63	120.2 (4)
C23—C24—C25	120.7 (4)	C64—C65—C66	119.6 (5)
C23—C24—H52A	119.6	C64—C65—H55A	120.2
C25—C24—H52A	119.6	С66—С65—Н55А	120.2

C24—C25—C20	121.6 (4)	C67—C66—C65	120.9 (5)
C24—C25—H27A	119.2	C67—C66—H70A	119.6
C20—C25—H27A	119.2	C65—C66—H70A	119.6
C23—C26—H10D	109.5	C66—C67—C68	120.2 (5)
С23—С26—Н10Е	109.5	C66—C67—H74A	119.9
H10D—C26—H10E	109.5	C68—C67—H74A	119.9
C23—C26—H10F	109.5	C67—C68—C69	120.6 (5)
H10D—C26—H10F	109.5	C67—C68—H68A	119.7
H10E—C26—H10F	109.5	C69—C68—H68A	119.7
C28—C27—C32	117.1 (3)	C64—C69—C68	120.3 (4)
C28—C27—C36	123.0 (3)	C64—C69—H63A	119.8
C32—C27—C36	119.9 (3)	C68—C69—H63A	119.8
C29—C28—C27	122.5 (4)	N3—C70—C72	115.3 (3)
С29—С28—НЗА	118.8	N3—C70—C71	107.3 (4)
С27—С28—НЗА	118.8	C72—C70—C71	111.9 (4)
C28—C29—C30	119.6 (4)	N3—C70—H37A	107.4
С28—С29—Н47А	120.2	С72—С70—Н37А	107.4
С30—С29—Н47А	120.2	С71—С70—Н37А	107.4
C31—C30—C29	120.1 (4)	С70—С71—Н77А	109.5
C31—C30—H44A	119.9	С70—С71—Н77В	109.5
С29—С30—Н44А	119.9	H77A—C71—H77B	109.5
C30—C31—C32	121.3 (4)	С70—С71—Н77С	109.5
C30—C31—H20A	119.3	H77A—C71—H77C	109.5
C32—C31—H20A	119.3	H77B—C71—H77C	109.5
C33—C32—C27	119.1 (4)	C73—C72—C77	116.7 (4)
C33—C32—C31	121.6 (4)	C73—C72—C70	122.2 (4)
C27—C32—C31	119.3 (3)	C77—C72—C70	121.0 (3)
C34—C33—C32	121.1 (4)	C74—C73—C72	122.4 (4)
C34—C33—H18A	119.5	С74—С73—Н49А	118.8
C32—C33—H18A	119.5	С72—С73—Н49А	118.8
C33—C34—C35	121.4 (4)	C73—C74—C75	120.5 (4)
С33—С34—Н35А	119.3	C73—C74—H48A	119.8
С35—С34—Н35А	119.3	C75—C74—H48A	119.8
O1—C35—C36	122.7 (3)	C76—C75—C74	117.4 (4)
O1—C35—C34	116.9 (3)	C76—C75—C78	122.2 (4)
C36—C35—C34	120.4 (4)	C74—C75—C78	120.4 (4)
C35—C36—C27	118.2 (3)	C75—C76—C77	122.4 (4)
C35—C36—C37	121.6 (3)	C75—C76—H19A	118.8
C27—C36—C37	120.3 (3)	С77—С76—Н19А	118.8
N1—C37—C38	109.6 (3)	C76—C77—C72	120.7 (4)
N1—C37—C36	110.7 (3)	C76—C77—H42A	119.7
C38—C37—C36	113.8 (3)	C72—C77—H42A	119.7
N1—C37—H6A	107.5	C75—C78—H10A	109.5
С38—С37—Н6А	107.5	C75—C78—H10B	109.5
С36—С37—Н6А	107.5	H10A—C78—H10B	109.5
C39—C38—C43	118.0 (3)	C75—C78—H10C	109.5
C39—C38—C37	121.1 (3)	H10A—C78—H10C	109.5
C43—C38—C37	120.9 (3)	H10B—C78—H10C	109.5

C40—C39—C38	121.6 (4)	C18—N2—C7	112.9 (3)
С40—С39—Н36А	119.2	C18—N2—H1A	108.6
С38—С39—Н36А	119.2	C7—N2—H1A	108.5
C41—C40—C39	120.0 (4)	C70—N3—C63	115.2 (3)
C41—C40—H80A	120.0	C70 - N3 - H3C	107.9
C39-C40-H80A	120.0	C63 - N3 - H3C	108.0
C42-C41-C40	1197(4)	C37 - N1 - C44	114.0(3)
C42-C41-H38A	120.2	C37—N1—H4B	108.4
C40-C41-H38A	120.2	C44—N1—H4B	108.4
$C_{41} - C_{42} - C_{43}$	120.2 120.3 (4)	$C_{35}$ $-01$ $-H_1$	109.5
C41 - C42 - H46A	119.8	C61 - O3 - H2	109.5
C43 - C42 - H46A	119.8	$C_{9} = 0^{2} = H_{8}$	109.5
C+5 C+2 11+0/A	117.0	02-110	109.5
C6—C1—C2—C3	0.1 (6)	C37—C38—C39—C40	178.3 (4)
C7—C1—C2—C3	179.3 (4)	C38—C39—C40—C41	-1.0(7)
C1—C2—C3—C4	-1.1 (7)	C39—C40—C41—C42	-0.1 (6)
$C_2 - C_3 - C_4 - C_5$	2.3(7)	C40-C41-C42-C43	0.8 (6)
$C_{3}-C_{4}-C_{5}-C_{6}$	-2.4(7)	$C_{39}$ $C_{38}$ $C_{43}$ $C_{42}$	-0.6(5)
$C_{2}-C_{1}-C_{6}-C_{5}$	-0.2(6)	$C_{37}$ $C_{38}$ $C_{43}$ $C_{42}$	-1775(3)
$C_{7}$ $C_{1}$ $C_{6}$ $C_{5}$	-1794(4)	$C_{41}$ $C_{42}$ $C_{43}$ $C_{38}$	-0.4(6)
C4-C5-C6-C1	14(6)	N1-C44-C46-C47	1295(4)
C6-C1-C7-N2	67 8 (4)	$C_{45}$ $C_{44}$ $C_{46}$ $C_{47}$	-107.9(4)
$C_{2}$ $C_{1}$ $C_{7}$ $N_{2}$	-1114(4)	N1 - C44 - C46 - C51	-534(5)
C6-C1-C7-C8	-56.2(5)	$C_{45}$ $C_{44}$ $C_{46}$ $C_{51}$	69 2 (5)
$C_{2} - C_{1} - C_{7} - C_{8}$	1246(4)	$C_{1}^{-1}$ $C_{4}^{-1}$ $C_{$	-0.6(6)
$N_{2} - C_{1} - C_{8} - C_{9}$	-381(5)	$C_{44} - C_{46} - C_{47} - C_{48}$	176.7(4)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	85 <i>A</i> (5)	$C_{46} = C_{47} = C_{48} = C_{48}$	1/0.7(4)
C1 - C7 - C8 - C9	145.2(4)	$C_{40} = C_{47} = C_{48} = C_{49} = C_{49}$	-1.2(6)
12 - 07 - 08 - 017	-01.3(4)	C47 = C48 = C49 = C50	-180.0(5)
$C_1 - C_2 - C_3 - C_1 / C_1 - C_2 - C_2 - C_1 / C_2 - C_2 $	-176 A (A)	$C_{47} = C_{48} = C_{49} = C_{52}$	20(6)
$C_{1}^{-} = C_{2}^{-} = C_{2$	6 8 (6)	$C_{40} = C_{40} = C_{50} = C_{51}$	2.0(0)
$C_{1} = C_{2} = C_{2} = C_{2}$	0.0(0)	$C_{32} = C_{49} = C_{30} = C_{31}$	-21(6)
C7 C8 C9 C10	-173 A (A)	C47 = C30 = C31 = C40	2.1(0)
$C_{}C_{0}$	173.4(4)	C44 = C46 = C51 = C50	-176.0(2)
$C_{2} = C_{2} = C_{10} = C_{11}$	-21(8)	$C_{44} = C_{40} = C_{51} = C_{50}$	-170.0(3)
$C_{0} = C_{10} = C_{11} = C_{12}$	-2.1(8)	$C_{38} = C_{33} = C_{34} = C_{35}$	0.9(3)
$C_{9}$ $C_{10}$ $C_{11}$ $C_{12}$ $C_{13}$	-0.0(10) -178.7(6)	$C_{02}$ $C_{53}$ $C_{54}$ $C_{55}$ $C_{56}$	-178.3(3)
C10 - C11 - C12 - C13	-1/8.7(0)	$C_{55} - C_{54} - C_{55} - C_{50}$	-0.7(3)
C10-C11-C12-C17	1.9 (8)	$C_{54} = C_{53} = C_{50} = C_{57}$	-0.2(0)
C17 - C12 - C13 - C14	-1/9.3(7)	$C_{55} = C_{50} = C_{57} = C_{58}$	0.9(0)
C12 - C12 - C13 - C14	-0.2(10)	$C_{34} = C_{33} = C_{38} = C_{59}$	1/8.1(3)
C12 - C13 - C14 - C15	-0.4(12)	$C_{02} = C_{03} = C_{03} = C_{03}$	-2.5(5)
C13 - C14 - C15 - C16	0.8(12)	$C_{24} = C_{23} = C_{28} = C_{27}$	-0.2(3)
$C_{14} - C_{15} - C_{16} - C_{17} - C_{28}$	-0.0(10)	02 - 03 - 03 - 03 - 03 - 03 - 03 - 03 -	179.2 (3)
C15 - C10 - C17 - C12	-1/9.7(5)	$C_{50} - C_{57} - C_{58} - C_{59}$	-1/8.9(4)
C15 - C16 - C17 - C12	0.0 (8)	$C_{50} - C_{5} / - C_{58} - C_{53}$	-0.7(6)
C9—C8—C17—C16	1//./(4)	C53-C58-C59-C60	0.9 (6)
C/-C8-C1/-C16	-5.6 (6)	05/	1/9.2 (4)
C9—C8—C17—C12	-2.0 (6)	C58—C59—C60—C61	1.1 (6)

C7—C8—C17—C12	174.7 (4)	C59—C60—C61—O3	178.3 (4)
C13—C12—C17—C16	0.4 (7)	C59—C60—C61—C62	-1.7 (6)
C11—C12—C17—C16	179.8 (5)	O3—C61—C62—C53	-179.8 (3)
C13—C12—C17—C8	-179.9 (5)	C60—C61—C62—C53	0.1 (5)
C11—C12—C17—C8	-0.5 (6)	O3—C61—C62—C63	3.8 (5)
N2-C18-C20-C21	110.8 (4)	C60—C61—C62—C63	-176.2 (3)
C19—C18—C20—C21	-126.5 (4)	C54—C53—C62—C61	-178.7 (3)
N2-C18-C20-C25	-67.5 (4)	C58—C53—C62—C61	2.0 (5)
C19—C18—C20—C25	55.2 (5)	C54—C53—C62—C63	-2.3 (5)
C25—C20—C21—C22	-0.6 (6)	C58—C53—C62—C63	178.4 (3)
C18—C20—C21—C22	-179.1 (4)	C61—C62—C63—N3	-27.7 (4)
C20—C21—C22—C23	0.8 (7)	C53—C62—C63—N3	156.1 (3)
C21—C22—C23—C24	-0.3 (7)	C61—C62—C63—C64	95.0 (4)
C21—C22—C23—C26	179.0 (5)	C53—C62—C63—C64	-81.3 (4)
C22—C23—C24—C25	-0.3 (7)	N3—C63—C64—C69	75.9 (4)
C26—C23—C24—C25	-179.6 (4)	C62—C63—C64—C69	-48.2 (4)
C23—C24—C25—C20	0.4 (6)	N3—C63—C64—C65	-101.5 (4)
C21—C20—C25—C24	0.1 (6)	C62—C63—C64—C65	134.4 (4)
C18—C20—C25—C24	178.5 (4)	C69—C64—C65—C66	0.7 (5)
C32—C27—C28—C29	2.1 (5)	C63—C64—C65—C66	178.1 (3)
C36—C27—C28—C29	-178.3 (3)	C64—C65—C66—C67	-2.2 (6)
C27—C28—C29—C30	-0.2 (6)	C65—C66—C67—C68	2.3 (7)
C28—C29—C30—C31	-1.1 (6)	C66—C67—C68—C69	-1.0 (8)
C29—C30—C31—C32	0.4 (6)	C65—C64—C69—C68	0.6 (6)
C28—C27—C32—C33	179.9 (3)	C63—C64—C69—C68	-176.8 (4)
C36—C27—C32—C33	0.4 (5)	C67—C68—C69—C64	-0.5 (7)
C28—C27—C32—C31	-2.8 (5)	N3—C70—C72—C73	-39.3 (5)
C36—C27—C32—C31	177.7 (3)	C71—C70—C72—C73	83.6 (5)
C30—C31—C32—C33	178.8 (4)	N3—C70—C72—C77	144.6 (3)
C30—C31—C32—C27	1.6 (6)	C71—C70—C72—C77	-92.5 (5)
C27—C32—C33—C34	-2.1 (5)	С77—С72—С73—С74	-0.6 (6)
C31—C32—C33—C34	-179.3 (4)	С70—С72—С73—С74	-176.9 (4)
C32—C33—C34—C35	1.6 (6)	C72—C73—C74—C75	1.7 (6)
C33—C34—C35—O1	-178.6 (4)	C73—C74—C75—C76	-1.7 (6)
C33—C34—C35—C36	0.6 (6)	C73—C74—C75—C78	179.4 (4)
O1—C35—C36—C27	177.0 (3)	C74—C75—C76—C77	0.6 (6)
C34—C35—C36—C27	-2.3 (5)	C78—C75—C76—C77	179.5 (4)
O1—C35—C36—C37	-3.0 (5)	C75—C76—C77—C72	0.5 (6)
C34—C35—C36—C37	177.8 (3)	C73—C72—C77—C76	-0.5 (5)
C28—C27—C36—C35	-177.8 (3)	С70—С72—С77—С76	175.8 (3)
C32—C27—C36—C35	1.8 (5)	C20-C18-N2-C7	-59.9 (4)
C28—C27—C36—C37	2.2 (5)	C19—C18—N2—C7	174.8 (3)
C32—C27—C36—C37	-178.3 (3)	C8—C7—N2—C18	-74.3 (4)
C35—C36—C37—N1	-26.7 (4)	C1—C7—N2—C18	160.3 (3)
C27—C36—C37—N1	153.4 (3)	C72—C70—N3—C63	-55.4 (4)
C35—C36—C37—C38	97.3 (4)	C71—C70—N3—C63	179.3 (4)
C27—C36—C37—C38	-82.6 (4)	C62—C63—N3—C70	-84.7 (4)
N1—C37—C38—C39	-104.1 (4)	C64—C63—N3—C70	150.5 (3)

# supporting information

C36—C37—C38—C39	131.2 (4)	C38—C37—N1—C44	156.8 (3)
N1—C37—C38—C43	72.7 (4)	C36—C37—N1—C44	-76.8 (3)
C36—C37—C38—C43	-51.9 (4)	C46—C44—N1—C37	-68.9 (4)
C43—C38—C39—C40	1.3 (6)	C45—C44—N1—C37	166.5 (3)

Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H···A	D···· $A$	D—H···A
01—H1…N1	0.82	1.86	2.571 (4)	144
O2—H8…N2	0.82	1.92	2.629 (5)	144
O3—H2…N3	0.82	1.87	2.597 (5)	147
C6—H53 <i>A</i> …O2	0.93	2.58	3.327 (6)	138