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## Structure Reports

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## 2-Methylbenzaldehyde 2-methylbenzylidenehydrazone

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Received 23 June 2008; accepted 25 June 2008
Key indicators: single-crystal X-ray study; $T=295 \mathrm{~K}$; mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$; $R$ factor $=0.039 ; w R$ factor $=0.116 ;$ data-to-parameter ratio $=17.9$.

The molecule of the title compound, $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2}$, is centrosymmetric and the dihedral angle between the benzene ring and the dimethylhydrazine mean plane is $16.11(15)^{\circ}$.

## Related literature

For background, see: Shan et al. (2003). For related structures, see: Fan et al. (2008); Shan et al. $(2004,2008)$.


## Experimental

Crystal data
$\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2} \quad M_{r}=236.31$

Monoclinic, $P 2_{1} / c$
$a=6.1578$ (11) A
$b=13.248$ (2) $\AA$
$c=8.8161(16) \AA$
$\beta=105.398(12)^{\circ}$
$V=693.4(2) \AA^{3}$

## $Z=2$

Mo $K \alpha$ radiation
$\mu=0.07 \mathrm{~mm}^{-1}$
$T=295$ (2) K
$0.32 \times 0.28 \times 0.12 \mathrm{~mm}$

## Data collection

Rigaku R-AXIS RAPID IP diffractometer
Absorption correction: none 5451 measured reflections

Refinement
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.038$
$w R\left(F^{2}\right)=0.115$
$S=1.10$
1503 reflections

1503 independent reflections
1168 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.020$

## 84 parameters

H -atom parameters constrained
$\Delta \rho_{\text {max }}=0.12 \mathrm{e}^{\AA^{-3}}$
$\Delta \rho_{\min }=-0.10 \mathrm{e}^{-3}$

Data collection: PROCESS-AUTO (Rigaku, 1998); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/ MSC, 2002); program(s) used to solve structure: SIR92 (Altomare et al., 1993); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2750).

## References

Altomare, A., Cascarano, G., Giacovazzo, C. \& Guagliardi, A. (1993). J. Appl. Cryst. 26, 343-350.
Fan, Z., Shan, S., Wang, S.-H. \& Wang, W.-L. (2008). Acta Cryst. E64, o1341.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
Rigaku (1998). PROCESS-AUTO. Rigaku Corporation, Tokyo, Japan.
Rigaku/MSC (2002). CrystalStructure. Rigaku/MSC, The Woodlands, Texas, USA.
Shan, S., Fan, Z., Hu, W.-X. \& Xu, D.-J. (2004). Acta Cryst. E60, o2473-o2475.
Shan, S., Tian, Y.-L., Wang, S.-H., Wang, W.-L. \& Xu, Y.-L. (2008). Acta Cryst. E64, o1153.
Shan, S., Xu, D.-J., Hung, C.-H., Wu, J.-Y. \& Chiang, M. Y. (2003). Acta Cryst. C59, o135-o136.
Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.

## supporting information

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## S1. Comment

As part of our ongong studies of hydrazone derivatives (Shan et al., 2003), the title compound, (I), has been prepared and its crystal structure is reported here (Fig. 1).
The molecule of (I) is centrosymmetric, with the mid-point of the $\mathrm{N}-\mathrm{N}$ bond located on an inversion center. The $\mathrm{N}=\mathrm{C} 7$ double bond distance of 1.2727 (14) $\AA$ is shorter than the $\mathrm{C}=\mathrm{N}$ bond distances found in related hydrazone structures, i.e. 1.295 (2) $\AA$ in (E)-3-methoxyacetophenone 4-nitrophenylhydrazone (Fan et al., 2008), 1.2977 (18) $\AA$ in (E)-2-furyl methyl ketone 2,4-dinitrophenylhydrazone (Shan et al., 2008) and 1.293 (2) $\AA$ in benzylideneacetone 2,4-dinitrophenylhydrazone (Shan et al. 2004). In (I), the terminal benzene ring is twisted with respect to the central dimethylhydrazine plane by $16.11(15)^{\circ}$. The crystal packing is controlled by van der Waals forces.

## S2. Experimental

Hydrazine hydrate $(0.10 \mathrm{~g}, 2 \mathrm{mmol})$ was dissolved in ethanol $(10 \mathrm{ml})$, then acetic acid $(0.1 \mathrm{ml})$ was added slowly to the ethanol solution with stirring. The solution was heated at 333 K for several minutes until the solution cleared. 2-Methylbenzaldehyde ( $0.24 \mathrm{~g}, 2 \mathrm{mmol}$ ) was then dropped slowly into the solution, and the mixture was kept at 333 K with continuous stirring for 2 h . After the solution had cooled to room temperature yellow powder appeared. The crude title compound was separated and washed with water three times. Recrystallization from an absolute ethanol yielded yellow plates of (I).

## S3. Refinement

Methyl H atoms were placed in calculated positions with $\mathrm{C}-\mathrm{H}=0.96 \AA$ and the torsion angle was refined to fit the electron density with $U_{\text {iso }}(\mathrm{H})=1.5 U_{\mathrm{eq}}(\mathrm{C})$. The other H atoms were placed in calculated positions with $\mathrm{C}-\mathrm{H}=0.93$ and refined as riding with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$.


## Figure 1

The molecular structure of (I) with $50 \%$ probability displacement ellipsoids for non-H atoms. Symmetry code: (i) $-\mathrm{x}, 1-\mathrm{y}$, -z.

## 2-Methylbenzaldehyde 2-methylbenzylidenehydrazone

## Crystal data

$\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2}$
$M_{r}=236.31$
Monoclinic, $P 2_{1} / c$
Hall symbol: -P 2ybc
$a=6.1578$ (11) $\AA$
$b=13.248$ (2) $\AA$
$c=8.8161$ (16) $\AA$
$\beta=105.398(12)^{\circ}$
$V=693.4$ (2) $\AA^{3}$
$Z=2$

## Data collection

Rigaku R-AXIS RAPID IP
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
Detector resolution: 10.00 pixels $\mathrm{mm}^{-1}$

## $\omega$ scans

5451 measured reflections

## Refinement

Refinement on $F^{2}$
Least-squares matrix: full
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.039$
$w R\left(F^{2}\right)=0.115$
$S=1.10$
$F(000)=252$
$D_{\mathrm{x}}=1.132 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation, $\lambda=0.71073 \AA$
Cell parameters from 2246 reflections
$\theta=3.0-25.5^{\circ}$
$\mu=0.07 \mathrm{~mm}^{-1}$
$T=295 \mathrm{~K}$
Plate, yellow
$0.32 \times 0.28 \times 0.12 \mathrm{~mm}$

1503 independent reflections
1168 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.020$
$\theta_{\text {max }}=27.0^{\circ}, \theta_{\text {min }}=2.9^{\circ}$
$h=-7 \rightarrow 7$
$k=-15 \rightarrow 16$
$l=-11 \rightarrow 11$

1503 reflections
84 parameters
0 restraints
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{0}^{2}\right)+(0.0537 P)^{2}+0.0481 P\right]$
where $P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3$
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\text {max }}=0.12 \mathrm{e} \AA^{-3}$
$\Delta \rho_{\text {min }}=-0.11 \mathrm{e} \AA^{-3}$
Extinction correction: SHELXL97 (Sheldrick, 2008), $\mathrm{Fc}^{*}=\mathrm{kFc}\left[1+0.001 \mathrm{xFc}^{2} \lambda^{3} / \sin (2 \theta)\right]^{-1 / 4}$

Extinction coefficient: 0.099 (12)

## Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving 1.s. planes.
Refinement. Refinement of $F^{2}$ against ALL reflections. The weighted $R$-factor $w R$ and goodness of fit $S$ are based on $F^{2}$, conventional $R$-factors $R$ are based on $F$, with $F$ set to zero for negative $F^{2}$. The threshold expression of $F^{2}>\sigma\left(F^{2}\right)$ is used only for calculating $R$-factors(gt) etc. and is not relevant to the choice of reflections for refinement. $R$-factors based on $F^{2}$ are statistically about twice as large as those based on $F$, and $R$ - factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\dot{A}^{2}$ )

|  | $x$ | $y$ | $z$ | $U_{\text {iso }} *^{\prime} / U_{\mathrm{eq}}$ |
| :--- | :--- | :--- | :--- | :--- |
| N | $0.03959(16)$ | $0.45944(7)$ | $0.05193(11)$ | $0.0545(3)$ |
| C 1 | $0.33920(18)$ | $0.40583(8)$ | $0.26999(12)$ | $0.0495(3)$ |
| C 2 | $0.56653(19)$ | $0.41758(9)$ | $0.35123(13)$ | $0.0553(3)$ |
| C 3 | $0.6648(2)$ | $0.34407(11)$ | $0.46155(15)$ | $0.0731(4)$ |
| H 3 | 0.8162 | 0.3499 | 0.5151 | $0.088^{*}$ |
| C 4 | $0.5435(3)$ | $0.26319(11)$ | $0.49318(18)$ | $0.0813(5)$ |
| H 4 | 0.6129 | 0.2158 | 0.5682 | $0.098^{*}$ |
| C5 | $0.3198(3)$ | $0.25229(10)$ | $0.41412(16)$ | $0.0754(4)$ |
| H5 | 0.2377 | 0.1977 | 0.4356 | $0.090^{*}$ |
| C6 | $0.2186(2)$ | $0.32277(9)$ | $0.30309(14)$ | $0.0617(4)$ |
| H6 | 0.0678 | 0.3151 | 0.2492 | $0.074^{*}$ |
| C7 | $0.22666(19)$ | $0.47938(8)$ | $0.15044(13)$ | $0.0514(3)$ |
| H7 | 0.2935 | 0.5419 | 0.1466 | $0.062^{*}$ |
| C8 | $0.7057(2)$ | $0.50502(11)$ | $0.32086(16)$ | $0.0720(4)$ |
| H8A | 0.6477 | 0.5668 | 0.3517 | $0.108^{*}$ |
| H8B | 0.6986 | 0.5078 | 0.2108 | $0.108^{*}$ |
| H8C | 0.8593 | 0.4963 | 0.3807 | $0.108^{*}$ |

## Atomic displacement parameters $\left(\AA^{2}\right)$

|  | $U^{11}$ | $U^{22}$ | $U^{33}$ | $U^{12}$ | $U^{13}$ | $U^{23}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| N | $0.0577(6)$ | $0.0502(6)$ | $0.0522(6)$ | $0.0060(4)$ | $0.0086(4)$ | $0.0054(4)$ |
| C 1 | $0.0564(6)$ | $0.0494(6)$ | $0.0427(6)$ | $0.0049(5)$ | $0.0129(5)$ | $-0.0008(5)$ |
| C 2 | $0.0563(7)$ | $0.0624(7)$ | $0.0471(6)$ | $0.0058(5)$ | $0.0137(5)$ | $-0.0003(5)$ |
| C 3 | $0.0655(8)$ | $0.0849(10)$ | $0.0623(8)$ | $0.0139(7)$ | $0.0055(6)$ | $0.0119(7)$ |
| C 4 | $0.0945(11)$ | $0.0741(9)$ | $0.0686(9)$ | $0.0164(8)$ | $0.0097(8)$ | $0.0235(7)$ |
| C 5 | $0.0976(11)$ | $0.0571(8)$ | $0.0700(9)$ | $-0.0035(7)$ | $0.0198(8)$ | $0.0121(6)$ |
| C 6 | $0.0682(8)$ | $0.0567(7)$ | $0.0570(7)$ | $-0.0038(6)$ | $0.0112(6)$ | $0.0030(5)$ |
| C 7 | $0.0549(6)$ | $0.0492(6)$ | $0.0495(6)$ | $0.0012(5)$ | $0.0126(5)$ | $0.0009(5)$ |


| C 8 | $0.0568(7)$ | $0.0860(10)$ | $0.0711(8)$ | $-0.0059(6)$ | $0.0135(6)$ | $0.0053(7)$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |

Geometric parameters ( $\AA,{ }^{\circ}$ )

| $\mathrm{N}-\mathrm{C} 7$ | $1.2727(14)$ | $\mathrm{C} 4-\mathrm{C} 5$ | $1.376(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{N}-\mathrm{N}^{\mathrm{i}}$ | $1.4121(17)$ | $\mathrm{C} 4-\mathrm{H} 4$ | 0.9300 |
| $\mathrm{C} 1-\mathrm{C} 6$ | $1.4007(16)$ | $\mathrm{C} 5-\mathrm{C} 6$ | $1.3764(17)$ |
| $\mathrm{C} 1-\mathrm{C} 2$ | $1.4016(16)$ | $\mathrm{C} 5-\mathrm{H} 5$ | 0.9300 |
| $\mathrm{C} 1-\mathrm{C} 7$ | $1.4672(15)$ | $\mathrm{C} 6-\mathrm{H} 6$ | 0.9300 |
| $\mathrm{C} 2-\mathrm{C} 3$ | $1.3956(17)$ | $\mathrm{C} 7-\mathrm{H} 7$ | 0.9300 |
| $\mathrm{C} 2-\mathrm{C} 8$ | $1.5065(17)$ | $\mathrm{C} 8-\mathrm{H} 8 \mathrm{~A}$ | 0.9600 |
| $\mathrm{C} 3-\mathrm{C} 4$ | $1.3761(19)$ | $\mathrm{C} 8-\mathrm{H} 8 \mathrm{~B}$ | 0.9600 |
| $\mathrm{C} 3-\mathrm{H} 3$ | 0.9300 | $\mathrm{C} 8-\mathrm{H} 8 \mathrm{C}$ | 0.9600 |
|  |  |  |  |
| $\mathrm{C} 7-\mathrm{N}-\mathrm{N}$ | C |  | 120.3 |
| $\mathrm{C} 6-\mathrm{C} 1-\mathrm{C} 2$ | $112.25(11)$ | $\mathrm{C} 2-\mathrm{C} 5-\mathrm{H} 5$ | 120.3 |
| $\mathrm{C} 6-\mathrm{C} 1-\mathrm{C} 7$ | $119.57(10)$ | $\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 1$ | $121.08(12)$ |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{C} 7$ | $119.74(10)$ | $\mathrm{C} 5-\mathrm{C} 6-\mathrm{H} 6$ | 119.5 |
| $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 1$ | $120.70(10)$ | $\mathrm{N}-\mathrm{C} 7-\mathrm{C} 1$ | 119.5 |
| $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 8$ | $117.90(12)$ | $\mathrm{N}-\mathrm{C} 7-\mathrm{H} 7$ | $121.44(11)$ |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 8$ | $119.92(11)$ | $\mathrm{C} 1-\mathrm{C} 7-\mathrm{H} 7$ | 119.3 |
| $\mathrm{C} 4-\mathrm{C} 3-\mathrm{C} 2$ | $122.18(10)$ | $\mathrm{C} 2-\mathrm{C} 8-\mathrm{H} 8 \mathrm{~A}$ | 119.3 |
| $\mathrm{C} 4-\mathrm{C} 3-\mathrm{H} 3$ | $121.77(13)$ | $\mathrm{C} 2-\mathrm{C} 8-\mathrm{H} 8 \mathrm{~B}$ | 109.5 |
| $\mathrm{C} 2-\mathrm{C} 3-\mathrm{H} 3$ | 119.1 | $\mathrm{H} 8 \mathrm{~A}-\mathrm{C} 8-\mathrm{H} 8 \mathrm{~B}$ | 109.5 |
| $\mathrm{C} 5-\mathrm{C} 4-\mathrm{C} 3$ | 119.1 | $\mathrm{C} 2-\mathrm{C} 8-\mathrm{H} 8 \mathrm{C}$ | 109.5 |
| $\mathrm{C} 5-\mathrm{C} 4-\mathrm{H} 4$ | $120.20(12)$ | $\mathrm{H} 8 \mathrm{C}-\mathrm{C} 8-\mathrm{H} 8 \mathrm{C}$ | 109.5 |
| $\mathrm{C} 3-\mathrm{C} 4-\mathrm{H} 4$ | 119.9 | $\mathrm{H} 8 \mathrm{~B}-\mathrm{C} 8-\mathrm{H} 8 \mathrm{C}$ | 109.5 |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ | 119.9 | 109.5 |  |

Symmetry code: (i) $-x,-y+1,-z$.

