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2-[5-(4-Hydroxyphenyl)-1-phenyl-1*H*-pyrazol-3-yl]phenol

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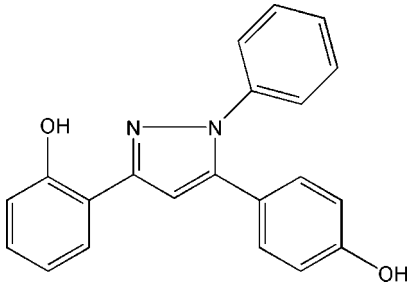
Key indicators: single-crystal X-ray study; T = 298 K; mean σ(C–C) = 0.003 Å; R factor = 0.047; wR factor = 0.129; data-to-parameter ratio = 16.4.

The title compound, $C_{21}H_{16}N_2O_2$, was derived from 1-(2-hydroxyphenyl)-3-(4-methoxyphenyl)propane-1,3-dione. The pyrazole ring and one of the hydroxy-substituted benzene rings are approximately coplanar, forming a dihedral angle of 7.5 (3)°. The relative conformation of these rings may be influenced by an intramolecular O—H···N hydrogen bond. In the crystal structure, intermolecular O—H···O hydrogen bonds involving different hydroxy groups of symmetry-related molecules form extended chains along [201].

Related literature

For related literature, see: Ahmad *et al.* (1990, 1997); Beeam *et al.* (1984); Elguero (1983); Trofinenko (1972).



Experimental

Crystal data

$C_{21}H_{16}N_2O_2$ M_r = 328.36
Monoclinic, $P2_1/c$
a = 10.793 (3) Å
b = 12.948 (3) Å
c = 11.705 (3) Å
β = 93.508 (14)°
V = 1632.7 (7) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.09 mm^{-1}
T = 298 (2) K
0.44 × 0.40 × 0.26 mm

Data collection

Siemens P4 diffractometer
Absorption correction: none
5767 measured reflections
3720 independent reflections
2353 reflections with $I > 2\sigma(I)$
R_{int} = 0.024
3 standard reflections
every 97 reflections
intensity decay: 3.6%

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.047
$wR(F^2)$ = 0.128
S = 1.03
3720 reflections
227 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.15 e Å^{-3}
$\Delta\rho_{min}$ = −0.16 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O1—H1B···O2^{i}	0.82	2.05	2.824 (2)	158
O2—H2B···N2	0.82	1.87	2.595 (2)	147

Symmetry code: (i) $x-1, -y+\frac{1}{2}, z-\frac{1}{2}$.

Data collection: *XSCANS* (Siemens, 1999); cell refinement: *XSCANS*; data reduction: *XSCANS*; program(s) used to solve structure: *SHELXTL-Plus* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL-Plus*; molecular graphics: *SHELXTL-Plus* and *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXTL-Plus*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2632).

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supporting information

Acta Cryst. (2008). E**64**, o1264 [doi:10.1107/S1600536808017054]

2-[5-(4-Hydroxyphenyl)-1-phenyl-1*H*-pyrazol-3-yl]phenol

Amir Badshah, Aurangzeb Hasan, Cecilia R. Barbarín and Mehwash Zia

S1. Comment

Pyrazoles are important because of their potential for biological activity (Beeam *et al.*, 1984). Both traditional and new scientific methods have been used used to prepare new materials for medicine (Elguero *et al.*, 1983) and agriculture (Trofinenko, 1972). Neutral and anionic pyrazoles are excellent ligands and their co-ordination chemistry has been extensively studied (Bonati, 1980). In the molecular structure of the title compound (III) (Fig. 1 and Fig. 3) there is an intramolecular hydrogen bond between the OH group of one phenolic group and the N atom of the pyrazole group (see Table 1 for hydrogen bond details). One of the phenyl groups is approximately coplanar with the pyrazole groups (dihedral angle = 7.5 (3)°), possibly due to the intramolecular hydrogen bond formation. The other two phenyl groups are rotated by 66.4 (12)°. In the crystal structure an intermolecular hydrogen bond between non equivalent hydroxy groups of symmetry related molecules, forms extended chains along [201] (Fig. 2).

S2. Experimental

Compound (I) [see Fig. 3] was prepared by a modified Baker Venkataram rearrangement as reported earlier (Ahmad *et al.*, 1990, 1997). Purification was carried out by recrystallization using absolute ethanol. Compound (II) was synthesized by adding 0.1 mole of phenyl hydrazine in 0.1 mole of compound (II) dissolved in 200 ml of absolute ethanol. The mixture was refluxed for 7 h. Solvent was removed under reduced pressure. Highly viscous residue was recrystallized using absolute ethanol. Compound (III) was synthesized by demethylation of compound (II) using 48% hydrogen bromide in acetic acid. Single crystals suitable for X-ray analysis were obtained by recrystallization from an ethanol solution of (III) at room temperature (Yield: 96%, m.p: 490K).

S3. Refinement

All H atoms were placed in idealized positions and treated as riding atoms, with C—H = 0.93Å, O-H = 0.82Å and U_{iso}(H) = 1.2U_{eq}(C) or 1.5U_{eq}(O).

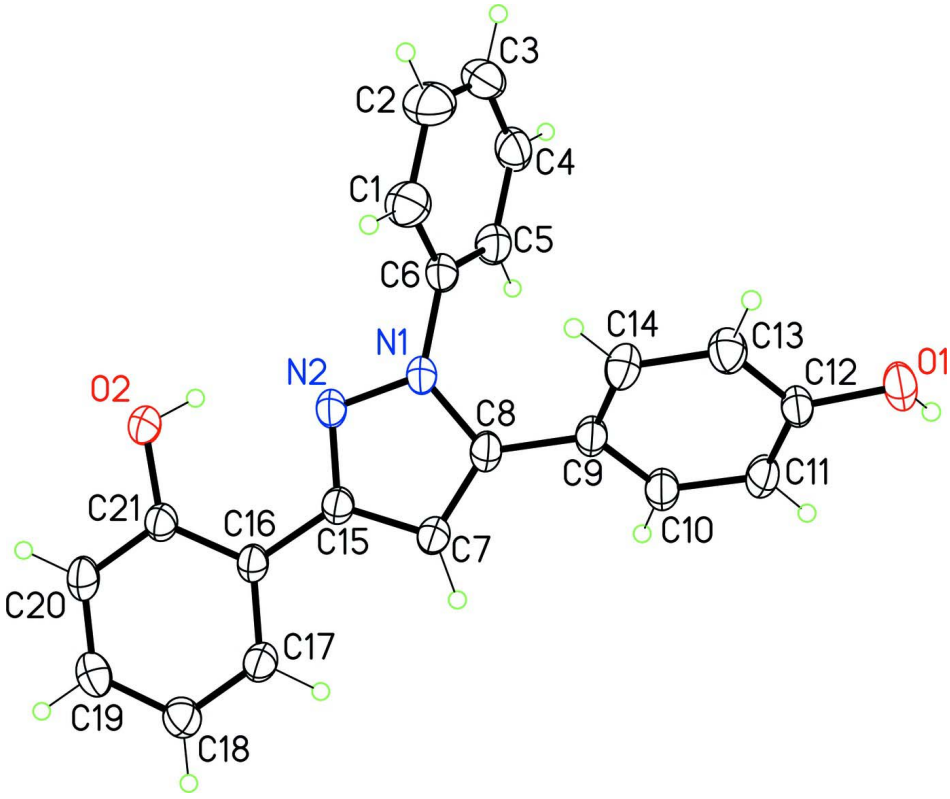

Figure 1

The molecular structure of (III) showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii.

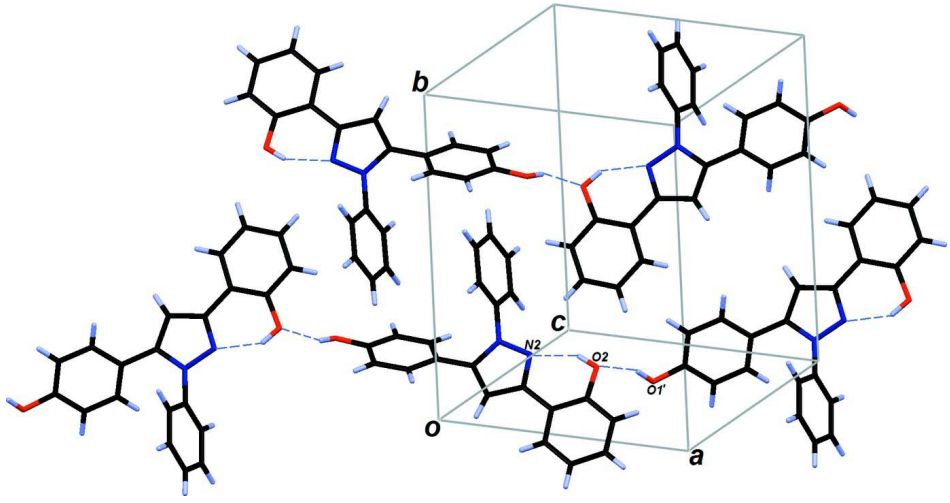

Figure 2

Part of the crystal structure of (III) showing the hydrogen bonds as dashed lines.

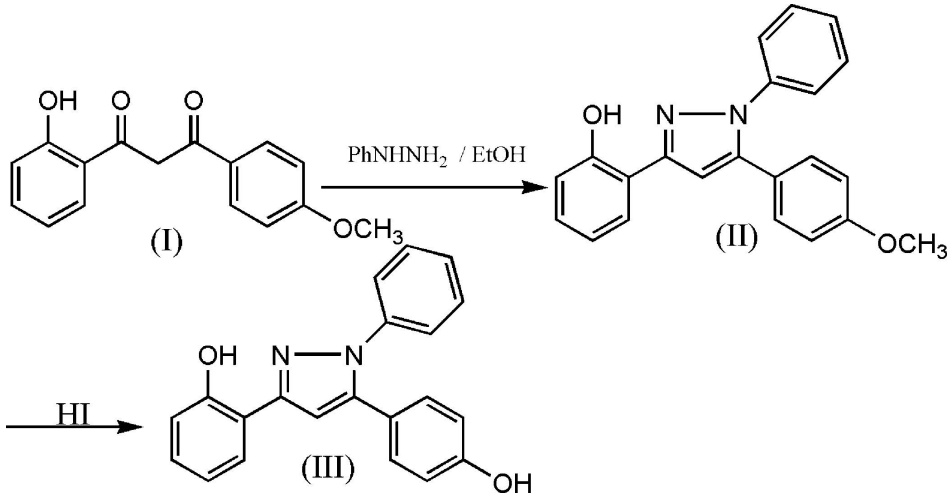

Figure 3

Reaction scheme.

2-[5-(4-Hydroxyphenyl)-1-phenyl-1*H*-pyrazol-3-yl]phenol

Crystal data

$C_{21}H_{16}N_2O_2$
$M_r = 328.36$
Monoclinic, $P2_1/c$
Hall symbol: -P 2ybc
$a = 10.793$ (3) Å
$b = 12.948$ (3) Å
$c = 11.705$ (3) Å
$\beta = 93.508$ (14)°
$V = 1632.7$ (7) Å^3
$Z = 4$
$F(000) = 688$
$D_x = 1.336$ Mg m^{-3}
Melting point: 490 K
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 84 reflections
$\theta = 4.6$–12.4°
$\mu = 0.09$ mm^{-1}
$T = 298$ K
Prismatic, colourless
0.44 × 0.40 × 0.26 mm

Data collection

Siemens P4 diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
$2\theta/\omega$ scans
5767 measured reflections
3720 independent reflections
2353 reflections with $I > 2\sigma(I)$
$R_{int} = 0.024$
$\theta_{max} = 27.5°$, $\theta_{min} = 1.9°$
$h = -14 \rightarrow 4$
$k = -16 \rightarrow 1$
$l = -15 \rightarrow 15$
3 standard reflections every 97 reflections
intensity decay: 3.7%

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.047$
$wR(F^2) = 0.128$
$S = 1.03$
3720 reflections
227 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0468P)^2 + 0.4416P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max} = 0.15$ e Å^{-3}
$\Delta\rho_{min} = -0.16$ e Å^{-3}

Extinction correction: *SHELXTL-Plus* (Sheldrick, 2008), $Fc^*=kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Extinction coefficient: 0.0155 (18)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	*y*	*z*	U_{iso}*/U_{eq}
O1	−0.19429 (12)	0.33852 (11)	−0.38082 (11)	0.0598 (4)
H1B	−0.2627	0.3197	−0.3621	0.090*
O2	0.55378 (12)	0.16872 (10)	0.17109 (13)	0.0603 (4)
H2B	0.5054	0.2064	0.1339	0.090*
N1	0.26395 (13)	0.27005 (12)	−0.01205 (13)	0.0467 (4)
N2	0.34918 (13)	0.21948 (12)	0.05691 (13)	0.0462 (4)
C1	0.3431 (2)	0.43942 (17)	−0.05478 (19)	0.0628 (6)
H1A	0.4075	0.4086	−0.0915	0.075*
C2	0.3343 (3)	0.54518 (19)	−0.0493 (2)	0.0775 (7)
H2A	0.3933	0.5863	−0.0821	0.093*
C3	0.2391 (2)	0.58993 (18)	0.0041 (2)	0.0721 (7)
H3A	0.2326	0.6615	0.0064	0.087*
C4	0.1531 (2)	0.53006 (17)	0.0543 (2)	0.0653 (6)
H4A	0.0891	0.5610	0.0915	0.078*
C5	0.16142 (18)	0.42387 (16)	0.04968 (17)	0.0546 (5)
H5A	0.1035	0.3827	0.0837	0.066*
C6	0.25650 (16)	0.38005 (14)	−0.00589 (15)	0.0460 (4)
C7	0.21875 (16)	0.10653 (14)	−0.03158 (15)	0.0437 (4)
H7A	0.1815	0.0444	−0.0540	0.052*
C8	0.18333 (15)	0.20293 (15)	−0.06634 (15)	0.0433 (4)
C9	0.08425 (15)	0.23722 (14)	−0.14967 (15)	0.0441 (4)
C10	−0.03190 (17)	0.19422 (15)	−0.15016 (17)	0.0523 (5)
H10A	−0.0474	0.1428	−0.0975	0.063*
C11	−0.12620 (17)	0.22564 (16)	−0.22714 (17)	0.0543 (5)
H11A	−0.2041	0.1950	−0.2266	0.065*
C12	−0.10491 (16)	0.30191 (14)	−0.30416 (15)	0.0460 (4)
C13	0.01140 (17)	0.34379 (16)	−0.30736 (16)	0.0516 (5)
H13A	0.0269	0.3941	−0.3614	0.062*
C14	0.10531 (17)	0.31154 (16)	−0.23075 (16)	0.0507 (5)
H14A	0.1841	0.3402	−0.2337	0.061*
C15	0.32262 (15)	0.11962 (14)	0.04482 (14)	0.0407 (4)
C16	0.40064 (15)	0.04285 (14)	0.10482 (14)	0.0404 (4)
C17	0.36890 (17)	−0.06076 (15)	0.10014 (15)	0.0476 (4)
H17A	0.2955	−0.0804	0.0603	0.057*
C18	0.44262 (18)	−0.13518 (16)	0.15259 (17)	0.0554 (5)

H18A	0.4201	−0.2044	0.1471	0.066*
C19	0.55033 (18)	−0.10638 (17)	0.21343 (16)	0.0546 (5)
H19A	0.6000	−0.1563	0.2503	0.066*
C20	0.58441 (18)	−0.00551 (16)	0.21988 (16)	0.0525 (5)
H20A	0.6572	0.0133	0.2613	0.063*
C21	0.51143 (16)	0.06923 (14)	0.16520 (15)	0.0447 (4)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0470 (8)	0.0663 (9)	0.0638 (8)	0.0073 (7)	−0.0160 (6)	0.0117 (7)
O2	0.0474 (8)	0.0503 (8)	0.0796 (10)	0.0043 (6)	−0.0253 (7)	−0.0134 (7)
N1	0.0380 (8)	0.0451 (9)	0.0548 (9)	0.0054 (7)	−0.0141 (7)	−0.0037 (7)
N2	0.0375 (8)	0.0480 (9)	0.0514 (8)	0.0067 (7)	−0.0119 (7)	−0.0054 (7)
C1	0.0587 (13)	0.0636 (14)	0.0668 (13)	−0.0021 (11)	0.0084 (10)	0.0014 (11)
C2	0.0867 (18)	0.0632 (15)	0.0824 (16)	−0.0137 (14)	0.0036 (14)	0.0141 (13)
C3	0.0878 (18)	0.0479 (12)	0.0775 (15)	0.0045 (12)	−0.0202 (14)	0.0046 (11)
C4	0.0574 (13)	0.0601 (13)	0.0764 (14)	0.0163 (11)	−0.0138 (11)	−0.0140 (11)
C5	0.0446 (10)	0.0554 (12)	0.0631 (12)	0.0036 (9)	−0.0028 (9)	−0.0045 (10)
C6	0.0423 (10)	0.0457 (10)	0.0483 (10)	0.0043 (8)	−0.0104 (8)	−0.0023 (8)
C7	0.0368 (9)	0.0468 (10)	0.0467 (9)	−0.0006 (8)	−0.0051 (7)	−0.0039 (8)
C8	0.0324 (8)	0.0524 (10)	0.0443 (9)	0.0035 (8)	−0.0032 (7)	−0.0031 (8)
C9	0.0348 (9)	0.0504 (10)	0.0460 (9)	0.0036 (8)	−0.0063 (7)	−0.0021 (8)
C10	0.0423 (10)	0.0558 (11)	0.0572 (11)	−0.0031 (9)	−0.0089 (8)	0.0118 (9)
C11	0.0363 (9)	0.0598 (12)	0.0651 (12)	−0.0072 (9)	−0.0108 (9)	0.0073 (10)
C12	0.0400 (9)	0.0490 (10)	0.0475 (10)	0.0079 (8)	−0.0103 (8)	−0.0008 (8)
C13	0.0477 (11)	0.0601 (12)	0.0465 (10)	−0.0009 (9)	−0.0014 (8)	0.0095 (9)
C14	0.0359 (9)	0.0654 (12)	0.0500 (10)	−0.0050 (9)	−0.0021 (8)	0.0021 (9)
C15	0.0351 (9)	0.0465 (10)	0.0400 (8)	0.0025 (8)	−0.0022 (7)	−0.0052 (8)
C16	0.0344 (8)	0.0487 (10)	0.0375 (8)	0.0031 (8)	−0.0034 (7)	−0.0033 (7)
C17	0.0415 (10)	0.0524 (11)	0.0478 (10)	−0.0047 (8)	−0.0051 (8)	0.0032 (8)
C18	0.0518 (11)	0.0526 (12)	0.0612 (11)	0.0003 (9)	−0.0005 (9)	0.0094 (9)
C19	0.0480 (11)	0.0617 (13)	0.0540 (11)	0.0108 (10)	0.0012 (9)	0.0122 (10)
C20	0.0407 (10)	0.0663 (13)	0.0488 (10)	0.0059 (9)	−0.0097 (8)	−0.0004 (9)
C21	0.0391 (9)	0.0491 (10)	0.0449 (9)	0.0044 (8)	−0.0052 (8)	−0.0084 (8)

Geometric parameters (Å, º)

O1—C12	1.362 (2)	C8—C9	1.471 (2)
O1—H1B	0.8200	C9—C10	1.371 (3)
O2—C21	1.367 (2)	C9—C14	1.380 (3)
O2—H2B	0.8200	C10—C11	1.379 (3)
N1—N2	1.3549 (19)	C10—H10A	0.9300
N1—C8	1.360 (2)	C11—C12	1.366 (3)
N1—C6	1.428 (2)	C11—H11A	0.9300
N2—C15	1.330 (2)	C12—C13	1.370 (3)
C1—C6	1.363 (3)	C13—C14	1.376 (3)
C1—C2	1.374 (3)	C13—H13A	0.9300

C1—H1A	0.9300	C14—H14A	0.9300
C2—C3	1.364 (4)	C15—C16	1.455 (2)
C2—H2A	0.9300	C16—C17	1.385 (3)
C3—C4	1.369 (3)	C16—C21	1.394 (2)
C3—H3A	0.9300	C17—C18	1.371 (3)
C4—C5	1.379 (3)	C17—H17A	0.9300
C4—H4A	0.9300	C18—C19	1.377 (3)
C5—C6	1.371 (3)	C18—H18A	0.9300
C5—H5A	0.9300	C19—C20	1.358 (3)
C7—C8	1.360 (3)	C19—H19A	0.9300
C7—C15	1.401 (2)	C20—C21	1.380 (3)
C7—H7A	0.9300	C20—H20A	0.9300

C12—O1—H1B	109.5	C11—C10—H10A	119.3
C21—O2—H2B	109.5	C12—C11—C10	119.82 (17)
N2—N1—C8	111.18 (15)	C12—C11—H11A	120.1
N2—N1—C6	119.34 (14)	C10—C11—H11A	120.1
C8—N1—C6	128.64 (14)	O1—C12—C11	123.06 (17)
C15—N2—N1	105.81 (13)	O1—C12—C13	117.22 (17)
C6—C1—C2	119.5 (2)	C11—C12—C13	119.71 (16)
C6—C1—H1A	120.3	C12—C13—C14	120.10 (18)
C2—C1—H1A	120.3	C12—C13—H13A	120.0
C3—C2—C1	120.0 (2)	C14—C13—H13A	120.0
C3—C2—H2A	120.0	C13—C14—C9	120.92 (17)
C1—C2—H2A	120.0	C13—C14—H14A	119.5
C2—C3—C4	120.4 (2)	C9—C14—H14A	119.5
C2—C3—H3A	119.8	N2—C15—C7	110.14 (15)
C4—C3—H3A	119.8	N2—C15—C16	119.87 (15)
C3—C4—C5	120.0 (2)	C7—C15—C16	129.96 (16)
C3—C4—H4A	120.0	C17—C16—C21	117.38 (16)
C5—C4—H4A	120.0	C17—C16—C15	120.54 (15)
C6—C5—C4	118.9 (2)	C21—C16—C15	122.04 (16)
C6—C5—H5A	120.5	C18—C17—C16	121.86 (18)
C4—C5—H5A	120.5	C18—C17—H17A	119.1
C1—C6—C5	121.20 (19)	C16—C17—H17A	119.1
C1—C6—N1	119.94 (18)	C17—C18—C19	119.33 (19)
C5—C6—N1	118.86 (18)	C17—C18—H18A	120.3
C8—C7—C15	106.22 (15)	C19—C18—H18A	120.3
C8—C7—H7A	126.9	C20—C19—C18	120.44 (18)
C15—C7—H7A	126.9	C20—C19—H19A	119.8
N1—C8—C7	106.65 (14)	C18—C19—H19A	119.8
N1—C8—C9	122.37 (17)	C19—C20—C21	120.24 (18)
C7—C8—C9	130.90 (17)	C19—C20—H20A	119.9
C10—C9—C14	117.98 (16)	C21—C20—H20A	119.9
C10—C9—C8	120.49 (17)	O2—C21—C20	117.24 (16)
C14—C9—C8	121.51 (16)	O2—C21—C16	122.02 (16)
C9—C10—C11	121.40 (18)	C20—C21—C16	120.73 (18)
C9—C10—H10A	119.3		

C8—N1—N2—C15	0.7 (2)	C10—C11—C12—O1	−178.41 (18)
C6—N1—N2—C15	171.10 (16)	C10—C11—C12—C13	2.6 (3)
C6—C1—C2—C3	−0.4 (4)	O1—C12—C13—C14	178.78 (17)
C1—C2—C3—C4	1.2 (4)	C11—C12—C13—C14	−2.2 (3)
C2—C3—C4—C5	−0.9 (3)	C12—C13—C14—C9	−0.2 (3)
C3—C4—C5—C6	−0.1 (3)	C10—C9—C14—C13	2.0 (3)
C2—C1—C6—C5	−0.7 (3)	C8—C9—C14—C13	−179.23 (17)
C2—C1—C6—N1	179.36 (19)	N1—N2—C15—C7	−0.6 (2)
C4—C5—C6—C1	0.9 (3)	N1—N2—C15—C16	177.64 (15)
C4—C5—C6—N1	−179.09 (17)	C8—C7—C15—N2	0.4 (2)
N2—N1—C6—C1	76.2 (2)	C8—C7—C15—C16	−177.66 (17)
C8—N1—C6—C1	−115.2 (2)	N2—C15—C16—C17	175.03 (17)
N2—N1—C6—C5	−103.7 (2)	C7—C15—C16—C17	−7.1 (3)
C8—N1—C6—C5	64.8 (3)	N2—C15—C16—C21	−7.4 (3)
N2—N1—C8—C7	−0.4 (2)	C7—C15—C16—C21	170.51 (18)
C6—N1—C8—C7	−169.74 (17)	C21—C16—C17—C18	0.1 (3)
N2—N1—C8—C9	−177.64 (16)	C15—C16—C17—C18	177.79 (17)
C6—N1—C8—C9	13.1 (3)	C16—C17—C18—C19	1.1 (3)
C15—C7—C8—N1	0.0 (2)	C17—C18—C19—C20	−1.1 (3)
C15—C7—C8—C9	176.91 (18)	C18—C19—C20—C21	−0.1 (3)
N1—C8—C9—C10	−138.9 (2)	C19—C20—C21—O2	−177.28 (18)
C7—C8—C9—C10	44.6 (3)	C19—C20—C21—C16	1.4 (3)
N1—C8—C9—C14	42.4 (3)	C17—C16—C21—O2	177.27 (17)
C7—C8—C9—C14	−134.1 (2)	C15—C16—C21—O2	−0.4 (3)
C14—C9—C10—C11	−1.6 (3)	C17—C16—C21—C20	−1.4 (3)
C8—C9—C10—C11	179.66 (18)	C15—C16—C21—C20	−179.04 (17)
C9—C10—C11—C12	−0.7 (3)		

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O1—H1*B*···O2^{i}	0.82	2.05	2.824 (2)	158
O2—H2*B*···N2	0.82	1.87	2.595 (2)	147

Symmetry code: (i) $x-1, -y+1/2, z-1/2$.