

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

2-Nitro-*N*-(4-pyridinio)benzenesulfonamideJiang-Sheng Li,^{a*} Dao-Wu Yang^a and Wei-Dong Liu^b

^aSchool of Chemical and Environmental Engineering, Changsha University of Science & Technology, Changsha 410076, People's Republic of China, and ^bHunan Research Institute of Chemical Industry, Changsha 410007, People's Republic of China

Correspondence e-mail: jansenlee1103@yahoo.com.cn

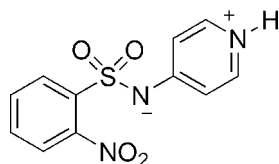
Received 20 November 2007; accepted 29 November 2007

Key indicators: single-crystal X-ray study; $T = 294$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; R factor = 0.058; wR factor = 0.174; data-to-parameter ratio = 12.5.

The title compound, $\text{C}_{11}\text{H}_9\text{N}_3\text{O}_4\text{S}$, crystallizes with two molecules in the asymmetric unit; each molecule exists as a zwitterion in the solid state. Intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds link the molecules into chains. Weak $\text{C}-\text{H}\cdots\text{O}$ interactions further stabilize the crystal structure.

Related literature

For zwitterionic forms of *N*-arylbenzenesulfonamides, see: Yu *et al.* (2007); Amendola *et al.* (2005); Lindley *et al.* (1977); Schaumann *et al.* (1975).



Experimental

Crystal data

 $\text{C}_{11}\text{H}_9\text{N}_3\text{O}_4\text{S}$ $M_r = 279.27$ Triclinic, $P\bar{1}$ $a = 7.768$ (3) Å $b = 12.570$ (5) Å $c = 13.436$ (5) Å $\alpha = 75.534$ (7)° $\beta = 85.400$ (7)° $\gamma = 88.741$ (6)° $V = 1266.3$ (8) Å³ $Z = 4$ Mo $K\alpha$ radiation $\mu = 0.27$ mm⁻¹ $T = 294$ (2) K

0.30 × 0.24 × 0.20 mm

Data collection

Bruker SMART 1K CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.924$, $T_{\max} = 0.948$

6351 measured reflections
4357 independent reflections
2918 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.058$ $wR(F^2) = 0.174$ $S = 1.02$

4357 reflections

349 parameters

H atoms treated by a mixture of independent and constrained refinement

 $\Delta\rho_{\text{max}} = 0.40$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|---|----------|-------------|-------------|---------------|
| $\text{N4}-\text{H4A}\cdots\text{O1}^{\text{i}}$ | 0.95 (4) | 2.67 (4) | 3.215 (4) | 117 (3) |
| $\text{N4}-\text{H4A}\cdots\text{N2}^{\text{i}}$ | 0.95 (4) | 2.08 (4) | 2.969 (4) | 155 (3) |
| $\text{N1}-\text{H1A}\cdots\text{N5}^{\text{ii}}$ | 0.88 (5) | 2.02 (5) | 2.898 (4) | 172 (4) |
| $\text{C14}-\text{H14}\cdots\text{O1}^{\text{i}}$ | 0.93 | 2.54 | 3.140 (5) | 122 |
| $\text{C14}-\text{H14}\cdots\text{O6}^{\text{iii}}$ | 0.93 | 2.58 | 3.249 (5) | 130 |
| $\text{C16}-\text{H16}\cdots\text{O1}^{\text{iv}}$ | 0.93 | 2.54 | 3.265 (5) | 135 |

Symmetry codes: (i) $x - 1, y - 1, z$; (ii) $x + 1, y, z$; (iii) $x - 1, y, z$; (iv) $x, y - 1, z$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT2635).

References

- Amendola, V., Boiocchi, M., Fabbrizzi, L. & Palchetti, A. (2005). *Chem. Eur. J.* **11**, 120–127.
- Bruker (1997). *SMART, SAINT and SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Lindley, P. F., Mahmoud, M. M., Dodd, C., Smith, C. H., Boyd, G. V. & Norris, T. (1977). *Acta Cryst.* **B33**, 2160–2164.
- Schaumann, E., Rohr, A., Sieveking, S. & Walter, W. (1975). *Angew. Chem. Int. Ed. Engl.* **14**, 493.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Yu, H.-J., Chen, J.-Z., Simpson, J., Li, J.-S. & Bai, G.-Y. (2007). *Acta Cryst.* **E63**, o3720.

supporting information

Acta Cryst. (2008). E64, o204 [https://doi.org/10.1107/S1600536807064410]

2-Nitro-*N*-(4-pyridinio)benzenesulfonamidate

Jiang-Sheng Li, Dao-Wu Yang and Wei-Dong Liu

S1. Comment

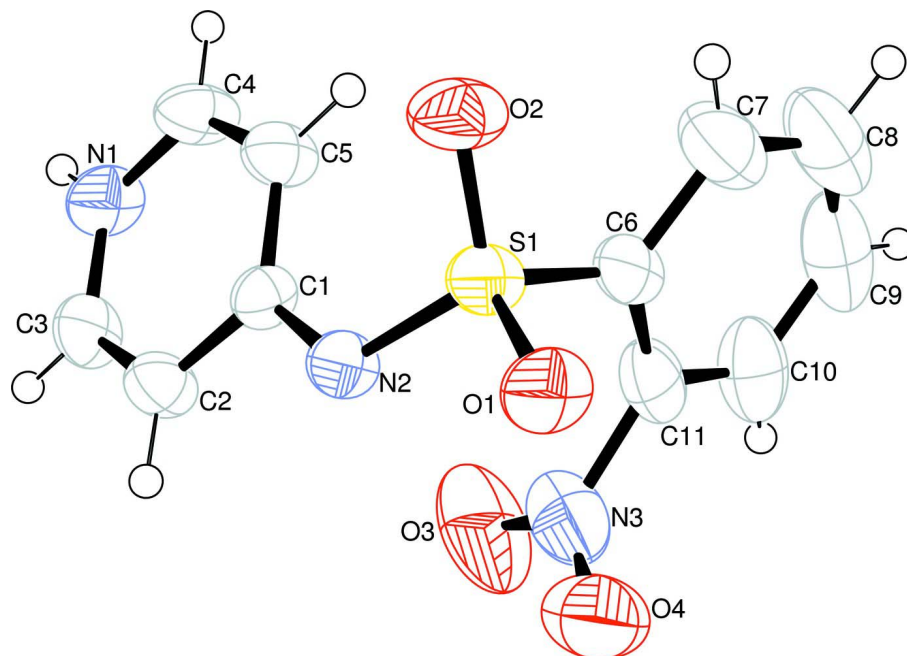
The title compound crystallizes with two molecules in the asymmetric unit; each molecule displays a zwitterion structure with the pyridine N protonated and the amide N deprotonated (Figs. 1 and 2). The relatively short C1—N2 [1.382 (4) Å] and C12—N5 [1.374 (4) Å] distances indicate that the N2 and N5 lone-pair electrons weakly conjugate with pyridinium rings. The benzene ring forms an angle of 85.1 (2) and 89.3 (2)° with the pyridinium ring in the two molecules of the asymmetric unit. In the addition, the nitro group is inclined to the benzene ring with 73.3 (3)°, and 80.0 (4)°. In the crystal, intermolecular N—H···N and C—H···O hydrogen bonds (Table 1) link the molecules into chains.

S2. Experimental

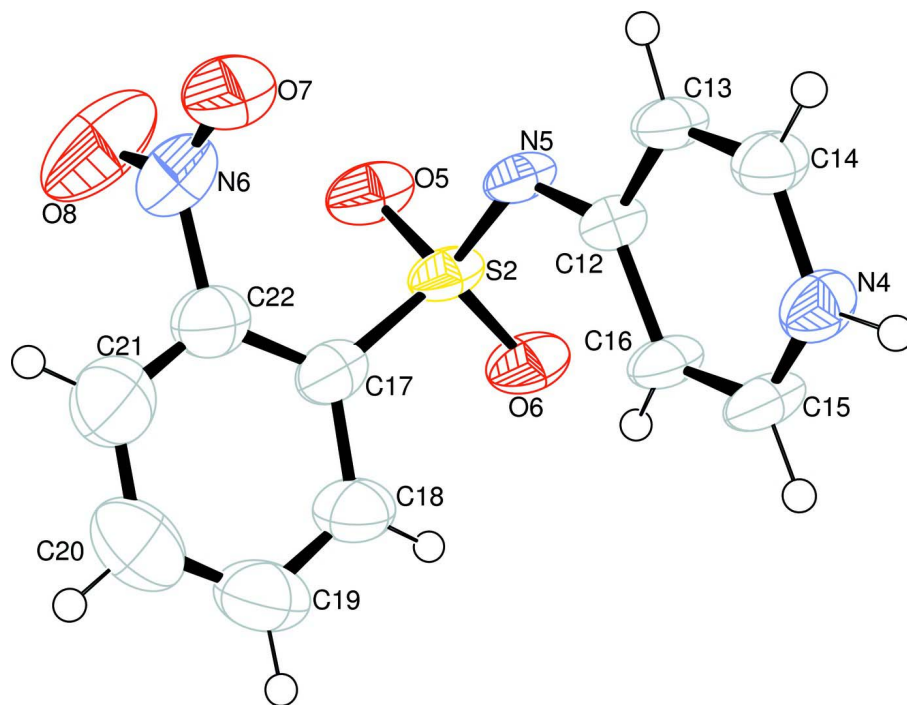
A solution of 2-nitrobenzenesulfonyl chloride (2.2 g, 10 mmol) in CH₂Cl₂ (10 ml) was added dropwise to a suspension of 4-aminopyridine (0.9 g, 10 mmol) in CH₂Cl₂ (10 ml) at room temperature with stirring. The reaction mixture was stirring overnight. The yellow solid obtained was washed with warm water in a yield of 70.3%. Yellow blocks were grown from its formic solution.

S3. Refinement

The N-bound H atoms were located in a difference map and their coordinates were refined with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{N})$. The C-bound H atoms were positioned geometrically (C—H = 0.93 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$.

**Figure 1**

Perspective view of one of the two molecules in the asymmetric unit with the atom-numbering scheme and 50% probability displacement ellipsoids.

**Figure 2**

Perspective view of the other molecule in the asymmetric unit with the atom-numbering scheme and 50% probability displacement ellipsoids.

2-Nitro-*N*-(4-pyridinio)benzenesulfonamidate*Crystal data*C₁₁H₉N₃O₄S $M_r = 279.27$ Triclinic, *P*1Hall symbol: -*P* 1 $a = 7.768$ (3) Å $b = 12.570$ (5) Å $c = 13.436$ (5) Å $\alpha = 75.534$ (7)° $\beta = 85.400$ (7)° $\gamma = 88.741$ (6)° $V = 1266.3$ (8) Å³ $Z = 4$ $F(000) = 576$ $D_x = 1.465$ Mg m⁻³Mo *K*α radiation, $\lambda = 0.71073$ Å

Cell parameters from 2204 reflections

 $\theta = 3.0$ – 24.8 ° $\mu = 0.27$ mm⁻¹ $T = 294$ K

Block, yellow

 $0.30 \times 0.24 \times 0.20$ mm*Data collection*Bruker SMART 1K CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 φ and ω scans

Absorption correction: multi-scan

(SADABS; Sheldrick, 1996)

 $T_{\min} = 0.924$, $T_{\max} = 0.948$

6351 measured reflections

4357 independent reflections

2918 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.026$ $\theta_{\text{max}} = 25.0$ °, $\theta_{\text{min}} = 1.6$ ° $h = -9 \rightarrow 8$ $k = -14 \rightarrow 11$ $l = -15 \rightarrow 11$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.058$ $wR(F^2) = 0.174$ $S = 1.02$

4357 reflections

349 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sitesH atoms treated by a mixture of independent
and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0887P)^2 + 0.7778P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} = 0.001$ $\Delta\rho_{\text{max}} = 0.40$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³*Special details*

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|--------------|-------------|-------------|----------------------------------|
| S1 | 0.35624 (11) | 1.06378 (7) | 0.83977 (8) | 0.0438 (3) |
| S2 | 0.16866 (12) | 0.54595 (7) | 0.82807 (9) | 0.0484 (3) |
| O1 | 0.3338 (3) | 1.1734 (2) | 0.8574 (2) | 0.0530 (7) |

| | | | | |
|-----|-------------|------------|------------|-------------|
| O2 | 0.2218 (3) | 0.9840 (2) | 0.8878 (2) | 0.0605 (8) |
| O3 | 0.7821 (5) | 1.1047 (4) | 0.6529 (3) | 0.1145 (16) |
| O4 | 0.6473 (5) | 1.2478 (4) | 0.6875 (3) | 0.0885 (12) |
| O5 | 0.1958 (4) | 0.6616 (2) | 0.8216 (3) | 0.0646 (9) |
| O6 | 0.2869 (3) | 0.4682 (2) | 0.8871 (2) | 0.0619 (8) |
| O7 | -0.1188 (6) | 0.6900 (4) | 0.6559 (3) | 0.1066 (15) |
| O8 | 0.1034 (8) | 0.7942 (3) | 0.6026 (4) | 0.141 (2) |
| N1 | 0.7802 (4) | 0.7241 (3) | 0.8638 (3) | 0.0519 (9) |
| H1A | 0.828 (6) | 0.660 (4) | 0.864 (3) | 0.062* |
| N2 | 0.5495 (4) | 1.0267 (2) | 0.8668 (2) | 0.0408 (7) |
| N3 | 0.6518 (5) | 1.1607 (4) | 0.6602 (3) | 0.0714 (11) |
| N4 | -0.2797 (4) | 0.2206 (3) | 0.9043 (3) | 0.0471 (8) |
| H4A | -0.344 (5) | 0.154 (3) | 0.914 (3) | 0.056* |
| N5 | -0.0324 (4) | 0.5224 (2) | 0.8627 (3) | 0.0439 (8) |
| N6 | 0.0329 (7) | 0.7048 (4) | 0.6261 (4) | 0.0840 (13) |
| C1 | 0.6149 (4) | 0.9243 (3) | 0.8638 (3) | 0.0367 (8) |
| C2 | 0.7930 (5) | 0.9064 (3) | 0.8816 (3) | 0.0467 (10) |
| H2 | 0.8574 | 0.9625 | 0.8941 | 0.056* |
| C3 | 0.8711 (5) | 0.8074 (3) | 0.8804 (3) | 0.0498 (10) |
| H3 | 0.9878 | 0.7978 | 0.8910 | 0.060* |
| C4 | 0.6109 (5) | 0.7371 (3) | 0.8454 (3) | 0.0529 (11) |
| H4 | 0.5509 | 0.6790 | 0.8332 | 0.063* |
| C5 | 0.5256 (5) | 0.8354 (3) | 0.8445 (3) | 0.0469 (9) |
| H5 | 0.4097 | 0.8429 | 0.8311 | 0.056* |
| C6 | 0.3505 (5) | 1.0794 (3) | 0.7031 (3) | 0.0489 (10) |
| C7 | 0.2000 (6) | 1.0521 (4) | 0.6646 (5) | 0.0733 (14) |
| H7 | 0.1070 | 1.0224 | 0.7104 | 0.088* |
| C8 | 0.1865 (8) | 1.0684 (5) | 0.5589 (6) | 0.0930 (19) |
| H8 | 0.0834 | 1.0522 | 0.5352 | 0.112* |
| C9 | 0.3248 (9) | 1.1085 (5) | 0.4890 (5) | 0.0883 (19) |
| H9 | 0.3158 | 1.1162 | 0.4190 | 0.106* |
| C10 | 0.4776 (7) | 1.1375 (4) | 0.5235 (4) | 0.0729 (14) |
| H10 | 0.5709 | 1.1654 | 0.4773 | 0.088* |
| C11 | 0.4862 (5) | 1.1233 (3) | 0.6295 (3) | 0.0535 (11) |
| C12 | -0.1050 (5) | 0.4203 (3) | 0.8786 (3) | 0.0387 (8) |
| C13 | -0.2877 (4) | 0.4112 (3) | 0.8968 (3) | 0.0435 (9) |
| H13 | -0.3524 | 0.4731 | 0.9014 | 0.052* |
| C14 | -0.3706 (5) | 0.3136 (3) | 0.9076 (3) | 0.0445 (9) |
| H14 | -0.4905 | 0.3107 | 0.9174 | 0.053* |
| C15 | -0.1061 (5) | 0.2233 (3) | 0.8910 (3) | 0.0494 (10) |
| H15 | -0.0461 | 0.1586 | 0.8906 | 0.059* |
| C16 | -0.0145 (5) | 0.3191 (3) | 0.8780 (3) | 0.0452 (9) |
| H16 | 0.1054 | 0.3184 | 0.8689 | 0.054* |
| C17 | 0.1985 (5) | 0.5308 (3) | 0.6978 (3) | 0.0510 (10) |
| C18 | 0.2909 (6) | 0.4405 (4) | 0.6757 (4) | 0.0650 (12) |
| H18 | 0.3386 | 0.3891 | 0.7288 | 0.078* |
| C19 | 0.3118 (7) | 0.4269 (5) | 0.5753 (4) | 0.0785 (15) |
| H19 | 0.3684 | 0.3649 | 0.5631 | 0.094* |

| | | | | |
|-----|------------|------------|------------|-------------|
| C20 | 0.2496 (7) | 0.5043 (5) | 0.4941 (5) | 0.0835 (16) |
| H20 | 0.2678 | 0.4960 | 0.4272 | 0.100* |
| C21 | 0.1591 (7) | 0.5951 (5) | 0.5133 (4) | 0.0792 (15) |
| H21 | 0.1146 | 0.6473 | 0.4596 | 0.095* |
| C22 | 0.1365 (6) | 0.6065 (4) | 0.6129 (4) | 0.0609 (12) |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|------------|--------------|--------------|--------------|
| S1 | 0.0292 (5) | 0.0336 (5) | 0.0692 (7) | 0.0070 (4) | -0.0021 (4) | -0.0151 (4) |
| S2 | 0.0308 (5) | 0.0362 (5) | 0.0800 (8) | 0.0050 (4) | -0.0071 (5) | -0.0173 (5) |
| O1 | 0.0437 (15) | 0.0371 (14) | 0.080 (2) | 0.0118 (12) | -0.0007 (13) | -0.0210 (13) |
| O2 | 0.0355 (15) | 0.0474 (16) | 0.096 (2) | -0.0039 (12) | 0.0047 (14) | -0.0152 (14) |
| O3 | 0.042 (2) | 0.187 (5) | 0.107 (3) | 0.030 (3) | -0.0068 (19) | -0.025 (3) |
| O4 | 0.094 (3) | 0.098 (3) | 0.073 (2) | -0.037 (2) | -0.0075 (19) | -0.016 (2) |
| O5 | 0.0410 (16) | 0.0427 (16) | 0.114 (3) | -0.0023 (12) | -0.0067 (16) | -0.0272 (16) |
| O6 | 0.0389 (16) | 0.0580 (18) | 0.091 (2) | 0.0156 (13) | -0.0159 (14) | -0.0204 (15) |
| O7 | 0.102 (3) | 0.114 (3) | 0.088 (3) | 0.060 (3) | 0.004 (2) | -0.005 (2) |
| O8 | 0.167 (5) | 0.049 (2) | 0.198 (5) | 0.014 (3) | -0.043 (4) | -0.007 (3) |
| N1 | 0.047 (2) | 0.0334 (18) | 0.076 (3) | 0.0107 (15) | -0.0029 (17) | -0.0165 (16) |
| N2 | 0.0327 (16) | 0.0312 (16) | 0.061 (2) | 0.0069 (12) | -0.0074 (14) | -0.0149 (14) |
| N3 | 0.052 (3) | 0.106 (4) | 0.051 (2) | -0.006 (2) | -0.0016 (18) | -0.009 (2) |
| N4 | 0.0427 (19) | 0.0346 (17) | 0.064 (2) | -0.0025 (14) | -0.0061 (16) | -0.0116 (15) |
| N5 | 0.0337 (17) | 0.0295 (16) | 0.069 (2) | 0.0076 (13) | -0.0044 (14) | -0.0143 (14) |
| N6 | 0.090 (4) | 0.068 (3) | 0.083 (3) | 0.025 (3) | -0.012 (3) | 0.001 (2) |
| C1 | 0.0348 (19) | 0.0311 (18) | 0.043 (2) | 0.0036 (15) | -0.0001 (15) | -0.0084 (15) |
| C2 | 0.043 (2) | 0.038 (2) | 0.064 (3) | 0.0032 (17) | -0.0134 (19) | -0.0195 (18) |
| C3 | 0.040 (2) | 0.044 (2) | 0.067 (3) | 0.0144 (18) | -0.0130 (19) | -0.0144 (19) |
| C4 | 0.042 (2) | 0.033 (2) | 0.088 (3) | 0.0000 (17) | -0.003 (2) | -0.023 (2) |
| C5 | 0.033 (2) | 0.039 (2) | 0.070 (3) | 0.0034 (16) | -0.0053 (18) | -0.0173 (18) |
| C6 | 0.033 (2) | 0.033 (2) | 0.084 (3) | 0.0124 (16) | -0.016 (2) | -0.0191 (19) |
| C7 | 0.050 (3) | 0.073 (3) | 0.106 (4) | 0.008 (2) | -0.025 (3) | -0.033 (3) |
| C8 | 0.079 (4) | 0.104 (5) | 0.116 (5) | 0.022 (3) | -0.053 (4) | -0.053 (4) |
| C9 | 0.109 (5) | 0.093 (4) | 0.082 (4) | 0.051 (4) | -0.051 (4) | -0.049 (3) |
| C10 | 0.075 (3) | 0.079 (3) | 0.068 (3) | 0.034 (3) | -0.019 (3) | -0.024 (3) |
| C11 | 0.047 (2) | 0.057 (3) | 0.061 (3) | 0.017 (2) | -0.018 (2) | -0.022 (2) |
| C12 | 0.039 (2) | 0.0328 (19) | 0.046 (2) | 0.0066 (15) | -0.0058 (16) | -0.0128 (16) |
| C13 | 0.032 (2) | 0.037 (2) | 0.064 (3) | 0.0085 (15) | -0.0013 (17) | -0.0165 (17) |
| C14 | 0.034 (2) | 0.041 (2) | 0.059 (3) | 0.0043 (16) | -0.0023 (17) | -0.0135 (18) |
| C15 | 0.044 (2) | 0.031 (2) | 0.075 (3) | 0.0121 (17) | -0.0098 (19) | -0.0155 (18) |
| C16 | 0.0316 (19) | 0.033 (2) | 0.071 (3) | 0.0094 (15) | -0.0041 (17) | -0.0143 (18) |
| C17 | 0.033 (2) | 0.036 (2) | 0.079 (3) | -0.0037 (16) | 0.0024 (19) | -0.0073 (19) |
| C18 | 0.051 (3) | 0.061 (3) | 0.081 (4) | 0.011 (2) | 0.008 (2) | -0.018 (2) |
| C19 | 0.067 (3) | 0.079 (4) | 0.093 (4) | 0.006 (3) | 0.016 (3) | -0.034 (3) |
| C20 | 0.065 (3) | 0.106 (5) | 0.080 (4) | -0.015 (3) | 0.018 (3) | -0.029 (3) |
| C21 | 0.061 (3) | 0.077 (4) | 0.089 (4) | -0.003 (3) | 0.005 (3) | -0.004 (3) |
| C22 | 0.053 (3) | 0.049 (3) | 0.077 (3) | -0.003 (2) | 0.004 (2) | -0.014 (2) |

Geometric parameters (Å, °)

| | | | |
|-----------|-------------|-------------|-----------|
| S1—O2 | 1.459 (3) | C5—H5 | 0.9300 |
| S1—O1 | 1.460 (3) | C6—C7 | 1.399 (6) |
| S1—N2 | 1.605 (3) | C6—C11 | 1.407 (6) |
| S1—C6 | 1.801 (5) | C7—C8 | 1.395 (8) |
| S2—O5 | 1.453 (3) | C7—H7 | 0.9300 |
| S2—O6 | 1.455 (3) | C8—C9 | 1.384 (9) |
| S2—N5 | 1.606 (3) | C8—H8 | 0.9300 |
| S2—C17 | 1.804 (5) | C9—C10 | 1.396 (8) |
| O3—N3 | 1.232 (5) | C9—H9 | 0.9300 |
| O4—N3 | 1.237 (5) | C10—C11 | 1.397 (6) |
| O7—N6 | 1.217 (6) | C10—H10 | 0.9300 |
| O8—N6 | 1.219 (6) | C12—C13 | 1.423 (5) |
| N1—C3 | 1.348 (5) | C12—C16 | 1.441 (5) |
| N1—C4 | 1.356 (5) | C13—C14 | 1.368 (5) |
| N1—H1A | 0.88 (5) | C13—H13 | 0.9300 |
| N2—C1 | 1.382 (4) | C14—H14 | 0.9300 |
| N3—C11 | 1.501 (6) | C15—C16 | 1.377 (5) |
| N4—C15 | 1.346 (5) | C15—H15 | 0.9300 |
| N4—C14 | 1.360 (5) | C16—H16 | 0.9300 |
| N4—H4A | 0.95 (4) | C17—C22 | 1.402 (6) |
| N5—C12 | 1.374 (4) | C17—C18 | 1.411 (6) |
| N6—C22 | 1.499 (6) | C18—C19 | 1.397 (7) |
| C1—C5 | 1.416 (5) | C18—H18 | 0.9300 |
| C1—C2 | 1.427 (5) | C19—C20 | 1.380 (8) |
| C2—C3 | 1.376 (5) | C19—H19 | 0.9300 |
| C2—H2 | 0.9300 | C20—C21 | 1.396 (7) |
| C3—H3 | 0.9300 | C20—H20 | 0.9300 |
| C4—C5 | 1.387 (5) | C21—C22 | 1.378 (7) |
| C4—H4 | 0.9300 | C21—H21 | 0.9300 |
| O2—S1—O1 | 116.20 (17) | C6—C7—H7 | 119.3 |
| O2—S1—N2 | 115.50 (16) | C9—C8—C7 | 120.7 (5) |
| O1—S1—N2 | 106.12 (15) | C9—C8—H8 | 119.6 |
| O2—S1—C6 | 105.37 (18) | C7—C8—H8 | 119.6 |
| O1—S1—C6 | 106.63 (16) | C8—C9—C10 | 120.1 (5) |
| N2—S1—C6 | 106.27 (16) | C8—C9—H9 | 120.0 |
| O5—S2—O6 | 116.83 (17) | C10—C9—H9 | 120.0 |
| O5—S2—N5 | 106.34 (16) | C9—C10—C11 | 118.1 (5) |
| O6—S2—N5 | 114.96 (17) | C9—C10—H10 | 121.0 |
| O5—S2—C17 | 106.20 (18) | C11—C10—H10 | 121.0 |
| O6—S2—C17 | 106.00 (18) | C10—C11—C6 | 123.5 (4) |
| N5—S2—C17 | 105.63 (17) | C10—C11—N3 | 114.9 (4) |
| C3—N1—C4 | 120.7 (3) | C6—C11—N3 | 121.6 (4) |
| C3—N1—H1A | 122 (3) | N5—C12—C13 | 118.1 (3) |
| C4—N1—H1A | 117 (3) | N5—C12—C16 | 126.5 (3) |
| C1—N2—S1 | 122.7 (2) | C13—C12—C16 | 115.4 (3) |

| | | | |
|---------------|------------|-----------------|------------|
| O3—N3—O4 | 125.2 (5) | C14—C13—C12 | 121.7 (3) |
| O3—N3—C11 | 117.7 (5) | C14—C13—H13 | 119.2 |
| O4—N3—C11 | 117.1 (4) | C12—C13—H13 | 119.2 |
| C15—N4—C14 | 120.3 (3) | N4—C14—C13 | 120.7 (3) |
| C15—N4—H4A | 122 (2) | N4—C14—H14 | 119.6 |
| C14—N4—H4A | 117 (2) | C13—C14—H14 | 119.6 |
| C12—N5—S2 | 122.8 (2) | N4—C15—C16 | 122.0 (3) |
| O7—N6—O8 | 124.3 (5) | N4—C15—H15 | 119.0 |
| O7—N6—C22 | 117.2 (5) | C16—C15—H15 | 119.0 |
| O8—N6—C22 | 118.4 (5) | C15—C16—C12 | 119.9 (3) |
| N2—C1—C5 | 127.5 (3) | C15—C16—H16 | 120.1 |
| N2—C1—C2 | 116.3 (3) | C12—C16—H16 | 120.1 |
| C5—C1—C2 | 116.3 (3) | C22—C17—C18 | 115.8 (4) |
| C3—C2—C1 | 121.0 (3) | C22—C17—S2 | 123.5 (3) |
| C3—C2—H2 | 119.5 | C18—C17—S2 | 120.7 (3) |
| C1—C2—H2 | 119.5 | C19—C18—C17 | 121.1 (4) |
| N1—C3—C2 | 120.7 (4) | C19—C18—H18 | 119.4 |
| N1—C3—H3 | 119.6 | C17—C18—H18 | 119.4 |
| C2—C3—H3 | 119.6 | C20—C19—C18 | 120.8 (5) |
| N1—C4—C5 | 121.3 (3) | C20—C19—H19 | 119.6 |
| N1—C4—H4 | 119.4 | C18—C19—H19 | 119.6 |
| C5—C4—H4 | 119.4 | C19—C20—C21 | 119.4 (5) |
| C4—C5—C1 | 120.0 (3) | C19—C20—H20 | 120.3 |
| C4—C5—H5 | 120.0 | C21—C20—H20 | 120.3 |
| C1—C5—H5 | 120.0 | C22—C21—C20 | 119.2 (5) |
| C7—C6—C11 | 116.2 (4) | C22—C21—H21 | 120.4 |
| C7—C6—S1 | 119.4 (4) | C20—C21—H21 | 120.4 |
| C11—C6—S1 | 124.3 (3) | C21—C22—C17 | 123.6 (4) |
| C8—C7—C6 | 121.4 (5) | C21—C22—N6 | 115.2 (4) |
| C8—C7—H7 | 119.3 | C17—C22—N6 | 121.2 (4) |
| O2—S1—N2—C1 | 46.5 (3) | O4—N3—C11—C10 | 105.2 (5) |
| O1—S1—N2—C1 | 176.9 (3) | O3—N3—C11—C6 | 108.8 (5) |
| C6—S1—N2—C1 | -69.9 (3) | O4—N3—C11—C6 | -73.6 (5) |
| O5—S2—N5—C12 | -178.8 (3) | S2—N5—C12—C13 | -172.5 (3) |
| O6—S2—N5—C12 | -47.9 (4) | S2—N5—C12—C16 | 7.1 (5) |
| C17—S2—N5—C12 | 68.6 (3) | N5—C12—C13—C14 | 176.4 (3) |
| S1—N2—C1—C5 | -4.5 (5) | C16—C12—C13—C14 | -3.2 (5) |
| S1—N2—C1—C2 | 175.0 (3) | C15—N4—C14—C13 | 0.5 (6) |
| N2—C1—C2—C3 | 180.0 (3) | C12—C13—C14—N4 | 2.0 (6) |
| C5—C1—C2—C3 | -0.5 (6) | C14—N4—C15—C16 | -1.5 (6) |
| C4—N1—C3—C2 | 1.7 (6) | N4—C15—C16—C12 | 0.1 (6) |
| C1—C2—C3—N1 | -1.0 (6) | N5—C12—C16—C15 | -177.3 (4) |
| C3—N1—C4—C5 | -0.9 (6) | C13—C12—C16—C15 | 2.2 (5) |
| N1—C4—C5—C1 | -0.6 (6) | O5—S2—C17—C22 | -40.9 (4) |
| N2—C1—C5—C4 | -179.3 (4) | O6—S2—C17—C22 | -165.8 (3) |
| C2—C1—C5—C4 | 1.2 (5) | N5—S2—C17—C22 | 71.8 (4) |
| O2—S1—C6—C7 | 22.1 (4) | O5—S2—C17—C18 | 137.9 (3) |

| | | | |
|---------------|------------|-----------------|------------|
| O1—S1—C6—C7 | -101.9 (3) | O6—S2—C17—C18 | 13.0 (4) |
| N2—S1—C6—C7 | 145.2 (3) | N5—S2—C17—C18 | -109.4 (3) |
| O2—S1—C6—C11 | -161.2 (3) | C22—C17—C18—C19 | -2.3 (6) |
| O1—S1—C6—C11 | 74.7 (3) | S2—C17—C18—C19 | 178.8 (3) |
| N2—S1—C6—C11 | -38.2 (3) | C17—C18—C19—C20 | 3.1 (7) |
| C11—C6—C7—C8 | -0.4 (6) | C18—C19—C20—C21 | -2.4 (8) |
| S1—C6—C7—C8 | 176.5 (4) | C19—C20—C21—C22 | 1.1 (8) |
| C6—C7—C8—C9 | 2.5 (8) | C20—C21—C22—C17 | -0.6 (7) |
| C7—C8—C9—C10 | -2.7 (8) | C20—C21—C22—N6 | -178.2 (5) |
| C8—C9—C10—C11 | 0.8 (7) | C18—C17—C22—C21 | 1.1 (6) |
| C9—C10—C11—C6 | 1.3 (6) | S2—C17—C22—C21 | -180.0 (4) |
| C9—C10—C11—N3 | -177.5 (4) | C18—C17—C22—N6 | 178.6 (4) |
| C7—C6—C11—C10 | -1.5 (6) | S2—C17—C22—N6 | -2.5 (6) |
| S1—C6—C11—C10 | -178.3 (3) | O7—N6—C22—C21 | 98.9 (5) |
| C7—C6—C11—N3 | 177.2 (4) | O8—N6—C22—C21 | -78.3 (6) |
| S1—C6—C11—N3 | 0.5 (6) | O7—N6—C22—C17 | -78.8 (6) |
| O3—N3—C11—C10 | -72.4 (5) | O8—N6—C22—C17 | 104.0 (6) |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|------------------------------------|-------------|---------------|-----------------------|-------------------------|
| N4—H4 <i>A</i> ...O1 ⁱ | 0.95 (4) | 2.67 (4) | 3.215 (4) | 117 (3) |
| N4—H4 <i>A</i> ...N2 ⁱ | 0.95 (4) | 2.08 (4) | 2.969 (4) | 155 (3) |
| N1—H1 <i>A</i> ...N5 ⁱⁱ | 0.88 (5) | 2.02 (5) | 2.898 (4) | 172 (4) |
| C14—H14...O1 ⁱ | 0.93 | 2.54 | 3.140 (5) | 122 |
| C14—H14...O6 ⁱⁱⁱ | 0.93 | 2.58 | 3.249 (5) | 130 |
| C16—H16...O1 ^{iv} | 0.93 | 2.54 | 3.265 (5) | 135 |

Symmetry codes: (i) $x-1, y-1, z$; (ii) $x+1, y, z$; (iii) $x-1, y, z$; (iv) $x, y-1, z$.