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#### Key indicators

Single-crystal X-ray study  
 T = 123 K  
 Mean  $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$   
 Disorder in solvent or counterion  
 R factor = 0.064  
 wR factor = 0.174  
 Data-to-parameter ratio = 16.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## Hydrochlorothiazide *N*-methyl-2-pyrrolidone disolvate

Hydrochlorothiazide forms a 1:2 solvate with *N*-methylpyrrolidone (systematic name: 6-chloro-3,4-dihydro-2*H*-1,2,4-benzothiadiazine-7-sulfonamide 1,1-dioxide *N*-methyl-2-pyrrolidone disolvate),  $\text{C}_7\text{H}_8\text{ClN}_3\text{O}_4\text{S}_2 \cdot 2\text{C}_5\text{H}_9\text{NO}$ . The compound crystallizes with one hydrochlorothiazide and two solvent molecules, one of which is disordered, in the asymmetric unit. The crystal structure is isostructural with the previously reported hydrochlorothiazide *N,N*-dimethylacetamide disolvate.

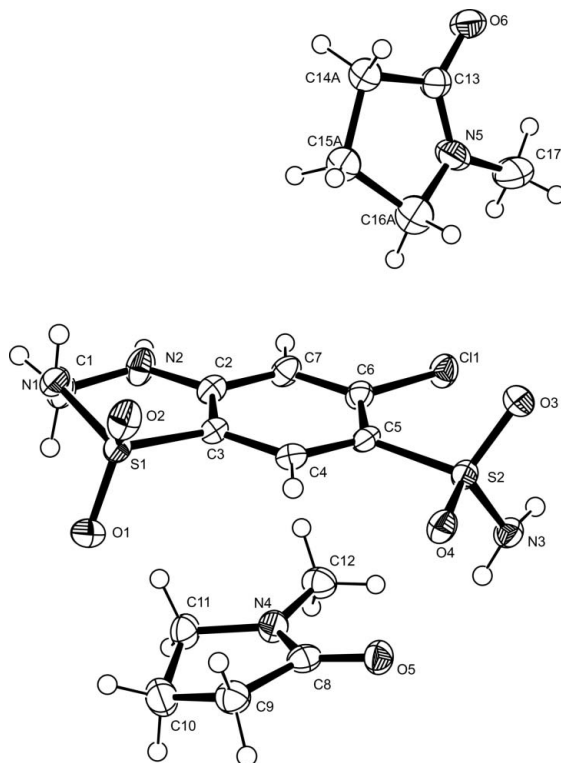
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#### Comment

Hydrochlorothiazide (HCT) is a thiazide diuretic which is known to crystallize in at least two non-solvated forms, form I (Dupont & Dideberg, 1972) and form II (Florence *et al.*, 2005). The title compound of this report, (I), was produced during an automated parallel crystallization study of HCT (Johnston, Florence, Shankland *et al.*, 2006). The sample was identified as a novel form using multi-sample foil transmission X-ray powder diffraction analysis (Florence *et al.*, 2003). Subsequent manual recrystallization from a saturated *N*-methylpyrrolidone (NMP) solution by slow evaporation at 298 K yielded samples of (I) suitable for single-crystal diffraction (Fig. 1).



It is notable that the crystal structure of (I) is isostructural with that of the previously reported HCT *N,N*-dimethylacetamide (DMA) disolvate (Johnston, Florence & Kennedy, 2006), with the same space group and very similar unit-cell parameters and packing arrangements. Adjacent HCT chains pack as layers in the *ab* plane and form an alternating stacked arrangement with layers of solvent molecules in the direction of the *c* axis (Fig. 2). The structures differ slightly in the extent of solvent disorder, with both solvent molecules disordered in the HCT–DMA disolvate, compared with a single molecule in (I). The structure also contains four  $\text{N}-\text{H} \cdots \text{O}$  hydrogen bonds, with N1, N2 and N3 of HCT donating contacts to adjacent O atoms of NMP (Table 1).


**Figure 1**

The asymmetric unit contents of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Minor disorder components have been omitted for clarity.

## Experimental

A single-crystal sample of the title compound was recrystallized from a saturated *N*-methylpyrrolidone solution by isothermal solvent evaporation at 298 K.

### Crystal data

$C_7H_8ClN_3O_4S_2 \cdot 2C_5H_9NO$   
 $M_r = 496.00$   
 Monoclinic,  $P2_1/c$   
 $a = 17.0756$  (6) Å  
 $b = 7.4819$  (3) Å  
 $c = 17.9978$  (6) Å  
 $\beta = 105.211$  (2)°  
 $V = 2218.81$  (14) Å<sup>3</sup>

$Z = 4$   
 $D_x = 1.485$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 $\mu = 0.41$  mm<sup>-1</sup>  
 $T = 123$  (2) K  
 Cut from prism, colourless  
 $0.32 \times 0.20 \times 0.12$  mm

### Data collection

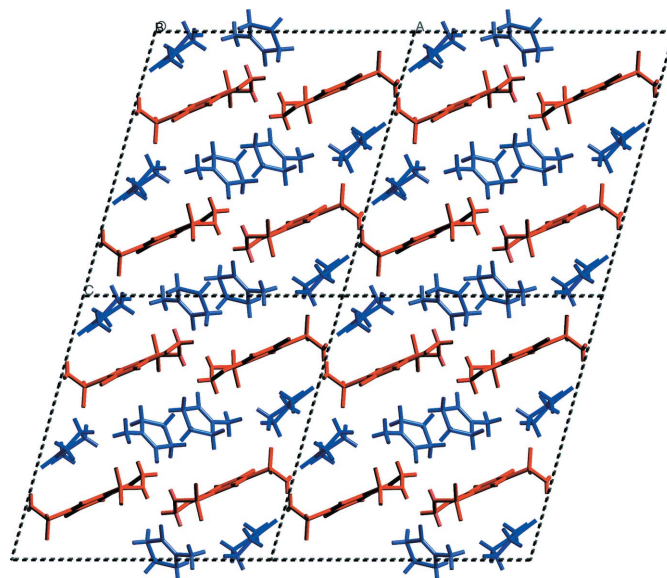
Nonius KappaCCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: none  
 24804 measured reflections

4852 independent reflections  
 3673 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.093$   
 $\theta_{max} = 27.1^\circ$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.064$   
 $wR(F^2) = 0.174$   
 $S = 1.08$   
 4852 reflections  
 296 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.1041P)^2 + 0.892P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} = 0.001$   
 $\Delta\rho_{max} = 0.70$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.55$  e Å<sup>-3</sup>


**Figure 2**

The crystal packing in the structure of (I), viewed down the *b* axis, showing the alternating layers of HCT and NMP molecules.

**Table 1**

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1N...O6 <sup>i</sup>	0.79 (5)	2.01 (5)	2.799 (4)	177 (5)
N2—H2N...O6 <sup>ii</sup>	0.80 (4)	2.35 (4)	2.929 (4)	130 (3)
N3—H3N...O5	0.84 (3)	2.12 (4)	2.891 (4)	153 (4)
N3—H4N...O5 <sup>iii</sup>	0.86 (4)	2.04 (4)	2.884 (4)	169 (3)
C1—H1A...O6 <sup>iv</sup>	0.99	2.58	3.075 (4)	111
C7—H7...O2 <sup>v</sup>	0.95	2.33	3.249 (4)	164
C11—H11B...O3 <sup>v</sup>	0.99	2.52	3.423 (4)	152

Symmetry codes: (i)  $-x + 1, -y + 1, -z$ ; (ii)  $x, -y + \frac{3}{2}, z + \frac{1}{2}$ ; (iii)  $-x + 2, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (iv)  $x, y + 1, z$ ; (v)  $x, -y + \frac{1}{2}, z + \frac{1}{2}$ .

Three C atoms (C14, C15 and C16) and the associated H atoms of one solvent molecule were treated as disordered over two sites. Isotropic refinement gave a refined occupancy of 0.52 (3):0.48 (3). All amine H atoms were found by difference synthesis and refined isotropically. All other H atoms were positioned geometrically at distances of 0.95 (CH), 0.98 (CH<sub>3</sub>) or 0.99 Å (CH<sub>2</sub>); a riding model was used during refinement, with  $U_{iso}(H) = 1.5U_{eq}(C)$  for methyl H atoms and  $U_{iso}(H) = 1.2U_{eq}(C)$  for all others.

Data collection: *COLLECT* (Nonius, 1988) and *DENZO* (Otwinowski & Minor, 1997); cell refinement: *DENZO* and *COLLECT*; data reduction: *DENZO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

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## References

Dupont, L. & Dideberg, O. (1972). *Acta Cryst.* **B28**, 2340–2347.

- Florence, A. J., Baumgartner, B., Weston, C., Shankland, N., Kennedy, A. R., Shankland, K. & David, W. I. F. (2003). *J. Pharm. Sci.* **92**, 1930–1938.
- Florence, A., Johnston, A., Fernandes, P., Shankland, K., Stevens, H. N. E., Osmunsden, S. & Mullen, A. B. (2005). *Acta Cryst.* **E61**, o2798–o2800.
- Johnson, C. K. (1976). *ORTEP II*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Johnston, A., Florence, A. J. & Kennedy, A. R. (2006). *Acta Cryst.* **E62**, o2926–o2928.
- Johnston, A., Florence, A. J., Shankland, N., Kennedy, A. R., Shankland, K. & Price, S. L. (2006). *Cryst. Growth Des.* Submitted.
- Nonius (1988). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography, Part A*, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.

## supporting information

*Acta Cryst.* (2006). E62, o5169–o5171 [https://doi.org/10.1107/S1600536806042838]

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6-chloro-3,4-dihydro-2*H*-1,2,4-benzothiadiazine-7-sulfonamide 1,1-dioxide *N*-methyl-2-pyrrolidone disolvate*Crystal data*

$C_7H_8ClN_3O_4S_2 \cdot 2C_5H_9NO$

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Monoclinic,  $P2_1/c$

Hall symbol:  $-P\ 2ybc$

$a = 17.0756$  (6) Å

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$Z = 4$

$F(000) = 1040$

$D_x = 1.485$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 5023 reflections

$\theta = 1.0$ – $27.1$ °

$\mu = 0.41$  mm<sup>-1</sup>

$T = 123$  K

Cut from prism, colourless

$0.32 \times 0.20 \times 0.12$  mm

*Data collection*

Nonius KappaCCD area-detector  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\varphi$  and  $\omega$  scans

24804 measured reflections

4852 independent reflections

3673 reflections with  $I > 2\sigma(I)$

$R_{int} = 0.093$

$\theta_{max} = 27.1$ °,  $\theta_{min} = 1.2$ °

$h = -21 \rightarrow 21$

$k = -9 \rightarrow 9$

$l = -22 \rightarrow 22$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.064$

$wR(F^2) = 0.174$

$S = 1.08$

4852 reflections

296 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites

H atoms treated by a mixture of independent  
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.1041P)^2 + 0.892P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} = 0.001$

$\Delta\rho_{max} = 0.70$  e Å<sup>-3</sup>

$\Delta\rho_{min} = -0.55$  e Å<sup>-3</sup>

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C11	0.83783 (4)	0.43550 (10)	0.16447 (4)	0.0287 (2)	
S1	0.65205 (4)	-0.18853 (10)	0.27397 (4)	0.0237 (2)	
S2	0.89965 (4)	0.01810 (10)	0.15514 (4)	0.0210 (2)	
O1	0.69144 (13)	-0.2471 (3)	0.35033 (12)	0.0372 (6)	
O2	0.63256 (13)	-0.3193 (3)	0.21369 (13)	0.0310 (5)	
O3	0.88310 (12)	0.0813 (3)	0.07722 (11)	0.0273 (5)	
O4	0.90847 (11)	-0.1695 (3)	0.16923 (11)	0.0247 (5)	
O5	0.98030 (12)	-0.0406 (3)	0.35009 (12)	0.0286 (5)	
O6	0.54188 (14)	1.0488 (3)	-0.13077 (13)	0.0361 (6)	
N1	0.56794 (15)	-0.0889 (3)	0.27559 (16)	0.0248 (5)	
N2	0.62120 (16)	0.2066 (4)	0.28470 (16)	0.0304 (6)	
N3	0.97927 (15)	0.1167 (4)	0.20314 (16)	0.0249 (5)	
N4	0.90110 (15)	0.0930 (4)	0.41855 (14)	0.0288 (6)	
N5	0.63002 (18)	0.9495 (4)	-0.02142 (15)	0.0378 (7)	
C1	0.58148 (19)	0.0752 (4)	0.32033 (18)	0.0289 (7)	
H1A	0.5288	0.1229	0.3247	0.035*	
H1B	0.6154	0.0494	0.3729	0.035*	
C2	0.68438 (17)	0.1647 (4)	0.25432 (16)	0.0238 (6)	
C3	0.70834 (16)	-0.0145 (4)	0.24753 (16)	0.0214 (6)	
C4	0.77418 (16)	-0.0536 (4)	0.21818 (16)	0.0219 (6)	
H4	0.7905	-0.1744	0.2158	0.026*	
C5	0.81629 (16)	0.0808 (4)	0.19233 (16)	0.0210 (6)	
C6	0.79023 (16)	0.2571 (4)	0.19616 (16)	0.0229 (6)	
C7	0.72558 (17)	0.2987 (4)	0.22588 (17)	0.0256 (6)	
H7	0.7090	0.4197	0.2270	0.031*	
C8	0.93403 (18)	-0.0472 (4)	0.39273 (17)	0.0264 (7)	
C9	0.90442 (19)	-0.2160 (4)	0.42372 (18)	0.0309 (7)	
H9A	0.9502	-0.2975	0.4461	0.037*	
H9B	0.8643	-0.2798	0.3826	0.037*	
C10	0.8654 (2)	-0.1475 (5)	0.48563 (19)	0.0368 (8)	
H10A	0.8162	-0.2174	0.4857	0.044*	
H10B	0.9040	-0.1550	0.5373	0.044*	
C11	0.84395 (19)	0.0454 (4)	0.46353 (18)	0.0314 (7)	
H11A	0.7871	0.0562	0.4322	0.038*	
H11B	0.8518	0.1220	0.5097	0.038*	
C12	0.91140 (19)	0.2762 (4)	0.39683 (18)	0.0320 (7)	
H12A	0.9503	0.2801	0.3654	0.048*	
H12B	0.9317	0.3485	0.4433	0.048*	
H12C	0.8591	0.3237	0.3671	0.048*	
C13	0.56205 (19)	0.9407 (4)	-0.07679 (18)	0.0295 (7)	

C14A	0.5044 (7)	0.7798 (12)	-0.0695 (6)	0.032 (3)*	0.52 (3)
H14A	0.4508	0.8221	-0.0653	0.039*	0.52 (3)
H14B	0.4969	0.6957	-0.1132	0.039*	0.52 (3)
C15A	0.5546 (8)	0.6956 (12)	0.0066 (5)	0.035 (2)*	0.52 (3)
H15A	0.5568	0.5641	0.0011	0.042*	0.52 (3)
H15B	0.5292	0.7227	0.0488	0.042*	0.52 (3)
C16A	0.6444 (5)	0.7776 (18)	0.0258 (6)	0.045 (3)*	0.52 (3)
H16A	0.6656	0.8027	0.0814	0.054*	0.52 (3)
H16B	0.6823	0.6969	0.0089	0.054*	0.52 (3)
C14B	0.5226 (8)	0.7851 (14)	-0.0571 (7)	0.033 (3)*	0.48 (3)
H14C	0.5107	0.6995	-0.1006	0.040*	0.48 (3)
H14D	0.4707	0.8193	-0.0461	0.040*	0.48 (3)
C15B	0.5793 (10)	0.6982 (14)	0.0145 (6)	0.045 (3)*	0.48 (3)
H15C	0.5501	0.6709	0.0539	0.055*	0.48 (3)
H15D	0.6037	0.5868	0.0008	0.055*	0.48 (3)
C16B	0.6381 (6)	0.829 (2)	0.0403 (7)	0.048 (3)*	0.48 (3)
H16C	0.6930	0.7750	0.0543	0.058*	0.48 (3)
H16D	0.6292	0.8906	0.0860	0.058*	0.48 (3)
C17	0.6911 (2)	1.0796 (6)	-0.0191 (2)	0.0549 (11)	
H17A	0.7374	1.0236	-0.0326	0.082*	
H17B	0.7089	1.1299	0.0329	0.082*	
H17C	0.6692	1.1751	-0.0559	0.082*	
H1N	0.538 (3)	-0.080 (6)	0.234 (3)	0.061 (14)*	
H2N	0.613 (2)	0.310 (5)	0.290 (2)	0.035 (10)*	
H3N	0.994 (2)	0.092 (5)	0.250 (2)	0.044 (11)*	
H4N	0.986 (2)	0.224 (5)	0.189 (2)	0.032 (9)*	

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0261 (4)	0.0232 (4)	0.0392 (4)	-0.0011 (3)	0.0127 (3)	0.0036 (3)
S1	0.0181 (4)	0.0252 (4)	0.0282 (4)	0.0008 (3)	0.0066 (3)	0.0044 (3)
S2	0.0179 (4)	0.0233 (4)	0.0224 (4)	0.0000 (3)	0.0066 (3)	-0.0006 (3)
O1	0.0254 (11)	0.0485 (14)	0.0340 (12)	-0.0007 (11)	0.0010 (9)	0.0174 (11)
O2	0.0282 (11)	0.0253 (11)	0.0422 (12)	-0.0012 (9)	0.0139 (10)	-0.0033 (9)
O3	0.0282 (11)	0.0337 (12)	0.0205 (10)	0.0016 (9)	0.0073 (9)	0.0022 (9)
O4	0.0213 (10)	0.0242 (11)	0.0303 (11)	0.0019 (8)	0.0097 (9)	-0.0018 (9)
O5	0.0278 (11)	0.0301 (11)	0.0312 (11)	0.0022 (9)	0.0133 (10)	0.0012 (9)
O6	0.0296 (12)	0.0368 (13)	0.0369 (13)	-0.0007 (10)	-0.0003 (10)	0.0083 (10)
N1	0.0177 (12)	0.0294 (14)	0.0269 (13)	0.0008 (10)	0.0053 (11)	0.0001 (11)
N2	0.0270 (14)	0.0262 (15)	0.0425 (16)	0.0012 (11)	0.0171 (12)	-0.0038 (12)
N3	0.0198 (12)	0.0260 (14)	0.0283 (14)	-0.0016 (11)	0.0054 (11)	0.0013 (12)
N4	0.0261 (13)	0.0316 (14)	0.0295 (14)	-0.0009 (11)	0.0086 (11)	0.0002 (11)
N5	0.0433 (17)	0.0437 (17)	0.0254 (14)	0.0050 (13)	0.0071 (13)	0.0052 (12)
C1	0.0277 (15)	0.0299 (16)	0.0321 (16)	-0.0015 (13)	0.0131 (13)	-0.0050 (13)
C2	0.0195 (14)	0.0259 (15)	0.0258 (14)	0.0014 (12)	0.0053 (11)	-0.0024 (12)
C7	0.0211 (14)	0.0210 (14)	0.0345 (16)	0.0022 (12)	0.0072 (12)	-0.0014 (12)
C6	0.0194 (13)	0.0222 (14)	0.0263 (14)	-0.0019 (12)	0.0049 (11)	0.0010 (12)

C5	0.0142 (13)	0.0253 (14)	0.0220 (14)	0.0015 (11)	0.0021 (11)	-0.0014 (11)
C4	0.0200 (14)	0.0214 (14)	0.0224 (14)	0.0023 (11)	0.0022 (11)	0.0009 (11)
C3	0.0162 (13)	0.0245 (14)	0.0224 (14)	0.0007 (11)	0.0032 (11)	0.0014 (11)
C8	0.0231 (15)	0.0296 (16)	0.0228 (14)	0.0004 (12)	-0.0004 (12)	-0.0029 (12)
C9	0.0304 (17)	0.0294 (16)	0.0312 (16)	-0.0015 (13)	0.0049 (13)	0.0010 (13)
C10	0.0380 (19)	0.0414 (19)	0.0326 (17)	-0.0058 (16)	0.0122 (15)	0.0032 (15)
C11	0.0298 (16)	0.0400 (18)	0.0276 (16)	0.0007 (14)	0.0134 (14)	-0.0020 (14)
C12	0.0336 (17)	0.0277 (16)	0.0326 (16)	-0.0013 (14)	0.0047 (14)	0.0022 (13)
C13	0.0286 (16)	0.0328 (17)	0.0278 (16)	0.0057 (13)	0.0086 (14)	-0.0026 (13)
C17	0.037 (2)	0.074 (3)	0.045 (2)	-0.010 (2)	-0.0049 (17)	0.009 (2)

*Geometric parameters (Å, °)*

C11—C6	1.735 (3)	C4—H4	0.9500
S1—O1	1.431 (2)	C8—C9	1.520 (4)
S1—O2	1.434 (2)	C9—C10	1.529 (4)
S1—N1	1.625 (3)	C9—H9A	0.9900
S1—C3	1.756 (3)	C9—H9B	0.9900
S2—O4	1.427 (2)	C10—C11	1.516 (5)
S2—O3	1.436 (2)	C10—H10A	0.9900
S2—N3	1.588 (3)	C10—H10B	0.9900
S2—C5	1.787 (3)	C11—H11A	0.9900
O5—C8	1.239 (4)	C11—H11B	0.9900
O6—C13	1.241 (4)	C12—H12A	0.9800
N1—C1	1.453 (4)	C12—H12B	0.9800
N1—H1N	0.79 (4)	C12—H12C	0.9800
N2—C2	1.368 (4)	C13—C14B	1.435 (11)
N2—C1	1.437 (4)	C13—C14A	1.582 (11)
N2—H2N	0.79 (4)	C14A—C15A	1.547 (12)
N3—H3N	0.84 (4)	C14A—H14A	0.9900
N3—H4N	0.86 (4)	C14A—H14B	0.9900
N4—C8	1.330 (4)	C15A—C16A	1.602 (14)
N4—C12	1.448 (4)	C15A—H15A	0.9900
N4—C11	1.466 (4)	C15A—H15B	0.9900
N5—C13	1.318 (4)	C16A—H16A	0.9900
N5—C16B	1.411 (10)	C16A—H16B	0.9900
N5—C17	1.419 (5)	C14B—C15B	1.538 (14)
N5—C16A	1.525 (10)	C14B—H14C	0.9900
C1—H1A	0.9900	C14B—H14D	0.9900
C1—H1B	0.9900	C15B—C16B	1.390 (14)
C2—C7	1.397 (4)	C15B—H15C	0.9900
C2—C3	1.416 (4)	C15B—H15D	0.9900
C7—C6	1.382 (4)	C16B—H16C	0.9900
C7—H7	0.9500	C16B—H16D	0.9900
C6—C5	1.400 (4)	C17—H17A	0.9800
C5—C4	1.386 (4)	C17—H17B	0.9800
C4—C3	1.393 (4)	C17—H17C	0.9800

O1—S1—O2	118.35 (14)	C9—C10—H10A	110.9
O1—S1—N1	108.55 (14)	C11—C10—H10B	110.9
O2—S1—N1	107.21 (13)	C9—C10—H10B	110.9
O1—S1—C3	109.54 (14)	H10A—C10—H10B	108.9
O2—S1—C3	109.62 (13)	N4—C11—C10	103.1 (2)
N1—S1—C3	102.34 (13)	N4—C11—H11A	111.2
O4—S2—O3	118.80 (12)	C10—C11—H11A	111.2
O4—S2—N3	108.95 (14)	N4—C11—H11B	111.2
O3—S2—N3	107.43 (14)	C10—C11—H11B	111.2
O4—S2—C5	104.41 (12)	H11A—C11—H11B	109.1
O3—S2—C5	107.89 (13)	N4—C12—H12A	109.5
N3—S2—C5	109.06 (13)	N4—C12—H12B	109.5
C1—N1—S1	112.6 (2)	H12A—C12—H12B	109.5
C1—N1—H1N	115 (3)	N4—C12—H12C	109.5
S1—N1—H1N	112 (3)	H12A—C12—H12C	109.5
C2—N2—C1	122.4 (3)	H12B—C12—H12C	109.5
C2—N2—H2N	116 (3)	O6—C13—N5	124.6 (3)
C1—N2—H2N	120 (3)	O6—C13—C14B	132.1 (6)
S2—N3—H3N	115 (3)	N5—C13—C14B	103.4 (6)
S2—N3—H4N	116 (2)	O6—C13—C14A	120.7 (5)
H3N—N3—H4N	118 (4)	N5—C13—C14A	114.8 (5)
C8—N4—C12	124.0 (3)	C15A—C14A—C13	99.9 (7)
C8—N4—C11	113.9 (3)	C15A—C14A—H14A	111.8
C12—N4—C11	121.6 (3)	C13—C14A—H14A	111.8
C13—N5—C16B	116.6 (5)	C15A—C14A—H14B	111.8
C13—N5—C17	123.5 (3)	C13—C14A—H14B	111.8
C16B—N5—C17	119.8 (5)	H14A—C14A—H14B	109.5
C13—N5—C16A	111.1 (4)	C14A—C15A—C16A	108.1 (7)
C17—N5—C16A	123.5 (4)	C14A—C15A—H15A	110.1
N2—C1—N1	110.8 (2)	C16A—C15A—H15A	110.1
N2—C1—H1A	109.5	C14A—C15A—H15B	110.1
N1—C1—H1A	109.5	C16A—C15A—H15B	110.1
N2—C1—H1B	109.5	H15A—C15A—H15B	108.4
N1—C1—H1B	109.5	N5—C16A—C15A	101.5 (6)
H1A—C1—H1B	108.1	N5—C16A—H16A	111.5
N2—C2—C7	120.5 (3)	C15A—C16A—H16A	111.5
N2—C2—C3	121.8 (3)	N5—C16A—H16B	111.5
C7—C2—C3	117.6 (2)	C15A—C16A—H16B	111.5
C6—C7—C2	120.7 (3)	H16A—C16A—H16B	109.3
C6—C7—H7	119.7	C13—C14B—C15B	108.5 (8)
C2—C7—H7	119.7	C13—C14B—H14C	110.0
C7—C6—C5	121.8 (3)	C15B—C14B—H14C	110.0
C7—C6—C11	116.4 (2)	C13—C14B—H14D	110.0
C5—C6—C11	121.8 (2)	C15B—C14B—H14D	110.0
C4—C5—C6	117.9 (2)	H14C—C14B—H14D	108.4
C4—C5—S2	118.0 (2)	C16B—C15B—C14B	102.9 (9)
C6—C5—S2	124.1 (2)	C16B—C15B—H15C	111.2
C5—C4—C3	121.1 (3)	C14B—C15B—H15C	111.2



C5—C4—H4	119.4	C16B—C15B—H15D	111.2
C3—C4—H4	119.4	C14B—C15B—H15D	111.2
C4—C3—C2	120.7 (3)	H15C—C15B—H15D	109.1
C4—C3—S1	120.1 (2)	C15B—C16B—N5	105.9 (8)
C2—C3—S1	119.2 (2)	C15B—C16B—H16C	110.6
O5—C8—N4	125.6 (3)	N5—C16B—H16C	110.6
O5—C8—C9	126.0 (3)	C15B—C16B—H16D	110.6
N4—C8—C9	108.4 (3)	N5—C16B—H16D	110.6
C8—C9—C10	103.8 (3)	H16C—C16B—H16D	108.7
C8—C9—H9A	111.0	N5—C17—H17A	109.5
C10—C9—H9A	111.0	N5—C17—H17B	109.5
C8—C9—H9B	111.0	H17A—C17—H17B	109.5
C10—C9—H9B	111.0	N5—C17—H17C	109.5
H9A—C9—H9B	109.0	H17A—C17—H17C	109.5
C11—C10—C9	104.5 (2)	H17B—C17—H17C	109.5
C11—C10—H10A	110.9		
O1—S1—N1—C1	65.6 (2)	C12—N4—C8—O5	3.5 (5)
O2—S1—N1—C1	-165.5 (2)	C11—N4—C8—O5	175.8 (3)
C3—S1—N1—C1	-50.2 (2)	C12—N4—C8—C9	-175.9 (3)
C2—N2—C1—N1	-42.9 (4)	C11—N4—C8—C9	-3.6 (3)
S1—N1—C1—N2	65.8 (3)	O5—C8—C9—C10	168.1 (3)
C1—N2—C2—C7	-174.7 (3)	N4—C8—C9—C10	-12.6 (3)
C1—N2—C2—C3	7.9 (5)	C8—C9—C10—C11	22.9 (3)
N2—C2—C7—C6	179.0 (3)	C8—N4—C11—C10	18.3 (3)
C3—C2—C7—C6	-3.4 (4)	C12—N4—C11—C10	-169.2 (3)
C2—C7—C6—C5	0.9 (4)	C9—C10—C11—N4	-24.6 (3)
C2—C7—C6—C11	-178.7 (2)	C16B—N5—C13—O6	172.2 (9)
C7—C6—C5—C4	1.1 (4)	C17—N5—C13—O6	-3.6 (5)
C11—C6—C5—C4	-179.3 (2)	C16A—N5—C13—O6	-168.2 (6)
C7—C6—C5—S2	-179.6 (2)	C16B—N5—C13—C14B	-7.4 (10)
C11—C6—C5—S2	-0.1 (4)	C17—N5—C13—C14B	176.8 (5)
O4—S2—C5—C4	-6.0 (2)	C16A—N5—C13—C14B	12.2 (8)
O3—S2—C5—C4	121.3 (2)	C16B—N5—C13—C14A	-6.7 (10)
N3—S2—C5—C4	-122.4 (2)	C17—N5—C13—C14A	177.5 (5)
O4—S2—C5—C6	174.8 (2)	C16A—N5—C13—C14A	12.9 (8)
O3—S2—C5—C6	-58.0 (3)	O6—C13—C14A—C15A	-177.5 (5)
N3—S2—C5—C6	58.4 (3)	N5—C13—C14A—C15A	1.4 (7)
C6—C5—C4—C3	-0.4 (4)	C14B—C13—C14A—C15A	5 (3)
S2—C5—C4—C3	-179.7 (2)	C13—C14A—C15A—C16A	-14.2 (10)
C5—C4—C3—C2	-2.3 (4)	C13—N5—C16A—C15A	-20.6 (10)
C5—C4—C3—S1	175.8 (2)	C17—N5—C16A—C15A	174.8 (5)
N2—C2—C3—C4	-178.4 (3)	C14A—C15A—C16A—N5	21.0 (11)
C7—C2—C3—C4	4.2 (4)	O6—C13—C14B—C15B	177.0 (6)
N2—C2—C3—S1	3.5 (4)	N5—C13—C14B—C15B	-3.5 (8)
C7—C2—C3—S1	-173.9 (2)	C14A—C13—C14B—C15B	180 (4)
O1—S1—C3—C4	83.6 (3)	C13—C14B—C15B—C16B	12.5 (14)
O2—S1—C3—C4	-47.8 (3)	C14B—C15B—C16B—N5	-15.9 (16)

N1—S1—C3—C4	-161.3 (2)	C13—N5—C16B—C15B	16.2 (17)
O1—S1—C3—C2	-98.2 (3)	C17—N5—C16B—C15B	-167.9 (10)
O2—S1—C3—C2	130.4 (2)	C16A—N5—C16B—C15B	-61 (2)
N1—S1—C3—C2	16.8 (3)		

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N1—H1N $\cdots$ O6 <sup>i</sup>	0.79 (5)	2.01 (5)	2.799 (4)	177 (5)
N2—H2N $\cdots$ O6 <sup>ii</sup>	0.80 (4)	2.35 (4)	2.929 (4)	130 (3)
N3—H3N $\cdots$ O5	0.84 (3)	2.12 (4)	2.891 (4)	153 (4)
N3—H4N $\cdots$ O5 <sup>iii</sup>	0.86 (4)	2.04 (4)	2.884 (4)	169 (3)
C1—H1A $\cdots$ O6 <sup>ii</sup>	0.99	2.58	3.075 (4)	111
C7—H7 $\cdots$ O2 <sup>iv</sup>	0.95	2.33	3.249 (4)	164
C11—H11B $\cdots$ O3 <sup>v</sup>	0.99	2.52	3.423 (4)	152

Symmetry codes: (i)  $-x+1, -y+1, -z$ ; (ii)  $x, -y+3/2, z+1/2$ ; (iii)  $-x+2, y+1/2, -z+1/2$ ; (iv)  $x, y+1, z$ ; (v)  $x, -y+1/2, z+1/2$ .