

# 3,3'-[(2-Bromo-1,3-phenylene)bis(methylene)]-bis(1-butyl-2,3-dihydro-1*H*-imidazole-2-selone)

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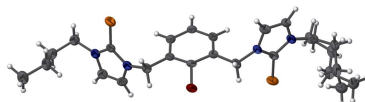
Keywords: crystal structure; bis(selone); imidazole-2-selone; Br...Br intermolecular interactions; C—H...Se interactions.

CCDC reference: 1535946

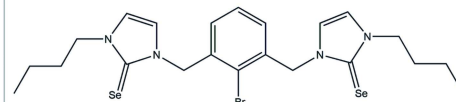
Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

In the title compound, C<sub>22</sub>H<sub>29</sub>BrN<sub>4</sub>Se<sub>2</sub>, the two Se atoms are directed in opposite directions with respect to the central benzene ring. The C=Se bond lengths at 1.848 (5) and 1.851 (5) Å are on the long side for a double bond but shorter than expected for a C—Se single bond. In the crystal, Br...Br intermolecular interactions [3.4685 (12) Å] link the molecules into a zigzag chain propagating along the *b*-axis direction. In addition, there are C—H...Se intermolecular interactions present, linking the chains to form slabs parallel to the *ab* plane. One of the two butyl side chains is disordered over two conformations with occupancies of 0.777 (9) and 0.223 (9).

## 3D view



## Chemical scheme

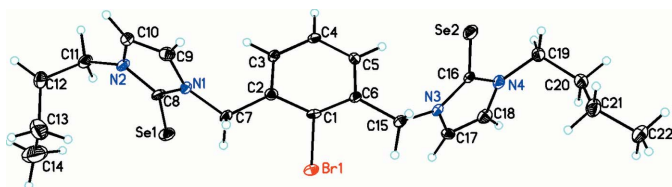


## Structure description

Selones, the Se-analogues of ketones, have generated considerable interest due to their promising utility in many domains, including homogeneous catalysis (see for example: Prabhu *et al.*, 2014; Babu *et al.*, 2016; Sharma *et al.*, 2017), photocatalysis (Jin *et al.*, 2013), and biological fields (see for example: Roy *et al.*, 2013; Palmer & Parkin, 2015; Banerjee *et al.*, 2015).

Recently, our group has reported the syntheses and crystal structures of [2 + 2] binuclear palladium(II) and platinum(II) self-assembled 24-membered metallomacrocycles resulting from the reaction Pd(COD)Cl<sub>2</sub> and Pt(COD)Cl<sub>2</sub> with 3,3'-[(2-bromo-1,3-phenylene)bis(methylene)]bis(1-mesityl-1,3-dihydro-2*H*-imidazole-2-selone), and of a mononuclear gold(III) complex formed with AuCl(SMe<sub>2</sub>) (Rani *et al.*, 2017). Herein, we report on the synthesis and crystal structure of the related title compound, 3,3'-[(2-bromo-1,3-phenylene)bis(methylene)]bis(1-butyl-2,3-dihydro-1*H*-imidazole-2-selone).

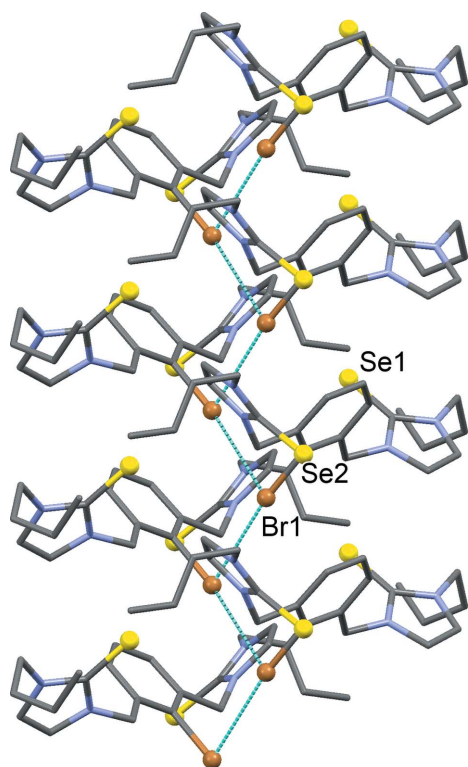
In the title compound, Fig. 1, one of the two butyl side chains (atoms C11–C14) is disordered over two conformations with occupancies of 0.777 (9)/0.223 (9). The molecule



**Figure 1**

The molecular structure of the title compound, with the atom labelling. Displacement ellipsoids are drawn at the 30% probability level. For the disordered *n*-butyl substituent (atoms C11–C14) only the major component is shown.

adopts a conformation in which the two Se atoms are directed in opposite directions with respect to the central benzene ring. Atom Br1 deviates from the plane of the benzene ring by 0.044 (6) Å, and the Se atoms deviate from the plane of the imidazole ring to which they are attached by 0.041 (7) Å for atom Se1 and 0.044 (7) Å for atom Se2. Each imidazole ring is almost perpendicular to the central benzene ring with dihedral angles of 87.9 (1) and 81.2 (1)° for rings N1/N2/C8–C10 and N3/N4/C16–C18, respectively, and they are inclined to one another by 57.0 (3)°. The C=Se bond lengths, Se1–C8 = 1.848 (5) and Se2–C16 = 1.851 (5) Å, are on the long side for a double bond but shorter than expected for a C–Se single bond (Murai *et al.*, 1995). The butyl side chains do not have extended conformations as indicated by their internal N–C–C and C–C–C–CH<sub>3</sub> torsion angles, which are N2–C11A–C12A–C13A = 71.2 (14) ° and C11A–C12A–



**Figure 2**

Diagram showing the Br...Br intermolecular interactions (as dashed lines), linking the molecules into zigzag chains propagating along the *b* axis direction. For the disordered *n*-butyl substituent only the major component is shown, and H atoms have been omitted for clarity.

**Table 1**

Hydrogen-bond geometry (Å, °).

| <i>D</i> –H... <i>A</i>      | <i>D</i> –H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> –H... <i>A</i> |
|------------------------------|-------------|---------------|-----------------------|-------------------------|
| C10–H10A...Se2 <sup>i</sup>  | 0.95        | 2.97          | 3.823 (5)             | 150                     |
| C12A–H12B...Se2 <sup>i</sup> | 0.99        | 2.79          | 3.590 (7)             | 139                     |
| C19–H19A...Se2 <sup>ii</sup> | 0.99        | 2.94          | 3.708 (5)             | 136                     |

Symmetry codes: (i)  $-x + \frac{1}{2}, -y + \frac{1}{2}, -z + \frac{1}{2}$ ; (ii)  $x, y - 1, z$ .

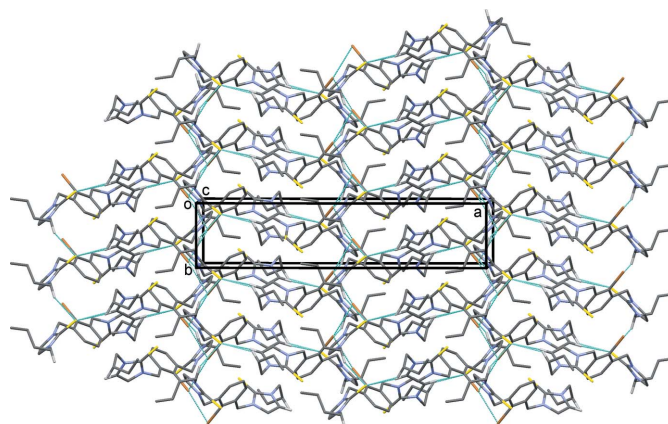
C13A–C14A = 66.6 (12) ° (major component), and N4–C19–C20–C21 = 58.1 (7) ° and C19–C20–C21–C22 = 178.1 (5) °.

In the crystal, molecules are linked by Br...Br<sup>iii</sup> [= 3.4685 (12) Å], intermolecular interactions forming zigzag chains propagating along the *b*-axis direction (Fig. 2; symmetry codes: (i)  $-x, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (ii)  $-x, y + \frac{1}{2}, -z + \frac{1}{2}$ ). In addition, there are C–H...Se2 intermolecular interactions present (Table 1). The latter link the chains to form slabs parallel to the *ab* plane (Fig. 3).

A search of the Cambridge Structural Database for analogous structures gave only one hit (QUNCOT: Ghavale *et al.*, 2015) where, instead of an imidazole-type moiety, benzimidazole moieties are present.

### Synthesis and crystallization

To a solution of 1,1'-[(2-bromo-1,3-phenylene)bis(methylene)]bis(3-mesityl-1*H*-imidazol-3-ium) dibromide (1.5 g, 2.53 mmol) in dry methanol (50 ml) was added Se powder (0.46 g, 5.82 mmol) followed by K<sub>2</sub>CO<sub>3</sub> (0.874 g, 6.33 mmol). The reaction mixture was refluxed for 48 h. A white coloured precipitate deposited near the walls. The reaction mixture was filtered through a Whatman filter paper to remove unconsumed selenium powder. The solvent was evaporated and the residue was redissolved in CHCl<sub>3</sub> to remove the leftover salt, which is insoluble in CHCl<sub>3</sub>. The solvent was evaporated and



**Figure 3**

A view along the *c* axis of the crystal packing of the title compound. The Br...Br and C–H...Se interactions are shown as dashed lines (see Table 1). For the disordered *n*-butyl substituent only the major component is shown, and, for clarity, only the H atoms involved in the C–H...Se interactions have been included.

**Table 2**  
Experimental details.

|   |  |
|---|--|
| Crystal data  |  |
| Chemical formula  | C <sub>22</sub> H <sub>29</sub> BrN <sub>4</sub> Se <sub>2</sub> |
| <i>M<sub>r</sub></i>  | 587.32   |
| Crystal system, space group   | Monoclinic, <i>I2/a</i>  |
| Temperature (K)   | 100  |
| <i>a</i> , <i>b</i> , <i>c</i> (Å)  | 23.767 (3), 5.1425 (3), 39.806 (10)                              |
| $\beta$ (°)   | 102.61 (2)   |
| <i>V</i> (Å <sup>3</sup> )  | 4747.8 (14)  |
| <i>Z</i>  | 8  |
| Radiation type  | Mo <i>K</i> $\alpha$   |
| $\mu$ (mm <sup>-1</sup> )   | 4.81   |
| Crystal size (mm)   | 0.23 × 0.17 × 0.09   |
| Data collection   |  |
| Diffractometer  | Rigaku CCD   |
| Absorption correction   | Multi-scan ( <i>CrysAlis PRO</i> ; Rigaku OD, 2015)              |
| <i>T<sub>min</sub></i> , <i>T<sub>max</sub></i>   | 0.581, 1.000   |
| No. of measured, independent and observed [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )] reflections                             | 34850, 7195, 4099  |
| <i>R<sub>int</sub></i>  | 0.113  |
| ( <i>sin</i> $\theta$ / $\lambda$ ) <sub>max</sub> (Å <sup>-1</sup> )   | 0.726  |
| Refinement  |  |
| <i>R</i> [ <i>F</i> <sup>2</sup> > 2 $\sigma$ ( <i>F</i> <sup>2</sup> )], <i>wR</i> ( <i>F</i> <sup>2</sup> ), <i>S</i> | 0.062, 0.146, 1.02   |
| No. of reflections  | 7195   |
| No. of parameters   | 302  |
| No. of restraints   | 311  |
| H-atom treatment  | H-atom parameters constrained                                    |
| $\Delta\rho_{\max}$ , $\Delta\rho_{\min}$ (e Å <sup>-3</sup> )  | 0.65, -1.00  |

Computer programs: *CrysAlis PRO* (Rigaku OD, 2015), *SHELXT* (Sheldrick, 2015a), *SHELXL2017* (Sheldrick, 2015b), *SHELXTL* (Sheldrick, 2008), *Mercury* (Macrae et al., 2008) and *PLATON* (Spek, 2009).

the sample was dried under *vacuo* (yield 1.20 g, 81%; m.p. 436 K). Colourless prismatic crystals were obtained by slow evaporation of a chloroform solution of the compound at room temperature.

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, 298 K):  $\delta$  7.24 (*t*, *J* = 7.5 Hz, 1H), 7.09–7.07 (*d*, *J* = 7.5 Hz, 2H), 6.88 (*d*, *J* = 2.0 Hz, 2H), 6.81 (*d*, *J* = 2.0 Hz, 2H), 5.51 (*s*, 4H), 4.18 (*t*, *J* = 7.5 Hz, 4H), 1.82–1.79 (*m*, 4H), 1.42–1.37 (*m*, 4H), 0.98 (*t*, *J* = 7.0 Hz, 6H). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>, 298 K): 156.8, 136.2, 129.8, 128.3, 124.3, 119.3, 118.8, 53.4, 49.9, 31.3, 19.9, 13.8. <sup>77</sup>Se NMR (95.4 MHz, CDCl<sub>3</sub>, 298 K):  $\delta$  0.5 p.p.m. Analysis calculated for C<sub>22</sub>H<sub>31</sub>N<sub>4</sub>BrSe<sub>2</sub> (587.3260): C 44.84, H 5.13, N 9.51. found C 45.03, H 5.01, N 8.31. ESI-MS: *m/z* calculated 626.9542; found 626.9500 [*M* + *K*]<sup>+</sup>. FT-IR (KBr, cm<sup>-1</sup>): 3158(*w*), 3077(*m*), 2957(*s*), 2931(*s*), 2871(*m*), 1677(*w*), 1566(*m*), 1458(*s*), 1407(*s*), 1357(*w*), 1284(*w*), 1236(*s*), 1218(*s*), 1181(*m*), 1132(*m*), 1055(*m*), 1025(*m*, C=Se), 761(*m*), 714(*m*), 668(*w*).

## Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2. One of the two butyl side chains is disordered over two conformations (C11A–C14A/C11B–C14B) with a refined occupancy ratio of 0.777 (9): 0.223 (9) and both were constrained to have similar metrical parameters.

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## References

- Banerjee, M., Karri, R., Rawat, K. S., Muthuvel, K., Pathak, B. & Roy, G. (2015). *Angew. Chem. Int. Ed.* **54**, 9323–9327.
- Ghavale, N., Manjare, S. T., Singh, H. B. & Butcher, R. J. (2015). *Dalton Trans.* **44**, 11893–11900.
- Jin, J., Shin, H.-W., Park, J. H., Park, J. H., Kim, E., Ahn, T. K., Ryu, D. H. & Son, S. U. (2013). *Organometallics*, **32**, 3954–3959.
- Macrae, C. F., Bruno, I. J., Chisholm, J. A., Edgington, P. R., McCabe, P., Pidcock, E., Rodriguez-Monge, L., Taylor, R., van de Streek, J. & Wood, P. A. (2008). *J. Appl. Cryst.* **41**, 466–470.
- Murai, T., Mizutani, T., Kanda, T. & Kato, S. (1995). *Heteroat. Chem.* **6**, 241–246.
- Babu, C. N., Srinivas, K. & Prabusankar, G. (2016). *Dalton Trans.* **45**, 6456–6465.
- Palmer, J. H. & Parkin, G. (2015). *J. Am. Chem. Soc.* **137**, 4503–4516.
- Prabhu, P., Singh, B. G., Noguchi, M., Phadnis, P. P., Jain, V. K., Iwaoka, M. & Priyadarsini, K. I. (2014). *Org. Biomol. Chem.* **12**, 2404–2412.
- Rani, V., Singh, H. B. & Butcher, R. J. (2017). *Eur. J. Inorg. Chem.* pp. 3720–3728.
- Rigaku OD (2015). *CrysAlis PRO*. Rigaku Oxford Diffraction, Woodlands, Texas, USA.
- Roy, G., Jayaram, P. N. & Mughes, G. (2013). *Chem. Asian J.* **8**, 1910–1921.
- Sharma, A. K., Joshi, H., Bhaskar, R. & Singh, A. K. (2017). *Dalton Trans.* **46**, 2228–2237.
- Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
- Sheldrick, G. M. (2015a). *Acta Cryst. A* **71**, 3–8.
- Sheldrick, G. M. (2015b). *Acta Cryst. C* **71**, 3–8.
- Spek, A. L. (2009). *Acta Cryst. D* **65**, 148–155.

## full crystallographic data

*IUCrData* (2017). 2, x171746 [https://doi.org/10.1107/S2414314617017461]

### 3,3'-[(2-Bromo-1,3-phenylene)bis(methylene)]bis(1-butyl-2,3-dihydro-1*H*-imidazole-2-selone)

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3,3'-[(2-Bromo-1,3-phenylene)bis(methylene)]bis(1-butyl-2,3-dihydro-1*H*-imidazole-2-selone)

#### Crystal data

C<sub>22</sub>H<sub>29</sub>BrN<sub>4</sub>Se<sub>2</sub>

*M<sub>r</sub>* = 587.32

Monoclinic, *I*2/*a*

*a* = 23.767 (3) Å

*b* = 5.1425 (3) Å

*c* = 39.806 (10) Å

$\beta$  = 102.61 (2)°

*V* = 4747.8 (14) Å<sup>3</sup>

*Z* = 8

*F*(000) = 2336

*D<sub>x</sub>* = 1.643 Mg m<sup>-3</sup>

Mo *K*α radiation,  $\lambda$  = 0.71073 Å

Cell parameters from 7207 reflections

$\theta$  = 2.2–28.9°

$\mu$  = 4.81 mm<sup>-1</sup>

*T* = 100 K

Prism, colorless

0.23 × 0.17 × 0.09 mm

#### Data collection

Rigaku CCD

diffractometer

Radiation source: fine-focus sealed X-ray tube

$\omega$  scans

Absorption correction: multi-scan

(CrysAlis PRO; Rigaku OD, 2015)

*T<sub>min</sub>* = 0.581, *T<sub>max</sub>* = 1.000

34850 measured reflections

7195 independent reflections

4099 reflections with *I* > 2σ(*I*)

*R<sub>int</sub>* = 0.113

$\theta_{\max}$  = 31.1°,  $\theta_{\min}$  = 2.2°

*h* = -34→32

*k* = -7→7

*l* = -54→57

#### Refinement

Refinement on *F*<sup>2</sup>

Least-squares matrix: full

*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.062

*wR*(*F*<sup>2</sup>) = 0.146

*S* = 1.02

7195 reflections

302 parameters

311 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

*w* = 1/[σ<sup>2</sup>(*F<sub>o</sub>*<sup>2</sup>) + (0.058*P*)<sup>2</sup>]

where *P* = (*F<sub>o</sub>*<sup>2</sup> + 2*F<sub>c</sub>*<sup>2</sup>)/3

(Δ/σ)<sub>max</sub> = 0.003

Δρ<sub>max</sub> = 0.65 e Å<sup>-3</sup>

Δρ<sub>min</sub> = -1.00 e Å<sup>-3</sup>

#### Special details

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** The H atoms were positioned geometrically and allowed to ride on their parent atoms: C—H = 0.95–0.99 Å with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C-methyl})$  and  $1.2U_{\text{eq}}(\text{C})$  for other C-bound H atoms.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

|      | <i>x</i>      | <i>y</i>      | <i>z</i>     | $U_{\text{iso}}^*/U_{\text{eq}}$ | Occ. (<1) |
|------|---------------|---------------|--------------|----------------------------------|-----------|
| Br1  | 0.03432 (2)   | 0.42252 (9)   | 0.23314 (2)  | 0.03175 (14)                     |           |
| Se1  | 0.13904 (2)   | −0.26010 (11) | 0.34617 (2)  | 0.04244 (16)                     |           |
| Se2  | 0.08179 (2)   | 0.17304 (11)  | 0.09546 (2)  | 0.04280 (17)                     |           |
| N1   | 0.18882 (16)  | 0.1338 (8)    | 0.31010 (10) | 0.0316 (9)                       |           |
| N2   | 0.25023 (16)  | −0.0141 (8)   | 0.35460 (10) | 0.0333 (9)                       |           |
| N3   | 0.00557 (16)  | −0.0457 (8)   | 0.13496 (10) | 0.0317 (9)                       |           |
| N4   | −0.00708 (16) | −0.2196 (8)   | 0.08424 (10) | 0.0331 (9)                       |           |
| C1   | 0.08406 (18)  | 0.1496 (9)    | 0.22385 (12) | 0.0278 (10)                      |           |
| C2   | 0.12891 (19)  | 0.0756 (9)    | 0.25105 (12) | 0.0283 (10)                      |           |
| C3   | 0.16548 (19)  | −0.1204 (9)   | 0.24409 (12) | 0.0312 (10)                      |           |
| H3A  | 0.196352      | −0.178879     | 0.261858     | 0.037*                           |           |
| C4   | 0.15725 (19)  | −0.2312 (9)   | 0.21144 (12) | 0.0314 (10)                      |           |
| H4A  | 0.182993      | −0.361719     | 0.207026     | 0.038*                           |           |
| C5   | 0.11175 (19)  | −0.1523 (9)   | 0.18535 (13) | 0.0318 (10)                      |           |
| H5A  | 0.106484      | −0.229836     | 0.163219     | 0.038*                           |           |
| C6   | 0.07374 (18)  | 0.0393 (9)    | 0.19125 (12) | 0.0288 (10)                      |           |
| C7   | 0.13510 (19)  | 0.1943 (10)   | 0.28618 (12) | 0.0353 (11)                      |           |
| H7A  | 0.131785      | 0.385518      | 0.283642     | 0.042*                           |           |
| H7B  | 0.102736      | 0.133754      | 0.296168     | 0.042*                           |           |
| C8   | 0.19483 (19)  | −0.0397 (10)  | 0.33649 (13) | 0.0327 (11)                      |           |
| C9   | 0.2400 (2)    | 0.2678 (10)   | 0.31233 (14) | 0.0378 (12)                      |           |
| H9A  | 0.246848      | 0.401618      | 0.297264     | 0.045*                           |           |
| C10  | 0.2781 (2)    | 0.1749 (10)   | 0.33961 (13) | 0.0362 (11)                      |           |
| H10A | 0.317070      | 0.228839      | 0.347217     | 0.043*                           |           |
| C11A | 0.2772 (6)    | −0.150 (4)    | 0.3877 (2)   | 0.0415 (18)                      | 0.777 (9) |
| H11A | 0.314525      | −0.227601     | 0.385894     | 0.050*                           | 0.777 (9) |
| H11B | 0.251583      | −0.292610     | 0.392113     | 0.050*                           | 0.777 (9) |
| C12A | 0.2866 (3)    | 0.0431 (17)   | 0.41745 (18) | 0.0459 (18)                      | 0.777 (9) |
| H12A | 0.312305      | −0.038945     | 0.437571     | 0.055*                           | 0.777 (9) |
| H12B | 0.307207      | 0.195545      | 0.410878     | 0.055*                           | 0.777 (9) |
| C13A | 0.2347 (4)    | 0.138 (2)     | 0.4285 (2)   | 0.058 (2)                        | 0.777 (9) |
| H13A | 0.207640      | 0.212700      | 0.408279     | 0.070*                           | 0.777 (9) |
| H13B | 0.246190      | 0.279971      | 0.445444     | 0.070*                           | 0.777 (9) |
| C14A | 0.2031 (3)    | −0.0743 (18)  | 0.4447 (2)   | 0.068 (3)                        | 0.777 (9) |
| H14A | 0.169133      | 0.001099      | 0.451154     | 0.102*                           | 0.777 (9) |
| H14B | 0.229140      | −0.144843     | 0.465166     | 0.102*                           | 0.777 (9) |
| H14C | 0.190994      | −0.214146     | 0.427911     | 0.102*                           | 0.777 (9) |
| C15  | 0.0234 (2)    | 0.1316 (10)   | 0.16346 (13) | 0.0360 (11)                      |           |
| H15A | −0.009932     | 0.163767      | 0.174104     | 0.043*                           |           |
| H15B | 0.033920      | 0.299565      | 0.154351     | 0.043*                           |           |
| C16  | 0.02429 (19)  | −0.0392 (9)   | 0.10495 (12) | 0.0313 (10)                      |           |
| C17  | −0.0368 (2)   | −0.2339 (10)  | 0.13295 (14) | 0.0360 (11)                      |           |

|      |             |              |               |             |           |
|------|-------------|--------------|---------------|-------------|-----------|
| H17A | -0.056450   | -0.279112    | 0.150487      | 0.043*      |           |
| C18  | -0.0446 (2) | -0.3398 (10) | 0.10158 (14)  | 0.0368 (11) |           |
| H18A | -0.071160   | -0.473955    | 0.092722      | 0.044*      |           |
| C19  | -0.0022 (2) | -0.2783 (11) | 0.04901 (13)  | 0.0414 (12) |           |
| H19A | 0.009974    | -0.461625    | 0.047847      | 0.050*      |           |
| H19B | 0.028133    | -0.166725    | 0.043003      | 0.050*      |           |
| C20  | -0.0580 (2) | -0.2359 (11) | 0.02291 (14)  | 0.0469 (14) |           |
| H20A | -0.051574   | -0.278474    | -0.000216     | 0.056*      |           |
| H20B | -0.087093   | -0.359333    | 0.027889      | 0.056*      |           |
| C21  | -0.0822 (2) | 0.0339 (12)  | 0.02194 (14)  | 0.0503 (14) |           |
| H21A | -0.053011   | 0.159054     | 0.017564      | 0.060*      |           |
| H21B | -0.090319   | 0.075274     | 0.044720      | 0.060*      |           |
| C22  | -0.1373 (3) | 0.0678 (15)  | -0.00551 (16) | 0.070 (2)   |           |
| H22A | -0.152009   | 0.244755     | -0.004373     | 0.105*      |           |
| H22B | -0.166194   | -0.057673    | -0.001584     | 0.105*      |           |
| H22C | -0.129043   | 0.038010     | -0.028284     | 0.105*      |           |
| C11B | 0.274 (2)   | -0.165 (14)  | 0.3813 (8)    | 0.048 (4)   | 0.223 (9) |
| H11C | 0.315587    | -0.187117    | 0.380908      | 0.057*      | 0.223 (9) |
| H11D | 0.256154    | -0.339009    | 0.377563      | 0.057*      | 0.223 (9) |
| C12B | 0.2706 (12) | -0.073 (6)   | 0.4171 (6)    | 0.056 (4)   | 0.223 (9) |
| H12C | 0.270102    | -0.228140    | 0.431643      | 0.067*      | 0.223 (9) |
| H12D | 0.306396    | 0.024516     | 0.426880      | 0.067*      | 0.223 (9) |
| C13B | 0.2211 (12) | 0.093 (8)    | 0.4200 (5)    | 0.059 (4)   | 0.223 (9) |
| H13C | 0.185392    | -0.011671    | 0.414007      | 0.070*      | 0.223 (9) |
| H13D | 0.217749    | 0.236373     | 0.403042      | 0.070*      | 0.223 (9) |
| C14B | 0.2259 (11) | 0.211 (6)    | 0.4562 (6)    | 0.059 (6)   | 0.223 (9) |
| H14D | 0.187460    | 0.220141     | 0.461396      | 0.089*      | 0.223 (9) |
| H14E | 0.242222    | 0.386475     | 0.456853      | 0.089*      | 0.223 (9) |
| H14F | 0.250942    | 0.101579     | 0.473348      | 0.089*      | 0.223 (9) |

*Atomic displacement parameters (Å<sup>2</sup>)*

|     | $U^{11}$    | $U^{22}$   | $U^{33}$   | $U^{12}$     | $U^{13}$     | $U^{23}$     |
|-----|-------------|------------|------------|--------------|--------------|--------------|
| Br1 | 0.0238 (2)  | 0.0309 (3) | 0.0410 (3) | 0.00448 (19) | 0.00805 (19) | 0.0010 (2)   |
| Se1 | 0.0290 (3)  | 0.0445 (3) | 0.0546 (4) | -0.0049 (2)  | 0.0109 (2)   | -0.0029 (3)  |
| Se2 | 0.0278 (3)  | 0.0556 (4) | 0.0433 (3) | -0.0039 (2)  | 0.0039 (2)   | 0.0106 (3)   |
| N1  | 0.0225 (18) | 0.040 (2)  | 0.031 (2)  | 0.0018 (16)  | 0.0020 (15)  | -0.0011 (18) |
| N2  | 0.0243 (18) | 0.038 (2)  | 0.035 (2)  | 0.0038 (17)  | 0.0025 (16)  | -0.0007 (18) |
| N3  | 0.0247 (19) | 0.036 (2)  | 0.033 (2)  | 0.0032 (16)  | 0.0024 (16)  | -0.0002 (17) |
| N4  | 0.028 (2)   | 0.037 (2)  | 0.033 (2)  | 0.0007 (17)  | 0.0033 (17)  | 0.0045 (18)  |
| C1  | 0.021 (2)   | 0.027 (2)  | 0.036 (2)  | 0.0039 (18)  | 0.0098 (18)  | 0.000 (2)    |
| C2  | 0.025 (2)   | 0.029 (2)  | 0.032 (2)  | 0.0003 (18)  | 0.0079 (18)  | 0.002 (2)    |
| C3  | 0.024 (2)   | 0.038 (3)  | 0.031 (2)  | 0.0067 (19)  | 0.0040 (19)  | 0.006 (2)    |
| C4  | 0.025 (2)   | 0.038 (3)  | 0.034 (3)  | 0.004 (2)    | 0.0113 (19)  | 0.002 (2)    |
| C5  | 0.029 (2)   | 0.036 (3)  | 0.032 (3)  | 0.002 (2)    | 0.0094 (19)  | 0.001 (2)    |
| C6  | 0.021 (2)   | 0.036 (3)  | 0.030 (2)  | 0.0012 (18)  | 0.0075 (17)  | 0.003 (2)    |
| C7  | 0.026 (2)   | 0.041 (3)  | 0.035 (3)  | 0.012 (2)    | -0.001 (2)   | -0.001 (2)   |
| C8  | 0.022 (2)   | 0.038 (3)  | 0.038 (3)  | 0.0029 (19)  | 0.0050 (19)  | 0.001 (2)    |

|      |            |            |           |             |             |            |
|------|------------|------------|-----------|-------------|-------------|------------|
| C9   | 0.030 (2)  | 0.039 (3)  | 0.045 (3) | -0.001 (2)  | 0.010 (2)   | 0.004 (2)  |
| C10  | 0.025 (2)  | 0.049 (3)  | 0.035 (3) | -0.002 (2)  | 0.0066 (19) | -0.005 (2) |
| C11A | 0.027 (3)  | 0.065 (4)  | 0.031 (4) | 0.007 (3)   | 0.003 (3)   | 0.012 (4)  |
| C12A | 0.032 (3)  | 0.071 (5)  | 0.031 (3) | -0.010 (3)  | 0.000 (3)   | 0.007 (3)  |
| C13A | 0.054 (4)  | 0.082 (5)  | 0.033 (4) | 0.004 (4)   | -0.002 (3)  | -0.008 (4) |
| C14A | 0.040 (4)  | 0.111 (7)  | 0.053 (5) | 0.000 (4)   | 0.011 (4)   | 0.007 (5)  |
| C15  | 0.030 (2)  | 0.042 (3)  | 0.034 (3) | 0.011 (2)   | 0.004 (2)   | -0.001 (2) |
| C16  | 0.022 (2)  | 0.040 (3)  | 0.031 (2) | 0.0041 (19) | 0.0039 (18) | 0.008 (2)  |
| C17  | 0.031 (2)  | 0.041 (3)  | 0.039 (3) | 0.001 (2)   | 0.013 (2)   | 0.008 (2)  |
| C18  | 0.033 (2)  | 0.035 (3)  | 0.041 (3) | -0.006 (2)  | 0.006 (2)   | 0.003 (2)  |
| C19  | 0.039 (3)  | 0.048 (3)  | 0.036 (3) | 0.008 (2)   | 0.008 (2)   | -0.002 (2) |
| C20  | 0.048 (3)  | 0.059 (3)  | 0.034 (3) | 0.007 (3)   | 0.009 (2)   | -0.003 (3) |
| C21  | 0.048 (3)  | 0.062 (4)  | 0.036 (3) | 0.008 (3)   | -0.001 (2)  | -0.004 (3) |
| C22  | 0.050 (4)  | 0.105 (6)  | 0.049 (4) | 0.026 (4)   | -0.001 (3)  | 0.002 (4)  |
| C11B | 0.035 (7)  | 0.069 (7)  | 0.033 (6) | 0.008 (6)   | -0.005 (6)  | 0.007 (6)  |
| C12B | 0.047 (7)  | 0.079 (7)  | 0.037 (5) | 0.005 (6)   | -0.002 (6)  | 0.005 (6)  |
| C13B | 0.049 (7)  | 0.085 (8)  | 0.037 (7) | 0.004 (7)   | 0.000 (6)   | -0.006 (7) |
| C14B | 0.045 (10) | 0.093 (12) | 0.040 (9) | -0.018 (10) | 0.008 (9)   | -0.009 (9) |

*Geometric parameters (Å, °)*

|                       |            |           |            |
|-----------------------|------------|-----------|------------|
| Br1—C1                | 1.922 (4)  | C12A—H12A | 0.9900     |
| Br1—Br1 <sup>i</sup>  | 3.4688 (7) | C12A—H12B | 0.9900     |
| Br1—Br1 <sup>ii</sup> | 3.4688 (7) | C13A—C14A | 1.545 (12) |
| Se1—C8                | 1.848 (5)  | C13A—H13A | 0.9900     |
| Se2—C16               | 1.851 (5)  | C13A—H13B | 0.9900     |
| N1—C8                 | 1.362 (6)  | C14A—H14A | 0.9800     |
| N1—C9                 | 1.383 (6)  | C14A—H14B | 0.9800     |
| N1—C7                 | 1.450 (5)  | C14A—H14C | 0.9800     |
| N2—C11B               | 1.34 (5)   | C15—H15A  | 0.9900     |
| N2—C8                 | 1.363 (6)  | C15—H15B  | 0.9900     |
| N2—C10                | 1.383 (6)  | C17—C18   | 1.338 (7)  |
| N2—C11A               | 1.508 (12) | C17—H17A  | 0.9500     |
| N3—C16                | 1.363 (6)  | C18—H18A  | 0.9500     |
| N3—C17                | 1.386 (6)  | C19—C20   | 1.512 (7)  |
| N3—C15                | 1.445 (6)  | C19—H19A  | 0.9900     |
| N4—C16                | 1.352 (6)  | C19—H19B  | 0.9900     |
| N4—C18                | 1.386 (6)  | C20—C21   | 1.500 (8)  |
| N4—C19                | 1.464 (6)  | C20—H20A  | 0.9900     |
| C1—C6                 | 1.388 (6)  | C20—H20B  | 0.9900     |
| C1—C2                 | 1.396 (6)  | C21—C22   | 1.520 (7)  |
| C2—C3                 | 1.397 (6)  | C21—H21A  | 0.9900     |
| C2—C7                 | 1.503 (6)  | C21—H21B  | 0.9900     |
| C3—C4                 | 1.393 (6)  | C22—H22A  | 0.9800     |
| C3—H3A                | 0.9500     | C22—H22B  | 0.9800     |
| C4—C5                 | 1.387 (6)  | C22—H22C  | 0.9800     |
| C4—H4A                | 0.9500     | C11B—C12B | 1.523 (12) |
| C5—C6                 | 1.391 (6)  | C11B—H11C | 0.9900     |

|   |             |                |            |
|---|-------------|----------------|------------|
| C5—H5A                                  | 0.9500      | C11B—H11D      | 0.9900     |
| C6—C15                                  | 1.518 (6)   | C12B—C13B      | 1.478 (10) |
| C7—H7A                                  | 0.9900      | C12B—H12C      | 0.9900     |
| C7—H7B                                  | 0.9900      | C12B—H12D      | 0.9900     |
| C9—C10                                  | 1.342 (7)   | C13B—C14B      | 1.545 (13) |
| C9—H9A                                  | 0.9500      | C13B—H13C      | 0.9900     |
| C10—H10A                                | 0.9500      | C13B—H13D      | 0.9900     |
| C11A—C12A                               | 1.524 (11)  | C14B—H14D      | 0.9800     |
| C11A—H11A                               | 0.9900      | C14B—H14E      | 0.9800     |
| C11A—H11B                               | 0.9900      | C14B—H14F      | 0.9800     |
| C12A—C13A                               | 1.479 (9)   |                |            |
|   |             |                |            |
| C1—Br1—Br1 <sup>i</sup>                 | 83.93 (13)  | C13A—C14A—H14B | 109.5      |
| C1—Br1—Br1 <sup>ii</sup>                | 167.60 (15) | H14A—C14A—H14B | 109.5      |
| Br1 <sup>i</sup> —Br1—Br1 <sup>ii</sup> | 95.68 (3)   | C13A—C14A—H14C | 109.5      |
| C8—N1—C9                                | 109.5 (4)   | H14A—C14A—H14C | 109.5      |
| C8—N1—C7                                | 125.2 (4)   | H14B—C14A—H14C | 109.5      |
| C9—N1—C7                                | 124.8 (4)   | N3—C15—C6      | 115.1 (4)  |
| C11B—N2—C8                              | 124 (3)     | N3—C15—H15A    | 108.5      |
| C11B—N2—C10                             | 126 (3)     | C6—C15—H15A    | 108.5      |
| C8—N2—C10                               | 109.8 (4)   | N3—C15—H15B    | 108.5      |
| C8—N2—C11A                              | 126.5 (8)   | C6—C15—H15B    | 108.5      |
| C10—N2—C11A                             | 123.5 (8)   | H15A—C15—H15B  | 107.5      |
| C16—N3—C17                              | 109.8 (4)   | N4—C16—N3      | 106.0 (4)  |
| C16—N3—C15                              | 125.2 (4)   | N4—C16—Se2     | 127.8 (4)  |
| C17—N3—C15                              | 124.8 (4)   | N3—C16—Se2     | 126.2 (4)  |
| C16—N4—C18                              | 109.3 (4)   | C18—C17—N3     | 106.8 (4)  |
| C16—N4—C19                              | 125.2 (4)   | C18—C17—H17A   | 126.6      |
| C18—N4—C19                              | 125.4 (4)   | N3—C17—H17A    | 126.6      |
| C6—C1—C2                                | 124.2 (4)   | C17—C18—N4     | 108.1 (4)  |
| C6—C1—Br1                               | 119.2 (3)   | C17—C18—H18A   | 126.0      |
| C2—C1—Br1                               | 116.6 (3)   | N4—C18—H18A    | 126.0      |
| C3—C2—C1                                | 116.5 (4)   | N4—C19—C20     | 112.8 (4)  |
| C3—C2—C7                                | 122.6 (4)   | N4—C19—H19A    | 109.0      |
| C1—C2—C7                                | 120.9 (4)   | C20—C19—H19A   | 109.0      |
| C4—C3—C2                                | 120.9 (4)   | N4—C19—H19B    | 109.0      |
| C4—C3—H3A                               | 119.6       | C20—C19—H19B   | 109.0      |
| C2—C3—H3A                               | 119.6       | H19A—C19—H19B  | 107.8      |
| C5—C4—C3                                | 120.4 (4)   | C21—C20—C19    | 115.2 (5)  |
| C5—C4—H4A                               | 119.8       | C21—C20—H20A   | 108.5      |
| C3—C4—H4A                               | 119.8       | C19—C20—H20A   | 108.5      |
| C4—C5—C6                                | 120.7 (5)   | C21—C20—H20B   | 108.5      |
| C4—C5—H5A                               | 119.7       | C19—C20—H20B   | 108.5      |
| C6—C5—H5A                               | 119.7       | H20A—C20—H20B  | 107.5      |
| C1—C6—C5                                | 117.3 (4)   | C20—C21—C22    | 112.9 (5)  |
| C1—C6—C15                               | 120.2 (4)   | C20—C21—H21A   | 109.0      |
| C5—C6—C15                               | 122.4 (4)   | C22—C21—H21A   | 109.0      |
| N1—C7—C2                                | 114.6 (4)   | C20—C21—H21B   | 109.0      |



|                |            |                     |             |
|----------------|------------|---------------------|-------------|
| N1—C7—H7A      | 108.6      | C22—C21—H21B        | 109.0       |
| C2—C7—H7A      | 108.6      | H21A—C21—H21B       | 107.8       |
| N1—C7—H7B      | 108.6      | C21—C22—H22A        | 109.5       |
| C2—C7—H7B      | 108.6      | C21—C22—H22B        | 109.5       |
| H7A—C7—H7B     | 107.6      | H22A—C22—H22B       | 109.5       |
| N2—C8—N1       | 105.7 (4)  | C21—C22—H22C        | 109.5       |
| N2—C8—Se1      | 127.1 (4)  | H22A—C22—H22C       | 109.5       |
| N1—C8—Se1      | 127.1 (3)  | H22B—C22—H22C       | 109.5       |
| C10—C9—N1      | 107.7 (4)  | N2—C11B—C12B        | 117 (3)     |
| C10—C9—H9A     | 126.2      | N2—C11B—H11C        | 108.0       |
| N1—C9—H9A      | 126.2      | C12B—C11B—H11C      | 108.0       |
| C9—C10—N2      | 107.2 (4)  | N2—C11B—H11D        | 108.0       |
| C9—C10—H10A    | 126.4      | C12B—C11B—H11D      | 108.0       |
| N2—C10—H10A    | 126.4      | H11C—C11B—H11D      | 107.2       |
| N2—C11A—C12A   | 109.7 (11) | C13B—C12B—C11B      | 117.6 (10)  |
| N2—C11A—H11A   | 109.7      | C13B—C12B—H12C      | 107.9       |
| C12A—C11A—H11A | 109.7      | C11B—C12B—H12C      | 107.9       |
| N2—C11A—H11B   | 109.7      | C13B—C12B—H12D      | 107.9       |
| C12A—C11A—H11B | 109.7      | C11B—C12B—H12D      | 107.9       |
| H11A—C11A—H11B | 108.2      | H12C—C12B—H12D      | 107.2       |
| C13A—C12A—C11A | 117.1 (6)  | C12B—C13B—C14B      | 113.8 (10)  |
| C13A—C12A—H12A | 108.0      | C12B—C13B—H13C      | 108.8       |
| C11A—C12A—H12A | 108.0      | C14B—C13B—H13C      | 108.8       |
| C13A—C12A—H12B | 108.0      | C12B—C13B—H13D      | 108.8       |
| C11A—C12A—H12B | 108.0      | C14B—C13B—H13D      | 108.8       |
| H12A—C12A—H12B | 107.3      | H13C—C13B—H13D      | 107.7       |
| C12A—C13A—C14A | 113.7 (7)  | C13B—C14B—H14D      | 109.5       |
| C12A—C13A—H13A | 108.8      | C13B—C14B—H14E      | 109.5       |
| C14A—C13A—H13A | 108.8      | H14D—C14B—H14E      | 109.5       |
| C12A—C13A—H13B | 108.8      | C13B—C14B—H14F      | 109.5       |
| C14A—C13A—H13B | 108.8      | H14D—C14B—H14F      | 109.5       |
| H13A—C13A—H13B | 107.7      | H14E—C14B—H14F      | 109.5       |
| C13A—C14A—H14A | 109.5      |                     |             |
|                |            |                     |             |
| C6—C1—C2—C3    | 0.8 (7)    | C8—N2—C10—C9        | 0.4 (6)     |
| Br1—C1—C2—C3   | -179.1 (3) | C11A—N2—C10—C9      | -174.9 (6)  |
| C6—C1—C2—C7    | -176.5 (4) | C8—N2—C11A—C12A     | -108.0 (11) |
| Br1—C1—C2—C7   | 3.5 (6)    | C10—N2—C11A—C12A    | 66.5 (10)   |
| C1—C2—C3—C4    | 0.7 (7)    | N2—C11A—C12A—C13A   | 71.2 (14)   |
| C7—C2—C3—C4    | 178.0 (4)  | C11A—C12A—C13A—C14A | 66.6 (12)   |
| C2—C3—C4—C5    | -1.3 (7)   | C16—N3—C15—C6       | -93.5 (5)   |
| C3—C4—C5—C6    | 0.3 (7)    | C17—N3—C15—C6       | 92.2 (5)    |
| C2—C1—C6—C5    | -1.8 (7)   | C1—C6—C15—N3        | -161.1 (4)  |
| Br1—C1—C6—C5   | 178.2 (3)  | C5—C6—C15—N3        | 20.2 (6)    |
| C2—C1—C6—C15   | 179.4 (4)  | C18—N4—C16—N3       | -0.8 (5)    |
| Br1—C1—C6—C15  | -0.6 (6)   | C19—N4—C16—N3       | 179.2 (4)   |
| C4—C5—C6—C1    | 1.2 (7)    | C18—N4—C16—Se2      | 178.5 (3)   |
| C4—C5—C6—C15   | 180.0 (4)  | C19—N4—C16—Se2      | -1.6 (7)    |

|                |             |                     |            |
|----------------|-------------|---------------------|------------|
| C8—N1—C7—C2    | -104.3 (5)  | C17—N3—C16—N4       | 1.1 (5)    |
| C9—N1—C7—C2    | 84.6 (6)    | C15—N3—C16—N4       | -174.0 (4) |
| C3—C2—C7—N1    | 13.2 (7)    | C17—N3—C16—Se2      | -178.2 (3) |
| C1—C2—C7—N1    | -169.6 (4)  | C15—N3—C16—Se2      | 6.8 (6)    |
| C11B—N2—C8—N1  | -175.0 (13) | C16—N3—C17—C18      | -0.9 (5)   |
| C10—N2—C8—N1   | 0.3 (5)     | C15—N3—C17—C18      | 174.1 (4)  |
| C11A—N2—C8—N1  | 175.4 (6)   | N3—C17—C18—N4       | 0.4 (5)    |
| C11B—N2—C8—Se1 | 5.9 (15)    | C16—N4—C18—C17      | 0.2 (5)    |
| C10—N2—C8—Se1  | -178.7 (4)  | C19—N4—C18—C17      | -179.7 (4) |
| C11A—N2—C8—Se1 | -3.6 (9)    | C16—N4—C19—C20      | -120.4 (5) |
| C9—N1—C8—N2    | -0.9 (5)    | C18—N4—C19—C20      | 59.5 (6)   |
| C7—N1—C8—N2    | -173.1 (4)  | N4—C19—C20—C21      | 58.1 (7)   |
| C9—N1—C8—Se1   | 178.1 (4)   | C19—C20—C21—C22     | 178.1 (5)  |
| C7—N1—C8—Se1   | 5.9 (7)     | C8—N2—C11B—C12B     | -92 (6)    |
| C8—N1—C9—C10   | 1.2 (6)     | C10—N2—C11B—C12B    | 94 (5)     |
| C7—N1—C9—C10   | 173.4 (4)   | N2—C11B—C12B—C13B   | 28 (8)     |
| N1—C9—C10—N2   | -0.9 (6)    | C11B—C12B—C13B—C14B | -171 (5)   |
| C11B—N2—C10—C9 | 175.7 (16)  |                     |            |

Symmetry codes: (i)  $-x, y-1/2, -z+1/2$ ; (ii)  $-x, y+1/2, -z+1/2$ .

#### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

| $D-H\cdots A$                         | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|---------------------------------------|-------|-------------|-------------|---------------|
| C10—H10A $\cdots$ Se2 <sup>iii</sup>  | 0.95  | 2.97        | 3.823 (5)   | 150           |
| C12A—H12B $\cdots$ Se2 <sup>iii</sup> | 0.99  | 2.79        | 3.590 (7)   | 139           |
| C19—H19A $\cdots$ Se2 <sup>iv</sup>   | 0.99  | 2.94        | 3.708 (5)   | 136           |

Symmetry codes: (iii)  $-x+1/2, -y+1/2, -z+1/2$ ; (iv)  $x, y-1, z$ .