

An orthorhombic polymorph of (*E*)-1-(benzo[*d*]-[1,3]dioxol-5-yl)-3-(2,3-dichlorophenyl)prop-2-en-1-one

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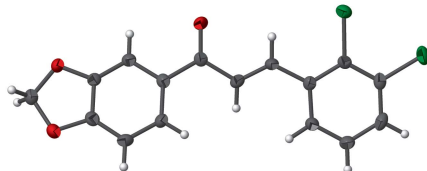
CCDC reference: 1583520

Structural data: full structural data are available from iucrdata.iucr.org

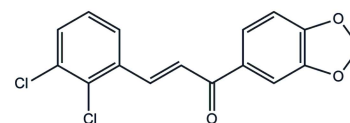
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The title compound, C₁₆H₁₀Cl₂O₃, is almost planar with a dihedral angle of 0.14 (16)^o between the benzodioxole ring system and the dichlorobenzene ring that are bridged by the olefinic double bond. The corresponding value reported for the monoclinic polymorph is [5.57 (9)^o (Lokeshwari *et al.* (2017). *IUCrData*, **2**, x170103)]. The carbonyl group lies almost in the plane of the olefinic double bond and is twisted slightly from the benzodioxole ring plane. In the crystal, the molecules are linked by weak C—H...O and C—H...Cl hydrogen bonds, forming a chain propagating along the *b*-axis direction.

3D view



Chemical scheme



Structure description

Chalcones and their derivatives exhibit a plethora of biological applications that include use as antioxidants, or antifungal, antibacterial and cardioprotective agents. As part of our ongoing work on such molecules (Rajendraprasad *et al.*, 2017; Naveen *et al.*, 2016*a,b*), we report here the synthesis and crystal structure of the title compound.

The structure of the title molecule is shown in Fig. 1. It is a polymorph having been reported previously in the monoclinic space-group *P*₂₁/*c* (Lokeshwari *et al.*, 2017). The molecule is nearly planar as seen by the dihedral angle of 0.14 (16)^o between the benzodioxole ring system and the dichlorobenzene ring; these are bridged by an olefinic double bond that adopts an *E* conformation. The corresponding dihedral angle reported for the monoclinic polymorph is 5.57 (9)^o (Lokeshwari *et al.*, 2017). The *trans* conformation of the C=C double bond in the central enone group is confirmed by the C—

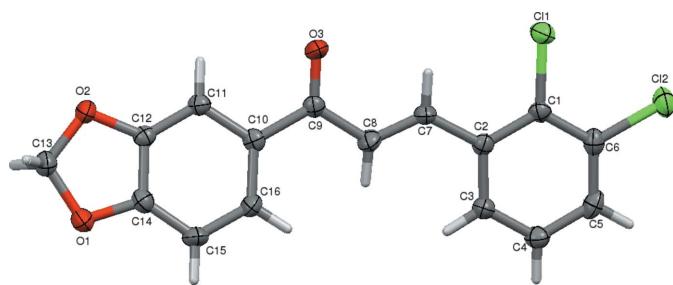


Figure 1
The molecular structure of the title compound with the atom-numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the 50% probability level.

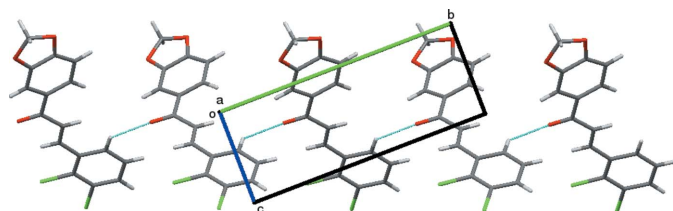


Figure 2
Packing of the molecules, viewed along the *b* axis, with hydrogen bonds shown as blue lines.

C=C—C torsion angle of $-174.9(4)^\circ$. The carbonyl group at C9 lies almost in the plane of the olefinic double bond and is twisted slightly from the benzodioxole ring as indicated by the C11—C10—C9—O3 and C7—C8—C9—O3 torsion angles of $13.7(5)$ and $5.0(6)^\circ$ respectively.

In the crystal, the molecules are linked by weak C—H...O and C—H...Cl hydrogen bonds (Table 1), forming a chain propagating along the *b*-axis direction, Fig. 2.

Synthesis and crystallization

A mixture of 2,3-dichlorobenzaldehyde (5 mmol), 1-(benzo[*d*][1,3]dioxol-5-yl)ethanone (5 mmol) and sodium hydroxide (5 mmol) in methanol (25 ml) was stirred at room temperature for 3 h. The progress of the reaction was monitored by TLC. After completion, the mixture was poured into ice-cold water and kept in the refrigerator for 18 h. The solid formed was filtered, and washed with cold 5% hydrochloric acid. Pale-green crystals were obtained from methanol solution by using the slow solvent evaporation technique, yield 88%, m.p. 401–402 K.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

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Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C3—H3...O3 ⁱ	0.93	2.46	3.185 (5)	135

Symmetry code: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, z + \frac{1}{2}$.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₆ H ₁₀ Cl ₂ O ₃
<i>M_r</i>	321.14
Crystal system, space group	Orthorhombic, <i>Pna</i> 2 ₁
Temperature (K)	296
<i>a</i> , <i>b</i> , <i>c</i> (\AA)	21.9756 (11), 12.7354 (6), 4.9889 (3)
<i>V</i> (\AA^3)	1396.23 (13)
<i>Z</i>	4
Radiation type	Cu <i>K</i> α
μ (mm^{-1})	4.25
Crystal size (mm)	0.27 \times 0.26 \times 0.22
Data collection	
Diffractometer	Bruker X8 Proteum
Absorption correction	Multi-scan (<i>SADABS</i> ; Bruker, 2013)
<i>T_{min}</i> , <i>T_{max}</i>	0.393, 0.455
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	6212, 1801, 1639
<i>R_{int}</i>	0.061
(<i>sin</i> θ / λ) _{max} (\AA^{-1})	0.585
Refinement	
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.044, 0.120, 1.06
No. of reflections	1801
No. of parameters	190
No. of restraints	1
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{\text{max}}$, $\Delta\rho_{\text{min}}$ (e \AA^{-3})	0.29, −0.37
Absolute structure	Flack (1983)
Absolute structure parameter	0.02 (2)

Computer programs: *APEX2* and *SAINT* (Bruker, 2013), *SHELXS97* (Sheldrick, 2008), *SHELXL97* (Sheldrick, 2008) and *Mercury* (Macrae *et al.*, 2008).

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full crystallographic data

IUCrData (2017). **2**, x171590 [https://doi.org/10.1107/S2414314617015905]

An orthorhombic polymorph of (*E*)-1-(benzo[*d*][1,3]dioxol-5-yl)-3-(2,3-dichlorophenyl)prop-2-en-1-one

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(*E*)-1-(Benzo[*d*][1,3]dioxol-5-yl)-3-(2,3-dichlorophenyl)prop-2-en-1-one

Crystal data

C₁₆H₁₀Cl₂O₃

M_r = 321.14

Orthorhombic, *Pna*2₁

Hall symbol: P 2c -2n

a = 21.9756 (11) Å

b = 12.7354 (6) Å

c = 4.9889 (3) Å

V = 1396.23 (13) Å³

Z = 4

F(000) = 656

D_x = 1.528 Mg m⁻³

Cu *Kα* radiation, λ = 1.54178 Å

Cell parameters from 1639 reflections

θ = 4.0–64.5°

μ = 4.25 mm⁻¹

T = 296 K

Prism, green

0.27 × 0.26 × 0.22 mm

Data collection

Bruker X8 Proteum
diffractometer

Radiation source: Bruker MicroStar microfocus
rotating anode

Helios multilayer optics monochromator

Detector resolution: 18.4 pixels mm⁻¹

φ and ω scans

Absorption correction: multi-scan
(SADABS; Bruker, 2013)

T_{min} = 0.393, *T_{max}* = 0.455

6212 measured reflections

1801 independent reflections

1639 reflections with *I* > 2σ(*I*)

R_{int} = 0.061

θ_{max} = 64.5°, θ_{min} = 4.0°

h = -25→24

k = -13→14

l = -4→5

Refinement

Refinement on *F*²

Least-squares matrix: full

R[*F*² > 2σ(*F*²)] = 0.044

wR(*F*²) = 0.120

S = 1.06

1801 reflections

190 parameters

1 restraint

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H atoms treated by a mixture of independent
and constrained refinement

w = 1/[σ²(*F_o*²) + (0.0639*P*)²]

where *P* = (*F_o*² + 2*F_c*²)/3

(Δ/σ)_{max} < 0.001

Δρ_{max} = 0.29 e Å⁻³

Δρ_{min} = -0.37 e Å⁻³

Absolute structure: Flack (1983)

Absolute structure parameter: 0.02 (2)

Special details

Geometry. Bond distances, angles etc. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell esds are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R-factors wR and all goodnesses of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The observed criterion of $F^2 > 2\sigma(F^2)$ is used only for calculating -R-factor-obs etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	0.40504 (4)	0.22650 (8)	1.0103 (2)	0.0306 (3)
C12	0.48016 (5)	0.35770 (9)	1.4215 (3)	0.0386 (3)
O1	0.05572 (13)	0.4848 (2)	-0.3392 (7)	0.0333 (10)
O2	0.05823 (13)	0.3043 (2)	-0.2738 (7)	0.0335 (10)
O3	0.23955 (15)	0.2272 (2)	0.3436 (7)	0.0399 (11)
C1	0.38991 (17)	0.3592 (3)	1.0505 (8)	0.0229 (11)
C2	0.34376 (17)	0.4072 (3)	0.8997 (9)	0.0227 (11)
C3	0.33324 (17)	0.5145 (3)	0.9436 (9)	0.0260 (11)
C4	0.36566 (19)	0.5703 (3)	1.1311 (9)	0.0283 (12)
C5	0.41077 (18)	0.5219 (4)	1.2808 (9)	0.0300 (14)
C6	0.42298 (18)	0.4164 (4)	1.2366 (8)	0.0268 (13)
C7	0.30817 (18)	0.3496 (3)	0.7011 (9)	0.0263 (12)
C8	0.26308 (18)	0.3868 (3)	0.5549 (8)	0.0283 (12)
C9	0.22949 (17)	0.3220 (3)	0.3630 (7)	0.0232 (11)
C10	0.18281 (17)	0.3718 (3)	0.1898 (8)	0.0220 (11)
C11	0.14240 (17)	0.3059 (3)	0.0514 (8)	0.0242 (11)
C12	0.10171 (17)	0.3520 (3)	-0.1190 (9)	0.0224 (11)
C13	0.02917 (19)	0.3861 (3)	-0.4182 (9)	0.0307 (14)
C14	0.10051 (17)	0.4598 (3)	-0.1576 (8)	0.0270 (14)
C15	0.13913 (18)	0.5262 (3)	-0.0252 (9)	0.0307 (14)
C16	0.17994 (18)	0.4807 (3)	0.1541 (9)	0.0270 (12)
H3	0.30360	0.54870	0.84320	0.0310*
H4	0.35730	0.64110	1.15780	0.0340*
H5	0.43250	0.55950	1.40880	0.0360*
H7	0.31840	0.27940	0.67500	0.0320*
H8	0.25220	0.45690	0.57450	0.0340*
H11	0.14330	0.23350	0.07480	0.0290*
H13A	-0.01410	0.38620	-0.37970	0.0370*
H13B	0.03460	0.37540	-0.60930	0.0370*
H15	0.13820	0.59830	-0.05360	0.0370*
H16	0.20590	0.52370	0.25230	0.0320*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0342 (5)	0.0241 (6)	0.0335 (6)	0.0047 (4)	-0.0025 (4)	0.0021 (5)

C12	0.0333 (5)	0.0449 (7)	0.0375 (6)	0.0001 (5)	-0.0139 (5)	0.0056 (5)
O1	0.0329 (16)	0.0280 (17)	0.0389 (18)	0.0053 (13)	-0.0114 (13)	-0.0013 (15)
O2	0.0322 (15)	0.0252 (17)	0.0431 (19)	0.0008 (13)	-0.0158 (14)	-0.0055 (15)
O3	0.0466 (18)	0.0182 (18)	0.055 (2)	-0.0044 (14)	-0.0244 (17)	0.0006 (14)
C1	0.0237 (19)	0.021 (2)	0.024 (2)	-0.0026 (16)	0.0011 (18)	0.0057 (17)
C2	0.0239 (19)	0.024 (2)	0.0203 (19)	-0.0035 (15)	-0.0004 (16)	0.0030 (18)
C3	0.0259 (18)	0.022 (2)	0.030 (2)	0.0005 (16)	-0.0036 (17)	0.0009 (19)
C4	0.034 (2)	0.023 (2)	0.028 (2)	0.0007 (18)	-0.0004 (19)	-0.001 (2)
C5	0.028 (2)	0.039 (3)	0.023 (2)	-0.0089 (18)	-0.0015 (16)	-0.006 (2)
C6	0.0214 (18)	0.036 (3)	0.023 (2)	-0.0028 (17)	-0.0016 (17)	0.002 (2)
C7	0.029 (2)	0.018 (2)	0.032 (2)	0.0014 (16)	-0.0055 (19)	-0.0016 (19)
C8	0.031 (2)	0.024 (2)	0.030 (2)	0.0005 (17)	-0.0033 (19)	-0.0078 (19)
C9	0.0225 (18)	0.025 (2)	0.022 (2)	-0.0022 (17)	0.0014 (16)	0.0039 (18)
C10	0.0230 (19)	0.026 (2)	0.017 (2)	0.0003 (16)	0.0024 (16)	0.0018 (18)
C11	0.0275 (19)	0.021 (2)	0.024 (2)	-0.0002 (17)	0.0006 (16)	-0.0002 (18)
C12	0.0231 (18)	0.022 (2)	0.022 (2)	0.0002 (16)	0.0018 (17)	-0.0020 (18)
C13	0.029 (2)	0.032 (3)	0.031 (2)	0.0012 (18)	-0.0066 (18)	-0.001 (2)
C14	0.0249 (19)	0.028 (3)	0.028 (2)	0.0035 (18)	-0.0004 (16)	0.0006 (19)
C15	0.037 (2)	0.018 (2)	0.037 (3)	-0.0012 (18)	-0.0029 (19)	0.0009 (19)
C16	0.027 (2)	0.024 (2)	0.030 (2)	-0.0023 (17)	-0.0055 (17)	-0.004 (2)

Geometric parameters (Å, °)

C11—C1	1.734 (4)	C10—C11	1.404 (5)
C12—C6	1.729 (4)	C10—C16	1.400 (5)
O1—C13	1.441 (5)	C11—C12	1.366 (6)
O1—C14	1.375 (5)	C12—C14	1.387 (5)
O2—C12	1.371 (5)	C14—C15	1.368 (6)
O2—C13	1.419 (5)	C15—C16	1.393 (6)
O3—C9	1.231 (5)	C3—H3	0.9300
C1—C2	1.403 (6)	C4—H4	0.9300
C1—C6	1.386 (6)	C5—H5	0.9300
C2—C3	1.403 (5)	C7—H7	0.9300
C2—C7	1.460 (6)	C8—H8	0.9300
C3—C4	1.374 (6)	C11—H11	0.9300
C4—C5	1.386 (6)	C13—H13A	0.9700
C5—C6	1.388 (7)	C13—H13B	0.9700
C7—C8	1.318 (6)	C15—H15	0.9300
C8—C9	1.464 (5)	C16—H16	0.9300
C9—C10	1.484 (5)		
C13—O1—C14	105.6 (3)	O1—C14—C12	109.5 (3)
C12—O2—C13	105.9 (3)	O1—C14—C15	128.2 (3)
C11—C1—C2	120.1 (3)	C12—C14—C15	122.2 (4)
C11—C1—C6	119.3 (3)	C14—C15—C16	116.9 (4)
C2—C1—C6	120.6 (4)	C10—C16—C15	121.5 (4)
C1—C2—C3	117.4 (4)	C2—C3—H3	119.00
C1—C2—C7	122.2 (3)	C4—C3—H3	119.00

C3—C2—C7	120.5 (4)	C3—C4—H4	120.00
C2—C3—C4	121.6 (4)	C5—C4—H4	120.00
C3—C4—C5	120.5 (4)	C4—C5—H5	121.00
C4—C5—C6	118.9 (4)	C6—C5—H5	121.00
C12—C6—C1	120.8 (4)	C2—C7—H7	117.00
C12—C6—C5	118.3 (3)	C8—C7—H7	117.00
C1—C6—C5	120.9 (4)	C7—C8—H8	119.00
C2—C7—C8	126.7 (4)	C9—C8—H8	119.00
C7—C8—C9	122.6 (4)	C10—C11—H11	121.00
O3—C9—C8	120.9 (3)	C12—C11—H11	121.00
O3—C9—C10	119.8 (3)	O1—C13—H13A	110.00
C8—C9—C10	119.3 (3)	O1—C13—H13B	110.00
C9—C10—C11	117.9 (3)	O2—C13—H13A	110.00
C9—C10—C16	121.9 (3)	O2—C13—H13B	110.00
C11—C10—C16	120.1 (4)	H13A—C13—H13B	108.00
C10—C11—C12	117.6 (4)	C14—C15—H15	122.00
O2—C12—C11	128.1 (3)	C16—C15—H15	122.00
O2—C12—C14	110.3 (3)	C10—C16—H16	119.00
C11—C12—C14	121.6 (4)	C15—C16—H16	119.00
O1—C13—O2	108.6 (3)		
C14—O1—C13—O2	1.6 (4)	C4—C5—C6—C1	-1.4 (6)
C13—O1—C14—C12	-1.1 (4)	C2—C7—C8—C9	-179.1 (4)
C13—O1—C14—C15	179.6 (4)	C7—C8—C9—O3	5.0 (6)
C13—O2—C12—C11	-178.6 (4)	C7—C8—C9—C10	-174.9 (4)
C13—O2—C12—C14	0.8 (4)	O3—C9—C10—C11	13.7 (5)
C12—O2—C13—O1	-1.5 (4)	O3—C9—C10—C16	-163.7 (4)
C11—C1—C2—C3	179.3 (3)	C8—C9—C10—C11	-166.4 (4)
C11—C1—C2—C7	-1.6 (6)	C8—C9—C10—C16	16.2 (6)
C6—C1—C2—C3	0.6 (6)	C9—C10—C11—C12	-176.5 (4)
C6—C1—C2—C7	179.7 (4)	C16—C10—C11—C12	1.0 (6)
C11—C1—C6—C12	1.7 (5)	C9—C10—C16—C15	174.8 (4)
C11—C1—C6—C5	-177.8 (3)	C11—C10—C16—C15	-2.6 (6)
C2—C1—C6—C12	-179.7 (3)	C10—C11—C12—O2	-179.9 (4)
C2—C1—C6—C5	0.9 (6)	C10—C11—C12—C14	0.8 (6)
C1—C2—C3—C4	-1.5 (6)	O2—C12—C14—O1	0.2 (5)
C7—C2—C3—C4	179.3 (4)	O2—C12—C14—C15	179.6 (4)
C1—C2—C7—C8	178.2 (4)	C11—C12—C14—O1	179.6 (4)
C3—C2—C7—C8	-2.8 (7)	C11—C12—C14—C15	-1.0 (6)
C2—C3—C4—C5	1.0 (7)	O1—C14—C15—C16	178.7 (4)
C3—C4—C5—C6	0.5 (6)	C12—C14—C15—C16	-0.5 (6)
C4—C5—C6—C12	179.1 (3)	C14—C15—C16—C10	2.3 (6)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C3—H3 \cdots O3 ⁱ	0.93	2.46	3.185 (5)	135

C7—H7…C11	0.93	2.62	3.061 (4)	109
C7—H7…O3	0.93	2.49	2.808 (5)	100

Symmetry code: (i) $-x+1/2, y+1/2, z+1/2$.