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4-[(4-Hydroxymethyl-2H-1,2,3-triazol-2-yl)methyl]-6,8-dimethyl-2H-chromen-2-one. Corrigendum

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In the paper by El-Khatatneh et al. [IUCrData (2016), 1, x161644], the scheme and chemical name in the title are corrected.

In the paper by El-Khatatneh et al. (2016), the chemical scheme should be as shown here.



The chemical name in the title is then corrected as '4-[(4-Hydroxymethyl-1H-1,2,3triazol-1-yl)methyl]-6,8-dimethyl-2H-chromen-2-one'.

References

El-Khatatneh, N., Chandra, Shamala, D., Shivashankar, K. & Mahendra, M. (2016). IUCrData, 1, x161644.



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4-[(4-Hydroxymethyl-2*H*-1,2,3-triazol-2-yl)methyl]-6,8-dimethyl-2*H*-chromen-2-one

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In the title compound, $C_{15}H_{15}N_3O_3$, the dihedral angle between the triazole ring and coumarin ring system [r.m.s. deviation = 0.040 Å] is 77.40 (6)°. The O atom of the hydroxymethyl group deviates from the triazole ring plane by 1.345 (1) Å. In the crystal, inversion dimers linked by pairs of $O-H\cdots O$ hydrogen bonds generate $R_2^2(22)$ loops; $C-H\cdots O$ and $C-H\cdots N$ interactions link the dimers into a three-dimensional network.



Structure description

Coumarin derivatives represent an important class of natural and synthetic heterocycles that are often linked to a broad array of biological activities (Gaspar *et al.*, 2015). As part of our ongoing studies of coumarin–triazole derivatives (El-Khatatneh *et al.*, 2016), the title compound (Fig. 1) was synthesized and its crystal structure is now reported.

The dihedral angle between the triazole ring and coumarin ring system [r.m.s. deviation = 0.040 Å] is 77.40 (6)°. Key inter-ring torsion angles include 97.34 (15)° for N19–N15–C14–C13) and -173.30 (13)° for C6–C13–C14–N15. The O atom of the hydroxymethyl group is displaced from the triazole ring plane by 1.345 (1) Å.

In the crystal, inversion dimers linked by pairs of $O-H\cdots O$ hydrogen bonds (Table 1) generate $R_2^2(22)$ loops. The dimers are linked by weak $C-H\cdots O$ and $C-H\cdots N$ hydrogen bonds, generating a three-dimensional network (Fig. 2).

Synthesis and crystallization

A mixture of propargyl alcohol (1.9 mmol), sodium azide (0.14 g, 2.0 mmol), copper(I) iodide (10 mol%) and triethylamine (0.19 g, 1.9 mmol) in 20 ml of acetone was taken in a



Figure 1

The molecular structure of the title compound with 50% probability displacement ellipsoids.

round-bottom flask and stirred for 1 h. To this mixture, 4-bromomethylcoumarin (1.9 mmol) was added and the stirring continued for 8 h (the reaction was monitored by TLC). After the completion of the reaction, the copper catalyst was filtered through celite and the product was extracted with diethyl ether (3.10 ml). The solvent was removed under vacuum. The crude product was dried and recrystallized from ethyl acetate solution to give colourless blocks.

Yield 92%; colourless solid; m.p. 210–212 °C; IR (KBr, cm⁻¹): 1742 cm⁻¹ (lactone C=O), 3311 cm⁻¹ (OH); ¹H NMR (400 MHz, CDCl₃): δ 1.70 (*s*, 1H, OH), 2.37 (*s*, 3H, C₆–CH₃) 2.42 (*s*, 3H, C₈–CH₃) 4.83 (*s*, 2H, –CH₂O–), 5.43 (*s*, 1H, C₃–H), 5.70 (*s*, 2H, –CH₂N–), 7.21-7.24 (*m*, 1H, C₇–H), 7.60 (*s*, 1H, C₅–H), 7.75 (*s*, 1H, Tr–H) p.p.m. ¹³C NMR (100 MHz, DMSO-*d*₆): δ 15.0, 20.3, 49.0, 55.0, 113.0, 116.5, 122.0, 123.8, 125.3, 133.1, 134.5, 148.6, 149.5, 150.6, 159.5 p.p.m. Analysis



Figure 2

The packing viewed along [100] with hydrogen bonds indicated by dashed lines.

Table 1	
Hydrogen-bond geometry (Å, °).	

$D-\mathrm{H}\cdots A$ $D-\mathrm{H}$ $\mathrm{H}\cdots$	$A \qquad D \cdots A \qquad D - H \cdots$	·A
$O21-H21\cdots O11^{i}$ 0.82 2.10	2.9155 (19) 176	
$C14 - H14A \cdots N19^{ii}$ 0.97 2.55	3.486 (2) 162	
$C14 - H14B \cdots N18^{iii}$ 0.97 2.41	3.344 (2) 162	
$C16-H16\cdots O21^{iii}$ 0.93 2.47	3.284 (2) 146	

Symmetry codes: (i) -x, -y + 2, -z + 1; (ii) -x + 1, -y + 2, -z + 2; (iii) x + 1, y, z.

 Table 2

 Experimental details

1	
Crystal data	
Chemical formula	$C_{15}H_{15}N_3O_3$
$M_{ m r}$	285.30
Crystal system, space group	Triclinic, P1
Temperature (K)	293
<i>a</i> , <i>b</i> , <i>c</i> (Å)	6.0265 (16), 11.062 (3), 11.848 (3)
α, β, γ (°)	108.812 (7), 103.950 (8),
	100.848 (8)
$V(Å^3)$	694.5 (3)
Z	2
Radiation type	Cu Ka
$\mu \text{ (mm}^{-1})$	0.80
Crystal size (mm)	$0.30 \times 0.20 \times 0.10$
Data collection	
Diffractometer	Bruker V& Proteum
No. of measured independent and	2019 2017 2142
No. of measured, independent and observed $[L > 2\pi(I)]$ reflections	8218, 2217, 2142
P	0.030
$(\operatorname{cin} \theta/1) = (\mathring{A}^{-1})$	0.050
$(\sin \theta/\lambda)_{\max}(\mathbf{A})$	0.387
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.042, 0.114, 1.04
No. of reflections	2217
No. of parameters	194
H-atom treatment	H-atom parameters constrained
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} ({ m e} { m \AA}^{-3})$	0.14, -0.14

Computer programs: APEX2 and SAINT (Bruker, 2009), SHELXS (Sheldrick, 2008), SHELXL2016/4 (Sheldrick, 2015), PLATON (Spek, 2009) and Mercury (Macrae et al., 2008).

 $C_{15}H_{15}N_3O_3$. Calculated for: C, 63.15; H, 5.30; N, 14.73%. Found: C, 63.08; H, 5.26; N, 14.68%.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

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full crystallographic data

IUCrData (2016). **1**, x161644 [https://doi.org/10.1107/S2414314616016448]

4-[(4-Hydroxymethyl-2*H*-1,2,3-triazol-2-yl)methyl]-6,8-dimethyl-2*H*chromen-2-one

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4-[(4-Hydroxymethyl-2H-1,2,3-triazol-2-yl)methyl]-6,8-dimethyl-2H-chromen-2-one

Crystal data

 $C_{15}H_{15}N_{3}O_{3}$ $M_{r} = 285.30$ Triclinic, *P*1 a = 6.0265 (16) Å b = 11.062 (3) Å c = 11.848 (3) Å $a = 108.812 (7)^{\circ}$ $\beta = 103.950 (8)^{\circ}$ $\gamma = 100.848 (8)^{\circ}$ $V = 694.5 (3) \text{ Å}^{3}$

Data collection

Bruker X8 Proteum diffractometer Radiation source: Bruker MicroStar microfocus rotating anode Helios multilayer optics monochromator Detector resolution: 18.4 pixels mm⁻¹ φ and ω scans 8218 measured reflections

Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.042$ $wR(F^2) = 0.114$ S = 1.042217 reflections 194 parameters 0 restraints Primary atom site location: structure-invariant direct methods Z = 2 F(000) = 300 $D_x = 1.364 \text{ Mg m}^{-3}$ Cu Ka radiation, $\lambda = 1.54178 \text{ Å}$ Cell parameters from 2217 reflections $\theta = 7.2-64.7^{\circ}$ $\mu = 0.80 \text{ mm}^{-1}$ T = 293 K Block, colourless $0.30 \times 0.20 \times 0.10 \text{ mm}$

2217 independent reflections 2142 reflections with $I > 2\sigma(I)$ $R_{int} = 0.030$ $\theta_{max} = 64.7^{\circ}, \ \theta_{min} = 7.2^{\circ}$ $h = -6 \rightarrow 7$ $k = -12 \rightarrow 12$ $l = -13 \rightarrow 13$

Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0622P)^2 + 0.126P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.14 \text{ e } \text{Å}^{-3}$ $\Delta\rho_{min} = -0.14 \text{ e } \text{Å}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	Occ. (<1)
09	0.37402 (19)	0.76949 (10)	0.45209 (9)	0.0574 (3)	
O21	-0.04447 (19)	1.36437 (12)	0.82987 (11)	0.0658 (3)	
H21	-0.055031	1.305321	0.763928	0.099*	
011	0.0867 (2)	0.85556 (13)	0.39863 (11)	0.0757 (4)	
C6	0.6271 (2)	0.84311 (13)	0.66749 (13)	0.0452 (3)	
N19	0.2818 (2)	1.08969 (13)	0.92565 (13)	0.0596 (4)	
N15	0.44456 (19)	1.12365 (11)	0.87345 (10)	0.0450 (3)	
C13	0.5024 (2)	0.94266 (13)	0.70280 (13)	0.0458 (3)	
C5	0.5580 (3)	0.75918 (14)	0.54097 (13)	0.0485 (3)	
N18	0.1788 (2)	1.18550 (14)	0.94803 (13)	0.0598 (4)	
C1	0.8117 (3)	0.82578 (14)	0.75347 (14)	0.0500 (4)	
H1	0.860125	0.880475	0.838505	0.060*	
C12	0.3208 (3)	0.94819 (14)	0.61462 (14)	0.0523 (4)	
H12	0.240245	1.011061	0.638265	0.063*	
C4	0.6663 (3)	0.66137 (15)	0.49641 (14)	0.0560 (4)	
C2	0.9229 (3)	0.72895 (15)	0.71413 (15)	0.0542 (4)	
C17	0.2756 (2)	1.28047 (13)	0.91088 (12)	0.0454 (3)	
C20	0.1846 (3)	1.39858 (15)	0.91808 (14)	0.0556 (4)	
H20A	0.178228	1.440812	1.002359	0.067*	
H20B	0.295413	1.462794	0.903203	0.067*	
C14	0.5903 (3)	1.03629 (14)	0.83948 (13)	0.0517 (4)	
H14A	0.597759	0.983779	0.891184	0.062*	
H14B	0.751444	1.090828	0.859203	0.062*	
C10	0.2480 (3)	0.85893 (15)	0.48426 (14)	0.0552 (4)	
C3	0.8477 (3)	0.64959 (16)	0.58574 (16)	0.0614 (4)	
H3	0.923601	0.585243	0.558673	0.074*	
C16	0.4471 (2)	1.24121 (13)	0.86378 (13)	0.0477 (3)	
H16	0.545138	1.286513	0.831682	0.057*	
C7	1.1162 (3)	0.70920 (19)	0.80743 (18)	0.0718 (5)	
H7A	1.168804	0.636256	0.764247	0.108*	0.31 (2)
H7B	1.055076	0.689718	0.869294	0.108*	0.31 (2)
H7C	1.248294	0.789024	0.848366	0.108*	0.31 (2)
H7D	1.145979	0.773742	0.890357	0.108*	0.69 (2)
H7E	1.259707	0.720281	0.785311	0.108*	0.69 (2)
H7F	1.066488	0.620975	0.806239	0.108*	0.69 (2)
C8	0.5915 (4)	0.57393 (19)	0.35871 (16)	0.0760 (5)	
H8A	0.690144	0.613072	0.319554	0.114*	
H8B	0.427523	0.565734	0.318815	0.114*	
H8C	0.609121	0.487213	0.349891	0.114*	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (A^2)

U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
0.0673 (7)	0.0594 (6)	0.0445 (5)	0.0320 (5)	0.0145 (5)	0.0135 (4)
0.0540 (6)	0.0732 (7)	0.0627 (7)	0.0362 (5)	0.0113 (5)	0.0124 (5)
0.0765 (8)	0.0868 (8)	0.0507 (6)	0.0440 (7)	0.0024 (6)	0.0120 (6)
0.0468 (7)	0.0422 (7)	0.0487 (7)	0.0192 (6)	0.0170 (6)	0.0156 (6)
0.0560 (7)	0.0609 (7)	0.0744 (9)	0.0269 (6)	0.0276 (6)	0.0316 (6)
0.0420 (6)	0.0456 (6)	0.0436 (6)	0.0198 (5)	0.0113 (5)	0.0105 (5)
0.0457 (7)	0.0434 (7)	0.0477 (7)	0.0186 (6)	0.0143 (6)	0.0143 (6)
0.0549 (8)	0.0481 (7)	0.0481 (8)	0.0228 (6)	0.0189 (6)	0.0197 (6)
0.0546 (7)	0.0692 (8)	0.0696 (8)	0.0322 (6)	0.0297 (6)	0.0294 (7)
0.0504 (8)	0.0486 (7)	0.0508 (8)	0.0237 (6)	0.0155 (6)	0.0144 (6)
0.0518 (8)	0.0524 (8)	0.0514 (8)	0.0272 (6)	0.0136 (6)	0.0136 (6)
0.0723 (10)	0.0526 (8)	0.0518 (8)	0.0309 (7)	0.0283 (7)	0.0180 (7)
0.0564 (8)	0.0527 (8)	0.0607 (9)	0.0290 (7)	0.0213 (7)	0.0218 (7)
0.0420 (7)	0.0472 (7)	0.0401 (7)	0.0183 (6)	0.0094 (5)	0.0082 (5)
0.0553 (8)	0.0534 (8)	0.0509 (8)	0.0274 (6)	0.0126 (6)	0.0080 (6)
0.0488 (8)	0.0502 (8)	0.0502 (8)	0.0271 (6)	0.0099 (6)	0.0093 (6)
0.0560 (8)	0.0583 (8)	0.0493 (8)	0.0268 (7)	0.0117 (7)	0.0163 (7)
0.0766 (11)	0.0591 (9)	0.0647 (9)	0.0434 (8)	0.0329 (8)	0.0237 (7)
0.0467 (7)	0.0452 (7)	0.0535 (8)	0.0203 (6)	0.0196 (6)	0.0156 (6)
0.0739 (11)	0.0756 (11)	0.0714 (11)	0.0479 (9)	0.0187 (9)	0.0242 (9)
0.1070 (15)	0.0748 (11)	0.0542 (10)	0.0497 (11)	0.0331 (10)	0.0167 (8)
	U^{11} 0.0673 (7) 0.0540 (6) 0.0765 (8) 0.0468 (7) 0.0560 (7) 0.0420 (6) 0.0457 (7) 0.0549 (8) 0.0546 (7) 0.0504 (8) 0.0518 (8) 0.0723 (10) 0.0564 (8) 0.0420 (7) 0.0553 (8) 0.0488 (8) 0.0560 (8) 0.0766 (11) 0.0467 (7) 0.0739 (11) 0.1070 (15)	U^{11} U^{22} 0.0673 (7) 0.0594 (6) 0.0540 (6) 0.0732 (7) 0.0765 (8) 0.0868 (8) 0.0468 (7) 0.0422 (7) 0.0560 (7) 0.0609 (7) 0.0420 (6) 0.0456 (6) 0.0457 (7) 0.0434 (7) 0.0549 (8) 0.0481 (7) 0.0546 (7) 0.0692 (8) 0.0546 (7) 0.0692 (8) 0.0518 (8) 0.0524 (8) 0.0526 (8) 0.0526 (8) 0.0564 (8) 0.0527 (8) 0.0420 (7) 0.0472 (7) 0.0553 (8) 0.0534 (8) 0.0560 (8) 0.0583 (8) 0.0766 (11) 0.0756 (11) 0.0739 (11) 0.0756 (11) 0.1070 (15) 0.0748 (11)	U^{11} U^{22} U^{33} 0.0673 (7) 0.0594 (6) 0.0445 (5) 0.0540 (6) 0.0732 (7) 0.0627 (7) 0.0765 (8) 0.0868 (8) 0.0507 (6) 0.0468 (7) 0.0422 (7) 0.0487 (7) 0.0560 (7) 0.0609 (7) 0.0744 (9) 0.0420 (6) 0.0456 (6) 0.0436 (6) 0.0457 (7) 0.0434 (7) 0.0477 (7) 0.0549 (8) 0.0481 (7) 0.0481 (8) 0.0546 (7) 0.0692 (8) 0.0696 (8) 0.0504 (8) 0.0486 (7) 0.0508 (8) 0.0518 (8) 0.0524 (8) 0.0514 (8) 0.0564 (8) 0.0527 (8) 0.0607 (9) 0.0420 (7) 0.0472 (7) 0.0401 (7) 0.0553 (8) 0.0534 (8) 0.0502 (8) 0.0560 (8) 0.0592 (8) 0.0502 (8) 0.0560 (8) 0.0534 (8) 0.0502 (8) 0.0560 (8) 0.0593 (8) 0.0493 (8) 0.0766 (11) 0.0591 (9) 0.0647 (9) 0.0467 (7) 0.0452 (7) 0.0535 (8) 0.0739 (11) 0.0748 (11) 0.0542 (10)	U^{11} U^{22} U^{33} U^{12} 0.0673 (7)0.0594 (6)0.0445 (5)0.0320 (5)0.0540 (6)0.0732 (7)0.0627 (7)0.0362 (5)0.0765 (8)0.0868 (8)0.0507 (6)0.0440 (7)0.0468 (7)0.0422 (7)0.0487 (7)0.0192 (6)0.0560 (7)0.0609 (7)0.0744 (9)0.0269 (6)0.0420 (6)0.0456 (6)0.0436 (6)0.0198 (5)0.0457 (7)0.0434 (7)0.0477 (7)0.0186 (6)0.0549 (8)0.0481 (7)0.0481 (8)0.0228 (6)0.0544 (7)0.0692 (8)0.0696 (8)0.0322 (6)0.0504 (8)0.0524 (8)0.0514 (8)0.0272 (6)0.0723 (10)0.0526 (8)0.0518 (8)0.0272 (6)0.0553 (8)0.0527 (8)0.0607 (9)0.0290 (7)0.0420 (7)0.0472 (7)0.0401 (7)0.0183 (6)0.0553 (8)0.0522 (8)0.0502 (8)0.0271 (6)0.0560 (8)0.0534 (8)0.0502 (8)0.0268 (7)0.0560 (8)0.0553 (8)0.0502 (8)0.0271 (6)0.0560 (8)0.0583 (8)0.0493 (8)0.0268 (7)0.0766 (11)0.0591 (9)0.0647 (9)0.0434 (8)0.0467 (7)0.0452 (7)0.0535 (8)0.0203 (6)0.0739 (11)0.0756 (11)0.0714 (11)0.0479 (9)0.1070 (15)0.0748 (11)0.0542 (10)0.0497 (11)	U^{11} U^{22} U^{33} U^{12} U^{13} 0.0673 (7)0.0594 (6)0.0445 (5)0.0320 (5)0.0145 (5)0.0540 (6)0.0732 (7)0.0627 (7)0.0362 (5)0.0113 (5)0.0765 (8)0.0868 (8)0.0507 (6)0.0440 (7)0.0024 (6)0.0468 (7)0.0422 (7)0.0487 (7)0.0192 (6)0.0170 (6)0.0560 (7)0.0609 (7)0.0744 (9)0.0269 (6)0.0276 (6)0.0420 (6)0.0456 (6)0.0436 (6)0.0198 (5)0.0113 (5)0.0457 (7)0.0434 (7)0.0477 (7)0.0186 (6)0.0143 (6)0.0549 (8)0.0481 (7)0.0481 (8)0.0228 (6)0.0189 (6)0.0546 (7)0.0692 (8)0.0696 (8)0.0322 (6)0.0297 (6)0.0504 (8)0.0524 (8)0.0514 (8)0.0272 (6)0.0136 (6)0.0723 (10)0.0526 (8)0.0518 (8)0.0309 (7)0.0283 (7)0.0420 (7)0.0472 (7)0.0401 (7)0.0183 (6)0.0094 (5)0.0553 (8)0.0524 (8)0.0509 (8)0.0271 (6)0.0126 (6)0.0488 (8)0.0502 (8)0.0502 (8)0.0271 (6)0.0126 (6)0.0560 (8)0.0583 (8)0.0493 (8)0.0268 (7)0.0117 (7)0.0766 (11)0.0591 (9)0.0647 (9)0.0434 (8)0.0329 (8)0.0467 (7)0.0452 (7)0.0535 (8)0.0203 (6)0.0196 (6)0.0739 (11)0.0756 (11)0.0714 (11)0.0497 (11)0.0331 (10)

Atomic displacement parameters $(Å^2)$

Geometric parameters (Å, °)

O9—C10	1.3712 (18)	C4—C8	1.505 (2)
O9—C5	1.3833 (18)	C2—C3	1.395 (2)
O21—C20	1.4118 (18)	C2—C7	1.503 (2)
O21—H21	0.8200	C17—C16	1.3635 (19)
O11—C10	1.2089 (19)	C17—C20	1.4952 (19)
C6—C5	1.392 (2)	C20—H20A	0.9700
C6—C1	1.404 (2)	C20—H20B	0.9700
C6—C13	1.4554 (18)	C14—H14A	0.9700
N19—N18	1.3135 (18)	C14—H14B	0.9700
N19—N15	1.3375 (17)	С3—Н3	0.9300
N15-C16	1.3395 (18)	C16—H16	0.9300
N15-C14	1.4513 (16)	С7—Н7А	0.9600
C13—C12	1.345 (2)	С7—Н7В	0.9600
C13—C14	1.5080 (19)	С7—Н7С	0.9600
C5—C4	1.392 (2)	C7—H7D	0.9600
N18—C17	1.352 (2)	С7—Н7Е	0.9600
C1—C2	1.3829 (19)	C7—H7F	0.9600
C1—H1	0.9300	C8—H8A	0.9600
C12—C10	1.442 (2)	C8—H8B	0.9600
С12—Н12	0.9300	C8—H8C	0.9600
C4—C3	1.384 (2)		

C10—O9—C5	122.09 (11)	N15-C14-H14B	108.6
C20—O21—H21	109.5	C13—C14—H14B	108.6
C5—C6—C1	117.93 (12)	H14A—C14—H14B	107.6
C5—C6—C13	118.08 (12)	O11—C10—O9	115.82 (13)
C1—C6—C13	123.98 (12)	O11—C10—C12	126.45 (14)
N18—N19—N15	106.63 (11)	O9—C10—C12	117.73 (13)
N19—N15—C16	110.78 (11)	C4—C3—C2	123.61 (13)
N19—N15—C14	119.15 (12)	C4—C3—H3	118.2
C16—N15—C14	130.04 (12)	C2—C3—H3	118.2
C12—C13—C6	119.78 (12)	N15—C16—C17	105.51 (12)
C12—C13—C14	123.64 (12)	N15-C16-H16	127.2
C6-C13-C14	116.58 (11)	C17—C16—H16	127.2
O9—C5—C6	120.68 (12)	С2—С7—Н7А	109.5
O9—C5—C4	116.33 (13)	С2—С7—Н7В	109.5
C6—C5—C4	122.99 (13)	H7A—C7—H7B	109.5
N19—N18—C17	109.68 (12)	С2—С7—Н7С	109.5
C2C1C6	121.27 (14)	H7A—C7—H7C	109.5
C2	119.4	H7B—C7—H7C	109.5
С6—С1—Н1	119.4	C2—C7—H7D	109.5
C13—C12—C10	121.56 (13)	H7A—C7—H7D	141.1
C13—C12—H12	119.2	H7B—C7—H7D	56.3
C10-C12-H12	119.2	H7C—C7—H7D	56.3
C3—C4—C5	116.29 (14)	С2—С7—Н7Е	109.5
C3—C4—C8	121.84 (14)	H7A—C7—H7E	56.3
C5—C4—C8	121.87 (14)	H7B—C7—H7E	141.1
C1—C2—C3	117.89 (14)	H7C—C7—H7E	56.3
C1—C2—C7	120.78 (14)	H7D—C7—H7E	109.5
C3—C2—C7	121.33 (13)	C2—C7—H7F	109.5
N18—C17—C16	107.39 (12)	H7A—C7—H7F	56.3
N18—C17—C20	121.83 (13)	H7B—C7—H7F	56.3
C16—C17—C20	130.68 (14)	H7C—C7—H7F	141.1
O21—C20—C17	112.70 (12)	H7D—C7—H7F	109.5
O21—C20—H20A	109.1	H7E—C7—H7F	109.5
C17—C20—H20A	109.1	C4—C8—H8A	109.5
O21—C20—H20B	109.1	C4—C8—H8B	109.5
C17—C20—H20B	109.1	H8A—C8—H8B	109.5
H20A—C20—H20B	107.8	C4—C8—H8C	109.5
N15-C14-C13	114.71 (11)	H8A—C8—H8C	109.5
N15-C14-H14A	108.6	H8B—C8—H8C	109.5
C13—C14—H14A	108.6		

Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H···A	D····A	<i>D</i> —H··· <i>A</i>
021—H21…O11 ⁱ	0.82	2.10	2.9155 (19)	176
C14—H14 A ···N19 ⁱⁱ	0.97	2.55	3.486 (2)	162

				data reports
C14—H14 <i>B</i> ····N18 ⁱⁱⁱ	0.97	2.41	3.344 (2)	162
С16—Н16…О21ііі	0.93	2.47	3.284 (2)	146

Symmetry codes: (i) -*x*, -*y*+2, -*z*+1; (ii) -*x*+1, -*y*+2, -*z*+2; (iii) *x*+1, *y*, *z*.