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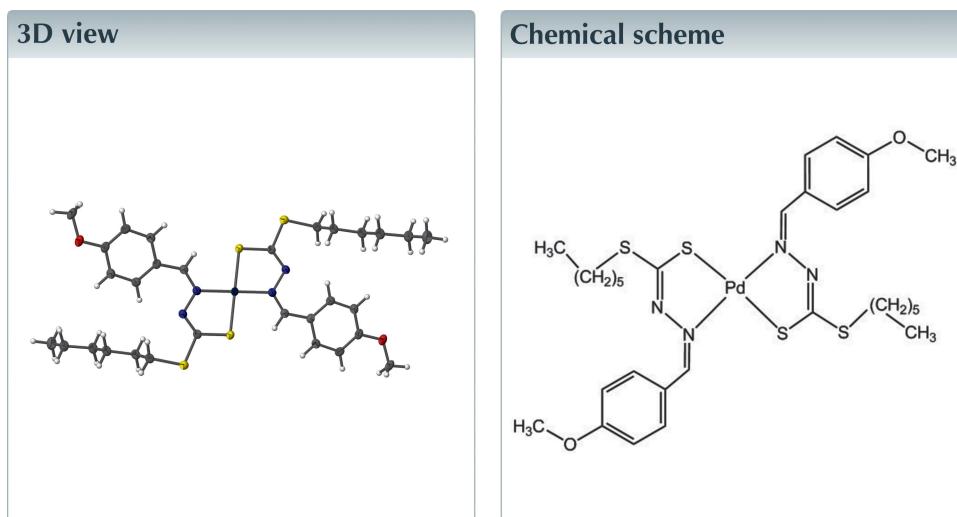
Structural data: full structural data are available from iucrdata.iucr.org

Bis[*S*-hexyl (*E*)-3-(4-methoxybenzylidene)dithiocarbazato- $\kappa^2 N^3, S$]palladium(II)

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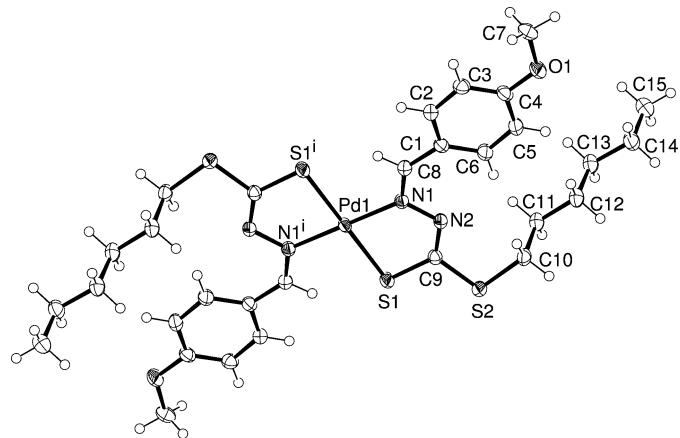
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In the title complex, $[Pd(C_{15}H_{21}N_2OS_2)_2]$, the Pd^{II} ion is located on a crystallographic inversion center. The two Schiff base ligands in a deprotonated imino thiolate form chelate the metal atom *via* the azomethine N and thiolate S atoms in a *trans*-square-planar configuration as imposed by the crystal symmetry. The complex has an approximately planar geometry with the exception of the hexyl chains; the mean plane of the five-membered chelate ring makes a dihedral angle of $15.43(5)^\circ$ with the benzene ring. In the crystal, the complex molecules are stacked along the *a* axis.



Structure description

The molecular structure of the title compound is illustrated in Fig. 1. In the complex, the $Pd-S$ and $Pd-N$ bond lengths are $2.2949(5)$ and $2.0358(14)$ Å, respectively, with the $S1-Pd1-N1$ chelating angle of $83.13(5)^\circ$. These values are in agreement with those observed in similar bis(dithiocarbazato) Pd^{II} complexes, either with a *trans* configuration of ligands (Khaledi & Mohd Ali, 2011; Tampouris *et al.*, 2007; Tarafder *et al.*, 2010) or with a *cis* configuration (Begum, Howlader, Sheikh *et al.*, 2015; Ali *et al.*, 2002; Liu *et al.*, 2011; Duan *et al.*, 1998; Tampouris *et al.*, 2007). The ligand recently reported (Begum, Howlader, Miyatake *et al.*, 2015) rotates about the $C9-N2$ bond by 180° in order to allow the N,S chelating behavior towards the metal. Upon coordination some salient features are observed, compared to the free ligand. The most significant one is an elongation of the $C9-S1$ bond length; the $C9-S1$ distance of $1.7311(17)$ Å in the present PdL_2 complex is longer than that of $1.6713(19)$ Å in the ligand HL , validating the coordination with the deprotonated thiolate S atom. Correspondingly, the present $N2-C9$ and $N1-N2$ bond

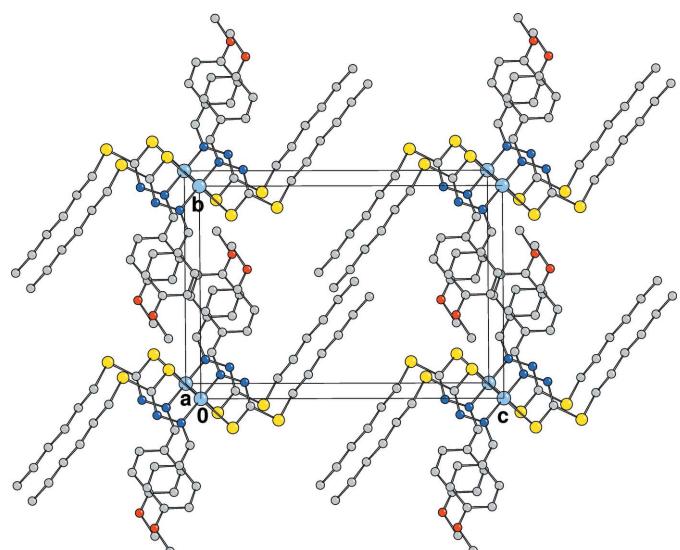
**Figure 1**

The molecular structure of the title compound, showing displacement ellipsoids at the 50% probability level [symmetry code: (i) $-x, -y, -z$].

lengths of 1.284 (3) and 1.411 (3) Å, respectively, are shorter and slightly longer than the values of 1.343 (3) and 1.377 (2) Å in the free ligand. In the crystal, the molecules are stacked along the *a* axis (Fig. 2).

Synthesis and crystallization

A solution of PdCl_2 (0.044 g, 0.25 mmol) in 25 ml methanol was added to a solution of *S*-hexyl (*E*)-3-(4-methoxybenzylidene)dithiocarbazate (0.155 g, 0.5 mmol) in 10 ml methanol. The resulting mixture was refluxed with constant stirring for 4 h. The orange-red precipitate formed was filtered off, washed with methanol and dried *in vacuo* over anhydrous CaCl_2 . Orange-red single crystals of the title compound suitable for X-ray diffraction were obtained by slow evaporation from a solution in a mixture of dichloromethane and acetonitrile (2:1) (m.p. 457 K).

**Figure 2**

A packing view of the title compound along the *a* axis.

Table 1
Experimental details.

Crystal data	[$\text{Pd}(\text{C}_{15}\text{H}_{21}\text{N}_2\text{OS}_2)_2$]
Chemical formula	$\text{C}_{72}\text{H}_{134}\text{N}_2\text{O}_4\text{Pd}_2$
M_r	725.33
Crystal system, space group	Triclinic, $P\bar{1}$
Temperature (K)	173
a, b, c (Å)	4.54371 (18), 11.6697 (6), 15.5677 (8)
α, β, γ (°)	90.8429 (14), 98.1733 (18), 100.3441 (11)
V (Å ³)	803.10 (7)
Z	1
Radiation type	Mo $K\alpha$
μ (mm ⁻¹)	0.87
Crystal size (mm)	0.20 × 0.09 × 0.02
Data collection	
Diffractometer	Rigaku R-AXIS RAPID
Absorption correction	Multi-scan (ABSCOR; Higashi, 1995)
T_{\min}, T_{\max}	0.891, 0.983
No. of measured, independent and observed [$F^2 > 2\sigma(F^2)$] reflections	6642, 2935, 2802
R_{int}	0.017
(sin θ/λ) _{max} (Å ⁻¹)	0.602
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.021, 0.057, 1.15
No. of reflections	2935
No. of parameters	189
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å ⁻³)	0.43, -0.23

Computer programs: RAPID-AUTO (Rigaku, 2001), SIR92 (Altomare *et al.*, 1993), SHELXL97 (Sheldrick, 2008), ORTEP-3 for Windows (Farrugia, 2012) and publCIF (Westrip, 2010).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 1.

Acknowledgements

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full crystallographic data

IUCrData (2016). **1**, x160521 [doi:10.1107/S2414314616005216]

Bis[*S*-hexyl (*E*)-3-(4-methoxybenzylidene)dithiocarbazato- κ^2N^3,S]palladium(II)

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Bis[*S*-hexyl (*E*)-3-(4-methoxybenzylidene)dithiocarbazato- κ^2N^3,S]palladium(II)

Crystal data

[Pd(C₁₅H₂₁N₂OS₂)₂]

$M_r = 725.33$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 4.54371$ (18) Å

$b = 11.6697$ (6) Å

$c = 15.5677$ (8) Å

$\alpha = 90.8429$ (14)°

$\beta = 98.1733$ (18)°

$\gamma = 100.3441$ (11)°

$V = 803.10$ (7) Å³

$Z = 1$

$F(000) = 376.00$

$D_x = 1.500$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71075$ Å

Cell parameters from 7004 reflections

$\theta = 3.1\text{--}27.5$ °

$\mu = 0.87$ mm⁻¹

$T = 173$ K

Platelet, orange

0.20 × 0.09 × 0.02 mm

Data collection

Rigaku R-AXIS RAPID
diffractometer

Detector resolution: 10.000 pixels mm⁻¹

ω scans

Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.891$, $T_{\max} = 0.983$

6642 measured reflections

2935 independent reflections

2802 reflections with $F^2 > 2\sigma(F^2)$

$R_{\text{int}} = 0.017$

$\theta_{\max} = 25.3$ °

$h = -5 \rightarrow 5$

$k = -14 \rightarrow 14$

$l = -18 \rightarrow 18$

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.021$

$wR(F^2) = 0.057$

$S = 1.15$

2935 reflections

189 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier

map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0251P)^2 + 0.3967P]$
where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.43$ e Å⁻³

$\Delta\rho_{\min} = -0.23$ e Å⁻³

Special details

Geometry. ENTER SPECIAL DETAILS OF THE MOLECULAR GEOMETRY

Refinement. Refinement was performed using all reflections. The weighted R-factor (wR) and goodness of fit (S) are based on F^2 . R-factor (gt) are based on F. The threshold expression of $F^2 > 2.0$ sigma(F^2) is used only for calculating R-factor (gt).

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Pd1	0.0000	0.0000	0.0000	0.02080 (8)
S1	-0.00459 (11)	-0.13653 (4)	0.10464 (3)	0.02995 (12)
S2	0.36335 (11)	-0.12720 (4)	0.27412 (3)	0.02920 (12)
O1	1.5353 (4)	0.49771 (13)	0.23168 (10)	0.0379 (4)
N1	0.3597 (4)	0.08710 (13)	0.08319 (10)	0.0223 (4)
N2	0.4365 (4)	0.04560 (14)	0.16655 (10)	0.0239 (4)
C1	0.7997 (4)	0.25574 (16)	0.11173 (12)	0.0235 (4)
C2	0.9437 (5)	0.34306 (17)	0.06407 (12)	0.0270 (4)
C3	1.1907 (5)	0.42501 (17)	0.10040 (13)	0.0282 (4)
C4	1.2993 (5)	0.42088 (17)	0.18824 (13)	0.0280 (4)
C5	1.1622 (5)	0.33362 (18)	0.23675 (13)	0.0330 (5)
C6	0.9169 (5)	0.25205 (18)	0.19998 (13)	0.0297 (5)
C7	1.6597 (5)	0.59746 (18)	0.18742 (15)	0.0383 (5)
C8	0.5364 (4)	0.17987 (16)	0.06361 (12)	0.0241 (4)
C9	0.2833 (4)	-0.05543 (16)	0.17755 (11)	0.0229 (4)
C10	0.6641 (5)	-0.02298 (18)	0.33661 (12)	0.0281 (4)
C11	0.5519 (5)	0.07304 (18)	0.38258 (12)	0.0283 (5)
C12	0.8067 (5)	0.15415 (18)	0.44003 (12)	0.0297 (5)
C13	0.6976 (5)	0.24810 (19)	0.48976 (13)	0.0320 (5)
C14	0.9466 (5)	0.32555 (19)	0.55097 (14)	0.0349 (5)
C15	0.8324 (6)	0.4187 (2)	0.59973 (16)	0.0448 (6)
H1	0.8693	0.3465	0.0041	0.0325*
H2	1.2847	0.4832	0.0659	0.0339*
H3	1.2389	0.3300	0.2965	0.0396*
H4	0.8266	0.1930	0.2344	0.0356*
H5	1.5003	0.6415	0.1674	0.0460*
H6	1.7440	0.5720	0.1374	0.0460*
H7	1.8201	0.6472	0.2272	0.0460*
H8	0.4834	0.2032	0.0061	0.0289*
H9	0.7793	-0.0649	0.3805	0.0337*
H10	0.8044	0.0130	0.2973	0.0337*
H11	0.4001	0.0372	0.4187	0.0340*
H12	0.4508	0.1192	0.3386	0.0340*
H13	0.9539	0.1922	0.4033	0.0357*
H14	0.9133	0.1072	0.4821	0.0357*
H15	0.6007	0.2975	0.4475	0.0384*
H16	0.5417	0.2100	0.5238	0.0384*
H17	1.1027	0.3639	0.5171	0.0418*
H18	1.0433	0.2764	0.5936	0.0418*
H19	0.6890	0.3813	0.6368	0.0537*
H20	0.7311	0.4663	0.5580	0.0537*
H21	1.0035	0.4684	0.6358	0.0537*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Pd1	0.02019 (11)	0.02079 (12)	0.01926 (11)	-0.00035 (8)	0.00101 (7)	-0.00108 (7)
S1	0.0326 (3)	0.0252 (3)	0.0254 (3)	-0.0065 (2)	-0.00314 (19)	0.00252 (19)
S2	0.0342 (3)	0.0249 (3)	0.0250 (3)	0.0003 (2)	-0.00123 (19)	0.00403 (19)
O1	0.0355 (8)	0.0308 (8)	0.0380 (9)	-0.0094 (7)	-0.0067 (7)	0.0002 (7)
N1	0.0230 (8)	0.0230 (8)	0.0203 (8)	0.0034 (7)	0.0022 (6)	0.0000 (6)
N2	0.0255 (8)	0.0242 (9)	0.0202 (8)	0.0023 (7)	0.0000 (6)	0.0011 (6)
C1	0.0220 (9)	0.0219 (10)	0.0256 (10)	0.0031 (8)	0.0019 (7)	-0.0016 (8)
C2	0.0291 (10)	0.0265 (10)	0.0240 (10)	0.0044 (8)	-0.0005 (8)	0.0005 (8)
C3	0.0278 (10)	0.0241 (10)	0.0316 (11)	0.0012 (8)	0.0046 (8)	0.0026 (8)
C4	0.0242 (9)	0.0227 (10)	0.0343 (11)	0.0013 (8)	-0.0005 (8)	-0.0034 (8)
C5	0.0362 (11)	0.0311 (11)	0.0259 (10)	-0.0019 (9)	-0.0049 (9)	0.0012 (9)
C6	0.0312 (10)	0.0277 (11)	0.0266 (10)	-0.0023 (9)	0.0014 (8)	0.0032 (8)
C7	0.0358 (11)	0.0259 (11)	0.0464 (13)	-0.0089 (9)	0.0019 (10)	-0.0016 (10)
C8	0.0237 (9)	0.0253 (10)	0.0225 (9)	0.0041 (8)	0.0010 (7)	0.0008 (8)
C9	0.0239 (9)	0.0240 (10)	0.0207 (9)	0.0050 (8)	0.0030 (7)	-0.0015 (7)
C10	0.0248 (9)	0.0315 (11)	0.0251 (10)	0.0029 (9)	-0.0029 (8)	0.0030 (8)
C11	0.0241 (9)	0.0353 (11)	0.0236 (10)	0.0024 (9)	0.0008 (8)	0.0008 (8)
C12	0.0246 (9)	0.0371 (12)	0.0250 (10)	0.0019 (9)	0.0002 (8)	0.0011 (9)
C13	0.0273 (10)	0.0373 (12)	0.0285 (11)	0.0024 (9)	-0.0008 (8)	-0.0007 (9)
C14	0.0330 (11)	0.0353 (12)	0.0319 (11)	-0.0007 (9)	-0.0009 (9)	-0.0013 (9)
C15	0.0492 (14)	0.0371 (13)	0.0427 (13)	0.0045 (11)	-0.0062 (11)	-0.0071 (10)

Geometric parameters (\AA , ^\circ)

Pd1—S1	2.2949 (5)	C14—C15	1.522 (4)
Pd1—S1 ⁱ	2.2949 (5)	C2—H1	0.950
Pd1—N1	2.0358 (14)	C3—H2	0.950
Pd1—N1 ⁱ	2.0358 (14)	C5—H3	0.950
S1—C9	1.7311 (17)	C6—H4	0.950
S2—C9	1.7564 (19)	C7—H5	0.980
S2—C10	1.8119 (18)	C7—H6	0.980
O1—C4	1.357 (3)	C7—H7	0.980
O1—C7	1.431 (3)	C8—H8	0.950
N1—N2	1.411 (3)	C10—H9	0.990
N1—C8	1.295 (3)	C10—H10	0.990
N2—C9	1.284 (3)	C11—H11	0.990
C1—C2	1.392 (3)	C11—H12	0.990
C1—C6	1.405 (3)	C12—H13	0.990
C1—C8	1.457 (3)	C12—H14	0.990
C2—C3	1.383 (3)	C13—H15	0.990
C3—C4	1.391 (3)	C13—H16	0.990
C4—C5	1.387 (3)	C14—H17	0.990
C5—C6	1.377 (3)	C14—H18	0.990
C10—C11	1.521 (3)	C15—H19	0.980
C11—C12	1.524 (3)	C15—H20	0.980

C12—C13	1.525 (4)	C15—H21	0.980
C13—C14	1.519 (3)		
S1—Pd1—S1 ⁱ	180.00 (3)	O1—C7—H5	109.469
S1—Pd1—N1	83.13 (5)	O1—C7—H6	109.467
S1—Pd1—N1 ⁱ	96.87 (5)	O1—C7—H7	109.464
S1 ⁱ —Pd1—N1	96.87 (5)	H5—C7—H6	109.479
S1 ⁱ —Pd1—N1 ⁱ	83.13 (5)	H5—C7—H7	109.470
N1—Pd1—N1 ⁱ	180.00 (9)	H6—C7—H7	109.477
Pd1—S1—C9	95.72 (7)	N1—C8—H8	113.450
C9—S2—C10	102.89 (9)	C1—C8—H8	113.448
C4—O1—C7	118.18 (16)	S2—C10—H9	108.858
Pd1—N1—N2	120.58 (11)	S2—C10—H10	108.853
Pd1—N1—C8	123.64 (13)	C11—C10—H9	108.859
N2—N1—C8	115.75 (14)	C11—C10—H10	108.849
N1—N2—C9	112.98 (14)	H9—C10—H10	107.706
C2—C1—C6	117.37 (16)	C10—C11—H11	109.100
C2—C1—C8	115.14 (16)	C10—C11—H12	109.105
C6—C1—C8	127.46 (17)	C12—C11—H11	109.106
C1—C2—C3	122.68 (17)	C12—C11—H12	109.103
C2—C3—C4	118.89 (18)	H11—C11—H12	107.845
O1—C4—C3	124.42 (18)	C11—C12—H13	108.895
O1—C4—C5	116.13 (18)	C11—C12—H14	108.898
C3—C4—C5	119.45 (17)	C13—C12—H13	108.898
C4—C5—C6	121.28 (19)	C13—C12—H14	108.893
C1—C6—C5	120.31 (19)	H13—C12—H14	107.732
N1—C8—C1	133.10 (18)	C12—C13—H15	108.748
S1—C9—S2	112.82 (10)	C12—C13—H16	108.753
S1—C9—N2	127.04 (14)	C14—C13—H15	108.757
S2—C9—N2	120.13 (13)	C14—C13—H16	108.759
S2—C10—C11	113.56 (14)	H15—C13—H16	107.646
C10—C11—C12	112.47 (17)	C13—C14—H17	108.973
C11—C12—C13	113.37 (17)	C13—C14—H18	108.975
C12—C13—C14	114.00 (18)	C15—C14—H17	108.977
C13—C14—C15	113.04 (19)	C15—C14—H18	108.973
C1—C2—H1	118.658	H17—C14—H18	107.774
C3—C2—H1	118.661	C14—C15—H19	109.470
C2—C3—H2	120.558	C14—C15—H20	109.468
C4—C3—H2	120.548	C14—C15—H21	109.470
C4—C5—H3	119.366	H19—C15—H20	109.477
C6—C5—H3	119.354	H19—C15—H21	109.472
C1—C6—H4	119.844	H20—C15—H21	109.470
C5—C6—H4	119.851		
S1—Pd1—N1—N2	-7.46 (11)	N2—N1—C8—C1	-2.8 (3)
S1—Pd1—N1—C8	170.36 (13)	C8—N1—N2—C9	-170.01 (16)
N1—Pd1—S1—C9	4.28 (5)	N1—N2—C9—S1	-3.2 (3)
S1—Pd1—N1 ⁱ —N2 ⁱ	-172.54 (11)	N1—N2—C9—S2	175.43 (14)

S1—Pd1—N1 ⁱ —C8 ⁱ	9.64 (13)	C2—C1—C6—C5	0.9 (3)
N1 ⁱ —Pd1—S1—C9	−175.72 (5)	C6—C1—C2—C3	−0.7 (3)
S1 ⁱ —Pd1—N1—N2	172.54 (11)	C2—C1—C8—N1	174.9 (2)
S1 ⁱ —Pd1—N1—C8	−9.64 (13)	C8—C1—C2—C3	177.76 (17)
N1—Pd1—S1 ⁱ —C9 ⁱ	175.72 (5)	C6—C1—C8—N1	−6.8 (4)
S1 ⁱ —Pd1—N1 ⁱ —N2 ⁱ	7.46 (11)	C8—C1—C6—C5	−177.33 (18)
S1 ⁱ —Pd1—N1 ⁱ —C8 ⁱ	−170.36 (13)	C1—C2—C3—C4	−0.4 (4)
N1 ⁱ —Pd1—S1 ⁱ —C9 ⁱ	−4.28 (5)	C2—C3—C4—O1	−178.54 (19)
Pd1—S1—C9—S2	179.30 (10)	C2—C3—C4—C5	1.4 (3)
Pd1—S1—C9—N2	−2.01 (17)	O1—C4—C5—C6	178.75 (18)
C9—S2—C10—C11	80.84 (13)	C3—C4—C5—C6	−1.2 (4)
C10—S2—C9—S1	−179.47 (11)	C4—C5—C6—C1	0.0 (4)
C10—S2—C9—N2	1.74 (18)	S2—C10—C11—C12	175.70 (11)
C7—O1—C4—C3	7.6 (3)	C10—C11—C12—C13	−177.57 (14)
C7—O1—C4—C5	−172.30 (17)	C11—C12—C13—C14	176.88 (15)
Pd1—N1—N2—C9	8.0 (2)	C12—C13—C14—C15	179.92 (15)
Pd1—N1—C8—C1	179.34 (14)		

Symmetry code: (i) $-x, -y, -z$.