

1. A print-out from punched cards
2. A punched-card deck
3. A magnetic tape from the punched cards.

This research has been supported by the National Institutes of Health, U.S. Public Health Service, under grant number GM-11987, and in part grant number NB-02763.

Reference

MOOTZ, E., MOOTZ, D. & JEFFREY, G. A. (1965), *A Knowledge Availability Survey of the Crystal Structural Data for Pyrimidine Derivatives*. Technical Report, University of Pittsburgh.

Book Reviews

Works intended for notice in this column should be sent direct to the Editor (A. J. C. Wilson, Department of Physics, The University, Birmingham 15, England). As far as practicable books will be reviewed in a country different from that of publication.

A systematic classification of types of intermetallic structures. By P. I. KRIPYAKEVICH [This is not a book, but a translation of research/review papers from *Zh. Strukt. Khim.*, 4, 117, 282.] 33 pp. 293 ref. New York: Consultants Bureau. Reprint in soft cover. Price \$ 10.

Before discussing the paper by Kripyakevich, I wish to draw attention to a statement on the face page of the translation by Consultants Bureau Enterprises Inc. 'No part of this publication may be reproduced in any form [my italics] without written permission from the publisher.' Far too many scientific publications are issued nowadays with this protection. It is essential to science that scientists continue to exercise their prerogative of being able to quote reasonable pieces of scientific data and diagrams from other scientific publications without let or hindrance, and writing letters to publishers for permission whenever one wishes to exercise this prerogative is certainly beyond all reason. We must, therefore, make the businessman-publishers of scientific papers and works understand quite clearly that in exchange for our patronage (to the mutual benefit of both themselves and ourselves), we require that they subscribe to the *Fair Copying Declaration* of the Royal Society of London.

The author sets forth a classification of binary, ternary and quaternary compounds having metallic properties, that is based on coordination numbers, configurations of coordination polyhedra, and their method of combination, which can therefore be applied to all structural types. The classification is based on the type of coordination polyhedron of the atoms having the *lowest* coordination number, or smallest effective atomic radius when there are components with the same coordination number. This allows all known structure types to be separated into one of 17 classes with C.N. ranging between 2 and 14. Within a given class, further subdivision proceeds according to the relative arrangement of the coordination polyhedra in the structure,

the C.N. of the remaining independent atoms, *etc.* If the smaller atoms have coordination polyhedra of two kinds, then the structure is classed according to that with the lower C.N. or the smaller effective radius if both have the same C.N. Thus for example Sn_4Ni_3 with C.N. 6 Ni in prisms and octahedra, is classified with the octahedra. The structure of a combination of atoms is characterized by a unit cell of specific symmetry, axial ratios and angles, occupied regular point-set(s) of a certain space group, and atomic coordinates. Thus when axial ratios and parameters vary beyond a certain range, so as to change the coordination significantly, a new 'structure' is obtained according to the classification, and so problems involving formally isotopic structures in which atoms have different coordination in degree and/or kind (*e.g.*, $F5_1$, $C11$, or $\text{PH}_4\text{I}-\text{BiIn}-\text{PbO}-\text{TeFe}$ types) are removed.

In its present form, the intending user must personally develop the classification along lines suggested by the author, if he is to derive any benefit from it. This must take some considerable time and effort. The language is not always easy and there is a fair amount of symbolism to remember. The author does not say a word on the usefulness of the classification, and based as it is on the simplest coordination unit in the structure, it does not appear to be always useful in showing up relationships between different structural types and derivatives, nor does it appear to be of any help to the crystallographer who seeks systematically to examine, at an early stage in a structural analysis, whether a compound has an already known structure. Under these circumstances, I feel that it may not be useful to those who have only a secondary interest in structural relationships, and we may hope for a further paper by the author developing themes for the use of the classification which he has outlined.

W. B. PEARSON

*Division of Pure Physics
National Research Council
Ottawa 2
Canada*