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Crystallographic data for certain alkaloids. III. By F. M. LOVELL. Viriamu Jones Laboratory, University

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The cell dimensions for these substances (Table 1) were obtained from oscillation and Weissenberg photographs using Cu $K\alpha$ radiation. Densities were determined by flotation. The maximum error in the numerical data given is of the order of 1%.

Ephedrine, $C_{10}H_{15}NO \cdot \frac{1}{2}H_2O$

Ephedrine is reported to occur in two forms: one is anhydrous and the other contains half a molecule of water of crystallization. Attempts to prepare the anhydrous form were unsuccessful. Oscillation photographs of the hydrated form showed the crystals to be orthorhombic with Laue symmetry *mmm*. This, in conjunction with the systematic absences, determines the space group uniquely as $C222_1$. The results obtained are compared in Table 1 with those given by Gossner & Neff (1935).

Hordenine sulphate, $(C_{10}H_{15}NO)_2 \cdot H_2SO_4 \cdot H_2O$

Crystals were obtained by recrystallization from methanol. Oscillation photographs showed that the crystals were monoclinic (Laue symmetry 2/m). Systematic absences indicated an *a* glide and a screw axis. The space group is therefore uniquely determined as $P2_1/a$. The monoclinic angle was measured from a Weissenberg photograph of the [010] zone of reflexions.

Homatropine, C₁₆H₂₁NO₃

Crystals were obtained by recrystallization from ethanol. Oscillation photographs determined the space group uniquely as $P2_1/c$. The monoclinic angle was again measured from a Weissenberg photograph.

Homatropine hydrobromide, C₁₆H₂₁NO₃. HBr

Crystals were obtained by recrystallization from water. Oscillation and Weissenberg photographs determined the space group uniquely as *Pcab*.

In both homatropine and homatropine hydrobromide the nitrogen atom may be optically active, in which case, since the space groups are centro-symmetric, the unit cells must be assumed to contain equal numbers of L- and D-molecules. The alkaloid homatropine is a synthetic product (Manske & Holmes, 1950) so that a racemate is probable.

Physostigmine, C₁₅H₂₁N₃O₂

The crystals obtained by recrystallization from ethanol were in the form of plates. Oscillation photographs showed the Laue symmetry to be *mmm* and the only absences to be screw absences along each axis. The space group is therefore determined uniquely as $P_{2,2,2,1}$.

Narcotine, C₂₂H₂₇NO₇

Long needle-shaped crystals were obtained by recrystallization of the commercial product from ethanol. As in physostigmine, the Laue symmetry and systematic absences determine the space group uniquely as $P2_12_12_1$.

Quinidine benzenate, $C_{20}H_{24}NO \cdot \frac{1}{4}C_6H_6$

Crystals were obtained from a solution of quinidine in benzene. The space group was determined uniquely from oscillation photographs as $P2_12_12_1$.

The formula given by Groth (1906-19) is

$$C_{20}H_{24}NO.\frac{1}{3}C_{6}H_{6}$$
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but the density calculated from this formula and the observed cell edges did not agree with that observed. Also the axial ratios

$$a:b:c = 0.735:1:1.24$$

did not agree with those given by Groth (a:b:c = 0.6916:1:1.0054).

When the formula $C_{20}H_{24}NO.\frac{1}{4}C_6H_6$ was adopted, agreement was found between observed and calculated densities. It is concluded that the material examined differed from that used for the observations quoted by Groth.

No further X-ray work on these substances is contemplated.

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Compound		Table 1			Density $(g.cm.^{-3})$			
	a (Å)	b (Å)	c (Å)	β (°)	Obs.	Calc.	\boldsymbol{Z}	Space group
Ephedrine	7.41	11.40	$24 \cdot 1$	_	1.12	1.13	8	$C222_{1}$
Ephedrine (Gossner & Neff)	7.41	11.25	$24 \cdot 1$		1.124	1.15	8	$C222_1$
Hordenine sulphate	24.1	8.29	11.75	103	1.31	1.29	4	$P2_1/a$
Homatropine	14.5	15.1	6.97	100	1.21	1.22	4	$P2_1/c$
Homatropine hydrobromide	10.3	16.4	19.25		1.48	1.46	8	Pcab
Physostigmine	7.24	14.25	14.50		1.20	1.22	4	$P2_{1}2_{1}2_{1}$
Narcotine	7.90	15.4	32.6	<u> </u>	1.38	1.38	8	$P2_{1}^{2}2_{1}^{2}2_{1}^{2}$
Quinidine benzenate	9.42	12.8	15.85		1.19	1.19	4	$P2_{1}^{2}2_{1}^{2}2_{1}^{2}$