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Mefenacet [2-(1,3-benzothiazol-2-yloxy)-*N*-methyl-*N*-phenylacetamide]

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Key indicators: single-crystal X-ray study; T = 173 K; mean σ (C–C) = 0.003 Å; R factor = 0.042; wR factor = 0.107; data-to-parameter ratio = 19.3.

The title compound, $C_{16}H_{14}N_2O_2S$, crystallizes with two independent molecules in the asymmetric unit. The dihedral angles between the plane of the benzothiazole ring system and the phenyl ring plane are 51.63 (7) and 60.46 (5)°. In the crystal structure, weak intermolecular C–H···O hydrogen bonds and C–H··· π interactions contribute to the stabilization of the packing.

Related literature

For information on the toxicity and herbicidal properties of the title compound, see: Lu *et al.* (2001). For related structures, see: Murru *et al.* (2009).



Experimental

Crystal data
$C_{16}H_{14}N_2O_2S$
$M_r = 298.35$
Monoclinic, $P2_1/c$

<i>a</i> =	11.2708	(6)	Ă
<i>b</i> =	15.7112	(9)	Å
<i>c</i> =	19.9579	(8)	Å

 $\beta = 122.997 \ (2)^{\circ}$ $V = 2964.0 \ (3) \ \text{\AA}^3$ Z = 8Mo $K\alpha$ radiation

Data collection

Bruker APEXII CCD diffractometer Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\rm min} = 0.959, T_{\rm max} = 0.980$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.042$ 381 parameters $wR(F^2) = 0.107$ H-atom parameters constrainedS = 1.02 $\Delta \rho_{max} = 0.27$ e Å $^{-3}$ 7338 reflections $\Delta \rho_{min} = -0.28$ e Å $^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

Cg,1 Cg2 and Cg3 are the centroids of the C27–C32, C17–C22 and S1/C1/C6/ N1/C7 rings, respectively.

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
C2-H2···O4 ⁱ	0.95	2.44	3.273 (2)	147
$C8-H8B\cdots O3^{ii}$	0.99	2.43	3.235 (2)	138
$C24 - H24B \cdots O2$	0.99	2.35	3.256 (2)	153
$C15-H15\cdots Cg1^{ii}$	0.95	2.97	3.84	154
$C26-H26B\cdots Cg2^{iii}$	0.98	3.00	3.74	134
$C29-H29\cdots Cg3^{iv}$	0.95	2.95	3.74	141
$C30-H30\cdots Cg2^{i}$	0.95	2.80	3.42	124
		3 . 1		

Symmetry codes: (i) $x + 1, -y + \frac{3}{2}, z + \frac{1}{2}$; (ii) -x + 1, -y + 1, -z + 1; (iii) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (iv) $x, -y + \frac{3}{2}, z - \frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2006); cell refinement: *SAINT* (Bruker, 2006); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL* and *DIAMOND* (Brandenburg, 1998); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WN2417).

References

Lu, Y., Han, S. & Zhang, C. (2001). Bull. Environ. Contam. Toxicol. 66, 17–23.Murru, S., Mondal, P., Yella, R. & Patel, B. K. (2009). Eur. J. Org. Chem. 31, 5406–5413.

Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany. Sheldrick, G. M. (2008). *Acta Cryst*. A**64**, 112–122.

organic compounds

 $\mu = 0.22 \text{ mm}^{-1}$

 $0.19 \times 0.17 \times 0.09 \text{ mm}$

30345 measured reflections

7338 independent reflections

5174 reflections with $I > 2\sigma(I)$

T = 173 K

 $R_{\rm int} = 0.045$

Brandenburg, K. (1998). DIAMOND. Crystal Impact GbR, Bonn, Germany. Bruker (2006). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.

supporting information

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Mefenacet [2-(1,3-benzothiazol-2-yloxy)-N-methyl-N-phenylacetamide]

Sanghun Cheon, Tae Ho Kim, Suk-Hee Moon and Jineun Kim

S1. Comment

Mefenacet (systematic name: 2-(1,3-benzothiazol-2-yloxy)-*N*- methylacetanilide), is a type of herbicide with low toxicity and high activity (Lu *et al.* 2001). However its crystal structure has not hitherto been reported.

In the asymmetric unit (Fig. 1), the dihedral angles between the plane of the benzothiazole ring system and the phenyl ring plane are $51.63 (7)^{\circ}$ and $60.46 (5)^{\circ}$. All bond lengths and bond angles are normal and comparable to those observed in similar crystal structures (Murru *et al.*, 2009).

In the crystal structure, as shown in Fig. 2, weak intermolecular C—H···O hydrogen bonds and C—H··· π interactions are observed (Table 1). These intermolecular interactions may contribute to the stabilization of the packing.

S2. Experimental

The title compound was purchased from the Dr. Ehrenstorfer GmbH Company. Slow evaporation of a solution in CH_2Cl_2 gave single crystals suitable for X-ray analysis.

S3. Refinement

All H-atoms were positioned geometrically and refined using a riding model with d(C-H) = 0.95 Å, $U_{iso}(H) = 1.2U_{eq}(C)$ for aromatic C, d(C-H) = 0.99 Å, $U_{iso}(H) = 1.2U_{eq}(C)$ for CH₂ and d(C-H) = 0.98 Å, $U_{iso}(H) = 1.5U_{eq}(C)$ for CH₃ groups.



Figure 1

The asymmetric unit of the title compound, with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen atoms are represented by small spheres of arbitrary radius.



Figure 2

Crystal packing of the title compound with intermolecular C—H···O and C—H··· π interactions shown as dashed lines. H atoms not involved in intermolecular interactions have been omitted for clarity. *Cg1—Cg3* are the centroids of the C27–C32, C17–C22, and S1/C1/C6/N1/C7 rings, respectively. [Symmetry codes: (i) x + 1, -y + 1.5, z + 1/2; (ii) -x + 1, -y + 1, -z + 1; (iii) x, -y + 1.5, z - 1/2; (iv) -x + 1, y - 1/2, -z + 1/2; (v) -x + 1, y - 1/2, -z + 1.5; (vi) x, -y + 1.5, z + 1/2.)

2-(1,3-benzothiazol-2-yloxy)-N-methyl-N-phenylacetamide

Crystal data

 $C_{16}H_{14}N_2O_2S$ F(000) = 1248 $M_r = 298.35$ $D_{\rm x} = 1.337 {\rm ~Mg} {\rm ~m}^{-3}$ Monoclinic, $P2_1/c$ Mo *K* α radiation, $\lambda = 0.71073$ Å Hall symbol: -P 2ybc Cell parameters from 5175 reflections $\theta = 2.2 - 25.3^{\circ}$ *a* = 11.2708 (6) Å b = 15.7112 (9) Å $\mu = 0.22 \text{ mm}^{-1}$ T = 173 K*c* = 19.9579 (8) Å $\beta = 122.997 (2)^{\circ}$ Block, colourless V = 2964.0 (3) Å³ $0.19 \times 0.17 \times 0.09 \text{ mm}$ Z = 8Data collection Bruker APEXII CCD Absorption correction: multi-scan diffractometer (SADABS; Sheldrick, 1996) $T_{\min} = 0.959, T_{\max} = 0.980$ Radiation source: fine-focus sealed tube Graphite monochromator 30345 measured reflections φ and ω scans 7338 independent reflections 5174 reflections with $I > 2\sigma(I)$

$R_{\rm int} = 0.045$	$k = -18 \rightarrow 20$
$\theta_{\rm max} = 28.3^{\circ}, \theta_{\rm min} = 1.8^{\circ}$	$l = -26 \rightarrow 26$
$h = -15 \rightarrow 15$	

Refinement

5	
Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.042$	Hydrogen site location: inferred from
$wR(F^2) = 0.107$	neighbouring sites
S = 1.02	H-atom parameters constrained
7338 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0435P)^2 + 0.7531P]$
381 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{\rm max} = 0.001$
Primary atom site location: structure-invariant	$\Delta ho_{ m max} = 0.27 \ { m e} \ { m \AA}^{-3}$
direct methods	$\Delta ho_{ m min} = -0.28 \ { m e} \ { m \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

	x	У	Ζ	$U_{ m iso}*/U_{ m eq}$
S1	0.94569 (4)	0.65520 (3)	0.70419 (3)	0.03607 (12)
S2	0.21697 (5)	0.53839 (3)	0.15747 (3)	0.03835 (13)
O1	0.71887 (12)	0.56299 (8)	0.65599 (8)	0.0371 (3)
O2	0.53812 (13)	0.61502 (8)	0.50454 (8)	0.0407 (3)
O3	0.37291 (13)	0.62655 (7)	0.28811 (7)	0.0358 (3)
O4	0.30217 (13)	0.78901 (8)	0.29306 (9)	0.0462 (3)
N1	0.69842 (14)	0.70886 (9)	0.66837 (9)	0.0310 (3)
N2	0.34408 (15)	0.59934 (9)	0.51009 (9)	0.0344 (3)
N3	0.38648 (14)	0.66403 (9)	0.17928 (8)	0.0310 (3)
N4	0.52868 (15)	0.83502 (9)	0.36531 (10)	0.0402 (4)
C1	0.92417 (17)	0.76410 (11)	0.71071 (10)	0.0289 (4)
C2	1.02172 (17)	0.82906 (12)	0.73152 (10)	0.0338 (4)
H2	1.1156	0.8173	0.7462	0.041*
C3	0.97758 (19)	0.91139 (12)	0.73012 (11)	0.0366 (4)
Н3	1.0418	0.9572	0.7434	0.044*
C4	0.84079 (19)	0.92827 (12)	0.70967 (11)	0.0376 (4)
H4	0.8133	0.9855	0.7096	0.045*
C5	0.74394 (18)	0.86369 (11)	0.68950 (11)	0.0356 (4)
Н5	0.6508	0.8760	0.6759	0.043*
C6	0.78526 (17)	0.78025 (11)	0.68938 (10)	0.0285 (4)
C7	0.76859 (17)	0.64308 (11)	0.67289 (10)	0.0304 (4)
C8	0.57044 (18)	0.55577 (11)	0.62402 (11)	0.0349 (4)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

H8A	0.5481	0.5861	0.6593	0.042*
H8B	0.5454	0.4950	0.6223	0.042*
C9	0.48359 (18)	0.59301 (10)	0.54072 (10)	0.0311 (4)
C10	0.2484 (2)	0.62796 (14)	0.42757 (11)	0.0465 (5)
H10A	0.3006	0.6636	0.4118	0.070*
H10B	0.1710	0.6610	0.4235	0.070*
H10C	0.2095	0.5784	0.3922	0.070*
C11	0.27959 (17)	0.56221 (11)	0.54862 (10)	0.0324 (4)
C12	0.2600 (2)	0.61034 (13)	0.59977 (12)	0.0417 (4)
H12	0.2915	0.6677	0.6111	0.050*
C13	0.1940(2)	0.57392 (16)	0.63416(13)	0.0555 (6)
H13	0.1807	0.6063	0.6698	0.067*
C14	0.1474(2)	0 49112 (16)	0.61706 (14)	0.0575 (6)
H14	0.1009	0.4668	0.6404	0.069*
C15	0.1676(2)	0.44331(14)	0.56647(14)	0.0540 (6)
H15	0.1361	0.3850	0.5553	0.0540 (0)
C16	0.1301 0.2320 (2)	0.3839 0.47876(12)	0.5555 0.53101 (12)	0.005
	0.2339 (2)	0.47670 (12)	0.33191 (12)	0.0423 (3)
H10	0.2481	0.4400	0.4908	0.051^{*}
C1/	0.23107(18)	0.50/33 (11)	0.07802 (11)	0.0351 (4)
	0.1640 (2)	0.53245 (13)	0.00209 (13)	0.0489 (5)
HI8	0.1012	0.4856	-0.0126	0.059*
C19	0.1912 (2)	0.56/94 (16)	-0.05143 (14)	0.0589 (6)
H19	0.1450	0.5458	-0.1041	0.071*
C20	0.2847 (3)	0.63537 (15)	-0.02975 (13)	0.0555 (6)
H20	0.3015	0.6584	-0.0679	0.067*
C21	0.3540 (2)	0.66979 (13)	0.04629 (12)	0.0431 (5)
H21	0.4188	0.7155	0.0610	0.052*
C22	0.32599 (18)	0.63536 (11)	0.10061 (11)	0.0323 (4)
C23	0.33790 (17)	0.61924 (10)	0.21272 (10)	0.0300 (4)
C24	0.48587 (18)	0.68520 (11)	0.33700 (11)	0.0337 (4)
H24A	0.5552	0.6842	0.3211	0.040*
H24B	0.5352	0.6675	0.3937	0.040*
C25	0.42889 (18)	0.77450 (11)	0.32799 (11)	0.0335 (4)
C26	0.4871 (2)	0.92358 (13)	0.36470 (17)	0.0638 (7)
H26A	0.4925	0.9557	0.3244	0.096*
H26B	0.5510	0.9490	0.4174	0.096*
H26C	0.3899	0.9252	0.3519	0.096*
C27	0.67722 (18)	0.81716 (11)	0.40232 (11)	0.0333 (4)
C28	0.73303 (19)	0.81424 (12)	0.35549 (12)	0.0387 (4)
H28	0.6737	0.8235	0 2995	0.046*
C29	0.87550 (19)	0.79774(12)	0.39043(12)	0.0416(5)
H29	0.9141	0.7954	0.3584	0.050*
C30	0.9111 0.9613 (2)	0.78468 (12)	0.5561 0.47130(13)	0.020
H30	1 0502	0.737	0.47130 (13)	0.052*
C21	0.0050 (2)	0.78758 (14)	0.51738 (12)	0.032
	0.9030 (2)	0.70730 (14)	0.51750(15)	0.0493 (3)
C22	0.7040	0.770(12)	0.3733	0.039
0.52	0.7027 (2)	0.00579 (15)	0.48334 (12)	0.0433 (3)
H32	0.7245	0.8057	0.5158	0.055*

supporting information

Atomic displacement parameters $(Å^2)$

	T T [1]	1.122	1733	1 712	<i>T 1</i> 13	173
	Un	022	<i>U</i> ³³	U ¹²	U	U ²³
S 1	0.0286 (2)	0.0329 (2)	0.0472 (3)	0.00576 (18)	0.0210 (2)	0.0021 (2)
S2	0.0339 (2)	0.0301 (2)	0.0484 (3)	-0.00813 (18)	0.0207 (2)	-0.0074 (2)
01	0.0329 (6)	0.0308 (6)	0.0433 (8)	-0.0001 (5)	0.0178 (6)	0.0011 (6)
O2	0.0446 (7)	0.0475 (8)	0.0369 (7)	-0.0051 (6)	0.0266 (6)	0.0006 (6)
O3	0.0383 (7)	0.0326 (7)	0.0316 (7)	-0.0099 (5)	0.0158 (6)	0.0015 (5)
O4	0.0305 (7)	0.0413 (8)	0.0622 (9)	-0.0038 (6)	0.0224 (7)	0.0042 (7)
N1	0.0274 (7)	0.0329 (8)	0.0328 (8)	0.0009 (6)	0.0166 (6)	-0.0005 (6)
N2	0.0366 (8)	0.0372 (8)	0.0296 (8)	0.0020 (6)	0.0182 (7)	0.0050 (6)
N3	0.0318 (7)	0.0276 (7)	0.0305 (8)	-0.0017 (6)	0.0150 (7)	0.0006 (6)
N4	0.0319 (8)	0.0315 (8)	0.0582 (11)	-0.0043 (6)	0.0251 (8)	-0.0048 (7)
C1	0.0275 (8)	0.0324 (9)	0.0260 (9)	0.0048 (7)	0.0141 (7)	0.0031 (7)
C2	0.0241 (8)	0.0420 (10)	0.0323 (10)	-0.0011 (7)	0.0134 (7)	-0.0003 (8)
C3	0.0364 (9)	0.0349 (10)	0.0366 (10)	-0.0042 (8)	0.0187 (8)	-0.0015 (8)
C4	0.0411 (10)	0.0322 (9)	0.0399 (11)	0.0034 (8)	0.0223 (9)	-0.0004 (8)
C5	0.0293 (9)	0.0365 (10)	0.0417 (11)	0.0053 (7)	0.0199 (8)	-0.0004 (8)
C6	0.0265 (8)	0.0328 (9)	0.0258 (9)	0.0027 (7)	0.0140 (7)	0.0006 (7)
C7	0.0278 (8)	0.0330 (9)	0.0295 (9)	-0.0002 (7)	0.0150 (7)	0.0007 (7)
C8	0.0327 (9)	0.0339 (10)	0.0350 (10)	-0.0049 (7)	0.0165 (8)	0.0027 (8)
C9	0.0384 (9)	0.0250 (8)	0.0311 (9)	-0.0031 (7)	0.0196 (8)	-0.0038 (7)
C10	0.0461 (11)	0.0539 (13)	0.0334 (11)	0.0082 (9)	0.0177 (9)	0.0084 (9)
C11	0.0283 (8)	0.0356 (9)	0.0315 (10)	0.0035 (7)	0.0151 (8)	0.0042 (7)
C12	0.0452 (11)	0.0407 (11)	0.0429 (11)	-0.0015 (9)	0.0264 (10)	-0.0051 (9)
C13	0.0553 (13)	0.0748 (16)	0.0501 (14)	-0.0038 (12)	0.0376 (12)	-0.0081 (12)
C14	0.0472 (12)	0.0788 (17)	0.0521 (14)	-0.0115 (12)	0.0306 (11)	0.0092 (12)
C15	0.0529 (13)	0.0469 (12)	0.0608 (15)	-0.0128 (10)	0.0300 (12)	0.0017 (11)
C16	0.0423 (11)	0.0399 (11)	0.0453 (12)	-0.0020 (8)	0.0241 (10)	-0.0051 (9)
C17	0.0275 (8)	0.0321 (9)	0.0388 (11)	0.0062 (7)	0.0135 (8)	-0.0040 (8)
C18	0.0395 (11)	0.0462 (12)	0.0494 (13)	0.0059 (9)	0.0168 (10)	-0.0157 (10)
C19	0.0596 (14)	0.0667 (15)	0.0420 (13)	0.0114 (12)	0.0223 (11)	-0.0172 (11)
C20	0.0678 (15)	0.0621 (15)	0.0446 (13)	0.0160 (12)	0.0358 (12)	0.0015 (11)
C21	0.0487 (11)	0.0435 (11)	0.0429 (12)	0.0087 (9)	0.0287 (10)	0.0040 (9)
C22	0.0307 (8)	0.0294 (9)	0.0328 (10)	0.0071 (7)	0.0147 (8)	0.0001 (7)
C23	0.0269 (8)	0.0242 (8)	0.0331 (10)	-0.0015 (6)	0.0125 (7)	0.0006 (7)
C24	0.0317 (9)	0.0327 (9)	0.0280 (9)	-0.0070 (7)	0.0107 (8)	0.0016 (7)
C25	0.0325 (9)	0.0352 (10)	0.0351 (10)	-0.0056 (7)	0.0199 (8)	0.0019 (8)
C26	0.0457 (12)	0.0342 (11)	0.114 (2)	-0.0049 (9)	0.0448 (14)	-0.0109 (12)
C27	0.0303 (9)	0.0295 (9)	0.0397 (11)	-0.0070 (7)	0.0187 (8)	-0.0045 (8)
C28	0.0376 (10)	0.0405 (10)	0.0360 (11)	-0.0051 (8)	0.0187 (9)	0.0042 (8)
C29	0.0371 (10)	0.0460 (11)	0.0475 (12)	-0.0053 (8)	0.0268 (10)	0.0003 (9)
C30	0.0320 (10)	0.0363 (10)	0.0516 (13)	-0.0067 (8)	0.0165 (9)	-0.0053 (9)
C31	0.0445 (11)	0.0561 (13)	0.0308 (11)	-0.0048 (10)	0.0095 (9)	-0.0039 (9)
C32	0.0510 (12)	0.0508 (12)	0.0408 (12)	-0.0065 (10)	0.0289 (10)	-0.0090 (9)
			× /	× /	· /	

Geometric parameters (Å, °)

S1—C1	1.7428 (17)	C11—C12	1.381 (3)
S1—C7	1.7472 (17)	C12—C13	1.382 (3)
S2—C17	1.738 (2)	C12—H12	0.9500
S2—C23	1.7465 (17)	C13—C14	1.375 (3)
O1—C7	1.344 (2)	C13—H13	0.9500
O1—C8	1.434 (2)	C14—C15	1.374 (3)
O2—C9	1.226 (2)	C14—H14	0.9500
O3—C23	1.336 (2)	C15—C16	1.380 (3)
O3—C24	1.438 (2)	C15—H15	0.9500
O4—C25	1.222 (2)	C16—H16	0.9500
N1—C7	1.274 (2)	C17—C18	1.386 (3)
N1—C6	1.395 (2)	C17—C22	1.402 (2)
N2—C9	1.345 (2)	C18—C19	1.380 (3)
N2	1.438 (2)	C18—H18	0.9500
N2C10	1.464 (2)	C19—C20	1.387 (3)
N3—C23	1.280 (2)	C19—H19	0.9500
N3—C22	1.403 (2)	C20—C21	1.384 (3)
N4—C25	1.346 (2)	C20—H20	0.9500
N4—C27	1.442 (2)	C21—C22	1.393 (3)
N4—C26	1.466 (2)	C21—H21	0.9500
C1—C2	1.388 (2)	C24—C25	1.512 (2)
C1—C6	1.404 (2)	C24—H24A	0.9900
C2—C3	1.381 (2)	C24—H24B	0.9900
C2—H2	0.9500	C26—H26A	0.9800
C3—C4	1.389 (2)	C26—H26B	0.9800
С3—Н3	0.9500	C26—H26C	0.9800
C4—C5	1.380 (3)	C27—C32	1.378 (3)
C4—H4	0.9500	C27—C28	1.383 (2)
C5—C6	1.392 (2)	C28—C29	1.383 (3)
С5—Н5	0.9500	C28—H28	0.9500
C8—C9	1.514 (2)	C29—C30	1.372 (3)
C8—H8A	0.9900	C29—H29	0.9500
C8—H8B	0.9900	C30—C31	1.373 (3)
C10—H10A	0.9800	C30—H30	0.9500
C10—H10B	0.9800	C31—C32	1.383 (3)
C10—H10C	0.9800	C31—H31	0.9500
C11—C16	1.381 (3)	С32—Н32	0.9500
C1—S1—C7	87.44 (8)	C14—C15—C16	119.9 (2)
C17—S2—C23	87.60 (9)	C14—C15—H15	120.0
C7—O1—C8	114.32 (13)	C16—C15—H15	120.0
C23—O3—C24	115.36 (13)	C15—C16—C11	119.62 (19)
C7—N1—C6	108.91 (14)	C15—C16—H16	120.2
C9—N2—C11	122.22 (14)	C11—C16—H16	120.2
C9—N2—C10	119.78 (15)	C18—C17—C22	121.31 (19)
C11—N2—C10	116.67 (14)	C18—C17—S2	128.87 (16)

C23—N3—C22	108.79 (14)	C22—C17—S2	109.82 (13)
C25—N4—C27	122.21 (15)	C19—C18—C17	117.9 (2)
C25—N4—C26	119.92 (16)	C19—C18—H18	121.1
C27—N4—C26	117.75 (15)	C17—C18—H18	121.1
C2—C1—C6	121.92 (16)	C18—C19—C20	121.3 (2)
C2—C1—S1	128.66 (13)	C18—C19—H19	119.4
C6-C1-S1	109.41 (12)	С20—С19—Н19	119.4
C3—C2—C1	117.71 (15)	C21—C20—C19	121.3 (2)
C3—C2—H2	121.1	$C_{21} = C_{20} = H_{20}$	119.3
C1-C2-H2	121.1	C19 - C20 - H20	119.3
$C_2 - C_3 - C_4$	121.0 (17)	C_{20} C_{21} C_{22} C_{22}	119.5 118.0(2)
$C_2 = C_3 = H_3$	119.5	C_{20} C_{21} C_{22}	121.0
C_{4} C_{3} H_{3}	119.5	$C_{20} = C_{21} = H_{21}$	121.0
$C_{4} = C_{3} = 113$	119.5	$C_{22} = C_{21} = H_{21}$	121.0 120.17(17)
$C_{5} = C_{4} = C_{5}$	110.2	$C_{21} = C_{22} = C_{17}$	120.17(17) 124.66(17)
$C_3 = C_4 = H_4$	119.5	$C_{21} = C_{22} = N_3$	124.00(17)
$C_3 = C_4 = H_4$	119.5	C17 - C22 - N3	113.17(10) 12(20(15))
C4 - C5 - C6	118.74 (10)	N3-C23-C3	120.29 (15)
C4—C5—H5	120.6	N3-C23-S2	118.62 (14)
C6—C5—H5	120.6	03-023-82	115.07 (12)
C5—C6—N1	125.27 (15)	03-C24-C25	110.61 (14)
C5—C6—C1	119.25 (15)	O3—C24—H24A	109.5
N1—C6—C1	115.47 (14)	C25—C24—H24A	109.5
N1—C7—O1	125.72 (15)	O3—C24—H24B	109.5
N1—C7—S1	118.76 (13)	C25—C24—H24B	109.5
O1—C7—S1	115.52 (12)	H24A—C24—H24B	108.1
O1—C8—C9	110.98 (14)	O4—C25—N4	123.28 (17)
O1—C8—H8A	109.4	O4—C25—C24	121.98 (15)
С9—С8—Н8А	109.4	N4—C25—C24	114.64 (15)
O1—C8—H8B	109.4	N4—C26—H26A	109.5
С9—С8—Н8В	109.4	N4—C26—H26B	109.5
H8A—C8—H8B	108.0	H26A—C26—H26B	109.5
O2—C9—N2	123.28 (16)	N4—C26—H26C	109.5
O2—C9—C8	121.62 (16)	H26A—C26—H26C	109.5
N2—C9—C8	115.11 (14)	H26B—C26—H26C	109.5
N2-C10-H10A	109.5	C32—C27—C28	120.39 (17)
N2-C10-H10B	109.5	$C_{32} - C_{27} - N_{4}$	120 35 (16)
H10A - C10 - H10B	109.5	$C_{28} C_{27} N_{4}$	119 26 (17)
N_2 —C10—H10C	109.5	C_{27} C_{28} C_{29}	119.20 (17)
H_{10A} $-C_{10}$ H_{10C}	109.5	$C_{27} = C_{28} = H_{28}$	120.1
HIOR CIO HIOC	109.5	C_{20} C_{28} H_{28}	120.1
$C_{16} = C_{11} = C_{12}$	109.5	$C_{29} = C_{28} = 1128$	120.1
$C_{10} = C_{11} = C_{12}$	120.03(17) 110.00(16)	C_{20} C_{29} C_{28} C_{20} C	120.13 (18)
$C_{10} = C_{11} = N_2$	119.09(10) 120.25(16)	C_{29} C_{29} H_{29}	119.9
$C_{12} \longrightarrow C_{11} \longrightarrow C_{12}$	120.23(10) 110.12(10)	$C_{20} = C_{27} = \Pi_{27}$	117.7 110.76 (19)
$C_{11} = C_{12} = C_{13}$	119.13 (19)	$C_{29} = C_{30} = C_{31}$	119.70 (18)
C12 - C12 - H12	120.4	$C_{29} - C_{30} - H_{30}$	120.1
C13 - C12 - H12	120.4	$C_{20} = C_{21} = C_{22}$	120.1
C14—C13—C12	120.3 (2)	C30-C31-C32	120.97 (19)
C14—C13—H13	119.8	C30—C31—H31	119.5

C12 C13 H13	110.8	C32 C31 H31	110.5
$C_{12} = C_{13} = 1113$	120.40 (10)	$C_{32} = C_{31} = C_{31}$	119.5
$C_{15} = C_{14} = C_{15}$	120.40 (19)	$C_{27} = C_{32} = C_{31}$	119.01 (18)
C13—C14—H14	119.8	$C_2 = C_3 $	120.5
C13—C14—H14	119.8	C31—C32—H32	120.5
C7 - S1 - C1 - C2	-178 11 (17)	C^{23} S^{2} C^{17} C^{18}	179 84 (17)
$C_7 = S_1 = C_1 = C_2$	0.60(13)	$C_{23} = S_2 = C_{17} = C_{18}$	1/9.04(17)
$C_{1} = C_{1} = C_{1} = C_{0}$	-0.5(2)	$C_{23} = S_{2} = C_{17} = C_{22}$	-1.2(2)
$C_0 - C_1 - C_2 - C_3$	0.5(3) 178 10(14)	$C_{22} = C_{17} = C_{18} = C_{19}$	1.3(3)
S1 - C1 - C2 - C3	1/0.19(14)	52	178.94(13)
C1 - C2 - C3 - C4	0.9(3)	C17 - C18 - C19 - C20	1.2(3)
$C_2 = C_3 = C_4 = C_5$	-0.5(3)	C18 - C19 - C20 - C21	-0.2(3)
03-04-05-06	-0.4 (3)	C19—C20—C21—C22	-0.7(3)
C4—C5—C6—N1	-178.22 (16)	C20—C21—C22—C17	0.6 (3)
C4—C5—C6—C1	0.8 (3)	C20—C21—C22—N3	-179.19 (17)
C7—N1—C6—C5	178.35 (17)	C18—C17—C22—C21	0.4 (3)
C7—N1—C6—C1	-0.7 (2)	S2—C17—C22—C21	-179.77 (13)
C2—C1—C6—C5	-0.4 (3)	C18—C17—C22—N3	-179.76 (15)
S1—C1—C6—C5	-179.29 (13)	S2—C17—C22—N3	0.03 (18)
C2—C1—C6—N1	178.75 (15)	C23—N3—C22—C21	179.63 (16)
S1—C1—C6—N1	-0.15 (19)	C23—N3—C22—C17	-0.2 (2)
C6—N1—C7—O1	-179.10 (16)	C22—N3—C23—O3	178.61 (15)
C6—N1—C7—S1	1.35 (19)	C22—N3—C23—S2	0.22 (18)
C8—O1—C7—N1	5.5 (2)	C24—O3—C23—N3	-6.3 (2)
C8—O1—C7—S1	-174.96 (12)	C24—O3—C23—S2	172.16 (11)
C1—S1—C7—N1	-1.25 (15)	C17—S2—C23—N3	-0.18 (14)
C1—S1—C7—O1	179.15 (14)	C17—S2—C23—O3	-178.75 (13)
C7—O1—C8—C9	69.22 (18)	C23—O3—C24—C25	86.56 (17)
C11—N2—C9—O2	170.94 (16)	C27—N4—C25—O4	-176.47 (17)
C10—N2—C9—O2	4.6 (3)	C26—N4—C25—O4	-0.5 (3)
C11—N2—C9—C8	-8.7(2)	C27—N4—C25—C24	7.3 (2)
C10—N2—C9—C8	-175.05 (16)	C26—N4—C25—C24	-176.75 (18)
01	8.8 (2)	O3-C24-C25-O4	10.1 (2)
01—C8—C9—N2	-171.57(14)	03—C24—C25—N4	-173.56(15)
C9—N2—C11—C16	-87.7(2)	C25—N4—C27—C32	-98.8(2)
C10 - N2 - C11 - C16	79.1 (2)	$C_{26} - N_{4} - C_{27} - C_{32}$	85.1 (2)
C9-N2-C11-C12	942(2)	$C_{25} N_{4} C_{27} C_{28}$	81.6 (2)
$C10 - N^2 - C11 - C12$	-990(2)	$C_{26} N_{4} C_{27} C_{28}$	-944(2)
C_{16} C_{11} C_{12} C_{13}	0.0(3)	$C_{20} = 104 + C_{27} + C_{20} + C_{2$	0.1(3)
N_{2} C_{11} C_{12} C_{13}	178.08(17)	N4 - C27 - C28 - C29	17957(16)
$C_{11} = C_{12} = C_{13}$	-0.6(3)	C_{27} C_{28} C_{29} C_{30}	-0.3(3)
$C_{12} = C_{12} = C_{13} = C_{14}$	0.0(3)	$C_{21} = C_{20} = C_{20} = C_{30}$	0.5(3)
$C_{12} = C_{13} = C_{14} = C_{15}$	-0.6(3)	$C_{20} = C_{27} = C_{30} = C_{31}$	-0.2(3)
C13 - C14 - C13 - C10	0.0(3)	$C_{27} = C_{30} = C_{31} = C_{32}$	0.2(3)
$C_{14} = C_{13} = C_{16} = C_{16}$	0.1(3)	120 - 127 - 132 - 131	0.1(3)
$\begin{array}{c} 12 - 11 - 10 - 15 \end{array}$	0.2(3)	104 - 027 - 032 - 031	-1/9.41(1/)
N2-C11-C16-C15	-1/7.86 (18)	C30—C31—C32—C27	0.0 (3)

Hydrogen-bond geometry (Å, °)

Cal Cal and Ca	3 are the centroids of	the C27_C32	C17_C22 and S1/C	1/C6/N1/C7 rings	respectively
$cg, r cg_2$ and cg .	s are the centrolus of	100027-052,	C17 = C22 and $S1/C$	1/CO/INI/C/ Imgs	, respectively.

D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	<i>D</i> —Н··· <i>A</i>
C2—H2…O4 ⁱ	0.95	2.44	3.273 (2)	147
C8—H8 <i>B</i> ···O3 ⁱⁱ	0.99	2.43	3.235 (2)	138
C24—H24 <i>B</i> ···O2	0.99	2.35	3.256 (2)	153
C15—H15…Cg1 ⁱⁱ	0.95	2.97	3.84	154
C26—H26 <i>B</i> ··· <i>Cg</i> 2 ⁱⁱⁱ	0.98	3.00	3.74	134
C29—H29…Cg3 ^{iv}	0.95	2.95	3.74	141
C30—H30…Cg2 ⁱ	0.95	2.80	3.42	124

Symmetry codes: (i) x+1, -y+3/2, z+1/2; (ii) -x+1, -y+1, -z+1; (iii) -x+1, y+1/2, -z+1/2; (iv) x, -y+3/2, z-1/2.