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# Identification of the site of oxidase substrate binding in Scytalidium thermophilum catalase 

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The catalase from Scytalidium thermophilum is a homotetramer containing a heme $d$ in each active site. Although the enzyme has a classical monofunctional catalase fold, it also possesses oxidase activity towards a number of small organics, including catechol and phenol. In order to further investigate this, the crystal structure of the complex of the catalase with the classical catalase inhibitor 3-amino-1,2,4-triazole (3TR) was determined at $1.95 \AA$ resolution. Surprisingly, no binding to the heme site was observed; instead, 3TR occupies a binding site corresponding to the NADPH-binding pocket in mammalian catalases at the entrance to a lateral channel leading to the heme. Kinetic analysis of site-directed mutants supports the assignment of this pocket as the binding site for oxidase substrates.

## 1. Introduction

Catalases (hydrogen-peroxide:hydrogen-peroxide oxidoreductases; EC 1.11.1.6) are redox enzymes that are responsible for the dismutation of hydrogen peroxide into water and molecular oxygen (Loewen, 1999). They are found in almost all aerobic organisms and play a crucial role in prokaryotic and eukaryotic cell detoxification (Maté et al., 2001). The crystal structures of 15 heme catalases, including that from the thermophilic fungus Scytalidium thermophilum (Yuzugullu et al., 2013), have been solved at high resolution (Díaz et al., 2012). The structures reveal a homotetrameric enzyme in which each of the four active sites consists of a pentacoordinated iron protoporphyrin IX prosthetic group with a tyrosinate axial ligand (Díaz et al., 2012; Yuzugullu et al., 2013). Some catalases also contain an NADPH cofactor tightly bound at the periphery of each subunit (Díaz et al., 2012).

In the resting state the heme is in a high-spin ferric state $\left(\mathrm{Fe}^{3+}\right)$, which is converted to compound I in a two-electron oxidation by hydrogen peroxide. One electron is removed from the Fe atom, forming an oxyferryl moiety $\left(\mathrm{Fe}^{4+}=\mathrm{O}\right)$ with one O atom from the hydrogen peroxide molecule, and the second electron is removed from the porphyrin, resulting in a $\pi$-cation radical (1). Compound I is reduced back to the native (ferric) state by a second molecule of hydrogen peroxide (2). Alternatively, under low hydrogen peroxide conditions, compound I can be reduced to compound II (3), which can react with another $\mathrm{H}_{2} \mathrm{O}_{2}$ to give the inactive compound III (4). For NADPH-binding catalases, it has been proposed that the enzyme is protected against compound III formation by the

Table 1
Oligonucleotides used in site-directed mutagenesis of catpo.

| Mutant | Sequence change | Oligonucleotide $\dagger$ |
| :--- | :--- | :--- |
| E316F | GAA $\rightarrow$ TTC | $5^{\prime}$-CCGACCAAAATCATCCCGTTCGAATACGCTCCGCTGACC-3' |
| E316H | GAA $\rightarrow$ CAC | $5^{\prime}$-CCGACCAAAATCATCCCGCACGAATACGCTCCGCTGACC-3' |
| H246W | CAC $\rightarrow$ TGG | $5^{\prime}-$ CTGATCAAATGGTGGTTCAAATCT-CGTCAGGGTAAAGCTAGTCTGG-3' |
| I313F | ATC $\rightarrow$ TTC | $5^{\prime}-$ GGACCCGACCAAATTCATCCCGGAAGAATACGC-3' |
| I314F | ATC $\rightarrow$ TTC | $5^{\prime}-$ GGACCCGACCAAAATCTTCCCGGAAGAATACGC-3' |
| L321A | CTG $\rightarrow$ GCG | $5^{\prime}-C C G G A A G A A T A C G C T C C G G C G A C C A A A C T G G G T C T G-3 ' ~$ |
| P158W | CCG $\rightarrow$ TGG | $5^{\prime}-$ CGTTGGTAACAACATCTGGGTTTTCTTCATCCAGGACGC-3' |
| Q293W | CAG $\rightarrow$ TGG | $5^{\prime}$-GGGACGTATGCGTATGGATCGTTGACGAATCTCAGGC-3' |
| V536A | GTT $\rightarrow$ GCG | $5^{\prime}-$ CAAAACCGCTGGTGCGTCTATCGTTGGTTCTGG-3' |
| V536W | GTT $\rightarrow$ TGG | $5^{\prime}$-CAAAACCGCTGGTTGGTCTATCGTTGGTTCTGGTCCG-3' |

$\dagger$ The underlined sequence is the codon that has been modified.

NADPH preventing or rescuing compound II formation (Sevinc et al., 1999; Putnam et al., 2000; Nicholls, 2012).

$$
\begin{gather*}
\text { Enz }\left(\mathrm{Por}-\mathrm{Fe}^{3+}\right)+\mathrm{H}_{2} \mathrm{O}_{2} \rightarrow \\
\mathrm{Cpd} \mathrm{I}\left(\mathrm{Por}^{+\bullet}-\mathrm{Fe}^{4+}=\mathrm{O}\right)+\mathrm{H}_{2} \mathrm{O}  \tag{1}\\
\text { Cpd I }\left(\mathrm{Por}^{+\bullet}-\mathrm{Fe}^{4+}=\mathrm{O}\right)+\mathrm{H}_{2} \mathrm{O}_{2} \rightarrow \\
\mathrm{Enz}\left(\mathrm{Por}-\mathrm{Fe}^{3+}\right)+\mathrm{H}_{2} \mathrm{O}+\mathrm{O}_{2} \tag{2}
\end{gather*}
$$

$$
\begin{align*}
& \text { Cpd I }\left(\text { Por }^{+\bullet}-\mathrm{Fe}^{4+}=\mathrm{O}\right)+A \mathrm{H}_{2} \rightarrow \\
& \text { Cpd II }\left(\text { Por }-\mathrm{Fe}^{4+}-\mathrm{OH}\right)+A \mathrm{H}^{\bullet}(3) \\
& \text { Cpd II }\left(\text { Por }-\mathrm{Fe}^{4+}-\mathrm{OH}\right)+\mathrm{H}_{2} \mathrm{O}_{2} \rightarrow \\
& \text { Cpd III }\left(\mathrm{Por}-\mathrm{Fe}^{3+}-\mathrm{OOH}\right)+\mathrm{H}_{2} \mathrm{O} . \tag{4}
\end{align*}
$$

The heme is deeply buried inside the protein, with a complex network of channels providing access to the exterior (Sevinc et al., 1999; Díaz et al., 2012). The main channel approaches the distal side of the heme, perpendicular to the plane of the heme, and is the access route for hydrogen peroxide (Díaz et al., 2012). A second channel, approaching the heme laterally, emerges on the enzyme surface at a location corresponding to the NADP(H)-binding pocket in catalases that bind a nicotinamide cofactor. For homologues that do not bind NADPH, there is some evidence that the channel is involved in either the exit of the reaction products (Díaz et al., 2012) or the entry of substrates/inhibitors (Sevinc et al., 1999). A third channel leading from the distal side of the heme to the central cavity of the tetramer is proposed to play a role in the oxidation of heme $b$ to heme $d$ for catalases that possess heme $d$ in their active site (Murshudov et al., 1996; Sevinc et al., 1999; Putnam et al., 2000), but no functional role has yet been presented for heme $b$ catalases (Chelikani et al., 2004).

We have previously shown that in addition to catalase activity, the catalase from $S$. thermophilum (CATPO) possesses a promiscuous phenolic oxidase activity in the absence of hydrogen peroxide (Ögel et al., 2006; Sutay Kocabas et al., 2008; Yuzugullu et al., 2013). This peroxideindependent secondary activity of catalases has also been identified in other catalases (Vetrano et al., 2005; Koclar Avci et al., 2013; Lončar \& Fraaije, 2015; Teng et al., 2016) and has been presumed to also occur at the heme active site. Here, we report a combined structural, spectroscopic and kinetic
analysis of CATPO that allows us to propose an alternative model.

## 2. Experimental procedures

### 2.1. Materials

Standard chemicals and biochemicals were obtained from Sigma and Merck. Molecular-size markers and DNA ladders were obtained from Bio-Rad and Biolab, respectively. Sitedirected mutagenesis was performed using the QuikChange approach (Agilent).

### 2.2. Strains, plasmids and growth conditions

Escherichia coli XL1-Blue (Stratagene) and BL21 Star (DE3) (Invitrogen) strains were used for cloning and expression, respectively. During cloning steps, E. coli cells were grown aerobically at $37^{\circ} \mathrm{C}$ in LB medium supplemented with $50 \mu \mathrm{~g} \mathrm{ml}^{-1}$ kanamycin. The plasmid pET28a-CATPO (Yuzugullu et al., 2013), which carries an N-terminal $6 \times$ Histag sequence and TEV protease cleavage site, was used as the source of the catpo gene.

### 2.3. Site-directed mutagenesis

Single-point mutations were introduced into the catpo coding region by QuikChange mutagenesis using Hot Start KOD DNA polymerase (Sigma). The PCR primers containing the desired mutations were purchased from Sentegen, Turkey and are listed in Table 1. Subsequent expression and purification were carried out as described previously (Yuzugullu et al., 2013).

### 2.4. Enzyme assays

Catalase and phenol oxidase activities were determined as described previously (Yuzugullu et al., 2013). One unit of catalase was defined as the amount of enzyme that catalyses the decomposition of $1 \mu \mathrm{~mol} \mathrm{H}_{2} \mathrm{O}_{2}$ per minute in a $10 \mathrm{~m} M$ $\mathrm{H}_{2} \mathrm{O}_{2}$ solution. The initial rates of $\mathrm{H}_{2} \mathrm{O}_{2}$ decomposition were used to determine the turnover number $\left(k_{\text {cat }}\right)$ and the apparent $K_{\mathrm{m}}$ values. Kinetic constants were derived by fitting $v$ versus [S] traces to the Michaelis-Menten equation using SigmaPlot 14.0 (Systat Software Inc.). The term ' $K_{\mathrm{m} \text { _app }}$ ' in the

Table 2
Crystallographic data-collection and refinement statistics.
Values in parentheses are for the outermost shell.

|  | E316F variant | H246W variant | V536W variant | 3TR complex |
| :---: | :---: | :---: | :---: | :---: |
| PDB code | 5 y 17 | 5xvz | 5xy4 | 5zz1 |
| Beamline | ID30B, ESRF | ID30B, ESRF | ID30B, ESRF | I03, DLS |
| Detector | PILATUS3 6M | PILATUS3 6M | PILATUS3 6M | ADSC Q315 |
| Oscillation angle ( ${ }^{\circ}$ ) | 0.1 | 0.05 | 0.05 | 0.5 |
| Exposure time (s) | 0.02 | 0.02 | 0.02 | 0.4 |
| Transmission (\%) | 22 | 11 | 13 | 50 |
| No. of images | 1140 | 2500 | 1860 | 720 |
| Wavelength (A) | 0.98 | 0.98 | 0.98 | 1.0 |
| Space group | I2 | I2 | I2 | I2 |
| Unit-cell parameters |  |  |  |  |
| $a(\AA)$ | 125.7 | 125.3 | 125.4 | 125.5 |
| $b$ (A) | 120.9 | 120.8 | 120.7 | 121.7 |
| $c(\mathrm{~A})$ | 183.8 | 185.2 | 184.7 | 185.5 |
| $\beta\left({ }^{\circ}\right)$ | 102.0 | 102.0 | 102.0 | 102.2 |
| Resolution ( A ) | $\begin{aligned} & 100.3-2.3 \\ & \quad(2.34-2.30) \end{aligned}$ | $\begin{aligned} & 100.5-1.9 \\ & (1.93-1.90) \end{aligned}$ | $\begin{aligned} & 90.3-1.8 \\ & (1.83-1.80) \end{aligned}$ | $\begin{aligned} & 29.4-1.91 \\ & (1.95-1.91) \end{aligned}$ |
| Mosaicity ( ${ }^{\circ}$ ) | 0.24 | 0.08 | 0.11 | 0.17 |
| $R_{\text {merge }} \dagger$ (\%) | 6.8 (27.4) | 6.9 (45.6) | 5.4 (52.0) | 5.7 (42.6) |
| $R_{\text {p.i.m. }} \ddagger(\%)$ | 5.2 (20.9) | 4.7 (35.7) | 4.7 (45.5) | 3.7 (23.5) |
| $\mathrm{CC}_{1 / 2}$ | 0.996 (0.941) | 0.998 (0.740) | 0.998 (0.643) | 0.998 (0.843) |
| Observed reflections | 236441 (12331) | 493165 (21545) | 419138 (20025) | 690894 (30098) |
| Unique reflections | 105712 (5468) | 200277 (9502) | 223187 (11150) | 206818 (9578) |
| Completeness (\%) | 88.8 (92.7) | 94.5 (90.9) | 89.9 (91.1) | 98.8 (92.5) |
| Multiplicity | 2.2 (2.3) | 2.5 (2.3) | 1.9 (1.8) | 3.3 (3.1) |
| $\langle I / \sigma(I)\rangle$ | 8.4 (3.1) | 6.6 (1.8) | 8.8 (1.6) | 10.2 (2.7) |
| Refinement |  |  |  |  |
| $R_{\text {work }}$ (\%) | 17.8 (24.5) | 16.2 (31.2) | 15.8 (30.0) | 14.0 (20.4) |
| $R_{\text {free }}$ § (\%) | 22.0 (28.6) | 19.1 (31.2) | 19.3 (33.6) | 16.5 (22.0) |
| No. of protein atoms | 21079 | 21452 | 21421 | 21337 |
| No. of solvent molecules | 925 | 1837 | 2031 | 1618 |
| No. of ligand atoms | 176 | 236 | 183 | 224 |
| No. of ion atoms | 10 | 12 | 10 | 5 |
| Average $B$ factor ( $\AA^{2}$ ) |  |  |  |  |
| Protein | 33.99 | 24.45 | 19.85 | 20.91 |
| Ligands | 24.44 | 27.68 | 19.08 | 16.02 |
| Solvent | 27.39 | 28.29 | 31.93 | 23.63 |
| Ions | 43.07 | 43.08 | 31.29 | 26.63 |
| R.m.s.d., bond lengths $\boldsymbol{\top}(\AA)$ | 0.0121 | 0.0147 | 0.0140 | 0.0150 |
| R.m.s.d., bond angles $\uparrow\left({ }^{\circ}\right.$ ) | 1.651 | 1.818 | 1.727 | 1.888 |
| Ramachandran plot $\dagger \dagger$ |  |  |  |  |
| Most favoured regions (\%) | 96.65 | 98.03 | 96.82 | 97.48 |
| Outliers (\%) | 0.71 | 0 | 0.63 | 0.55 |
| Alignment with wild-type structure $\ddagger \ddagger$ ( PDB entry 4aum; Yuzugullu et al., 2013) over all residues |  |  |  |  |
| R.m.s.d. ( $\AA$ ) | 0.232 | 0.226 | 0.242 | 0.242 |
| $Q$-score | 0.984 | 0.985 | 0.980 | 0.980 |

$\dagger R_{\text {merge }}=\sum_{h k l} \sum_{i}\left|I_{i}(h k l)-\langle I(h k l)\rangle\right| / \sum_{h k l} \sum_{i} I_{i}(h k l) . \quad \ddagger R_{\text {p.i.m. }}$ is the precision-indicating (multiplicity-weighted) $R_{\text {merge }}$ relative to $I^{+}$or $I^{-}$. § $R_{\text {free }}$ was calculated with $5 \%$ of the reflections that were set aside randomly. © Based on the ideal geometry values of Engh \& Huber (1991). $\dagger \dagger$ Ramachandran analysis using MolProbity (Chen et al., 2010) . 抹 R.m.s.d and $Q$-scores were calculated using GESAMT (Krissinel, 2012)
pH 7.0 at $60^{\circ} \mathrm{C}$ using a temperaturecontrolled spectrophotometer (Agilent Cary 50 or 60 ).

### 2.5. Crystallization, data collection and refinement

Crystals were obtained by hangingdrop vapor diffusion using a reservoir consisting of $6-16 \%(v / v)$ PEG 400, 0.2 M potassium chloride, 0.01 M calcium chloride, 0.05 M sodium cacodylate in the pH interval 5.0-5.6. The complex of 3TR with CATPO was prepared by soaking crystals for 20 min in mother liquor containing $40 \mathrm{~m} M$ 3TR. Crystals were flash-cooled in liquid nitrogen (Teng, 1990) after soaking for several minutes in a synthetic mother liquor containing $20 \%(v / v)$ PEG 400 as a cryoprotectant. Diffraction data were collected on beamlines ID29 and ID30B at the European Synchrotron Radiation Facility (ESRF; de Sanctis et al., 2012; McCarthy et al., 2018) and on beamline I03 at Diamond Light Source (DLS; Allan et al., 2015) at 100 K (Table 2) and were processed using $X D S$ (Kabsch, 2010). Subsequent scaling (Evans, 1997), structure-solution, modelbuilding and refinement steps were carried out using the CCP4 suite (Winn et al., 2011). Although the E316F mutant data extended to higher resolution, refinement was unstable. Examination of the data using AUSPEX (Thorn et al., 2017) showed a severe ice ring at $\sim 2.2 \AA$ resolution. Truncation of the data set to $2.3 \AA$ resolution resulted in stable refinement. The wild-type structure of CATPO (PDB entry 4aum; Yuzugullu et al., 2013) was used to obtain initial phases
context of catalases is the peroxide concentration at $V_{\max } / 2$ and is used because the catalase reaction does not saturate with substrate and therefore does not precisely follow Michaelis-Menten kinetics (Switala \& Loewen, 2002). One unit of phenol oxidase was defined as the amount of enzyme that catalyses the formation of one nanomole of product per minute. The effects of 3 -amino-1,2,4-triazole (3TR) and catechol on oxidase activity were also investigated. Experiments with these compounds were conducted in the same manner but in the presence of the inhibitor at stated concentrations in the reaction buffer. Protein concentration was estimated using the Bradford assay (Bradford, 1976). All assays were performed in triplicate in 100 mM sodium phosphate buffer
by molecular replacement using MOLREP (Vagin \& Teplyakov, 2010). Iterative model building and refinement were performed using Coot (Emsley et al., 2010) and REFMAC5 (Murshudov et al., 2011), with each chain treated as a single TLS domain (Winn et al., 2001) and local NCS restraints (Usón et al., 1999; Murshudov et al., 2011).

The final structures of the E316F, H246W and V536W variants and the CATPO-3TR complex were determined at $2.3,1.9,1.8$ and $1.91 \AA$ resolution, respectively. The asymmetric units of the four CATPO variants analysed in this study each contained a CATPO homotetramer. The N-terminal 20 residues of all subunits in each of the four variants were disordered, as in the wild-type enzyme (Yuzugullu et al., 2013),

Table 3
Kinetic constants.

| Variant | $\begin{aligned} & K_{\mathrm{m} \_ \text {app }}{ }^{\dagger} \\ & (\mathrm{m} M) \end{aligned}$ | $\begin{aligned} & k_{\mathrm{cat}} \\ & \left(\mathrm{~s}^{-1}\right) \end{aligned}$ | $\begin{aligned} & k_{\mathrm{cat}} / K_{\mathrm{m}=\text { app }} \\ & \left(\mathrm{s}^{-1} M^{-1}\right) \end{aligned}$ | $R_{\text {Z }} \ddagger$ | Heme type | Specific oxidase activity ( $\mathrm{nmol} \mathrm{mg}{ }^{-1} \mathrm{~min}^{-1}$ ) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CATPO | 10 | 203410 | $20.3 \times 10^{6}$ | 0.8 | $d$ | $213 \pm 5$ |
| E316F | 20 | 196000 | $9.8 \times 10^{6}$ | 0.8 | $d$ | $173 \pm 2$ |
| E316H | 33 | 261000 | $7.9 \times 10^{6}$ | 0.8 | $d$ | $163 \pm 2$ |
| H246W | 40 | 152000 | $3.8 \times 10^{6}$ | 0.8 | $d$ | $33 \pm 2$ |
| I313F | 10 | 205000 | $20.5 \times 10^{6}$ | 0.9 | $d$ | $232 \pm 16$ |
| I314F | 50 | 194000 | $3.9 \times 10^{6}$ | 0.8 | $d$ | $92 \pm 7$ |
| L321A | 11 | 274000 | $24.3 \times 10^{6}$ | 0.9 | $d$ | $107 \pm 1$ |
| V536A | 67 | 860500 | $12.9 \times 10^{6}$ | 0.8 | $d$ | $236 \pm 6$ |
| V536W | 600 | 2402000 | $4.0 \times 10^{6}$ | 0.6 | $d$ | $85 \pm 7$ |

$\dagger K_{\mathrm{m} \_ \text {app }}$ is the $\mathrm{H}_{2} \mathrm{O}_{2}$ concentration at $V_{\max } / 2$ and is used because the catalase reaction does not saturate with substrate and therefore does not precisely follow MichaelisMenten kinetics (Switala \& Loewen, 2002). $\ddagger R_{\mathrm{Z}}=A_{406} / A_{280}$.
and were not included in the refined structures. The structure factors and coordinates for each structure have been deposited in the Protein Data Bank with accession codes $5 y 17$ for the E316F mutant, 5xvz for the H246W mutant, 5xy4 for the V536W mutant and 5zz1 for the CATPO-3TR complex. All figures were prepared using PyMOL (http://www.pymol.org/).

## 3. Results and discussion

Despite extensive efforts, we have been unable to obtain a crystal of CATPO in complex with catechol owing to its rapid auto-oxidation at concentrations high enough for binding, given the low $K_{\mathrm{m}}$. However, we were able to obtain the structure of its complex with the widely used catalase inhibitor 3-amino-1,2,4-triazole (3TR; Margoliash et al., 1960; Nicholls, 1962) to $1.91 \AA$ resolution (Table 2). Surprisingly, and in contrast to other structural reports of complexes of 3TR with catalase [Putnam et al., 2000; Borovik et al., 2011; PDB entry

1th4 (R. Sugadev, M. N. Ponnuswamy, D. Kumaran, S. Swaminathan \& K. Sekar, unpublished work)], we did not observe 3TR bound at the heme (Fig. 1a). Instead, 3TR occupies a surface pocket at the end of the lateral channel leading from the heme, where its interactions with the protein are almost exclusively mediated by a series of well ordered water molecules (Fig. 1b), as well as a second binding site at the interdimer interfaces of the homotetramer (Fig. 1a). Interestingly, the pocket at the end of the lateral channel corresponds to the site of NADPH binding in mammalian catalases (Fig. 1c). However, CATPO contains a C-terminal extension, residues 533-537, that does not exist in the NADPH-binding catalases and that lies across the upper region of this pocket, preventing the binding of NADPH. The mediation of the 3TR-CATPO interaction by water molecules in the plane of the inhibitor suggests a flexible binding site with the possibility of accommodating a variety of planar ligands of different sizes, suggesting that this could also be the site of phenolic substrate binding.

To further investigate this possibility, oxidase assays were performed at increasing concentrations of the CATPO oxidase substrate catechol ( $0-300 \mathrm{mM}$ ) with and without 3TR (0$10 \mathrm{~m} M)$. The $K_{\mathrm{m} \_a p p}$ and $V_{\max }$ values for catechol were calculated as $92.5 \mathrm{~m} M$ and $12500 \mathrm{nmol} \mathrm{ml}^{-1} \mathrm{~min}^{-1}$, respectively. 3TR showed competitive inhibition with respect to catechol, with a $K_{\mathrm{i}}$ of $2.1 \times 10^{-2} M$ (Fig. 2). This indicates that 3TR and catechol bind at the same site, but does not allow us to conclude that this site is the 3TR binding site observed crystallographically. We therefore generated a series of sitedirected mutants around the putative 3TR/catechol binding pocket to further test this hypothesis.

The residues Pro158, His246, Gln293, Ile313, Ile314, Glu316, Leu321 and Val536 surround the 3TR binding pocket and are structurally homologous to the residues that bind

(a) The CATPO tetramer shown as a ribbon diagram, highlighting the heme and 3TR binding sites. The heme is colored red, 3TR in the oxidase pocket is colored pink and 3TR at the dimer interface is colored orange. (b) The 3TR binding site in the lateral channel of CATPO. Composite OMIT electron density, calculated using the CCP4 COMIT program (Winn et al., 2011), for 3TR and bound waters is drawn at 1 r.m.s.d. and shown as a blue wire mesh. Analysis of the hydrogen bonding suggests that 3 TR is bound as $2 H-1,2,4$-triazole-3-amine and at the pH of the crystals should be in its neutral form. (c) View of chain $A$ of the CATPO complex with 3TR (PDB entry 5zz1; grey) superposed onto human catalase (PDB entry 1dgh; blue). CATPO loop 533537 lies across the top of the NADPH-binding pocket, clashing with the position of the NADPH in the human enzyme.

NADPH in the mammalian catalases. They were individually mutated to residues of the opposite size (i.e. small to large and vice versa) and the resulting variants were characterized. Specifically, the variants P158W, H246W, Q293W, I313F, I314F, E316F, E316H, L321A, V536A and V536W were constructed. All except P158W and Q293W were expressed normally. The catalase turnover numbers ( $k_{\text {cat }}$ ) are essentially unaffected for the E316F, E316H, H246W, I313F, I314F and L321A variants, but a marked increase in the catalase $k_{\text {cat }}$ value was observed for V 536 W , accompanied by a large increase in $K_{\mathrm{m} \_a p p}$ for $\mathrm{H}_{2} \mathrm{O}_{2}$ (Table 3). The catechol oxidase activities of the H246W, I314F, L321A and V536W variants were noticeably reduced (50-92\%) with respect to the wild-type enzyme, whereas the E316F, E316H, I313F and V536A variants of CATPO had little effect on the oxidase activity.

To further probe the effect of the mutations, the crystal structures of three variants were determined: H246W (lowest oxidase activity), E316F (little effect on oxidase activity) and V536W (40\% reduced oxidase activity, but a major effect on catalase kinetic parameters) (Table 3). The resulting structures


Figure 2
An illustrative double-reciprocal plot (Lineweaver \& Burk, 1934) is presented showing classical competitive inhibition kinetics for 3TR with respect to the CATPO oxidase activity. Error bars show the standard deviation of the SigmaPlot fit of the raw data for each point. Full details of the analysis are provided in Supplementary Fig. S1.
were almost identical to the native enzyme (Table 2), aside from changes in the ordered solvent molecules found in the putative oxidase substrate-binding pocket, the lateral channel and the main channel through which peroxide is thought to reach the heme active site. Surprisingly, the most extreme example of this is in the E316F mutant, in which the upper main channel and the outer part of the lateral channel contain almost no ordered solvent molecules; however, this is likely to be at least partially attributable to the lower resolution of this data set (Figs. $3 a$ and $4 b$ ). The bulky phenylalanine side chain of the E 316 F variant is oriented away from the entrance to the lateral channel and putative oxidase substrate-binding pocket and thus it is not surprising that mutations at this position show minimal effect on the oxidase activity. In contrast, the bulky tryptophan residue of the V536W variant protrudes into the top of the putative oxidase substrate site, partially occluding it, consistent with the partial reduction in oxidase activity (Figs. $3 b$ and $4 d$ ).

A similar picture is seen for His246, which lies directly below and perpendicular to the 3 TR ring in the CATPO-3TR complex. In the H 246 W variant both of the alternate conformations of the tryptophan side chain that were observed protrude further into the $3 T \mathrm{R}$ binding site and are likely to considerably hinder the binding of any small organic substrate (Figs. $3 c$ and $4 c$ ). This is consistent with the marked reduction in oxidase activity for this mutant. Interestingly, in the V536W variant, although not in the wild-type enzyme, His246 also adopts an alternate conformation, although this does not protrude into the 3 TR binding site (Figs. $3 b$ and $4 d$ ). Both the V536W and H246W variants therefore support our assignment of this pocket as the oxidase substrate-binding site.

Val536 is part of the C-terminal extension of CATPO that blocks the upper part of the NADPH-binding pocket in mammalian catalases and thus it was surprising that changes in this residue had such marked effects on the catalase activity. As for the E316F variant, in the V536W variant the chain of ordered solvent molecules in the main channel is disrupted (Fig. 4d), suggesting that there may be some crosstalk between


Figure 3
Comparison of 3 TR binding sites in the lateral channel of the $\mathrm{E} 316 \mathrm{~F}(a)$, V536W $(b)$ and $\mathrm{H} 246 \mathrm{~W}(c)$ variants superposed onto the complex of CATPO with 3 TR . The corresponding $2 F_{\mathrm{o}}-F_{\mathrm{c}}$ electron density, contoured at 0.7 r.m.s.d., is shown for the three cases as a blue mesh. Changes in solvent organization are evident among the structures. Trp246 has two alternate conformations (Fig. 4).


Figure 4
Main and lateral channel solvent in CATPO-3TR (a) and the E316F (b), H246W (c) and V536W (d) variants. The corresponding electron densities are shown for the four cases. Possible hydrogen-bond interactions are shown as dashed lines. Lys312, Ile313 and Ile314 in the lateral channel were removed for clarity. Mutated residues are shown in red and the ligand 3TR in purple. The inset in (c) shows the two alternate conformations of Trp 246 face on with density. $2 F_{\mathrm{o}}-F_{\mathrm{c}}$ electron density is shown as a blue mesh contoured at 0.7 r.m.s.d.. The channels are shown as transparent surfaces in Supplementary Fig. S2.
the main and lateral channels. However, in contrast to the V536W mutation, the E316F mutation has almost no effect on the catalase activity (Table 3). In addition, there is no obvious structural change or alteration in relative $B$-factor distribution in either mutant compared with the apo wild-type structure (PDB entry 4aum; Yuzugullu et al., 2013) that could explain this observation, and further investigation will be required.

The currently accepted model for 3TR inhibition of catalases is via either a reversible binding mode at high 3TR concentration (Appleman et al., 1956; Margoliash et al., 1960) or a slow peroxide-dependent irreversible inactivation that results in the formation of a covalent adduct at the heme active site (Borovik et al., 2011). The covalent adduct is likely to be the state that was observed in the previously reported complexes of 3TR with other catalases (Putnam et al., 2000; Borovik et al., 2011; PDB entry 1th4). However, the 3TR binding site for reversible inhibition has not been identified to date, although its noncompetitive nature would be consistent with a non-heme-centred mode of action (Nicholls, 1962; Putnam et al., 2000). We therefore wondered whether this reversible inhibition could in fact be mediated via the oxidase substrate-binding site that we have identified. In this case, we
would predict that catechol binding would protect the CATPO catalase activity against 3 TR inhibition. We therefore carried out a competition assay at a constant concentration of 3 TR


Figure 5
The effect of increasing catechol concentrations (up to $5 \mathrm{~m} M$ ) on the inhibition of CATPO by 3 TR at a constant concentration of $40 \mathrm{~m} M$.
$(40 \mathrm{~m} M)$ that is sufficient to nearly totally inhibit the catalase activity and increasing amounts of catechol, and observed that the presence of catechol reduces 3TR inhibition in a dosedependent manner (Fig. 5). Interestingly, catechol itself inhibits the catalase activity, but in a much less potent fashion than 3TR. This suggests that, as proposed by others (Nicholls, 1962; Putnam et al., 2000), the reversible inhibition by 3TR is mediated via an allosteric effect and that other molecules binding to this pocket will also have an inhibitory action. In this context it is interesting to note that the 3TR complex also shows a reduction in the number of ordered water molecules in the main channel when compared with the apo wild-type structure (Supplementary Fig. S3).

## 4. Conclusions

In summary, based on our structural, mutation and kinetic data we propose that the pocket at the entrance to the lateral channel, occupied by the nicotinamide moiety of NADPH in mammalian catalases, is the site of both oxidase substrate and 3TR binding. The promiscuous nature of the CATPO oxidase is explained by the presence of a number of ordered water molecules that both mediate substrate binding by forming bridging hydrogen bonds and can be displaced to accommodate different sized and shaped substrates. Peroxideindependent phenolic substrate oxidation is then likely to occur in a similar manner to NADPH oxidation, via electron transfer from the substrate to a high-valent iron-oxo intermediate, presumably formed via reaction with oxygen.

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