

wealth of information in tabular and figure form with excellent text which can be readily appreciated by anybody. Nearly all the data have been deposited in the Brookhaven Protein Data Bank and can, therefore, be treated to further and different analyses by any interested person.

The second part of the *Atlas* was prepared by Max Perutz. Here he probes the molecular mechanism by comparing the structure and properties of abnormal and fetal hemoglobins with normal (A) hemoglobin. Fortunately for the study of hemoglobin, they are still a fairly rapidly evolving family of proteins. Hence, the rate of accepted point mutations is relatively high and the number of abnormal hemoglobulins found in the human population is appreciable. Well over 30 such hemoglobulins have been studied in some detail. The best known example is that of sickle-cell hemoglobin (S) where a glutamine on the  $\beta$  chain in position 3 on  $\alpha$ -helix A changes to a valine causing the formation of destructive crystalline fibers in the red cells of venous blood. Such single alterations give precise probes into the molecular properties causing change of oxygen affinity, alteration in the cooperativity between subunits and the Bohr effect. These abnormalities are discussed by Perutz in relation to the structure illustrated in the first part of the *Atlas*. There are shown a number of black and white on grey electron density maps as experimental evidence of the structural information. Unfortunately, the difference electron density maps on pages 92 and 93 are not accompanied, like the others, by explanatory diagrams. Some of the descriptions of the structural alterations (e.g. the heme is more upright or inclined) are dependent on the view of the observer and require unavailable information of a standard model. This section is, however, a fantastic compendium of information and its mostly successful analysis in terms of the properties of hemoglobin is a tremendous achievement.

An alternative approach to the study of hemoglobin is the analysis of globins from widely different species. This has been the procedure of Love and co-workers as well as a number of other laboratories around the world. Unlike the abnormal human hemoglobins, these structures are not isomorphous with oxy or deoxy human hemoglobin A and the amino acid differences are numerous rather than in a well defined and specific position. Consequently, the differences in properties and structure are on a much larger scale. Recognition of this alternative approach is given on page 102 of the *Atlas* where Table 16 gives invariant residues. The treatment here is brief indeed, and does not balance the detail in the first 101 pages. No doubt, however, data on the other globins will be the topic of future editions by the appropriate authors.

The Fermi-Perutz *Atlas* is far more than a substantial work on hemoglobin and myoglobin. It is a magnificent example of how a complex molecule has been investigated. The analysis of the regulation of subunit interactions is appropriate not only to all cooperative oligomeric enzymes but, for instance, bears also on the assembly of viruses where small switches provided by cations or the binding of phosphates (in the form of nucleic acids) can substantially alter the preferred subunit associations.

MICHAEL G. ROSSMANN

*Department of Biological Sciences  
Purdue University  
West Lafayette  
Indiana 47907 USA*

*Acta Cryst.* (1982). B38, 2097

**Ionophores and their structures.** By MAX DOBLER. Pp. 379. New York, London: John Wiley, 1981. Price £42.50.

Ionophores are defined as 'substances which promote the transfer of ions from an aqueous medium into a hydrophobic phase'. These include naturally occurring peptide and depsipeptide ionophores and their synthetic analogues, the macrolide antibiotics, naturally occurring carboxylic acid ionophores and synthetic ionophores, such as macrocyclic 'crown' polyethers, and 'cryptates'.

The early chapters summarize the biophysical and the chemical investigations of the compounds mainly with respect to their complexing ability with alkali-metal and alkaline-earth-metal cations. There is an extensive table of formation constants and selectivity sequences. Use of chiral complexes for separation of racemates is also mentioned. All is compressed into 38 pages which give a good independent survey.

The main part of the book is a compendium of the results of X-ray crystal structure analysis on the ionophore molecules and their complexes. These are drawn as stereo pairs by the excellent program *PLUTO* of the Cambridge Crystallographic Data Centre. The author has obtained unpublished coordinates, not only from those files, but also from original authors (and, in fact, the book has already saved me from writing a letter). Appendices show displayed structural formulae with the numbering and the orthogonal coordinates of the unique molecules in Å enabling the reader to carry out further geometrical calculations. Many of these calculations have already been done by the author who provides tables of dimensions of interest, such as torsion angles in complexed and uncomplexed molecules and important bond lengths. In the text he discusses the significance of the results and is particularly good in critical appraisal of apparent contradictions in the literature. The nature of the book probably owes something to the author's experience of having to determine the structure of a potassium valinomycin complex himself to obtain the coordinates years after a preliminary communication from another laboratory.

Coverage of the literature is reasonably complete to 1979. The price is not unreasonable in view of the large number of illustrations.

MARY R. TRUTER

*Molecular Structures Department  
Rothamsted Experimental Station  
Harpenden  
Hertfordshire AL 5 2JQ  
England*

*Acta Cryst.* (1982). B38, 2097-2098

**X-ray diffraction by disordered and ordered systems.** By D. W. L. HUKINS. Pp. ix + 164. Oxford: Pergamon, 1981. Price US\$28.75.

The aim of this little book is to show how the X-ray scattering phenomenon is used to achieve structural data on

a wide variety of states of matter (gas, liquid and solid). The book is written especially for students who are not physicists. The author illustrates the basic principles of the diffraction process with the minimum of formulae, therefore, and 'often supplements mathematics with descriptions in words'. Only the classical theory is developed while the quantum nature of the phenomenon and the dynamical theory are ignored. Whenever necessary, the author refers to the best papers and authors. Each chapter contains a list of these references.

The book starts (chapters I and II) by explaining two basic principles: the electrons in matter are responsible for scattering, and the scattered waves are the Fourier transform of the electron density distribution of the sample. Following chapters (V to X) then describe how the diffraction patterns are modified by various interference processes as the nature of the sample changes: isolated atom, group of atoms or molecule, ideal gas, diluted solution, liquid, amorphous solid, and ordered materials in one, two or three dimensions. For each of these cases, the author explains what are the limits of the experiments and what region in the pattern is significant. He gives the basic methods for the analysis of the data and consequently shows what kind of parameters can be measured. He treats the reader as a prospective experimenter and defines for him the scale of the observation (size and shape of the scattering bodies or collective arrangement of atoms or internal structure of the molecule . . .). Typical examples of one-, two- and three-dimensional systems are chosen from among biological materials. These give a good overview of contemporary crystallography in this field. More illustrations would have been useful.

The last chapter (XI) introduces electron microscopy. The author discusses the relationship which exists between the Abbé theory and the optical principles of the diffraction of X-rays. Useful comparisons show the complementarity of the two techniques.

The book is designed very much as a teaching text, with a style that is clear, progressive and simple. As the author insists on the phenomenology of the diffraction process, students should read this book to improve their crystallographic knowledge.

F. ROUSSEAU

*Laboratoire Chimie de Solide Minéral  
Université Nancy I  
BP 239  
54506 Vandoeuvre-les-Nancy CEDEX  
Boulevard des Aiguillettes  
Vandoeuvre  
France*

*Acta Cryst.* (1982). B38, 2098

### Book Received

*The following book has been received by the Editor. Brief and generally uncritical notices are given of works of marginal crystallographic interest; occasionally a book of fundamental interest is included under this heading because of difficulty in finding a suitable reviewer without great delay.*

**Crystallography: an introduction for earth science (and other solid state) students.** By E. J. W. WHITTAKER. Pp. xii + 254. Oxford: Pergamon Press, 1981. Price £8.35, US \$19.95 (softback); £13.50, US \$32.50 (hardback). A review of this book, by M. J. Buerger, has been published in the July 1982 issue of *Acta Crystallographica*, Section A, p. 560.