

**MS27-2-13 Crystal Structure of High-Pressure Modification of Gold(III) Trifluoride**  
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**Abstract**

Gold(III) trifluoride is a binary compound of gold and fluorine. At standard conditions the compound crystallizes in the hexagonal space group  $P6_122$  (or  $P6_522$ ) [1]. A few phase transitions at higher pressures were predicted and observed in the Raman spectra of  $AuF_3$  by Kurzydowski and coworkers [2]. To the best of our knowledge, no crystal structure of any of the high pressure modifications of  $AuF_3$  was so far determined experimentally.

Here we will present our latest data on the first experimental determination of the crystal structure of gold(III) trifluoride at higher pressure. The crystals of  $AuF_3$  were synthesized in a fluorine oven by interaction of gold metal powder with fluorine-argon mixture at increased temperatures. The crystals grown in the colder part of the fluorine oven were transferred and stored in a glove box and were used for single crystal X-ray diffraction later.

The diffraction experiment was carried out using a One20DAC diamond anvil cell by Almax easyLab. The pressure in the DAC was controlled using the ruby fluorescence method and was set to be equal to approximately 2 GPa. At the conditions of the experiment we observed the phase transition predicted and observed in the Raman spectra of  $AuF_3$  by Kurzydowski and coworkers [1]. We confirm that the high pressure polymorph modification of gold trifluoride crystallizes in the hexagonal space group  $P6_1$  (or  $P6_5$ ). The following unit cell parameters were determined from the diffraction data:  $a = 4.9195(4)$ ,  $c = 16.3956(13)$  Å,  $V = 343.64(6)$  Å<sup>3</sup>,  $Z = 6$  at 293 K and ca. 2 GPa.

**References**

[1] F. W. B. Einstein, P. R. Rao, James Trotter, Neil Bartlett. *J. Chem. Soc. A*, **1967**, 478-482.

[2] D. Kurzydowski, S. Kobayakov, Z. Mazej, S. B. Pillai, B. Chakraborty, P. K. Jha. *Chem. Commun.*, **2020**, 56, 4902-4905