

Poster Presentation

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Five coordinate copper(II) complexes of saccharin with pyridine and dmf

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The copper(II) saccharinate complex containing pyridine and dmf have been prepared and characterized by elemental analyses, IR, UV-Visible, magnetic measurements, thermal analysis and single crystal X-ray diffraction methods. The crystal structure of aqua-bis(pyridine)di(saccharinato)copper(II) [Cu(sac)₂(py)₂(H₂O)] and diaqua-(dimethylformamide)-di(saccharinato)copper(II) [Cu(sac)₂(H₂O)₂(dmf)] (sac = saccharinate anion; py = pyridine and dmf = dimethylformamide) have been determined. The compounds crystallize in the orthorhombic space group Iba₂ with Z = 4 and the Cu(II) ion presents a CuN₄O square pyramidal coordination sphere. Two one-electron electrochemical redox processes have been followed by both of the complexes.

Keywords: [Cu\(II\) complex](#), [crystal structure](#), [redox process](#)