

Low temperature structural investigations along the  $\text{Cu}_3\text{SbS}_3$  (skinnerite) -  $\text{Cu}_3\text{BiS}_3$  (wittichenite) join.

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The majority of minerals in the Cu-Sb-Bi-S system occur in the pseudoternary section defined by  $\text{Cu}_2\text{S}$  (chalcocite),  $\text{Sb}_2\text{S}_3$  (stibnite) and  $\text{Bi}_2\text{S}_3$  (bismuthinite). Within this section, stibnite and bismuthinite and  $\text{CuSbS}_2$  (chalcostibite) and  $\text{CuBiS}_2$  (emplectite) are isostructural pairs displaying complete solid solution along their respective joins. In contrast,  $\text{Cu}_3\text{SbS}_3$  (skinnerite) and  $\text{Cu}_3\text{BiS}_3$  (wittichenite) are not isostructural at ambient temperature and a miscibility gap from 10 to 50 atomic percent bismuth has previously been described at 648K. Prior crystallographic investigation of these end-member phases has indicated significant polymorphism, mostly at elevated temperatures.

End-member  $\text{Cu}_3\text{SbS}_3$  and  $\text{Cu}_3\text{BiS}_3$  along with intermediate compositions with Sb:Bi ratios of 75:25, 50:50 and 25:75 were synthesized in evacuated silica glass tubes at temperatures between 675 and 693K and their structures determined and refined at temperatures ranging from 293 to 100K. Pure  $\text{Cu}_3\text{SbS}_3$  is  $P2_1/c$  at 293K but converts to a  $P2_12_12_1$  form isostructural with pure  $\text{Cu}_3\text{BiS}_3$  below approximately 250K, as previously reported. Compositions at Sb:Bi ratios of 75:25 and 50:50 adopt the  $P2_12_12_1$   $\text{Cu}_3\text{BiS}_3$  structure over the entire temperature range examined. Pure  $\text{Cu}_3\text{BiS}_3$  converts to a previously unknown phase of space group  $\text{Pnm}2_1$  below approximately 250K and the composition at an Sb:Bi ratio of 25:75 adopts this structure below approximately 175K.

Equipment limitations prevented studies above 293K or below 100K, however we infer that a change from  $P2_12_12_1$  to  $\text{Pnm}2_1$  may occur around 50K for the 50:50 composition and that changes from  $P2_12_12_1$  to  $P2_1/c$  may occur for the 75:25 and 50:50 compositions at temperatures between 293 and 400K. Given the increasing interests in copper bearing sulfosalt minerals as energy technology materials, further investigations at wider temperature ranges are warranted.