

MS34-P6 Co-crystals of
**5,6-Dimethyl-2-thiouracil: further proof of
 the robustness of the ADA-DAD
 N—H...O/N—H...N/N—H...S-synthon**

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One of the main goals in crystal engineering is the search for reliable non-covalent intermolecular interaction motifs ("synthons") for the design of new solids with desired properties.^[1] The replacement of a strong hydrogen bond within a certain interaction with a weaker one, say N—H...O with N—H...S, may open up ways to new synthons.^[2] Thereby, the robustness of the synthon still may be preserved due to the cooperativity of hydrogen bonding interactions.^[3] Inspired by the cyanuric acid–melamine co-crystal, where an ADA-DAD N—H...O/N—H...N/N—H...O interaction is observed (A = acceptor, D = donor), the title compound 5,6-dimethyl-2-thiouracil (DMTU) was selected for co-crystallization. DMTU contains an ADA hydrogen-bonding site involving an S-atom as an acceptor and, therefore, should be capable of forming a mixed ADA-DAD N—H...O/N—H...N/N—H...S synthon with suitable cofomers like 2,4-diaminopyrimidine (DAPY), 2,4,6-triaminopyrimidine (TAPY) or 2,4,6-triamino-1,3,5-triazine (melamine; MELA). Co-crystallization experiments yielded two solvates of DMTU, (I) and (II), one salt-hydrate and two co-crystal solvates with DAPY, (III) – (V), one co-crystal solvate and one co-crystal salt-solvate with TAPY, (VI) and (VII), and one co-crystal solvate with MELA, (VIII) (Fig. 1). The two solvates show the formation of dimers or tetramers in the crystal packing formed by either "pure" R²₂(8) N—H...O hydrogen bonds in (I) or by a combination of "pure" R²₂(8) N—H...O and "mixed" R²₂(8) N—H...O and N—H...S hydrogen bonds in (II). While DMTU showed an AA-DD motif with DAPY in (III) the synthesis of the desired ADA-DAD synthon was successful in co-crystals (IV) – (VIII). Moreover, due to a proton transfer along an N—H...N-hydrogen bond in (VII), one of the two observed ADA-DAD interactions was changed into an AAA-DDD motif. The results demonstrate, that the ADA-DAD N—H...O/N—H...N/N—H...S motif is a reliable synthon for the design of co-crystals of DMTU.

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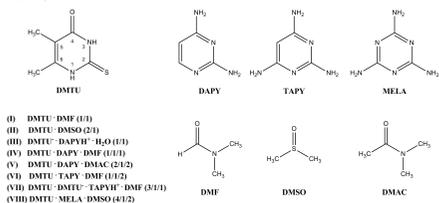


Figure 1. Structures of the compounds used and composition of the crystals.

Keywords: Co-crystals, crystal engineering, N—H...S hydrogen bonds