

MS17-P12 Neutron diffraction study of piezoelectric material under electric fieldTakuro Kawasaki¹, Takayoshi Ito², Yasuhiro Inamura¹, Takeshi Nakatani¹, Stefanus Harjo¹, Wu Gong¹, Kazuya Aizawa¹, Takaaki Iwahashi¹

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Piezoelectric material is widely used as an actuator and other various devices because the size of the material can be controlled by the applied electric field. Two types of microscopic deformation, lattice strain and domain switching, produces this macroscopic deformation. On the other hand, time-resolved measurement is attracting great interest in order to observe transient phenomena of materials under external fields. The behaviors of the lattice and the domain under the cyclic electric field can be studied by time-resolved in-situ diffraction experiment. In this study, the lattice strain and the domain characteristic of piezoelectric material under static and cyclic electric fields were studied by time-of-flight neutron diffraction. The diffraction intensities from PZT-based ceramics were measured at Engineering Materials Diffractometer TAKUMI in J-PARC. The intensities in the parallel direction and the vertical direction to the field were measured simultaneously by $\pm 90^\circ$ detector banks. The collected data were divided based on the voltage and the phase conditions of applied electric field and were converted to diffraction patterns using the data reduction software developed in J-PARC. The variation of the position and the intensity of the diffraction peaks according to the applied field were found in the patterns. The responses of the lattice and the domain to the field were successfully observed.

Keywords: neutron diffraction, piezoelectric material, lattice strain**MS17-P13** High-pressure study of $\text{Mn}(\text{BH}_4)_2$: a new polymorphs with high hydrogen densityNikolay Tumanov¹, Elsa Roedern², Dorrit B. Nielsen², Torben R. Jensen², Alexander V. Talyzin³, Radovan Černý⁴, Dmitry Chernyshov⁵, Vladimir Dmitriev⁵, Zbigniew Łodziana⁶, Yaroslav Filinchuk¹

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Hydrogen economy is one of the perspective trends in the future of renewable energy, but hydrogen storage is still a limiting factor. A possible way is to chemically bound hydrogen in light-weight compounds. In this context, metal borohydrides have been studied intensively in recent years. One of the problems in the practical application of borohydrides as hydrogen storage materials is often a too high or a too low decomposition temperature, but manganese borohydride is one of the borohydrides with moderate temperature of decomposition (160 °C). The structure and properties of metal borohydrides are usually studied at ambient conditions, but applying high hydrostatic pressure to the compounds not only adds an additional dimension to the phase diagram, revealing new high-pressure polymorphs of borohydrides, but may also allow to discover new high-pressure phases that are stable upon decompression. Such compounds are not only denser than those known to form at ambient conditions, but may also provide new or improved properties to a system, *e.g.* regarding, stability, decomposition and melting temperatures, etc. For instance, $\delta\text{-Mg}(\text{BH}_4)_2$, obtained by compression of $\text{Mg}(\text{BH}_4)_2$ to ~ 1 GPa, is more stable in air [1], than its ambient pressure phases.

The high-pressure behavior of $\text{Mn}(\text{BH}_4)_2$ was studied up to 15 GPa using diamond anvil cells and powder X-ray diffraction and Raman spectroscopy and two new polymorphs was discovered using combination of diffraction and DFT calculations. The first polymorph, $\delta\text{-Mn}(\text{BH}_4)_2$, forms near 1 GPa and is isostructural to a magnesium analog, $\delta\text{-Mg}(\text{BH}_4)_2$. This polymorph is stable upon decompression to ambient conditions. The same polymorph was obtained by compression of $\alpha\text{-Mn}(\text{BH}_4)_2$ in a large-volume steel press, as well as by high-energy ball milling. The thermal stability was studied by *in situ* powder X-ray diffraction and thermal analysis coupled with mass-spectroscopy. $\delta\text{-Mn}(\text{BH}_4)_2$ transforms back to $\alpha\text{-Mn}(\text{BH}_4)_2$ upon heating (67-109 °C) and decomposition is initiated at 130 °C, releasing hydrogen and some diborane. The high-pressure polymorph, $\delta\text{-Mn}(\text{BH}_4)_2$, transforms into a second, $\delta'\text{-Mn}(\text{BH}_4)_2$, in the pressure range of 8.6-11.8 GPa, with