MS15-P26 Fluorous low-dielectric-constant materials: non-porous coordination polymers *vs.* metal-organic frameworks

Simona Galli¹, Alessandro Cimino^{1,2}, Carlotta Giacobbe^{1,3}, Giovanni Palmisano¹, Angelo Maspero¹, Joshua F. Ivy², Chi Yang², Sammer Tekarli², Mohammad A. Omary²

- 1. Dipartimento di Scienza e Alta Tecnologia, Università dell'Insubria
- 2. Department of Chemistry, University of North Texas
- 3. ESRF, Grenoble, France.

email: simona.galli@uninsubria.it

Identifying novel low-dielectric-constant (low- κ) materials for integrated circuit (IC) use is a main concern for the microelectronics industry.

Metal-organic frameworks (MOFs) and functional non-porous coordination polymers (FN-PCPs) are studied as alternatives to currently applied low- κ dielectrics: their properties can be modulated through a sensible choice of nodes and spacers, with the added value of the high thermal stability imparted by N-donor struts. Compared to MOFs, FN-PCPs possess i) higher material density, granting a higher density of functionally-active centers; and ii) lower sensitivity to high- κ species, such as water (κ ~80), due to lack of porosity. Nevertheless, few reports, regarding exclusively MOFs, have appeared to date on dielectric properties.

Here we report on two classes of related CPs:

- 1. Tetrazolate-based CPs: These include FN-PCP-1 $(Ag_{\gamma}(FBTB),$ **FBTB** 1,4-bis-(tetrazol-5-ate)tetrafluorobenzene) and FMOF-3 (Cu(FBTB)), which are isostructural to $Ag_2(BTB)$ [1] and Cu(BTB) [2], respectively. As demonstrated by VT-PXRD, fluorination enhances thermal stability: FN-PCP-1 and FMOF-3 decompose in air at 260 and 400 °C, respectively (vs. 380 °C $T_{\rm dec}$ for ${\rm Ag_2(BTB)};~100$ °C $T_{\rm phase_change}$ for Cu(BTB)). Fluorination and lack of porosity favour water stability: contact angles up to ~75° were measured for FN-PCP-1 (vs. ~50° for Ag₂(BTB); FN-PCP-1 is recovered intact after 1 month exposure to water vapours while FMOF-3 undergoes a phase change after 24 h. Fluorination reduces κ_{RT} values to 2.59(3) for FN-PCP-1 and 2.44(5) for FMOF-3 (vs. 3.79(4) for Ag₂(BTB)). After 72 h exposure to water vapours FN-PCP-1 κ increases to only 2.63(5).
- 2. Triazolate-based CPs: These include FMOF-1 ($Ag_2(Ag_4Tz_6)$, Tz = 3,5-bis(trifluoromethyl)-1,2,4-triazolate) [3] and its polymorph FN-PCP-2 ($Ag_7(Tz)_7$). FMOF-1 is stable, under N₂, up to 380 °C. DFT simulations show that both FMOF-1 and FN-PCP2 possess non-polar building blocks. Measured contact angles up to ~160° for FMOF-1 suggest superhydrophobicity. A record low- κ_{RT} is obtained from preliminary studies that remain ongoing.

Work is in progress to complete this case study and exploit other metal-ligand combinations, maintaining the

above promising properties yet increasing the thermal stability up to the $\sim\!450$ °C needed for practical IC application.

References

- [1] A.Maspero et al. Inorg. Chim. Acta 2009, 362, 4340.
- [2] M.Dincă et al. J.Am.Chem.Soc. 2006, 128, 8904.
- [3] C.Yang et al. J.Am.Chem.Soc. 2007, 129, 15454.

Keywords: Powder X-ray diffraction, coordination polymers, metal-organic frameworks, low- κ dielectrics, thermal stability, hydrophobicity