

## Poster Presentation

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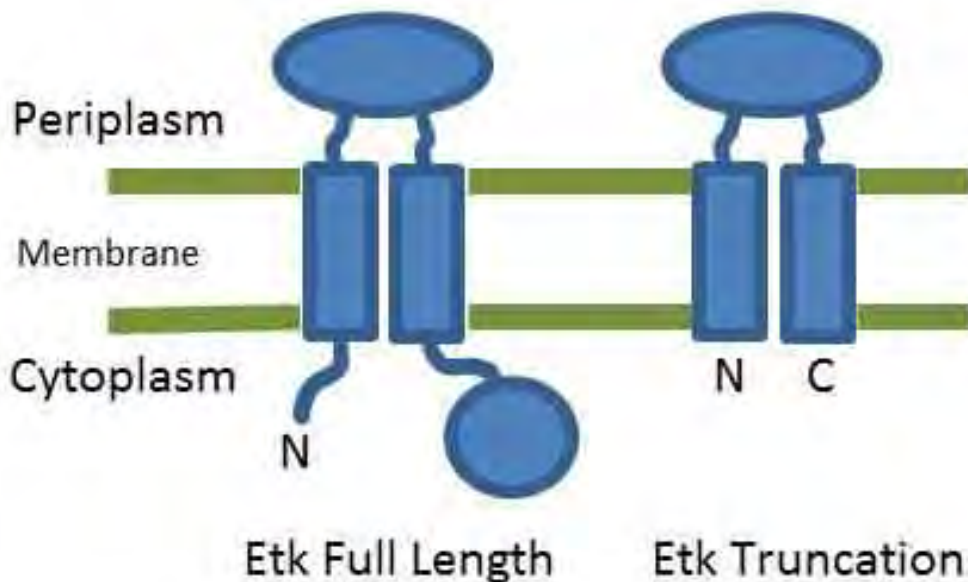
### Biophysical characterization and crystallization of the membrane protein Etk

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The inner membrane protein E. coli tyrosine kinase (Etk) is part of a large protein complex that assembles and exports capsular polysaccharide (CPS) in Gram-negative bacteria. Etk interacts with an outer membrane protein channel, YccZ (1), and regulates CPS export through autophosphorylation of a tyrosine cluster in its C-terminal tail. Previous work resulted in the structure of the isolated Etk C-terminal kinase domain (2). In the present study, the full-length protein has been purified and characterized in vitro. When purified in n-dodecyl- $\beta$ -D-maltoside (DDM), Etk full length retains autophosphorylation activity, but is not suitable for crystallization because it severely aggregates and degrades. Using the main degradation product, a truncation containing the N-terminal domain (interacts with YccZ) and both transmembrane helices was designed. Truncated Etk does not further degrade and exists as a mixture of monomers and dimers when solubilized by five detergents as determined by size-exclusion chromatography and analytical ultracentrifugation. Crystals have been successfully grown when the protein is solubilized in DDM or n-decyl- $\beta$ -D-maltoside (DM). The most promising crystals (DDM, 0.1 M MES pH 6.0, 1-5% PEG 3000, 20-30% PEG 200) have been reproduced and optimized with the assistance of a colorimetric assay (3). This assay relies on a reaction between 2,6-dimethylphenol, sulfuric acid, and the sugar head group of certain detergents to accurately quantify detergent in crystallization samples with minimal sample loss. Additive screening also revealed that MgCl<sub>2</sub> improves crystallization. Optimization of this crystallization condition has significantly improved reproducibility of these crystals, but x-ray diffraction is limited to 6.5 Å. Current work is focused on reproducing and optimizing a second crystallization lead (DM, 0.1 M KH<sub>2</sub>PO<sub>4</sub> pH 7.5, 32% PEG 400, 0.1 M KCl).

[1] R. Collins, K. Beis, C. Dong, C. Botting et al. *P. Natl. Acad. Sci.* (2007) 104, 2390-2395, [2] D. Lee, J. Zheng, Y. She, Z. Jia, *EMBO J.* (2008) 27, 1758-1766., [3] C. Prince, Z. Jia, *Acta Cryst. D* (2012) 68, 1694-1696.



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