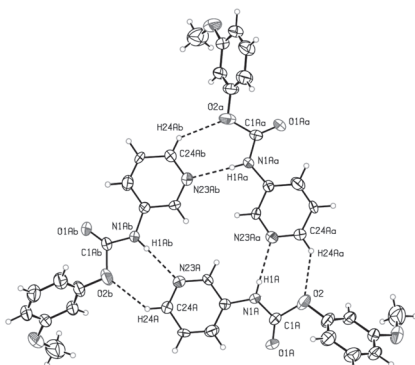


**MS43-P13** Carbamates I: Structural and conformational study of eight methoxyphenyl-*N*-pyridinylcarbamates. Pavle Mocić, John F. Gallagher, *School of Chemical Sciences, Dublin City University, Ireland*  
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A 3×3 isomer grid of nine methoxyphenyl-*N*-pyridinylcarbamates (C<sub>13</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub>) as **CxxOMe** (x = *para*-/*meta*-/*ortho*-) was synthesized and studied to determine the crystal structures and correlate structural relationships from both *ab initio* calculations and the solid-state using conformational analysis. Eight of nine isomers crystal structures were determined using single crystal X-ray diffraction. All isomers form N-H...N hydrogen bonds as primary interaction, with one isomer (**CmmOMe**) forming a relatively unusual disordered hydrogen bonded trimer *via* the N-H...N interactions, while the **CoxOMe** isomers form the N-H...N hydrogen bonded dimers. In all isomers additional C-H...O interactions aid aggregation, whereas in most of isomers the methoxy group is engaged in important C-H...O hydrogen bonding patterns.

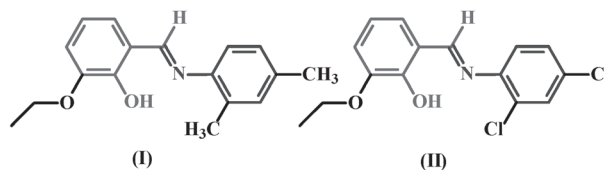


The solid state and the modelled conformations are mismatched in 3 of 8 molecules in which methoxy groups and *meta*-methoxyphenyl rings (**mOMe**) of the solid state structures adopt *meta*-stable or unstable conformations as compared to the optimised computational models.

**Keywords:** carbamates, conformational analysis, *ab initio*

**MS43-P14** (*E*)-2-[(2,4-dimethylphenylimino)methyl]-6-ethoxyphenol and (*E*)-2-[(2,4-dichlorophenylimino)methyl]-6-ethoxyphenol Hakký Yasin Odabaşođlu<sup>a</sup>, Mustafa Odabaşođlu<sup>b</sup>, Osman Ozan Avinç<sup>a</sup> & Orhan Büyükgüngör<sup>c</sup>  
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In industry, azomethine dyes have a wide range of application such as dyes and pigments with luminescent properties [1]. These dyes are known to be among most important dyes because of their wide applications, including color photographic system, dye diffusion thermal transfer print system and others [2]. In addition, azomethine dyes have been used widely as ligands in the field of coordination chemistry [3]. *o*-Hydroxy azomethine dyes that have a strong intra-molecular hydrogen bond. These compounds are of interest because of their thermochromism and photochromism in the solid state, which can involve reversible intra-molecular proton transfer from an oxygen atom to the neighboring nitrogen atom [4]. Taking into account these important features of the azomethine dyes, we aimed to investigate the conformation of the (I) and (II) by X-ray crystallography.



The crystal structure of the (I) exhibit O-H...N, C-H...O hydrogen bonds and C-H... $\pi$  interactions. There are two symmetry-independent molecules in the asymmetric unit. The dihedral angle between the aromatic rings in (I) are 4.95(2)° and 13.72(2)°. The crystal structure of the (II) has O-H...N, O-H...Cl and C-H... $\pi$  interactions. The dihedral angle between the aromatic rings in (II) is 12.47(3)°.

- [1] Taggi, A. E., Hazef, A. M., Wack, H., Young, B., Ferraris, D., Lectka, T. (2002). *J. Am. Chem. Soc.* 124, 6626-6635.
- [2] Ichijima, S. & Kobayashi, H. (2005). *Bull. Chem. Soc. Jpn.* 78, 1929-1938.
- [3] Calligaris, M., Nardin, G. M. J. & Randaccio, C. (1972). *Coord. Chem.* 7, 385-389.
- [4] Hadjoudis, E., Vitterakis, M., Moustakali-Mavridis, I. (1987). *Tetrahedron* 43, 1345-1360; Moustakali-Mavridis, I., Hadjoudis, E., Mavridis, A. (1978). *Acta Crystallogr.* B34, 3709-3715.

**Keywords:** Schiff's base, Azomethine dye, Salicyl aldehyde