

o.m2.o5 **Structure transformations in fluorinated complex copper oxides.** A.M. Abakumov¹, M.G. Rozova¹, B.Ph. Pavlyuk¹, R.V. Shpanchenko¹, E.V. Antipov¹, J. Hadermann², O.I. Lebedev², G. Van Tendeloo², ¹*Department of Chemistry, Moscow State University, 119899 Moscow, Russia;* ²*EMAT, University of Antwerp (RUCA), Groenenborgerlaan 171, B-2020 Antwerp, Belgium*

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The soft chemistry technique based on low temperature fluorination by XeF₂ was applied for the preparation of fluorinated cuprates with different structures: HgBa₂CuO_{4+δ} (Hg-1201), YBa₂Cu₃O_{6.11} (Y-123), Y₂Ba₄Cu₇O_{14.01} (Y-247), La₂CuO₄ (T-phase), LaHo_{0.75}Sr_{0.25}CuO_{3.9} (T*-phase), La_{6.7}Sr_{1.3}Cu₈O_{19.7} and LaACuGaO₅ (A=Ca,Sr). The structures of fluorinated phases were studied using X-ray or neutron powder diffraction and high-resolution electron microscopy.

Fluorination of the strongly reduced Hg-1201, phase with T_c=61 K resulted consequently in the increase of T_c up to 97 K, the decrease of T_c and the suppression of superconductivity due to overdoping. Twice the amount of extra anions in comparison with those for the oxygenated Hg-1201 phases with close T_c was found by neutron diffraction. The exchange of extra oxygen for the double amount of fluorine causes an extended shortening of the apical Cu-O bond distances, while the in-plane distances, as well as T_c, do not vary.

The insertion of fluorine in the strongly reduced Y-123 produces the superconducting YBa₂Cu₃O₆F₂ phase with T_c=75-94K. Fluorine atoms occupy positions into anion deficient Cu1 layers forming a distorted octahedral coordination of Cu1 atoms. The increase of CuO_(apical) distance by 0.5-0.7Å due to Jahn-Teller deformation leads to *c* parameter for the fluorinated phase ranged in 12.8-13.2Å. Similar transformation also takes place in Y₂Ba₄Cu₇O₁₄, where the increase of *c* parameter from 51.9Å to 53.5Å was observed.

The sequences of structural changes were found in T and T* phases at different degrees of fluorination. Small amount of fluorine intercalates between the doubled LaO layers in La₂CuO₄ causing the increase of the orthorhombic distortion and the appearance of superconductivity (T_c≈40K). Further fluorine insertion followed by a transformation to tetragonal K₂NiF₄-type structure. At elevated temperatures the fluorination of T and T* materials is accompanied by a redistribution of anions between rocksalt and fluorite-type positions in A₂O₂ blocks resulting in complicated superstructures due to anion ordering and cation displacements.

New fluorine-containing perovskite type tetragonal structures were observed after fluorination of the La_{6.7}Sr_{1.3}Cu₈O_{19.7} and LaACuGaO₅ (A=Ca, Sr) phases due to random distribution of oxygen and fluorine atoms between vacant anion sites. Distorted octahedral environment of Cu atoms (in plane d_{Cu-(O,F)} = 1.89Å and apical d_{Cu-(O,F)} = 2.03Å for fluorinated La_{6.7}Sr_{1.3}Cu₈O_{19.7}) can be explained by Jahn-Teller effect. These compounds do not exhibit superconductivity, possibly due to improper copper oxidation state or common occupations of anion positions in conducting layers by fluorine and oxygen atoms.